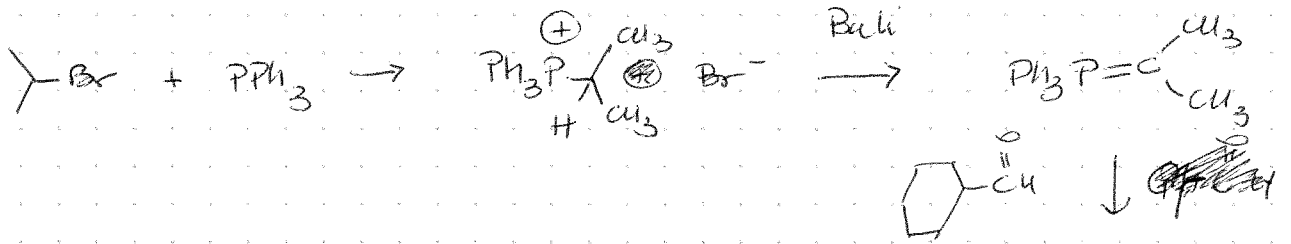
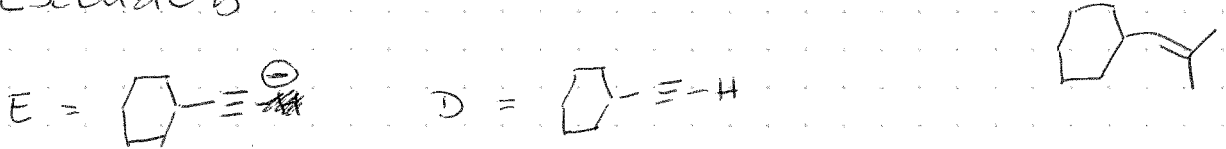


ESECUZIO 4

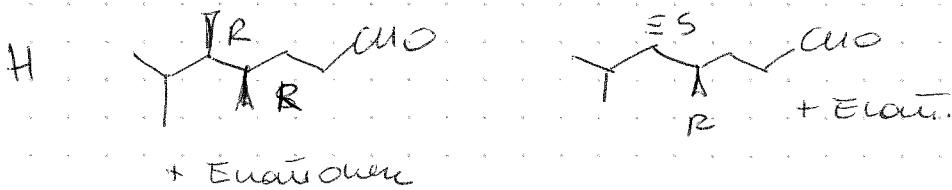
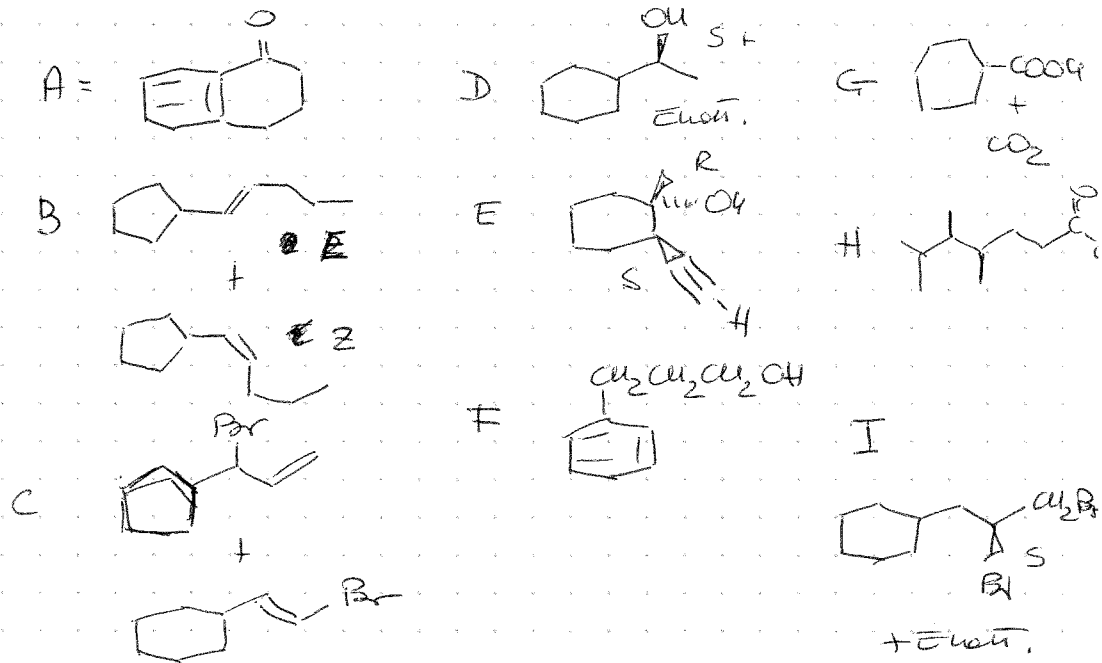
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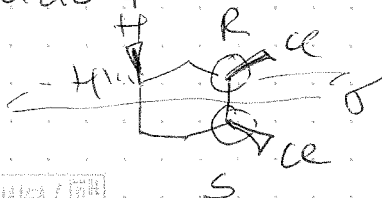
ESECUZIO 5



