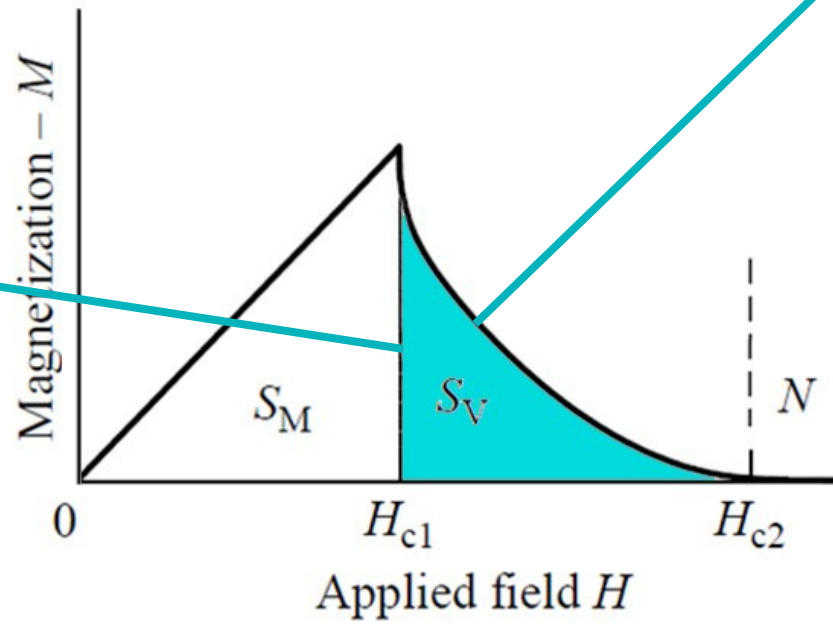
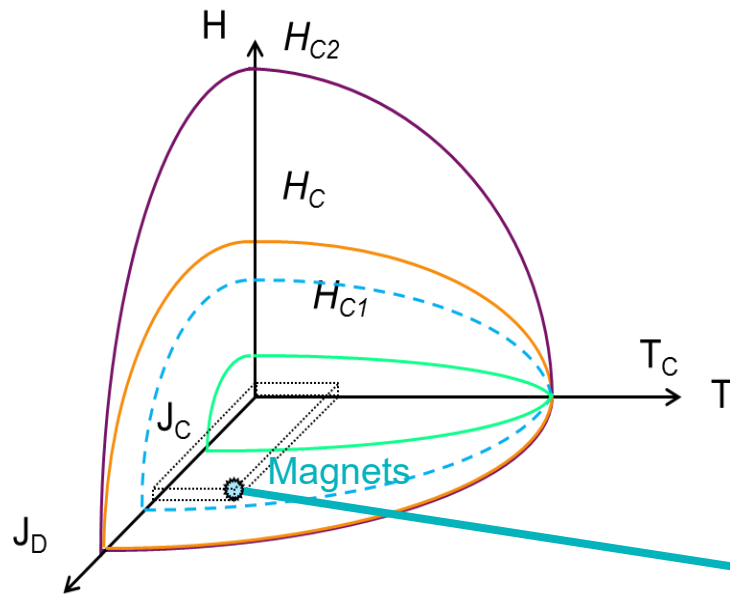


# Superconductive Materials

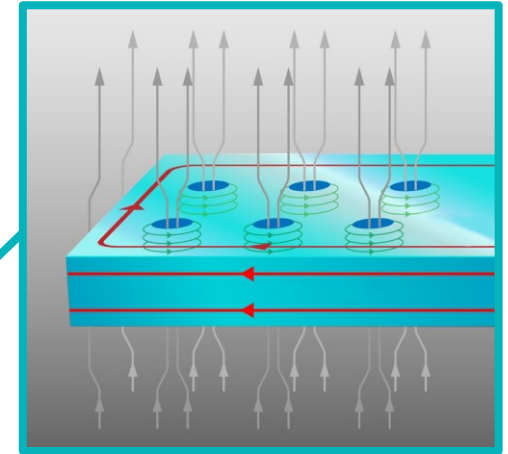
**Part 12**

**Materials for SRF - Bulk Nb**

# Superconducting Magnets

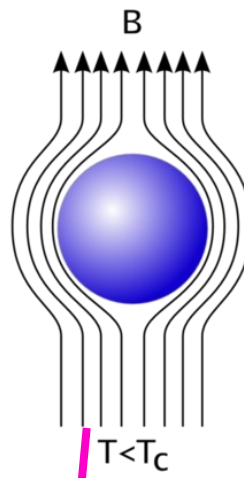
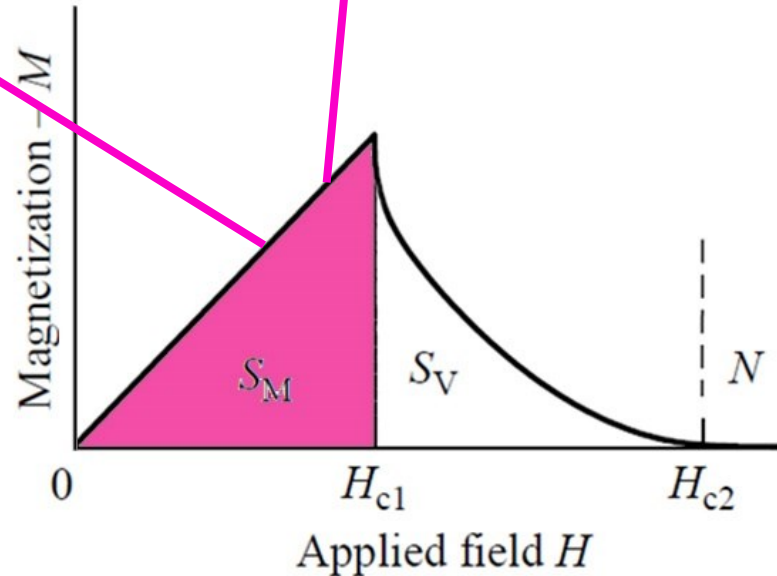
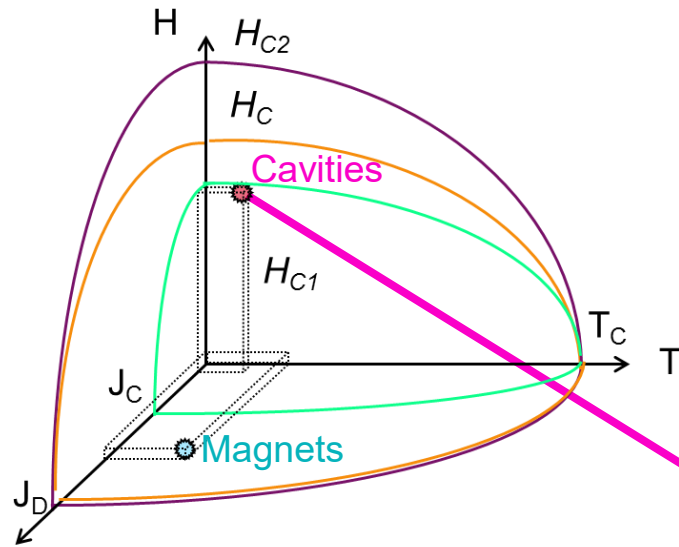


Mixed State, Vortex

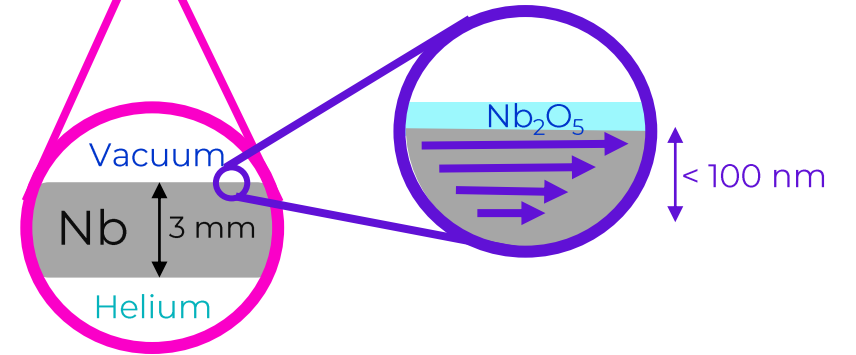


REBCO cable prototype, CERN

# SRF Cavities



Complete Flux Expulsion



RF interact only with 100s of nm on the **surface**

# Different regime means different materials

Good SC for magnet application are bad for cavities!



## Magnets – DC

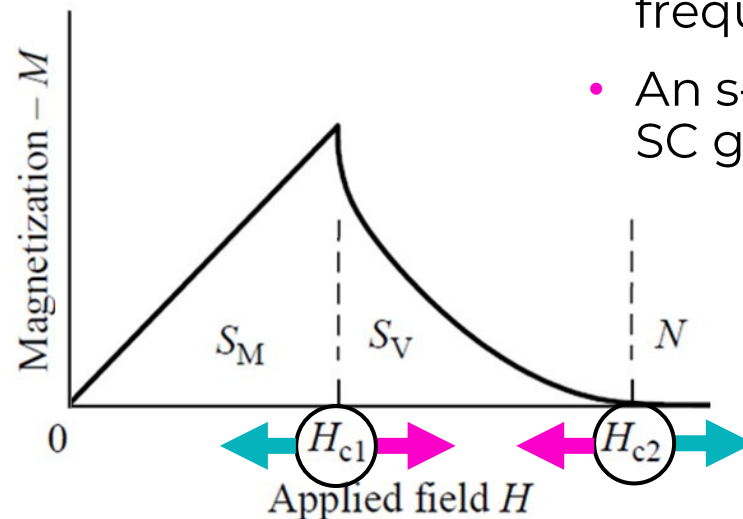
- High current densities with 0 resistance
- Mixed State

**Defects** are voluntarily introduced to **enhance pinning**



## Cavities – RF

- very high field with minimal dissipation (10-20 nΩ @1.3 GHz)
- Vortices cannot keep pinned at this frequency → **Meissner State**
- An s-wave Cooper pairing state with a full SC gap on the entire Fermi surface



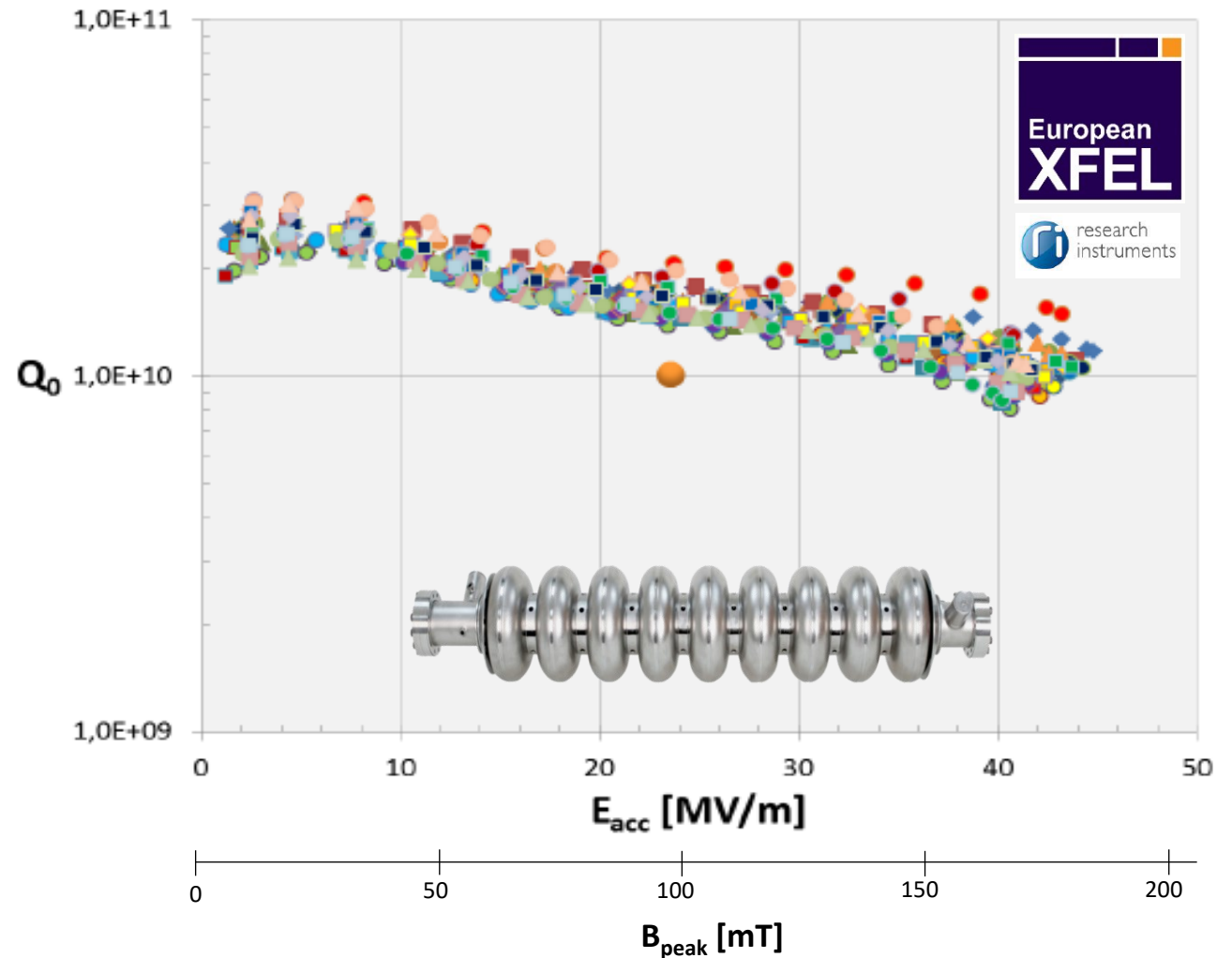
Emphasis is placed on **reducing** the number of **defects**

# State of the art: Bulk Nb

Performances closer to Nb theoretical limits

$H_{SH} \sim 200 \text{ mT}$

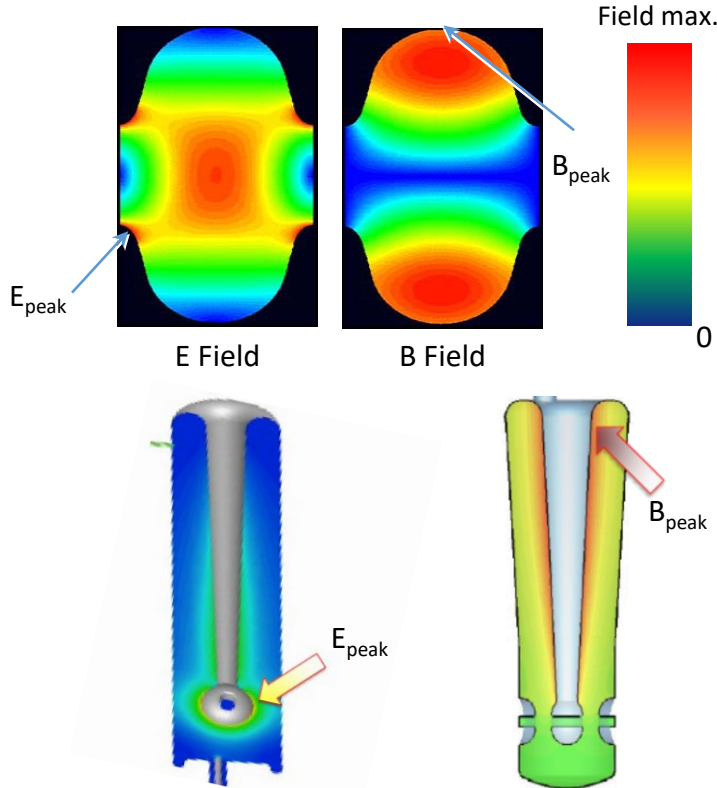
Because **Tc of Nb is 9.2 K**  
SRF cavities are operated at **2 K for High Q**



# Niobium Elliptical SRF Cavities

**Bulk Nb:** monopoly since > 30 years

**Nb/Cu:** applications at low accelerating field only



## Figures of merit:

$$E_{\text{acc}} \propto B_{\text{PEAK}}$$

*limitation : magnetic transition*

$$Q_0 \propto 1/R_S$$

*limitation : thermal transition*

Duty cycle ( $\Rightarrow$  100%)

*limitation : cryogenic power*

$$\beta = \frac{v}{c}$$

*particle speed /light speed (influences design)*

At  $\omega < 3$  GHz: cavities are mainly limited by  $B_{\text{PEAK}}$ !!!

C. Antoine, CEA Saclay

# The right SRF material should provide:

1. **Low surface resistance**, including low residual resistance at  $T \rightarrow 0$ ;
2. An **s-wave Cooper pairing state** with a full superconducting gap on the entire Fermi surface;
3. A **high lower critical magnetic field  $H_{c1}$**  at which the weakly dissipative Meissner state is destroyed due to penetration of vortices;
4. A **high superheating magnetic field** which defines the theoretical limit of the SRF breakdown;
5. **High thermal conductivity** to transfer the rf dissipated power through the cavity wall;
6. **Grain boundaries transparent to high rf** screening currents in polycrystalline cavities;
7. Comparatively **simple chemical composition**, so that the material is not contaminated by nonsuperconducting second phases, and the superconducting properties are not degraded by local chemical nonstoichiometry;
8. **Good mechanical properties and malleability** to minimize crack formation during cavity manufacturing (forging, deep drawing, spinning, etc.)

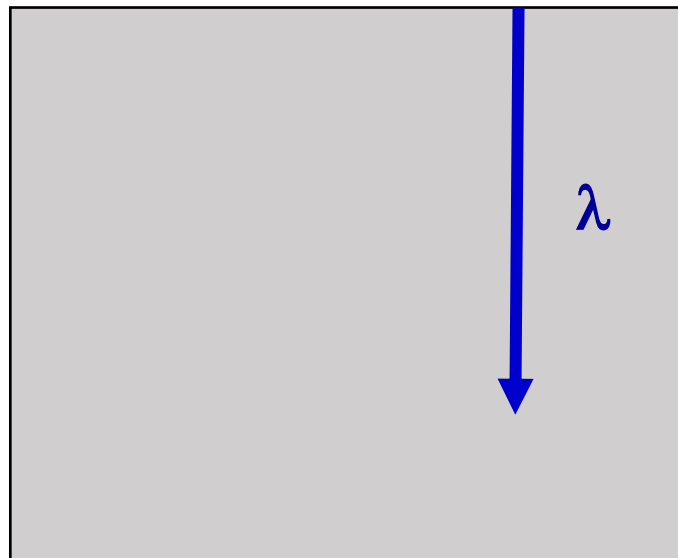
# Superconductors for SRF

Material	$T_c$ (K)	$\mu_0 H_{SH}$ (mT)@ 0 K
Pb	7,1	100
<b>Nb</b>	<b>9,2</b>	<b>219,0</b>
NbN	17,1	214,0
<b>Nb<sub>3</sub>Sn</b>	<b>18,3</b>	<b>425,0</b>
MgB <sub>2</sub>	39,0	170,0
Pnictides Ba <sub>0.6</sub> K <sub>0.4</sub> Fe <sub>2</sub> As <sub>2</sub>	38,0	756,0
Cuprates YBaCuO	93,0	1050,0

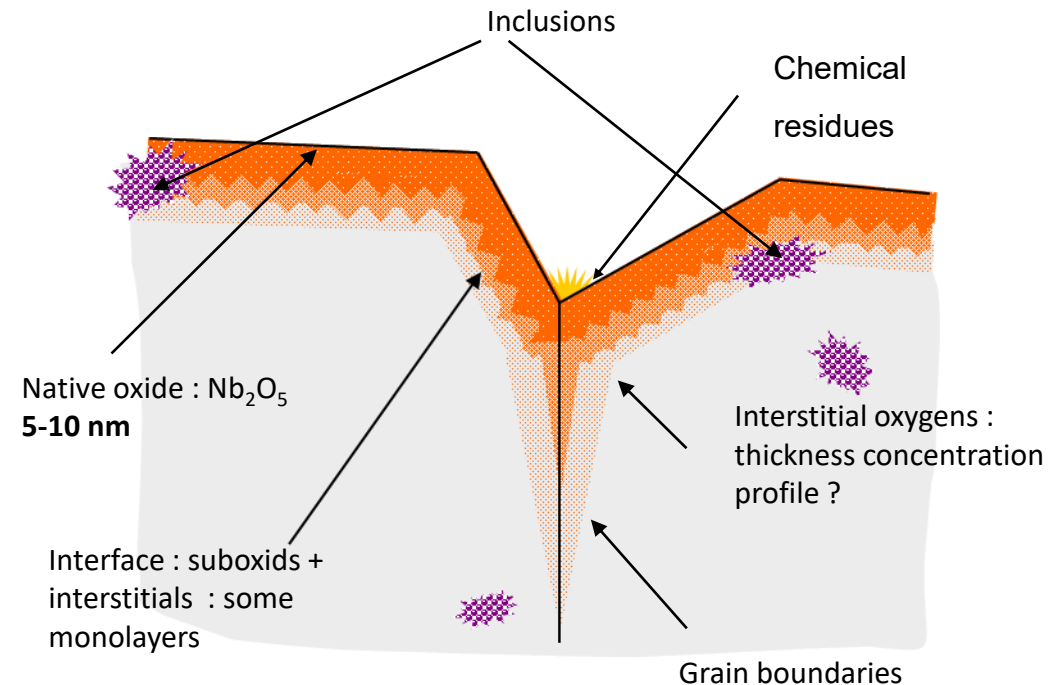


# We need a quasi perfect material on the surface...

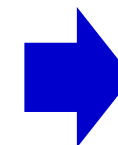
Surface as seen by theoretician...



Real Surface



Even after best surface treatments they are **still a lot of defects at the  $\mu\text{m}$  and  $\text{nm}$  level** that have a **strong influence on SC properties** but are difficult to include in models



**always trust experiments rather than models...**

*C. Antoine, CEA Saclay*

# Nb bulk cavities



# Nb bulk Cavities

State of the art in superconductive resonant cavities

Maximum accelerating field of 50 MV/m close to theoretical limit

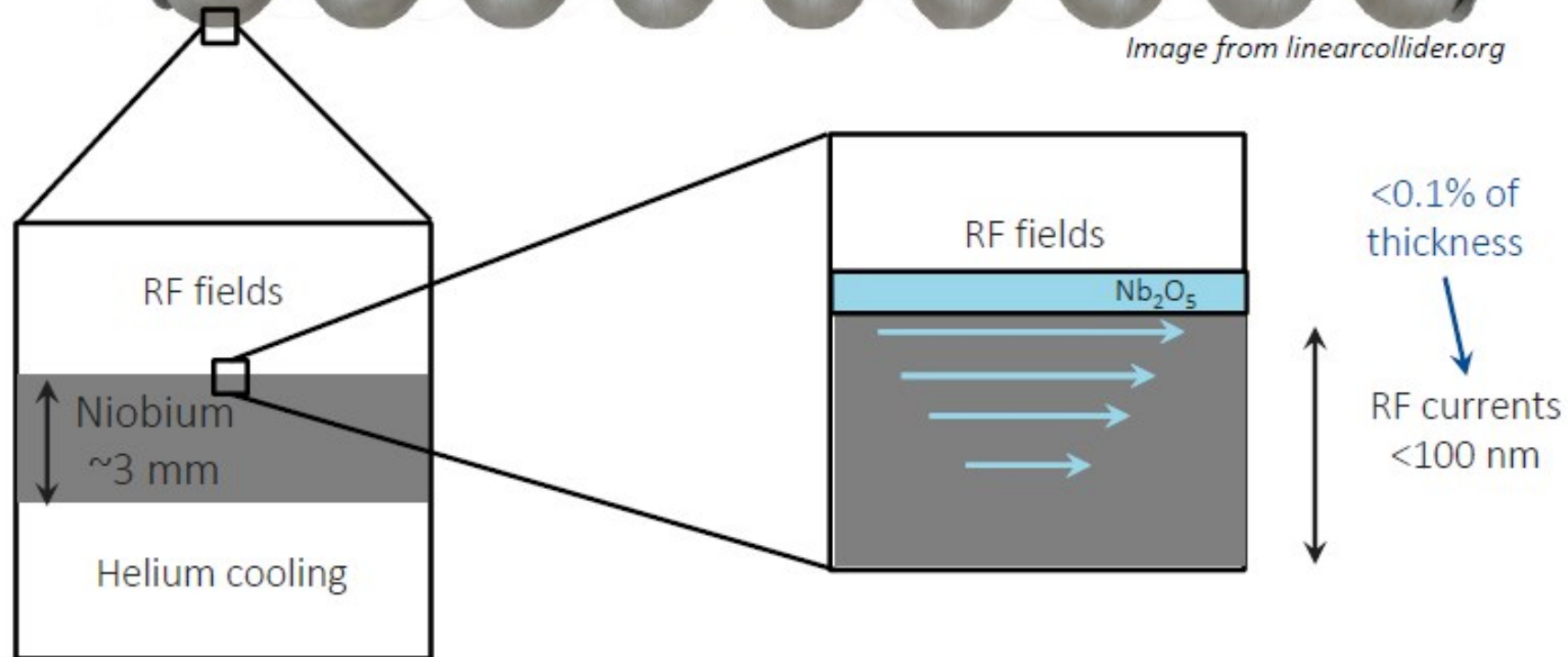
Cost reduction is an issue for the future accelerators



# Performance are determined by nanometer scale structure of inner surface



Image from linearcollider.org

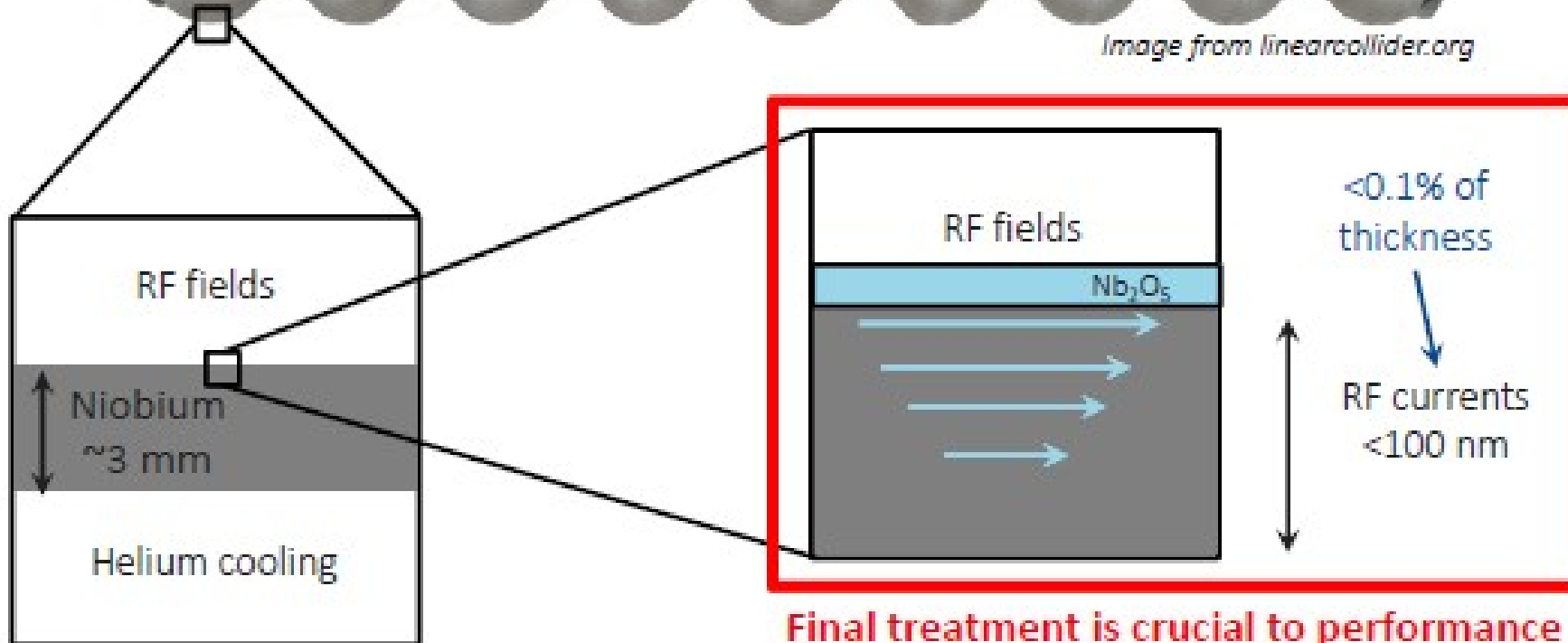


Grassellino, SRF2019 Tutorials

# Performance are determined by nanometer scale structure of inner surface



Image from [linearcollider.org](http://linearcollider.org)



**Final treatment is crucial to performance**

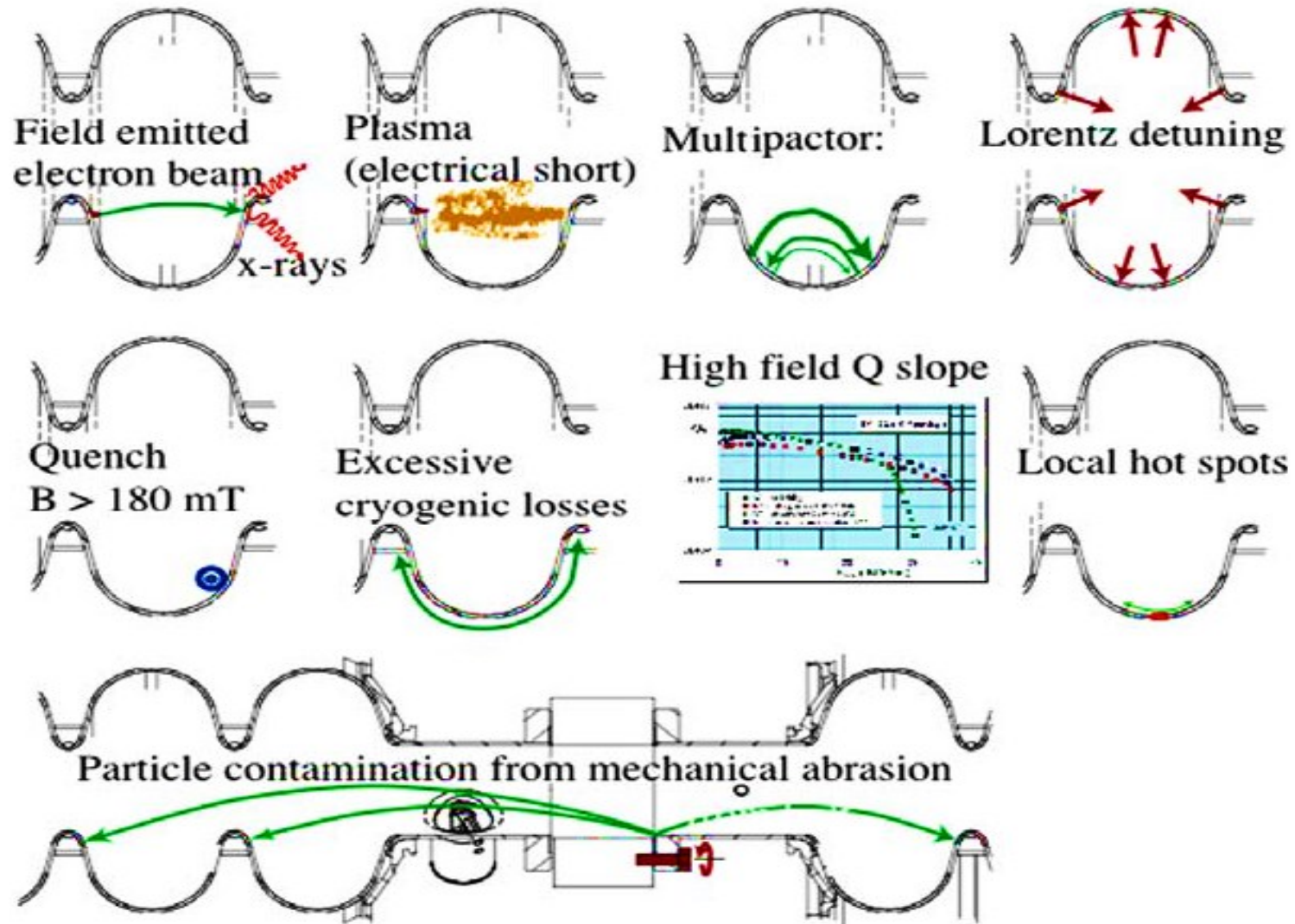
A. Grassellino, SRF2019 Tutorials

# SC cavities production: a long chain, but...

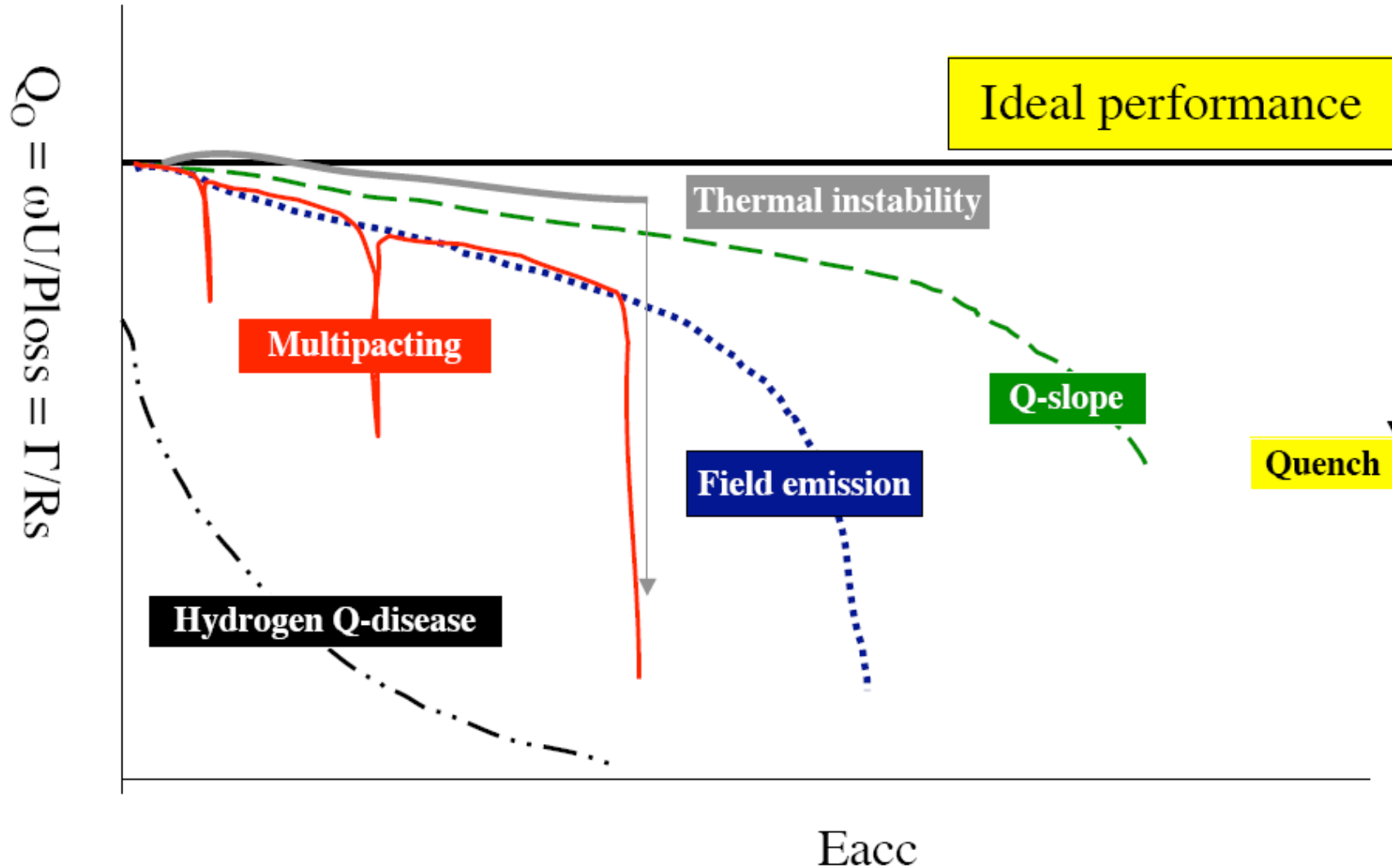
> A chain is as strong as its weakest link !!!