ESECUZIO 6



ESECUZIO 7

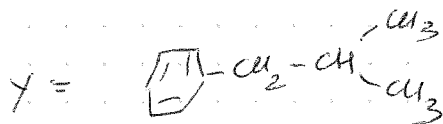


BRUNNEN

- 1 = piano riflessione
- 2 = achirale
- 3 = C = enantiotopici
- 4 = H = diastereotopici
- 4 R, S (meso)

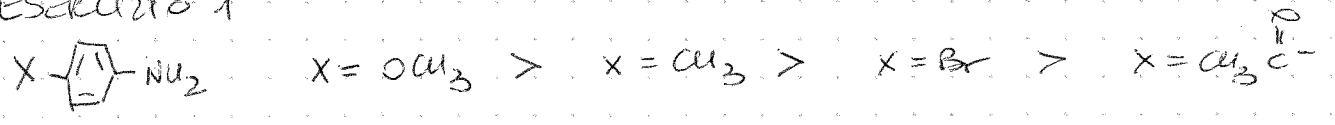
ESERCIZIO 8 $\text{Et}-\text{C}_6\text{H}_5-\text{COOH}$

ESERCIZIO 9 $\text{C}_{10}\text{H}_{12}\text{O}$



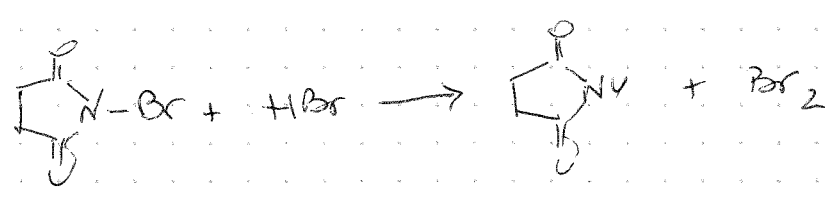
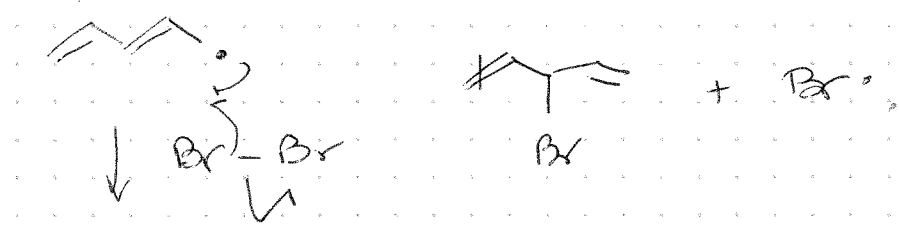
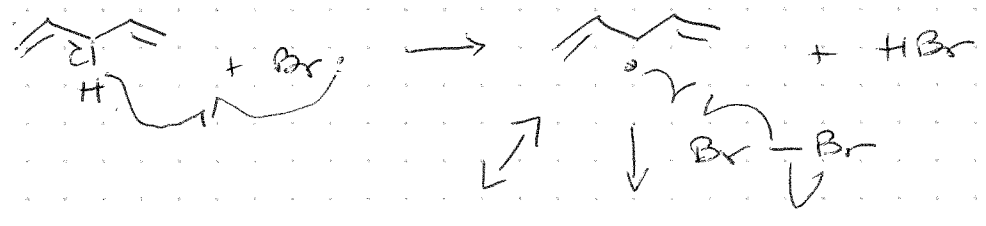
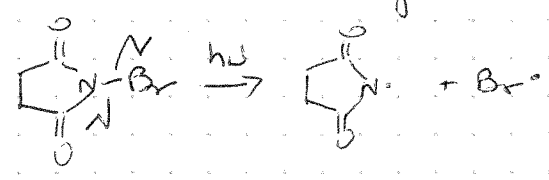
MODELLO 3 Soluzioni

ESERCIZIO 1



La basicità dipende dalla densità elettronica del doppio legame non condiviso sull'atomo di azoto. La natura elettronica o donatrice influenzano queste proprietà.

ESERCIZIO 2



ESERCIZIO 3