- Chain of
  - Material
  - Fabrication
  - Surface Preparation
    - incl. cleanroom, media, procedures, human factor
  - Vacuum
  - Quality assurance
- For high gradient / low loss SRF cavities all aspects have to be fulfilled

# SC cavities may have various “illness”



# Anomalous loss mechanism

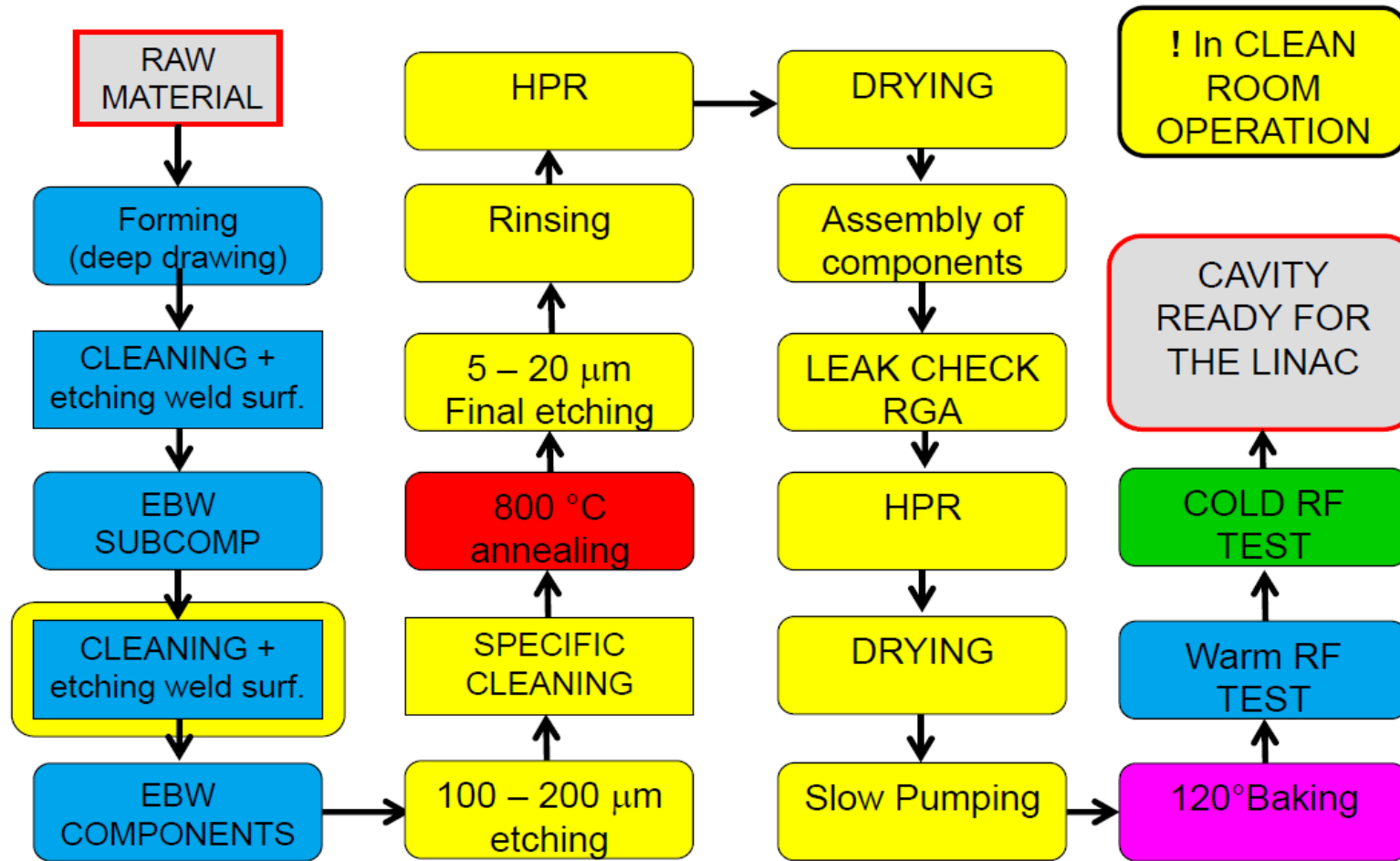




# Some general statements

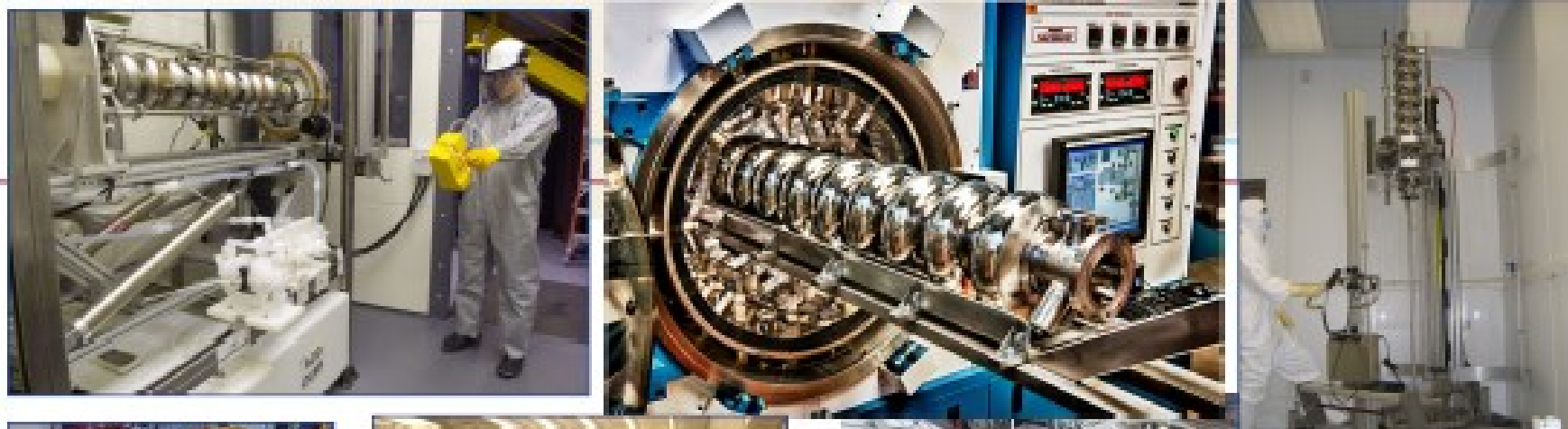
- Anomalous loss mechanisms:
  - **Quench** (local thermal instability)  
=> material + fabrication (=> cleanliness)
  - **Field emission**  
=> Cleanliness of surface treatment, assembly, handling + vacuum
  - **Q-drop** (without field emission) + Q-slope => ?
  - **Multipacting**  
=> Cavity shape + RF surface condition
  - **Hydrogen Q-disease**  
=> Chemical surface treatment
  - Increased residual surface resistance  
=> Cleanliness of surface treatment, assembly, handling + vacuum

# From raw material to RF cavity for EXFEL

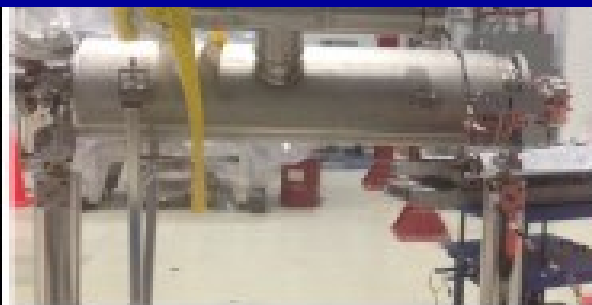


Paolo Michelato, SRF2013 tutorials, September 2013

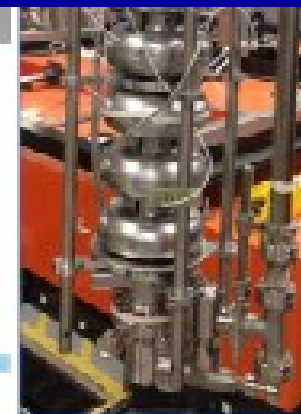




Large scale specialized infrastructure is required to make/study high Q cavities and assemble/test full accelerating cryomodules



Grassellino - SRF 2019  
Tutorials





TRIUMF

SLAC



Argonne  
NATIONAL LABORATORY

FRIB

Fermilab



BROOKHAVEN  
NATIONAL LABORATORY

Jefferson Lab

Science & Technology  
Facilities Council

cea

INFN

HZB  
Helmholtz  
Zentrum Berlin



Institute of High Energy Physics  
Chinese Academy of Sciences



# Nb bulk cavities: fabrication cycle



# From EM Design to Fabrication: few considerations

- From the EM point of view, the cavity is **design at the operative conditions** (usually 2 K, in vacuum, with tuner).
- When you **fabricate** the cavity, you are at **room temperature, in air** and the cavity needs to be **treated** before being operational
- You need then to consider
  - Thermal shrinkage from 2 K to 300 K (geometry, frequency)
  - **Pressure effect** (frequency)
  - **Dielectric constant effect** (frequency)
  - Over-metal for chemical treatment (geometry, frequency)
  - **Pre-tuning** (frequency)

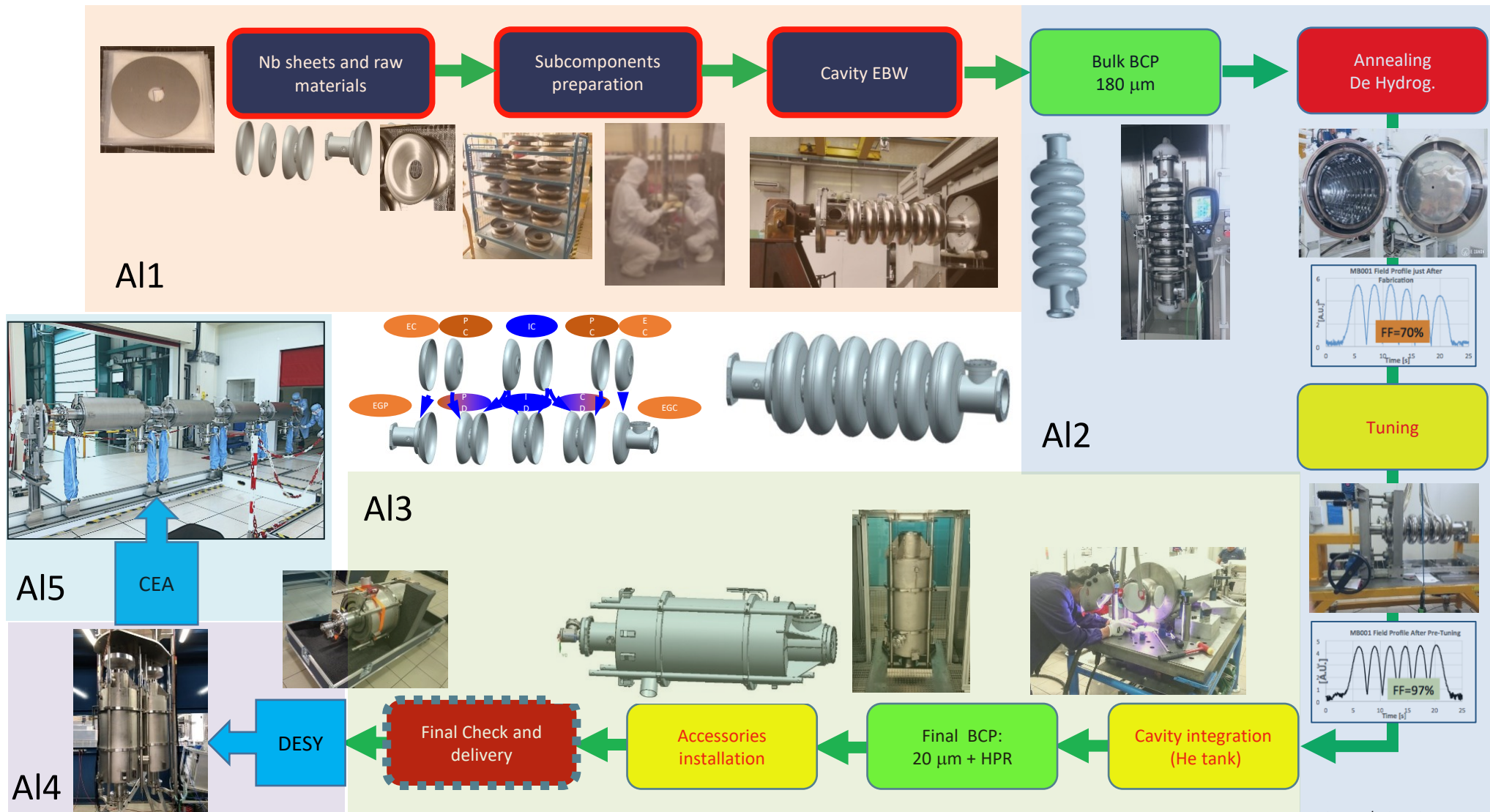
P.S. When you fabricate a cavity, be sure that the design couples with the feasibility of the processes you are going to apply

# The ESS example

Step	Df [MHz]	Cavity Frequency [MHz]	Comment
Goal Frequency		704.420 MHz	2 K in vacuum
Pre-load for tuner	-0.100 MHz	704.320 MHz	Unloaded cavity at cold
Room Temperature	-1.028 MHz	703.292 MHz	Shrinkage from 2 K to 300 K <sup>a</sup>
In Air	-0.234 MHz	703.058 MHz	Dielectric constant <sup>b</sup>
Etching	+0.480 MHz	703.538 MHz	Before Chemistry (150 $\mu\text{m}$ ) <sup>c</sup>
Weld Seam	+0.000 MHz	703.538 MHz	Weld Seam perturbation <sup>d</sup>

- <sup>a</sup> Integral shrinkage from 2 K to 300 K tabulated
- <sup>b</sup> Inversely proportional to square root of dielectric constant
- <sup>c</sup> Estimated from Slater's perturbation on cavity inner surface
- <sup>d</sup> Estimated from direct measurement. More significant for high frequency cavities

# The ESS fabrication cycle



D. Sertore, SRF Cavity Fabrication, EASISchool 3 Genoa, 2020



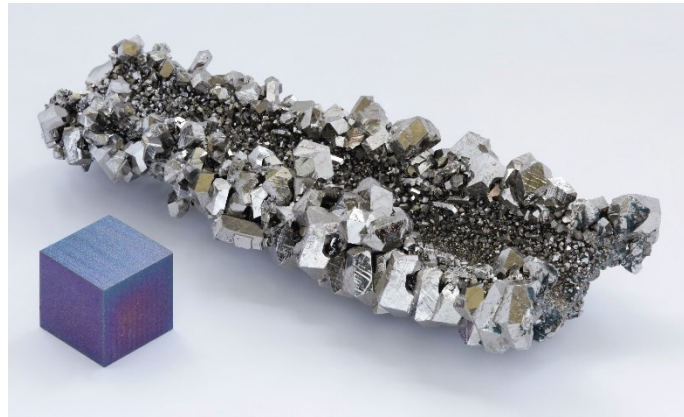
# Nb bulk cavities: from raw Nb to sheets



# Niobium

- Niobium is **THE** material for fabrication of superconducting cavities
  - Critical temperature  $T_c = 9.25$  K
  - High critical field ( $H_c(0\text{ K}) \cong 240$  mT)
  - Chemically inert (surface covered by Niobium pentoxide  $\text{Nb}_2\text{O}_5$ )
  - Easily machined and deep drawn
  - Available as bulk and sheets of any size and different shapes

Highest between pure metals



Los Alamos National Laboratory Chemistry Division

Periodic Table of the Elements

1A	2A											3A	4A	5A	6A	7A	8A
1 H Hydrogen 1.008	4 Be Beryllium 9.012											5 B Boron 10.81	6 C Carbon 12.01	7 N Nitrogen 14.01	8 O Oxygen 16.00	9 F Fluorine 19.00	10 Ne Neon 20.18
3 Li Lithium 6.94	11 Na Sodium 22.99	12 Mg Magnesium 24.31	13 Al Aluminum 26.98	14 Si Silicon 28.09	15 P Phosphorus 30.97	16 S Sulfur 32.06	17 Cl Chlorine 35.45	18 Ar Argon 39.95									
19 K Potassium 39.10	20 Ca Calcium 40.08	39 Y Yttrium 88.91	40 Zr Zirconium 91.22	41 Nb Niobium 92.91	42 Mo Molybdenum 95.96	43 Tc Technetium (98)	44 Ru Ruthenium 101.1	45 Rh Rhodium 102.9	46 Pd Palladium 106.4	47 Ag Silver 107.9	48 Cd Cadmium 112.4	49 In Indium 114.8	50 Sn Tin 118.7	51 Sb Antimony 121.8	52 Te Tellurium 127.6	53 I Iodine 126.9	54 Xe Xenon 131.3
37 Rb Rubidium 85.47	55 Cs Cesium 132.9	56 Ba Barium 137.3	72 Hf Hafnium 178.5	73 Ta Tantalum 180.9	74 W Tungsten 183.9	75 Re Rhenium 186.2	76 Os Osmium 190.2	77 Ir Iridium 192.2	78 Pt Platinum 195.1	79 Au Gold 197.0	80 Hg Mercury 200.5	81 Tl Thallium 204.4	82 Pb Lead 207.2	83 Bi Bismuth 209.0	84 Po Polonium (209)	85 At Astatine (210)	86 Rn Radon (222)
87 Fr Francium (223)	88 Ra Radium (226)	**	104 Rf Rutherfordium (261)	105 Db Dubnium (268)	106 Sg Seaborgium (271)	107 Bh Bohrium (270)	108 Hs Hassium (277)	109 Mt Meitnerium (276)	110 Ds Darmstadtium (281)	111 Rg Roentgenium (280)	112 Cn Copernicium (285)	113 Uut Ununtrium (284)	114 Fl Flerovium (289)	115 Uup Ununpentium (288)	116 Lv Livermorium (293)	117 Uus Ununseptium (294)	118 Uuo Ununoctium (294)
Lanthanide Series*																	
57 La Lanthanum 138.9	58 Ce Cerium 140.1	59 Pr Praseodymium 140.9	60 Nd Neodymium 144.2	61 Pm Promethium (145)	62 Sm Samarium 150.4	63 Eu Europium 152.0	64 Gd Gadolinium 157.3	65 Tb Terbium 158.9	66 Dy Dysprosium 162.5	67 Ho Holmium 164.9	68 Er Erbium 167.3	69 Tm Thulium 168.9	70 Yb Ytterbium 173.0	71 Lu Lutetium 175.0			
Actinide Series**																	
89 Ac Actinium (227)	90 Th Thorium 232	91 Pa Protactinium 231	92 U Uranium 238	93 Np Neptunium (237)	94 Pu Plutonium (244)	95 Am Americium (243)	96 Cm Curium (247)	97 Bk Berkelium (247)	98 Cf Californium (251)	99 Es Einsteinium (252)	100 Fm Fermium (257)	101 Md Mendelevium (258)	102 No Nobelium (259)	103 Lr Lawrencium (260)			

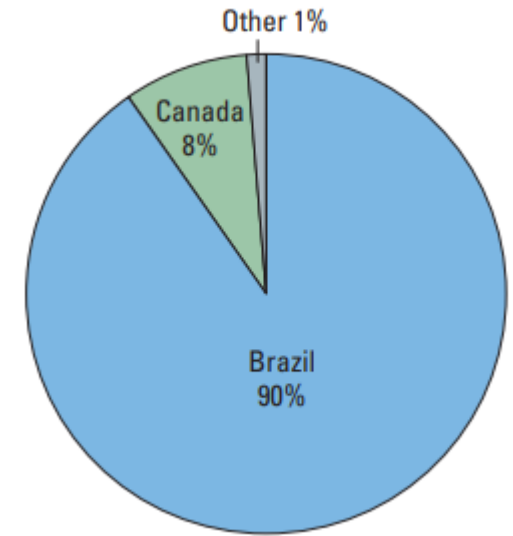
Los Alamos  
NATIONAL LABORATORY

CHEMISTRY

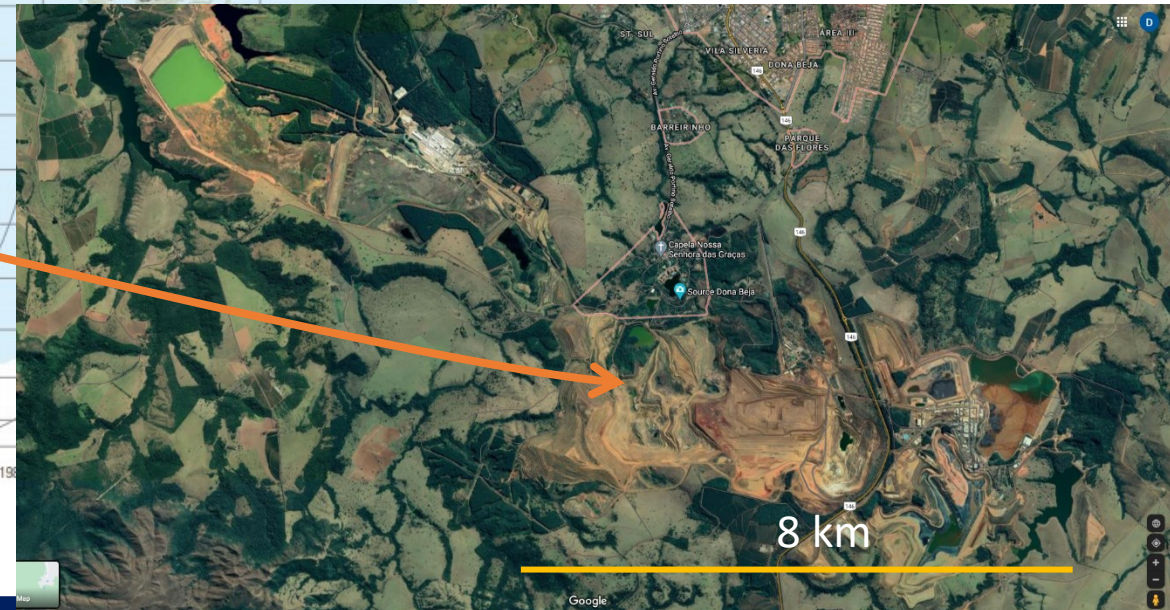
element names in blue are liquids at room temperature  
element names in red are gases at room temperature  
element names in black are solids at room temperature

# Niobium Production

A. Niobium production



CBMM, Arexà, Brazil



Base from U.S. Geological Survey Global 30 arc-second elevation data (1996) and from Natural Earth (2014); Robinson projection; World Geodetic System 1984

# Niobium Production

- The leading use of niobium (about 75 %) is in the production of **high strength steel** alloys used in pipelines, transportation infrastructure, and structural applications.
- Niobium is primarily **derived from the complex oxide minerals** of the pyrochlore group and carbonatites, **usually together with Tantalum**
- The **estimated global reserves** appear more than sufficient to meet global demand for the foreseeable future, possibly the **next 500 years**

# Niobium Production Process

- The ore (Pyrochlore in this case) is treated and **refined** at different stages until it reaches the **purity necessary** for the Electron Beam Refining.
- The **Electron Beam** process **reduces the impurities** present in the incoming Niobium

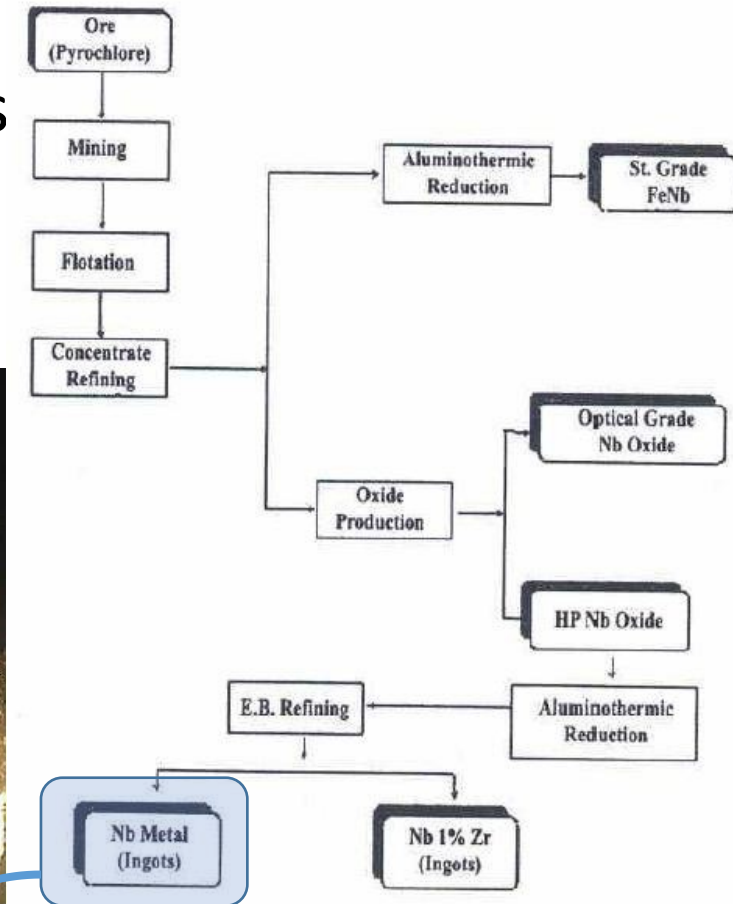


Fig. 3: Production flow chart at CBMM.

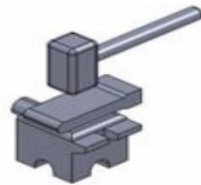
# Niobium sheets production

## 1. Introduction of production process

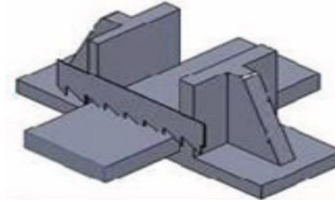
## Nb300 Sheet



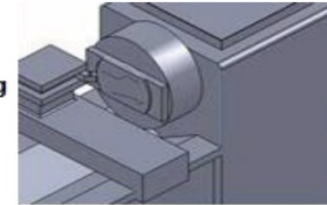
1. Ingot



2. Forging



3. Sawing



4. Mechanical Peeling



5. Rolling



6. Polishing



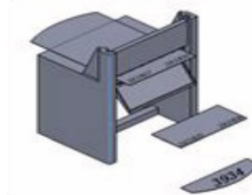
7. Acid Etching



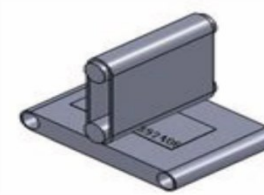
8. Annealing



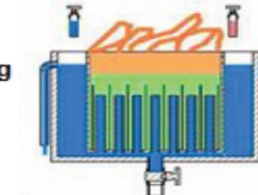
9. Rolling



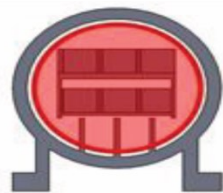
10. Cutting



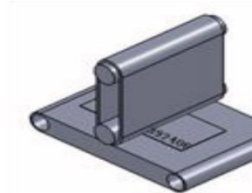
11. Polishing



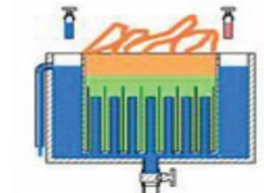
12. Acid Etching



13. Annealing



14. Polishing



15. Acid Etching



16. Inspection & Packing

[www.otic.com.cn](http://www.otic.com.cn)

# Nb Technical Specifications (typical)

Concentration of impurities in wt.ppm				Mechanical properties	
Ta*	≤ 500	H*	≤ 2	Yield strength**, $\sigma_{0,2}$	$50 < \sigma_{0,2} < 100$ N/mm <sup>2</sup> (Mpa)
W*	≤ 70	N*	≤ 10	Tensile strength**	> 100 N/mm <sup>2</sup> (Mpa)
Ti*	≤ 50	O*	≤ 10	Elongation at break**	30 %
Fe*	≤ 30	C*	≤ 10	Vickers hardness** HV 10	≤ 60
Mo*	≤ 50	RRR*	≥ 300	Absence of foreign material inclusions*	Proven by scanning
Ni*	≤ 30	Recrystal. degree. Grain size* ,** ?	≈ 50 μm	Texture *, ** ?	

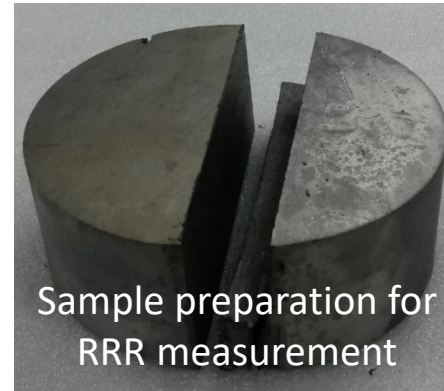
\* - relevant for performance

\*\* - relevant for successful fabrication

# Nb sheets QC at producer premises

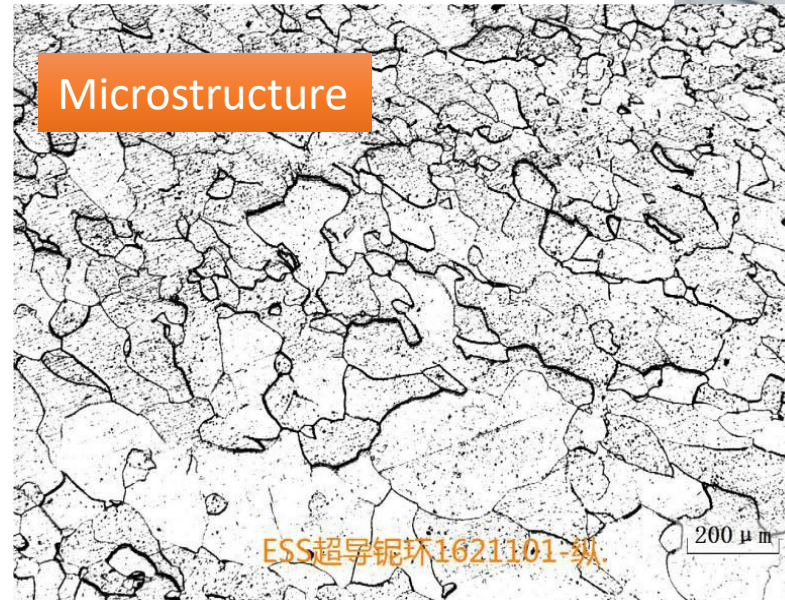
- **Ingot**

- RRR
- Mechanical Properties
- Gas analysis
- Crystallography



- **All sheets**

- Visual Inspection
  - Defects (i.e. scratches)
  - Delamination
- “Rust” test
  - Coarse check for Fe inclusions



Dimension and tolerances		Drawing number		DWG-MB-FG-Materials-WOC Pos.1							
		Size(mm)		CO 160(+2.0)/10130(0-2)/40(+2.0)							
Item No.	1	Annealing Charge No.	162-171019-5								
Ingot No.	ENT-162	RRR value of the ingot	360/385								
Chemical Composition		I: Ingot Analysis	P: Product Analysis								
Content	Ta	W	Ti	Fe	Si	Mo	Ni	H <sub>2</sub>	N <sub>2</sub>	O <sub>2</sub>	C
guaranteed	±0.05%	±0.007%	±0.005%	±0.003%	±0.003%	±0.005%	±0.003%	±2ppm	±10ppm	±10ppm	±10ppm
Result (I)	0.023	-0.0005	-0.0005	0.0005	0.0010	-0.0005	-0.0005	2	10	10	5
Specimen No	1621101							1	8	8	5
	P							-	-	-	-
Specimen No	1623101							2	10	10	5
	Mechanical tests										
Method of test		Hardness test (DIN EN ISO 6507)									
Shape of the test piece		Rectangle specimen									
Specimen condition		Annealed									
Specimen No	Dim of specimen (mm)			Sampling Location	Hardness, HV (min. load 10 N)						
	L	W	H		Requirement	±60					
1621101	10	10	3	*1	Min	54.1					
					Max	59.0					
1623101	10	10	3	*1	Min	53.8					
					Max	58.0					
Method of test		RRR values test									
Shape of the test piece		Cuboid specimen									
Specimen condition		Annealed									
Specimen No	Dim of specimen (mm)			Sampling Location	RRR values						
	L	W	H		Requirement	±300					
1621101 (A-B)	85	3	3.0	*1	-	288/309					
					-	281/313					
1623101 (1R-2R)	85	3	3.0	*1	-	281/313					
					-						

Final Report



# Residual Resistance Ratio (RRR)

- Electrical resistivity of metals at low temperatures is related to the impurity concentrations. The residual resistivity at  $T = 0$  K is caused mainly by scattering of electrons by impurities.

## Residual Resistivity Ratio

$$RRR = \frac{\rho(295 \text{ K})}{\rho(4.2 \text{ K})}$$

- **RRR** depends on **impurity content** in the material (typical RRR for cavity is around 300)
- **RRR** is linked to the Nb **thermal conductivity** by

$$\lambda(4.2 \text{ K}) \approx 0.25 RRR \left[ \frac{\text{W}}{\text{m K}} \right]$$

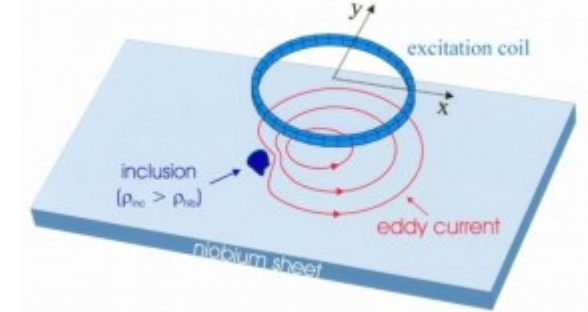
# Nb Eddy Current Scanning

When an **AC current flows** in a coil in close **proximity** to a **conducting surface** the magnetic field of the coil will **induce circulating (eddy) currents** in that surface.

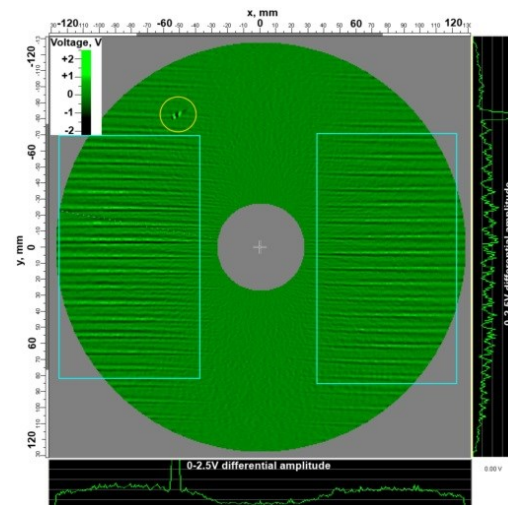
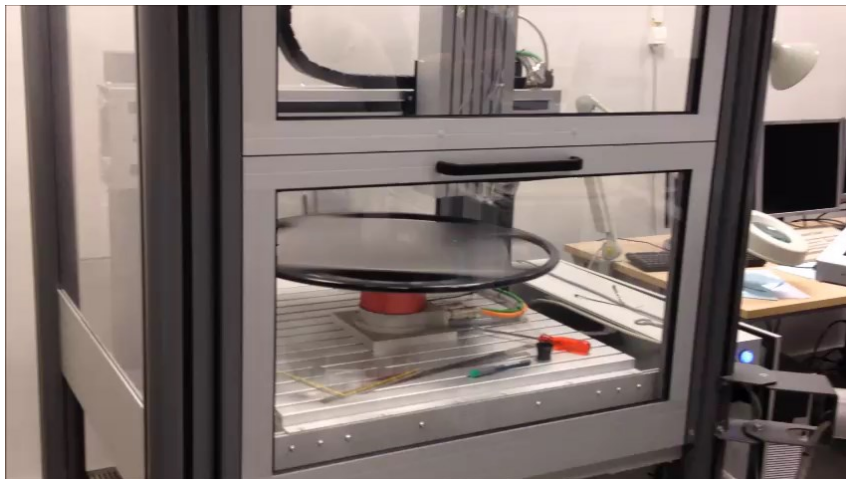
The **magnitude and phase** of the eddy currents will **affect the loading** on the coil and thus its impedance. If there is a **deep crack in the surface** immediately underneath the coil, it will **interrupt or reduce the eddy current flow**, thus decreasing the loading on the coil and increasing its effective impedance.

The operating frequency is between 100 kHz to few MHz and can span from the surface down into the materials for **some hundreds of microns**.

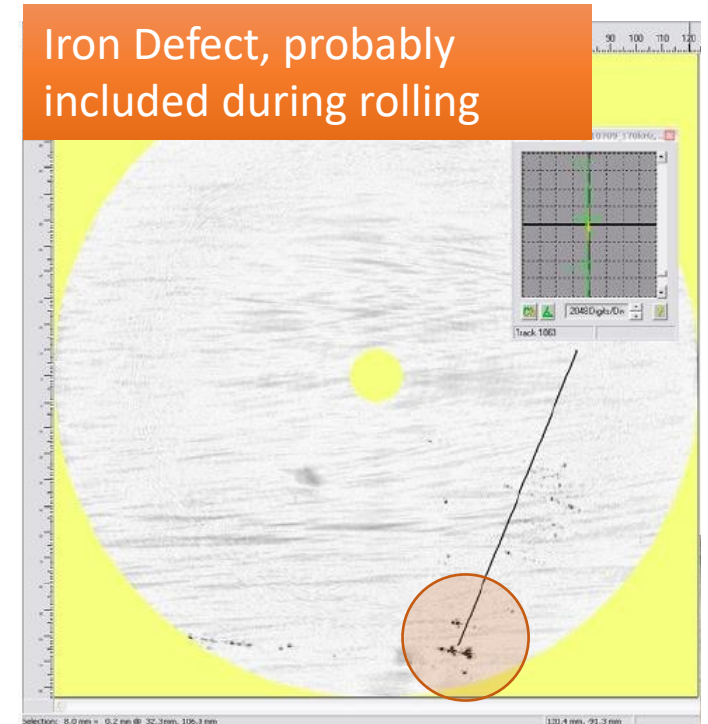
DESY Eddy Current principle



Principle of eddy current measurement



Iron Defect, probably included during rolling

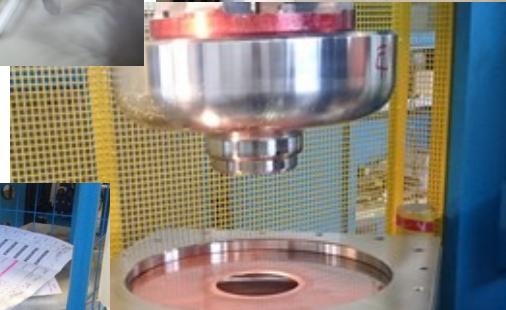
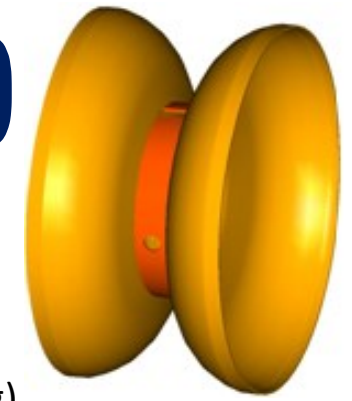


# Nb bulk cavities: from sheets to cavity





# Dumb-bell Fabrication (an example)



1. Nb sheets cutting
2. Deep drawing
3. Mechanical measurement
4. Cleaning (by ultrasonic cleaning +rinsing)
5. Trimming of iris region and reshaping of cups if needed
6. Cleaning
7. Rf measurement of cups
8. Buffered chemical polishing + rinsing (for welding of Iris)
9. Welding of Iris
10. Welding of stiffening rings
11. Mechanical measurement of dumb-bells
12. Reshaping of dumb bell if needed
13. Cleaning
14. Rf measurement of dumb-bell
15. Trimming of dumb-bells ( Equator regions )
16. Cleaning
17. Intermediate chemical etching ( BCP /20- 40  $\mu\text{m}$  ) + rinsing
18. Visual Inspection of the inner surface of the dumb-bell
19. Local grinding if needed + (second chemical treatment + inspection )

# Mechanical QC



Half Cell dime control



3D profile

E. ZANON											
Doc No		180929									
Remarks											
	M10	M11									
max	373.20	200.00	373.20	147.00	2.00	2.00	148.00	0.00	0.00	0.60	0.60
0°	372.11	99.81	372.36	147.23	2.55	2.52	147.58	0.46	0.17	0.68	0.45
90°	372.28	99.88	372.30	146.98	2.48	2.59	148.06				
180°				146.77	2.52	2.42					
270°				147.14	2.41	2.40					
Result	not good	good	not good	good	good	good	good	good	good	good	good
Average	372.20	99.85	372.33	147.03	2.49	2.48	147.82	0.46	0.17	0.68	0.45
Min-Max	0.17	0.07	0.06	0.46	0.14	0.19	0.48	0.00	0.00	0.00	0.00
Instructions											
Rose cells indicates measurements to be taken for each dimensions.											
Keep the IC with the lower serial number always in the bottom position!											
If shape 3D measurements are not done insert n.a. for M10 or M11											
Shape 2D measurements: good or not good											
Remarks on shape											
M10 - points with shape between 0.6 and 0.8 (max 10%) [%]											
1.32											
M11 - points with shape between 0.6 and 0.8 (max 10%) [%]											
0											
Inner shape in tolerance?											
	shape 2D	shape 3D									
M10	good	good									
M11	good	good									
Inspection	10/1/2019 11:07	MARAGNO NICOLA									
Description / Notes	Date	Name									

Mechanical measurements



Visual Inspection

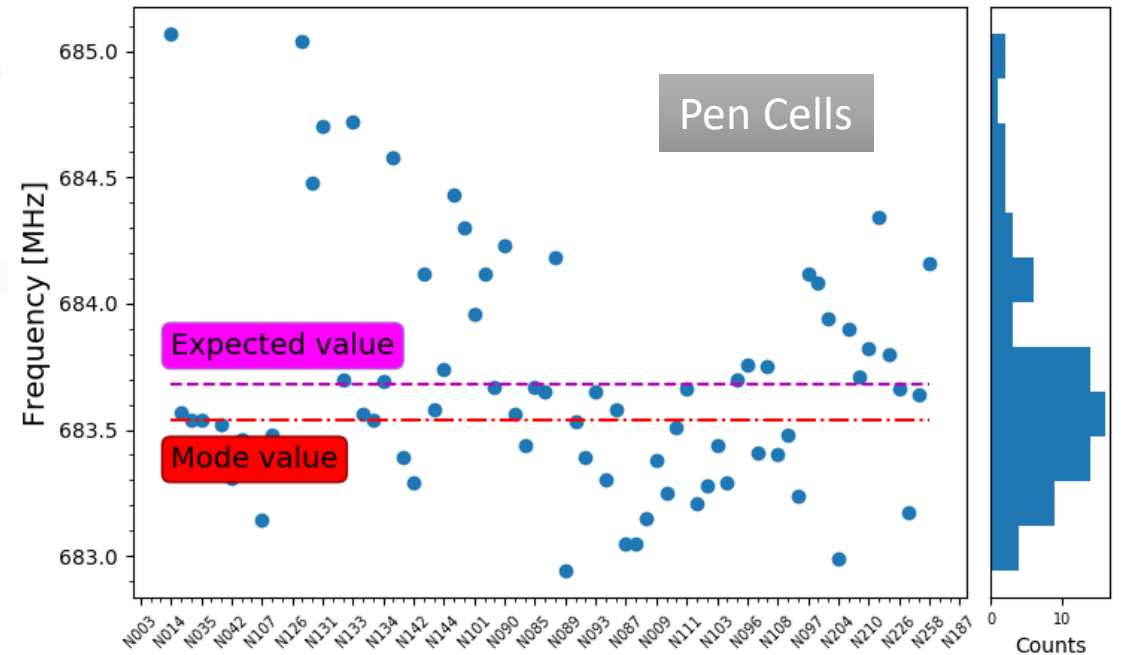
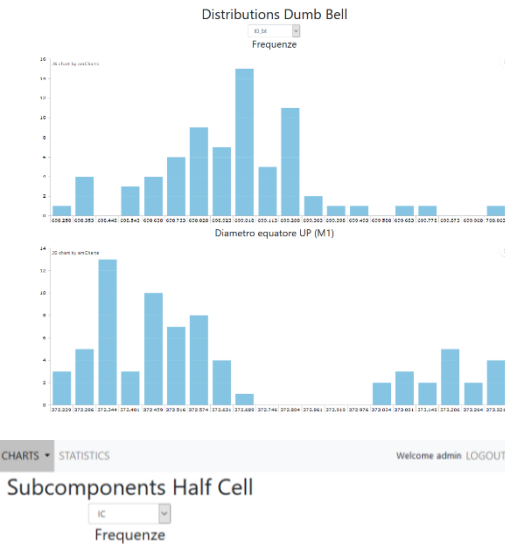
# Frequency QC

Measure frequency and length to determine how much to **trim** at the equator to obtain **target frequency and length** of the cavity fully welded



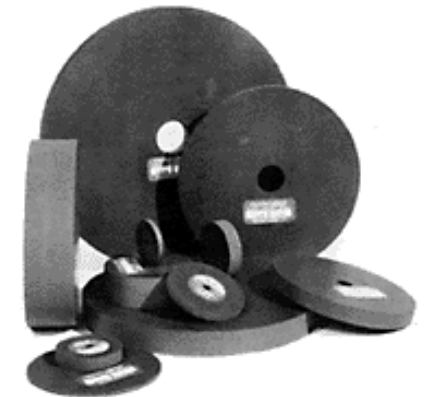
Co.Co.De for HC and DB RF measurements

Mfr Work No.	3282	QCP No.	3282.F.001	Step No.	205	Doc No.	170467																																																																																								
<b>Inspection sheet V_M01</b> Dimensional Control of Inner Cell Controllo dimensionale Inner Cell																																																																																															
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# Mechanical Grinding

**Mechanical grinding** of visible **local defects** (deeper than  $15\ \mu\text{m}$  for EXFEL) with aluminum oxide grinding discs or rubberized abrasive (CRATEX<sup>®</sup>)

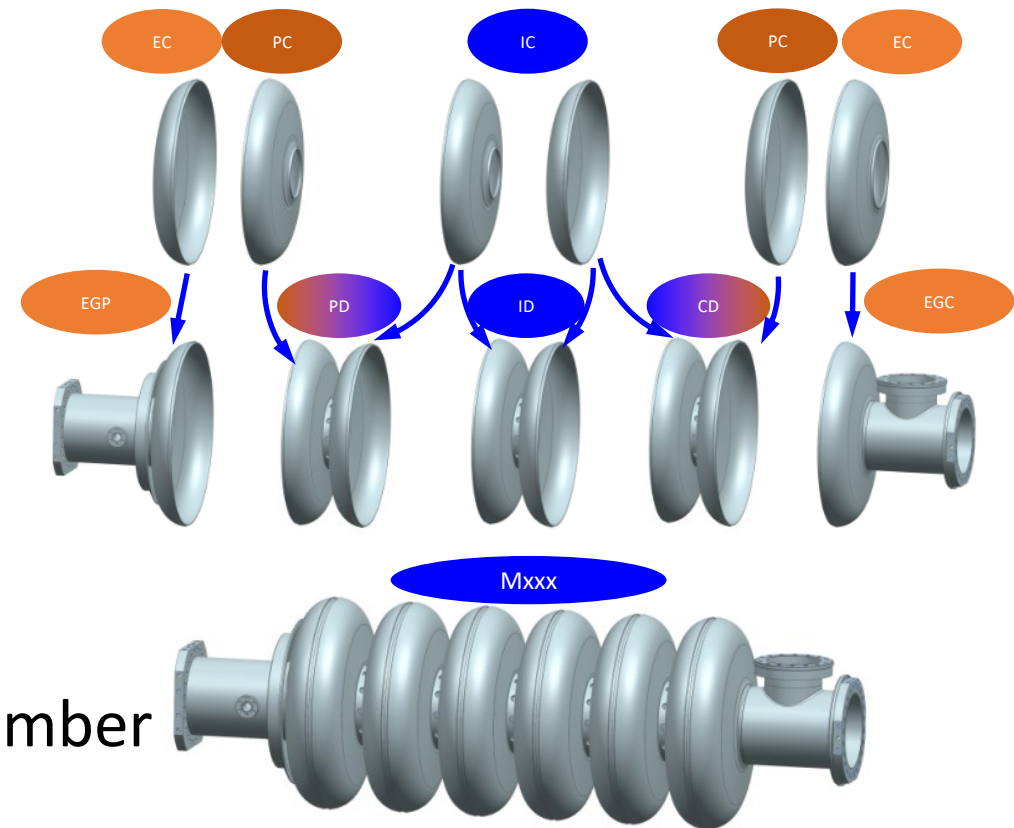


Rubberized abrasive



# Cavity parts EB welding

- Degreasing and rinsing of parts
- Drying under clean condition
- Chemical etching at the welding area (equator)
- Careful and intensive rinsing with Ultra Pure Water
- Dry under clean conditions
- Install parts to fixture under clean conditions
- Install parts into Electron Beam (EB) welding chamber (**no contamination on the weld area allowed**)
- Pump down to vacuum in the EBW chamber in the  $10^{-5}$  mbar range
- Welding and cool down of Nb to  $T < 150$  °C, then venting with  $N_2$
- Leak check of welds

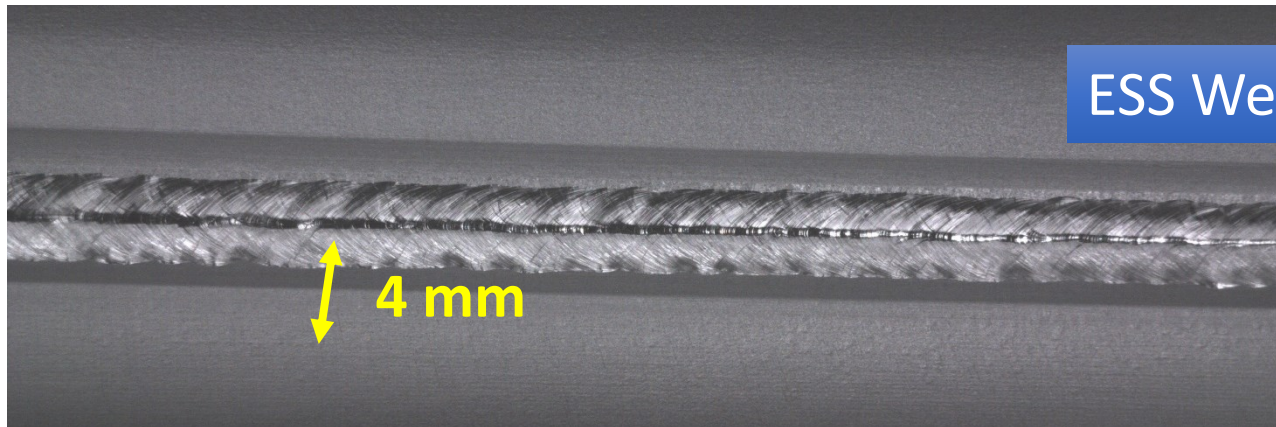


# Electron Beam Welding

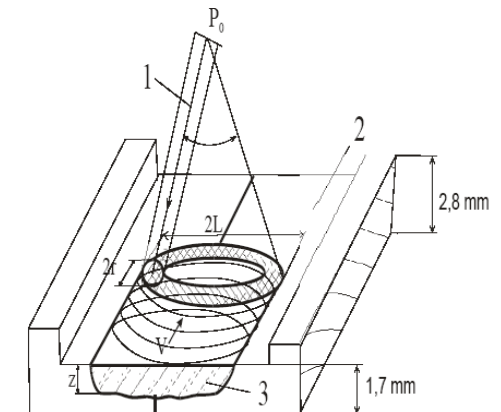
- **Welding under good vacuum**,  $10^{-5}$  mbar range
- Broad welding seam
  - Operate with defocussed beam
  - Smooth underbead
- **Overlap at end of welding** to avoid accumulation of impurities
- **Wait to cool down before opening chamber**



JLAB EBW



ESS Weld Seam



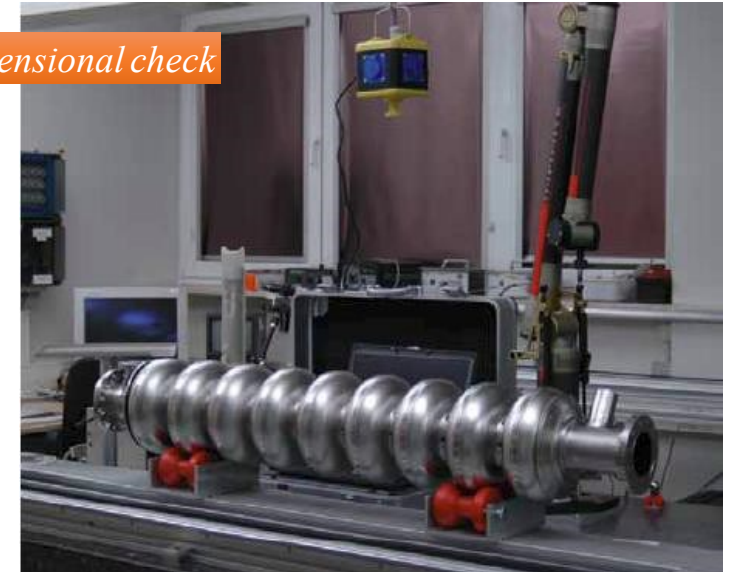
## Welding Scheme (circular raster)

1. Electron beam ( $P_0$ -power of the beam,  $r$ -spot radius on the surface,  $L$ -scanning amplitude,  $V$ -velocity of the beam movement)
2. Nb sheet
3. melting zone ( $z$ -depth of the melting zone)

# Mechanical and frequency QC

After EBW, the cavity is mechanical measured, inner inspected and the frequency is controlled

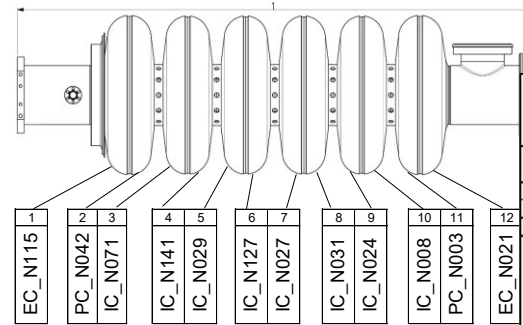
Dimensional check



Mfr' Work No.		QCP No.		Step No.		Doc No	
Comessa		3282.F.003		2		178835	
foreseeable* length [mm]		Remarks					
Maximum value	1262.40	* before equator welding the foreseeable cavity length after mechanical fabrication					
Minimum value	1256.40						
Expected value	1257.10						
Measured value	1256.41						

Inspection		ESS MB Cavity	
Description / Notes	Date	Name	3282.1.000.000
Inspection	25/10/2018 9:43	BATTISTA VINCENZO	



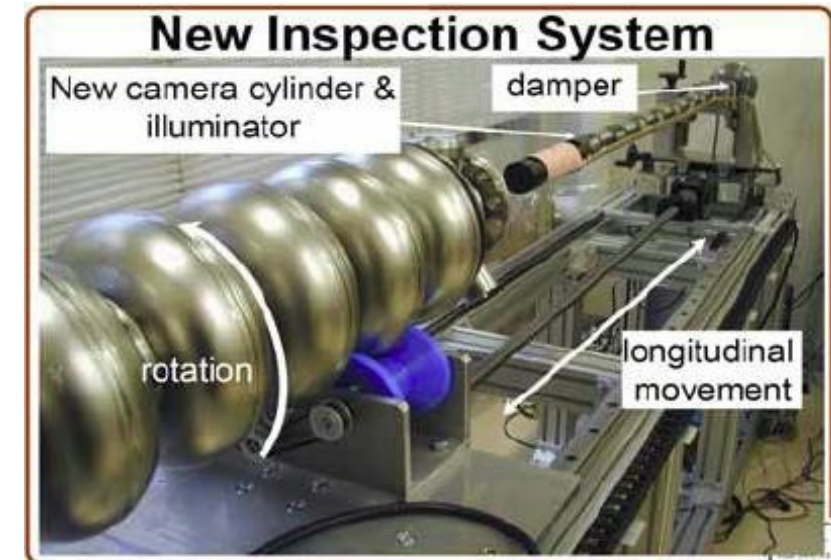
Inspection sheet X_F01		Modes						Measurement Condition	
FF and RF spectrum after Equatorial Welding		1π/6	2π/6	3π/6	4π/6	5π/6	π	cavity open, w/o frame, in air, HOR	
Frequency [MHz]		693.432	695.398	698.111	700.741	702.639	703.588		
Frequency @ 22 °C [MHz] (calc.)		693.438	695.404	698.117	700.747	702.645	703.594		
T [dB]		-43.048216	-37.360483	-34.301687	-32.596058	-32.49489	-36.058278		

Frequency [MHz]		Modes	
706.0		1π/6	π
704.0			
702.0			
700.0			
698.0			
696.0			
694.0			
692.0			
690.0			
688.0			

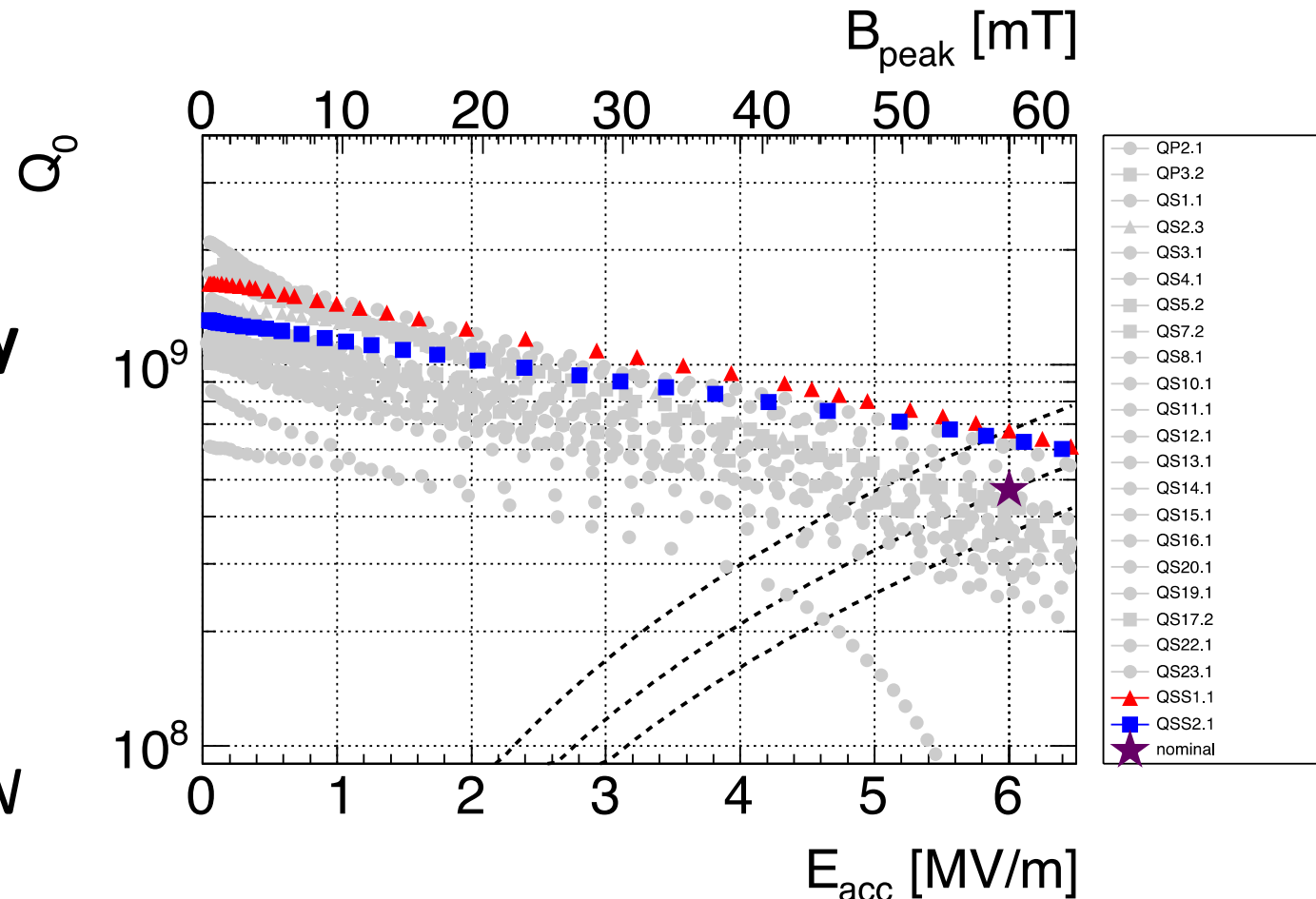
  

Temperature [°C]		Pressure [mbar]		Umidity [%]		ESS MB Cavity		M001	
23.0	976	55							
Frequency measurement	29/10/2018 11:53	D. Settore							
Description / Notes	Date/ timestamp	Name						Serial-No.	



# Advantages of seamless cavities

- Cheaper
- Avoid defects and irregularity of welding seams
- Increase RF performances  
*(real examples of ALPI @ INFN and HIE-ISOLDE @ CERN)*



HIE ISOLDE two seamless cavities performance at 4.5 K

Courtesy of Walter Venturini

# Seamless cavities by spinning

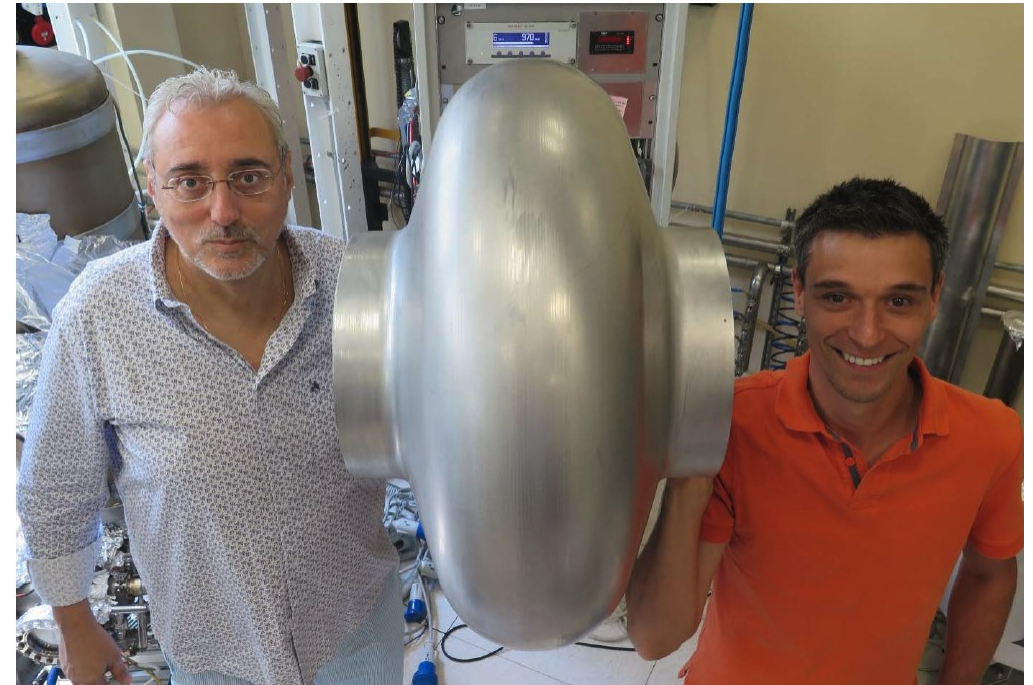
Hydroforming, explosive forming, **electroforming**, **electrodeposition** and **spinning** are the principal techniques explored for the production of seamless elliptical cavities

LNL have a long experience in spinning of 1,3 and 1,5 GHz elliptical cavities

Process invented and developed by Enzo Palmieri in the 90's



*First seamless multicell by spinning*



*400MHz Seamless Aluminum Cavity prototype (2017)*

# Spinning production steps

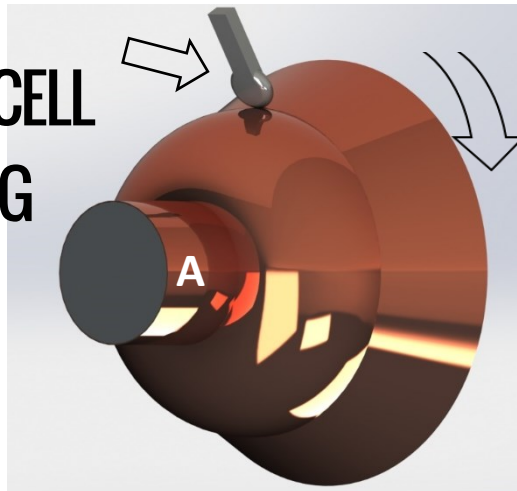
**Step 1**  
COPPER PLATE  
PREPARATION



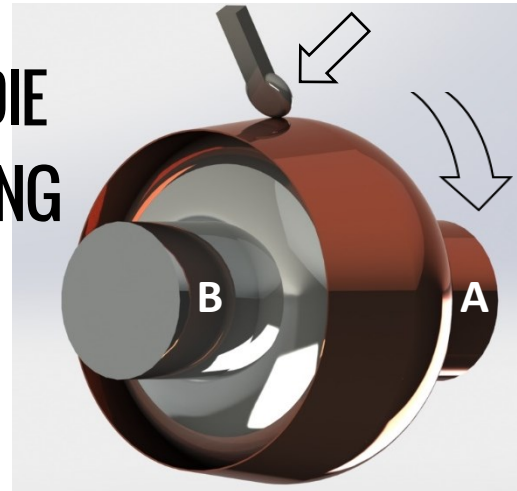
**Step 2**  
DEEP DRAWING



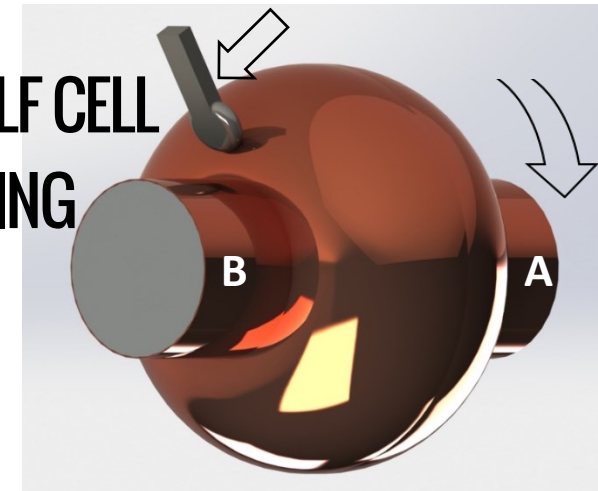
**Step 3**  
1<sup>st</sup> HALF CELL  
SPINNING



**Step 4**  
CONE DIE  
SPINNING



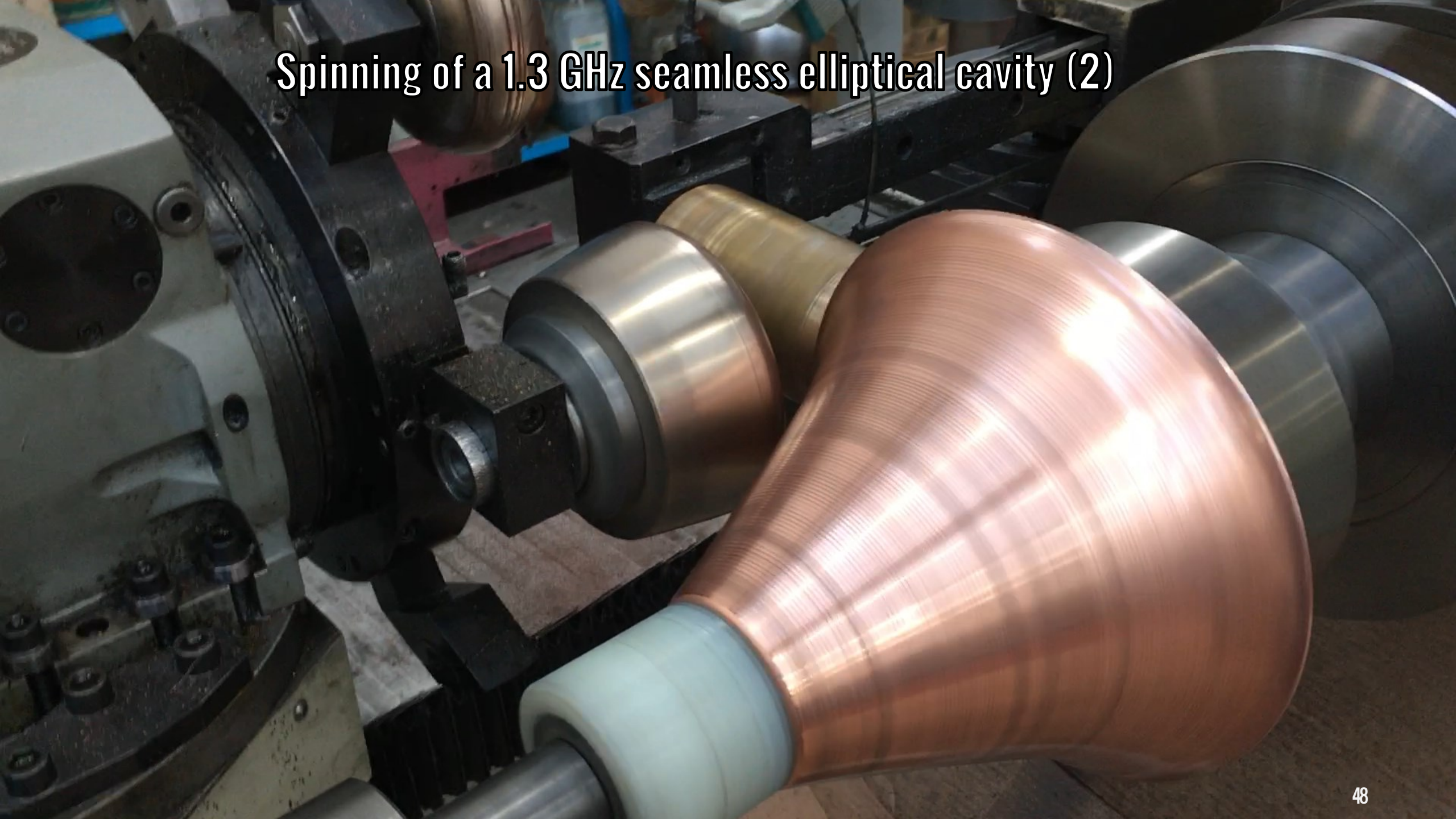
**Step 5**  
2<sup>nd</sup> HALF CELL  
SPINNING



# Spinning of a 1.3 GHz seamless elliptical cavity (1)



# Spinning of a 1.3 GHz seamless elliptical cavity (2)





## Spinning of a 1.3 GHz seamless elliptical cavity (3)



# Spinning of a 1.3 GHz seamless elliptical cavity (4)



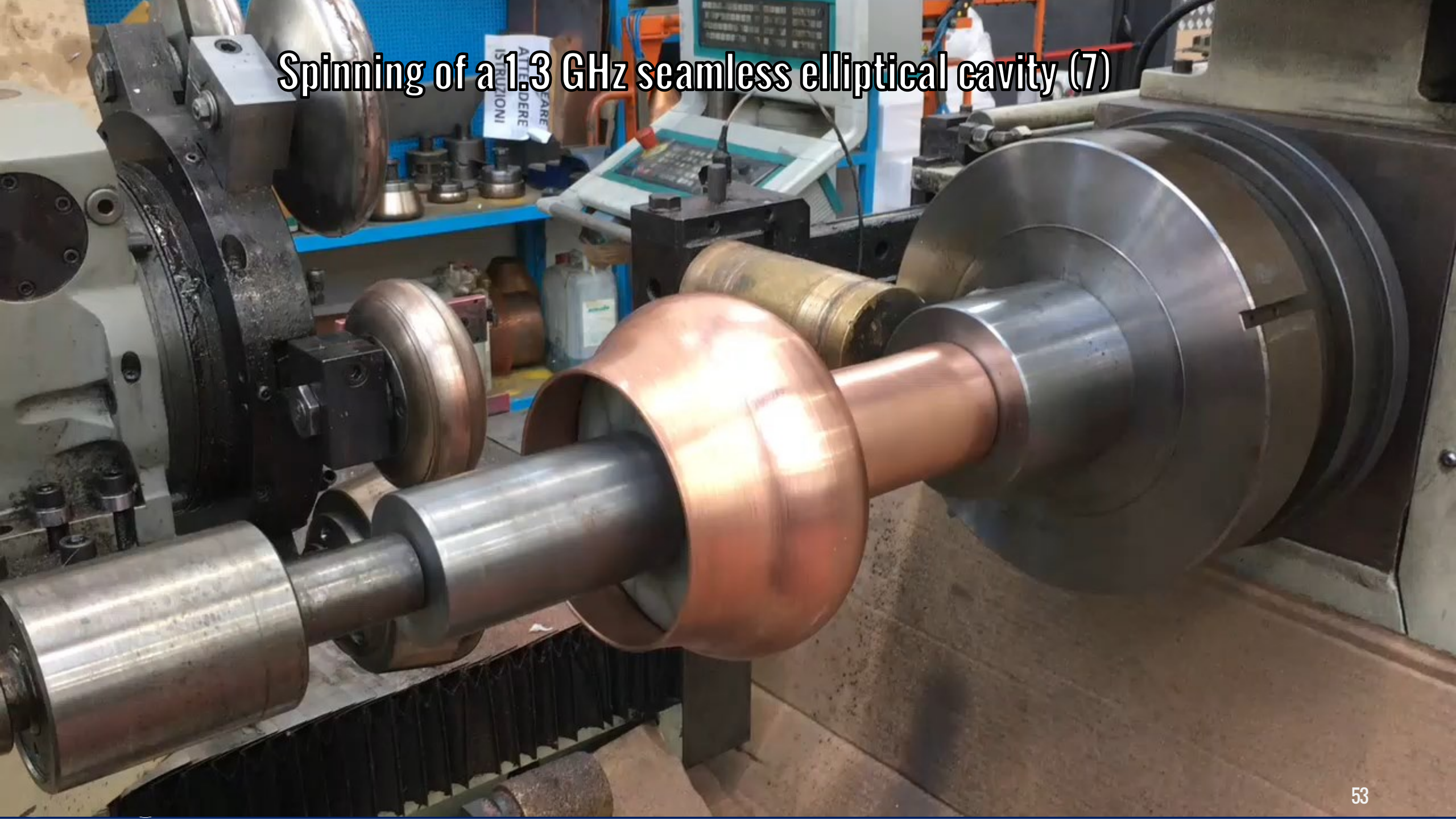
# Spinning of a 1.3 GHz seamless elliptical cavity (5)



# Spinning of a 1.3 GHz seamless elliptical cavity (6)



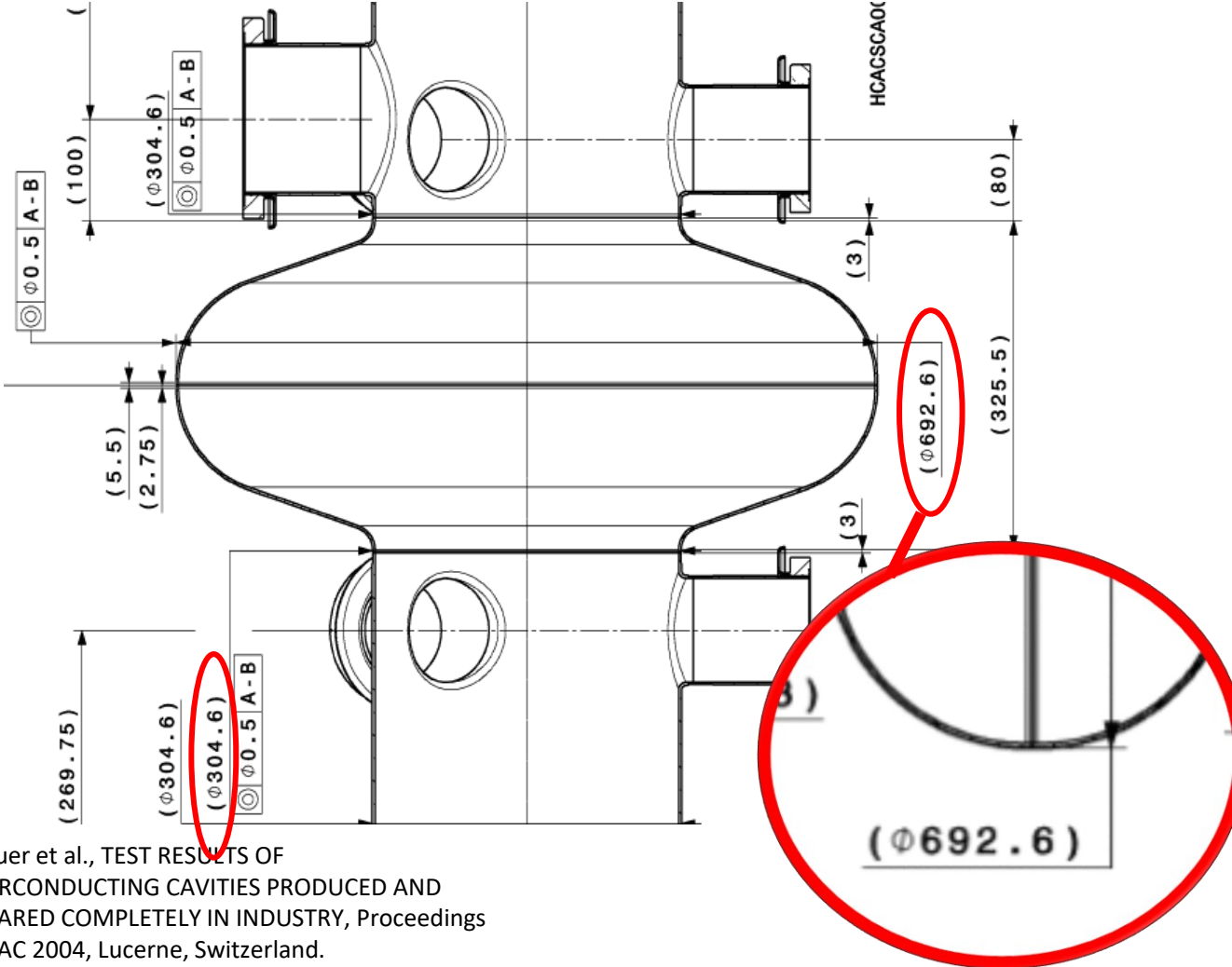
# Spinning of a 13 GHz seamless elliptical cavity (7)



# Spinning steps



# A 400 MHz elliptical cavity challenge (FCC studies)



S. Bauer et al., TEST RESULTS OF SUPERCONDUCTING CAVITIES PRODUCED AND PREPARED COMPLETELY IN INDUSTRY, Proceedings of EPAC 2004, Lucerne, Switzerland.

# A 400 MHz elliptical cavity challenge (FCC studies)





# STEP 1: OFE Copper sheet

- Dimension out of standard
- Difficulties to provide a cold rolled sheet
- **Thermal treated** (Hardness HV = 46)
- Received @ LNL on June 2018



Abnahmeprüfzeugnis DIN EN 10204-3.1  
Inspection Certificate DIN EN 10204-3.1



Besteller / Purchaser:

CERN  
ORGANISATION POUR LA RECHERCHE

1211 Geneve  
CH

**MKM Mansfelder Kupfer und  
Messing GmbH**

Lichtlöcherberg 40  
06333 Hettstedt / Germany  
Produktbereiche Bleche / Bänder

Überprüft nach AD 2000-Merkblatt W0/TRD100  
durch den TÜV Hannover/Sachsen-Anhalt e.V.  
Zertifiziert nach Richtlinie 97/23/EG durch die  
TÜV CERT-Zertifizierungsstelle für Druckgeräte der  
TÜV Nord Gruppe, Notifizierte Stelle, Kern-Nr.: 0045  
phone: +493476892473  
fax: +493476892403  
e-mail: Tom.Burchert@mkm.eu

# STEP 2 Deep Drawing



IRON DIES



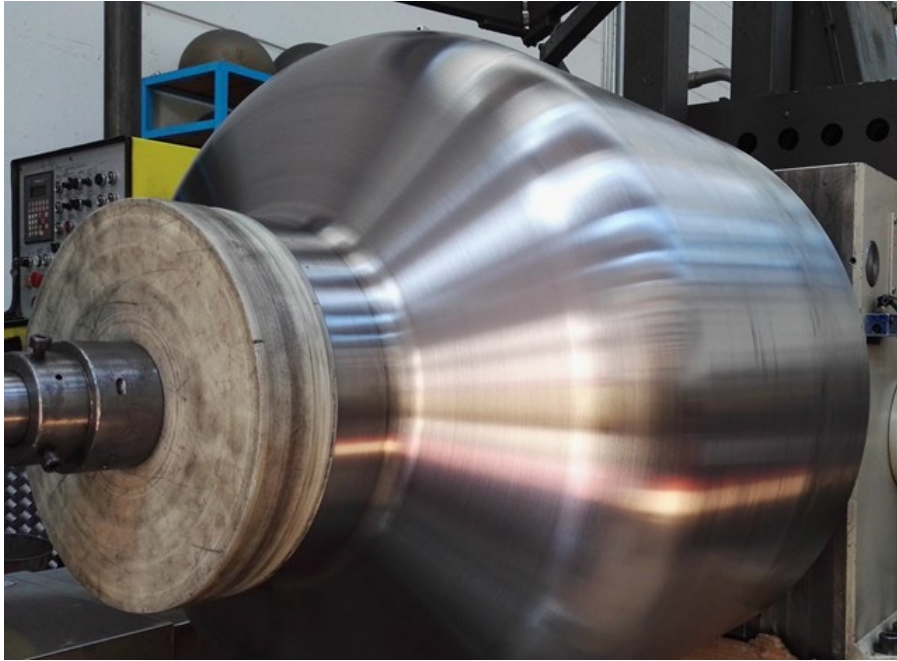
Ø 1150 mm



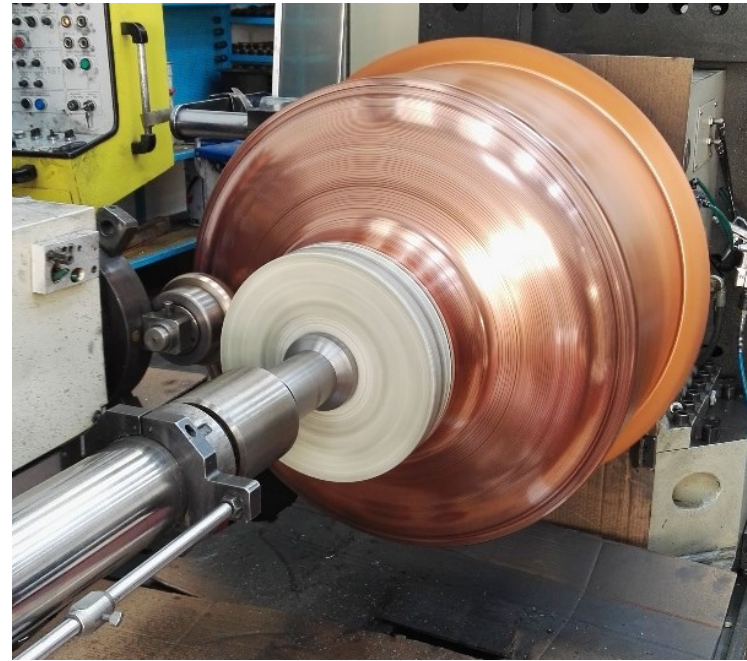
385 mm

Ø 693 mm

# STEP 3 Spinning of 1<sup>st</sup> Half Cell



IRON DIE



SPINNING OF THE FIRST HALF CELL



FIRST HALF CELL SPUN

- No thermal treatments are necessary for the spinning of the 1° half cell

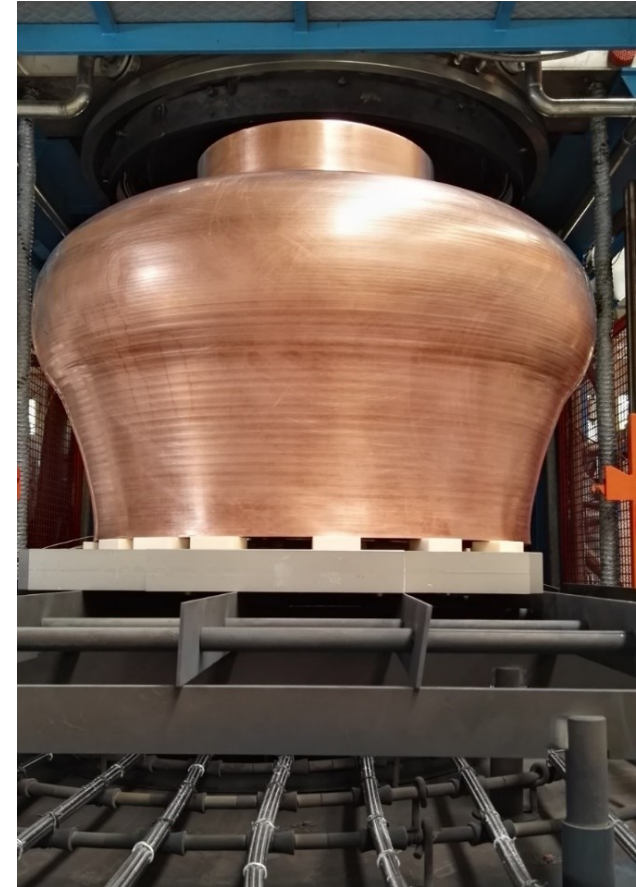
# STEP 4 Spinning of the intermediate conical die



NYLON CONICAL DIE



LOW SPINNABILITY



FIRST THERMAL ANNEALING

- A **thermal treatment** is necessary for the spinning of the conical die

# Thermal annealing

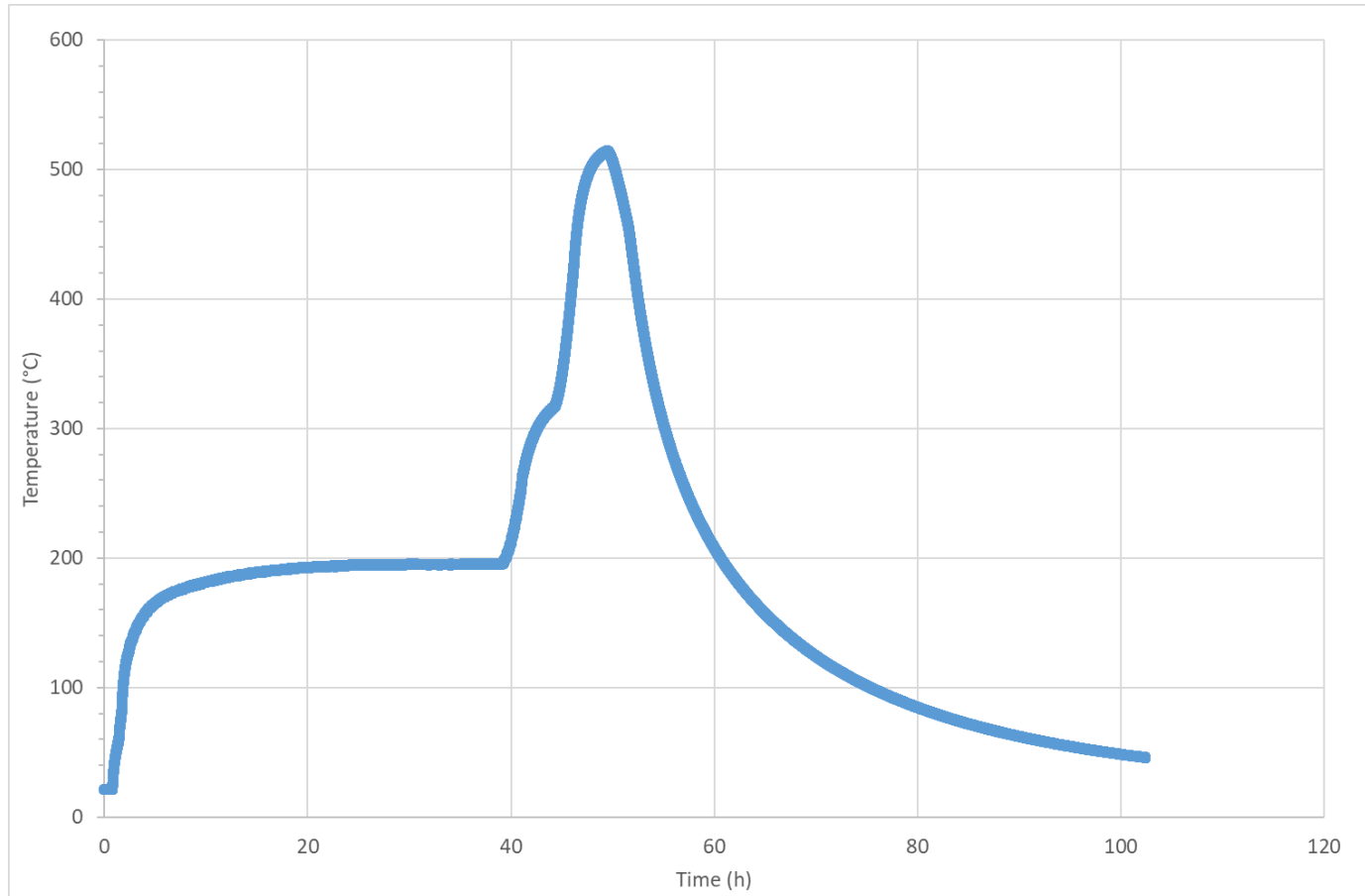


## UHV Furnace @ LNL

- Maximum mass to be loaded: 1000 kg
- Nominal vacuum conditions for routine treatments:  $5 \times 10^{-5} \div 1 \times 10^{-6}$  mbar
- Lowest achievable base pressure:  $10^{-7}$  mbar
- Maximum operational temperature:  $1300^{\circ}\text{C}$
- Useful maximum diameter of components to be treated: 1300 mm
- Maximum height of components to be treated: 1600 mm
- Useful volume: around  $2 \text{ m}^3$
- With height expansion (option): maximum height 2100 mm, maximum volume  $\sim 3 \text{ m}^3$
- Access: vertical loading with lifted platform
- Quick load cooling by argon/nitrogen inlet and dedicated heat exchanger



# Thermal annealing

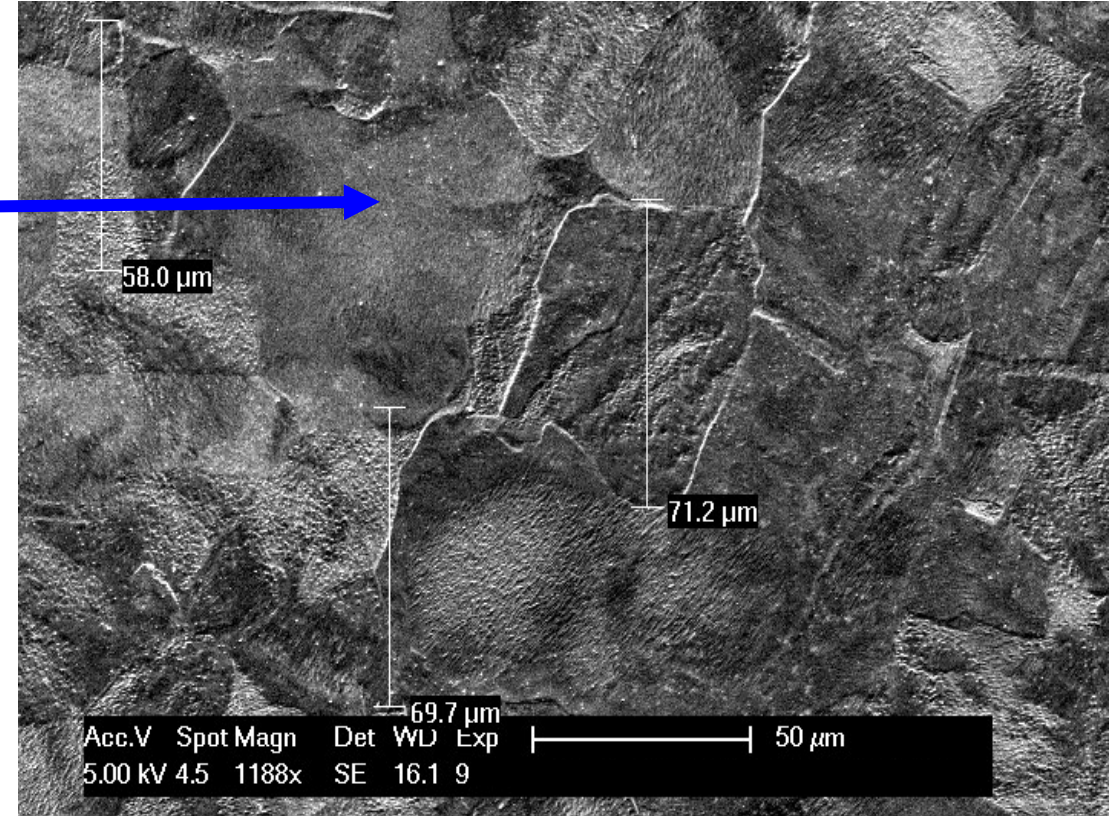
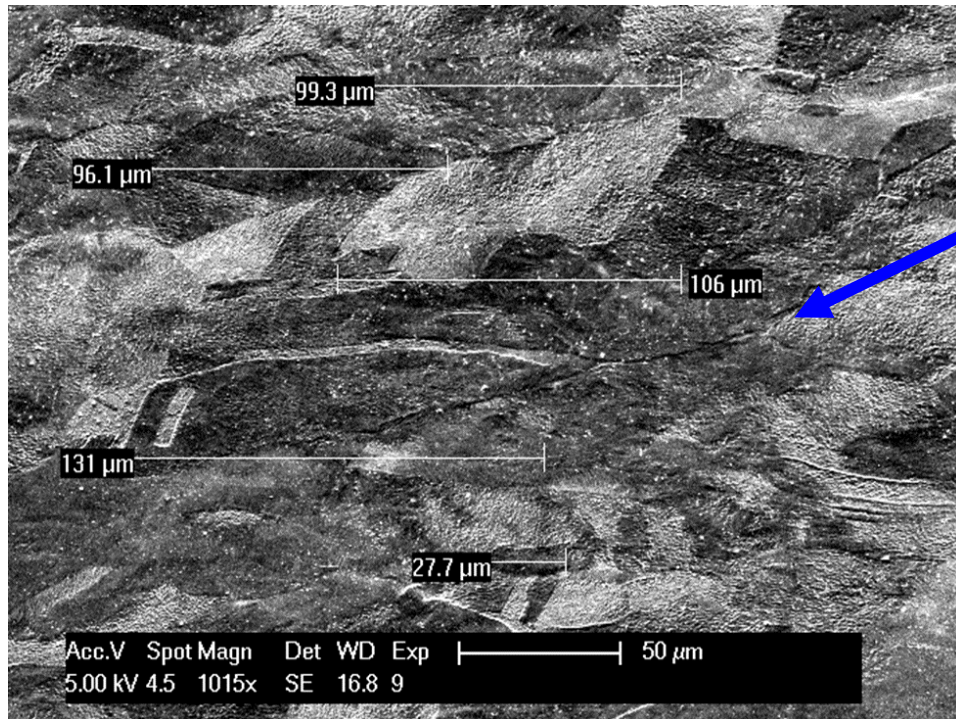


## Thermal Cycle (4 Days):

- Degassing at 200 °C 24h
- Thermal treatment **at 300 °C** for 3h
- Thermal treatment **at 500 °C** for 2h

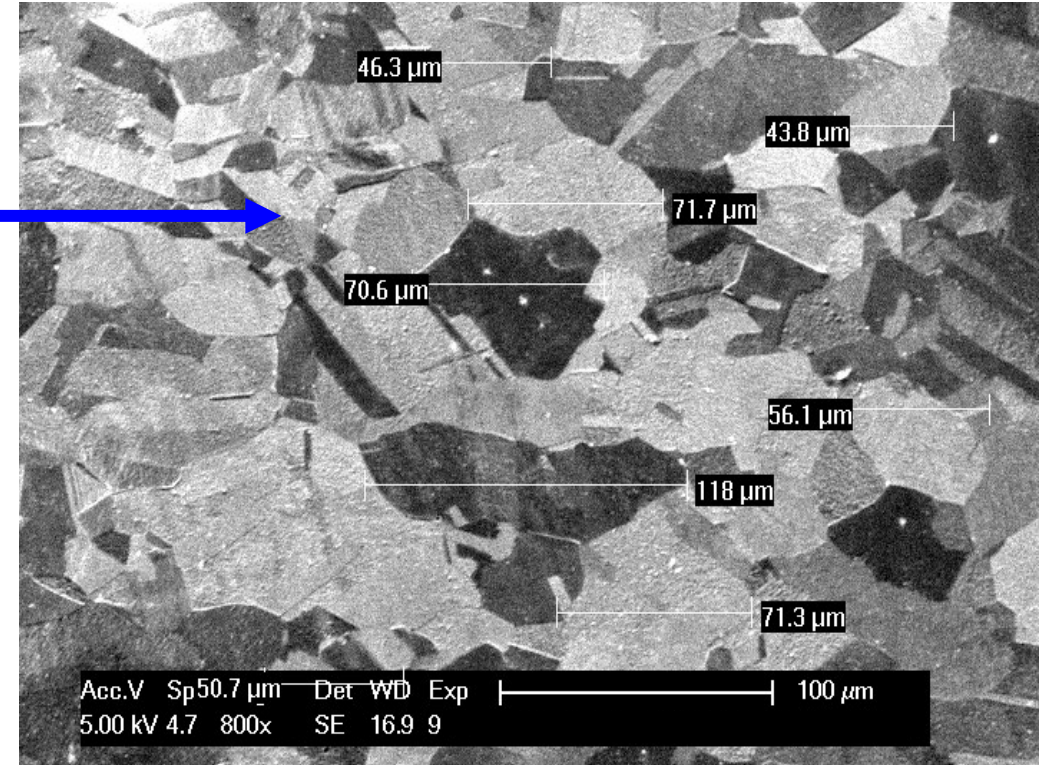
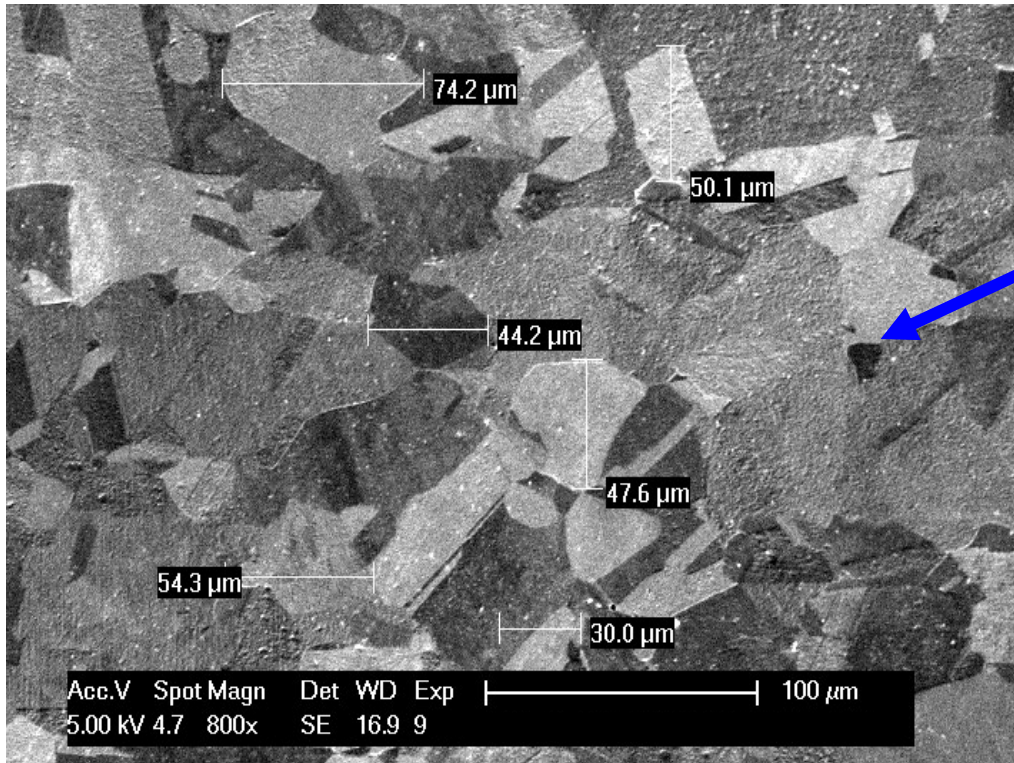
# Grains Before Annealing (After Deep drawing)

Grains Stretched



# Grains After Annealing

Grains **not** stretched





# STEP 4 Spinning of the intermediate conical die



NYLON CELL DIE +  
NYLON CONE DIE



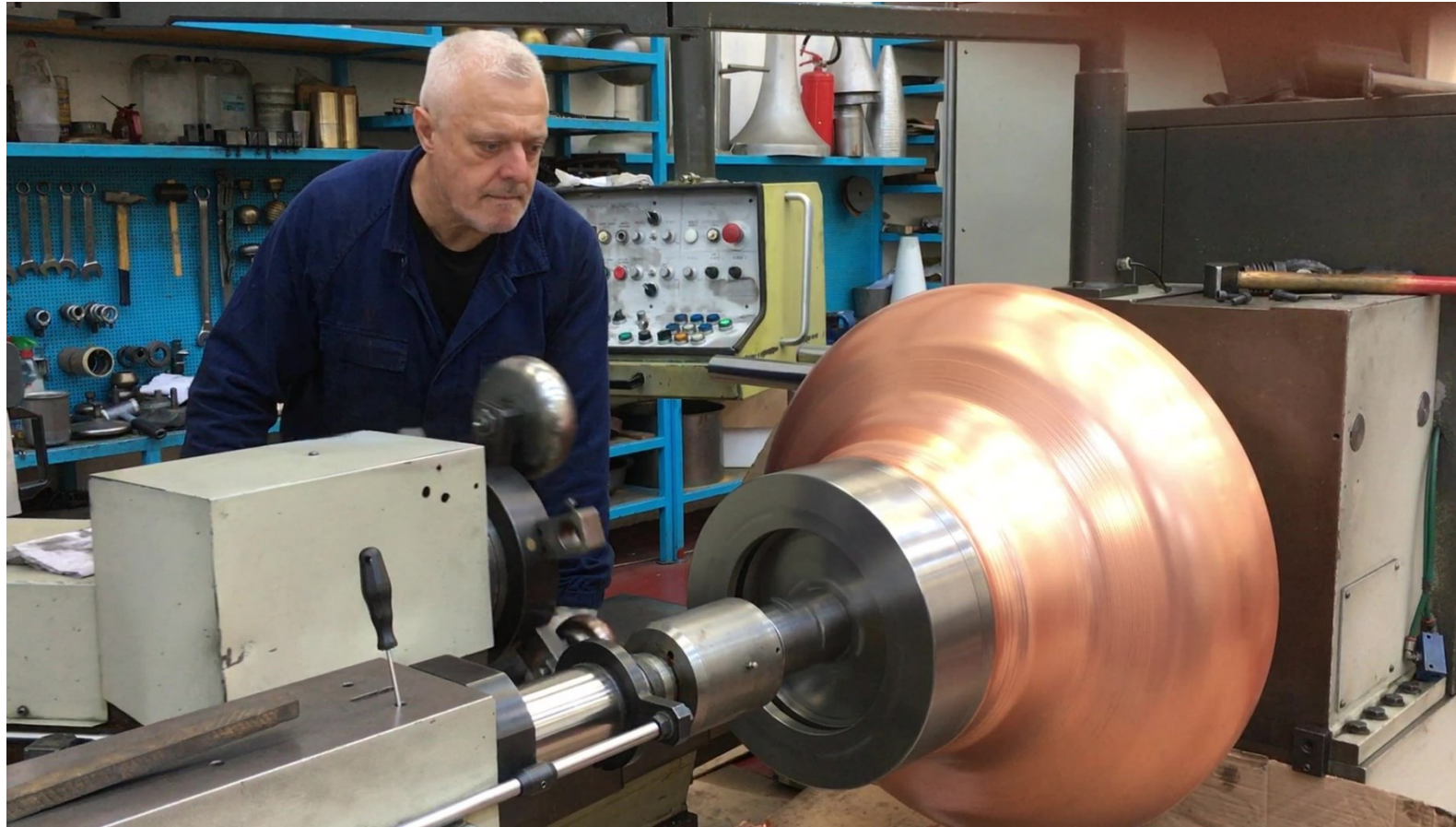
FIRST THERMAL ANNEALING



READY FOR THE SPINNING OF 2° HALF CELL

- **A thermal treatment** is mandatory for the spinning of the conical die

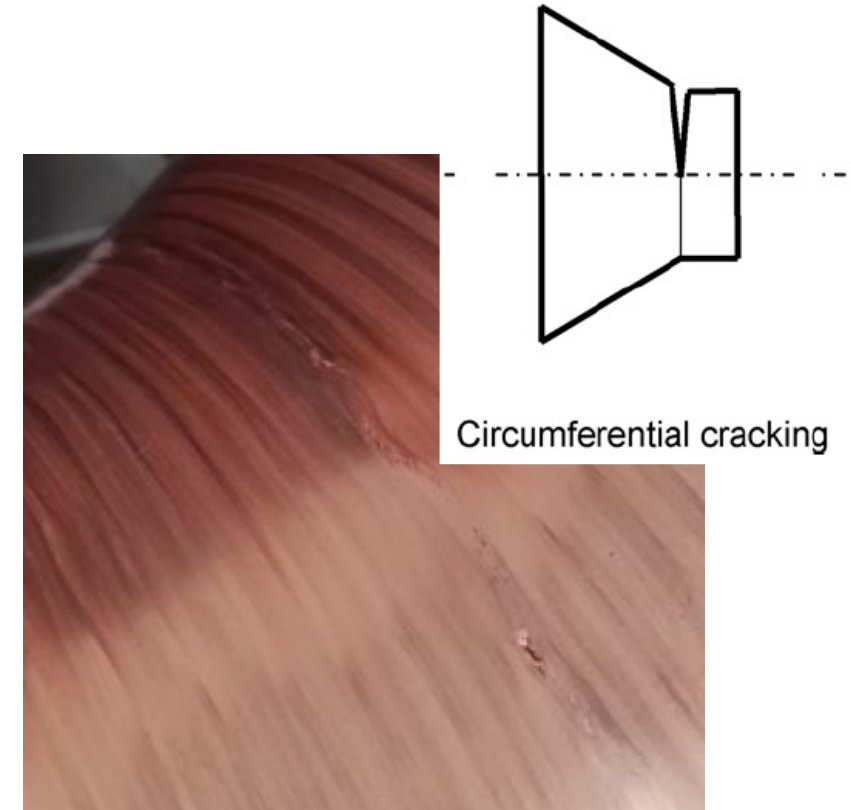
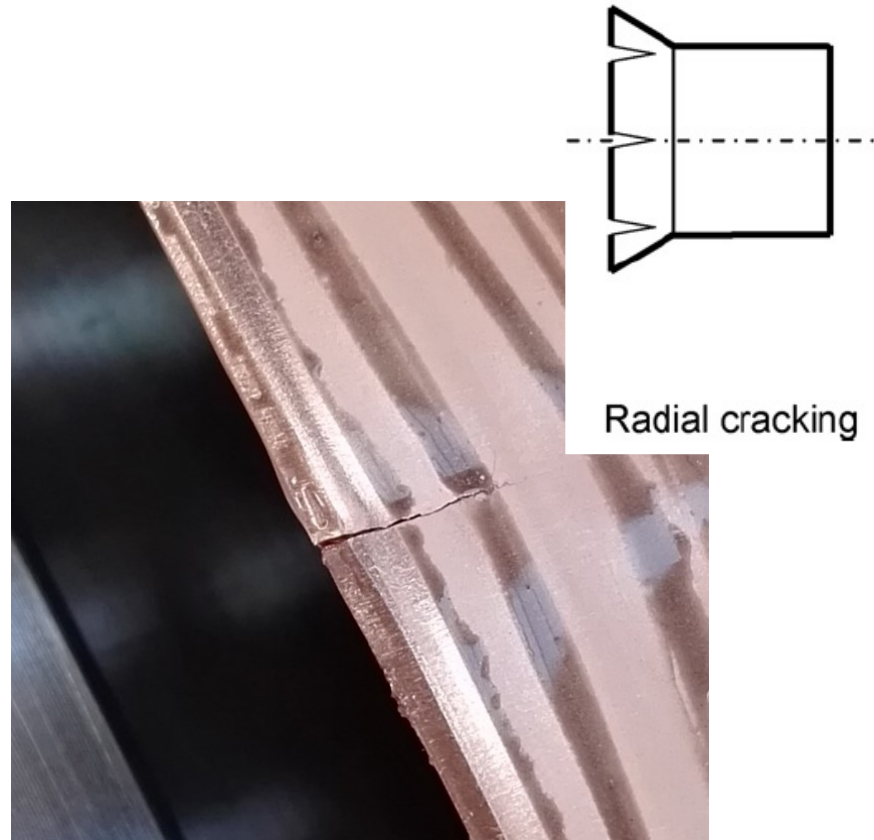
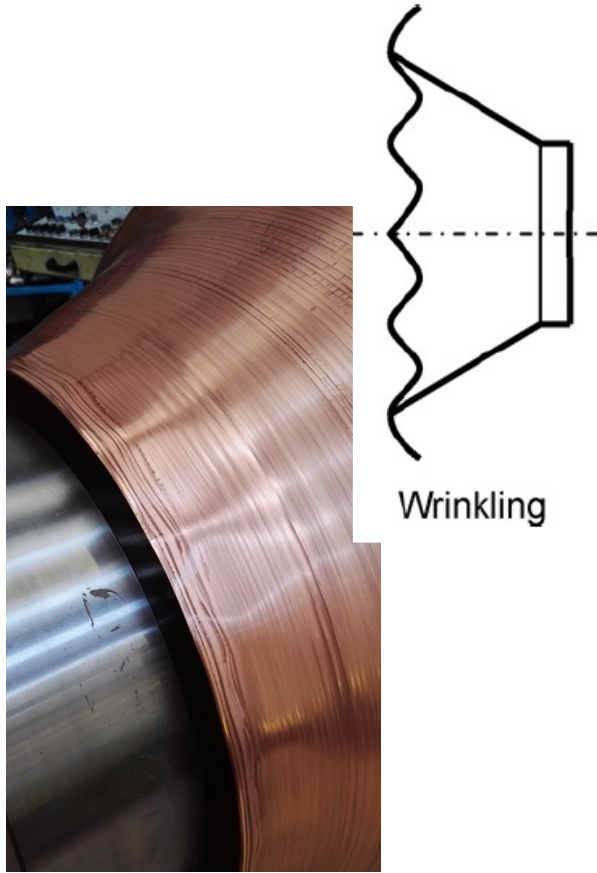
# STEP 5 Spinning of the 2<sup>nd</sup> half cell



**Here comes the troubles**

# Failure due to Copper Hardening

- The large amount of cold work introduced, produce copper hardening and consequent failures
- The sensibility of technician is fundamental to understand when the material need a thermal annealing



# Thermal annealings in Seamless 400 MHz

3 annealing necessary!



1st Thermal Annealing



2nd Thermal Annealing



3rd Thermal Annealing



Spinning completed!

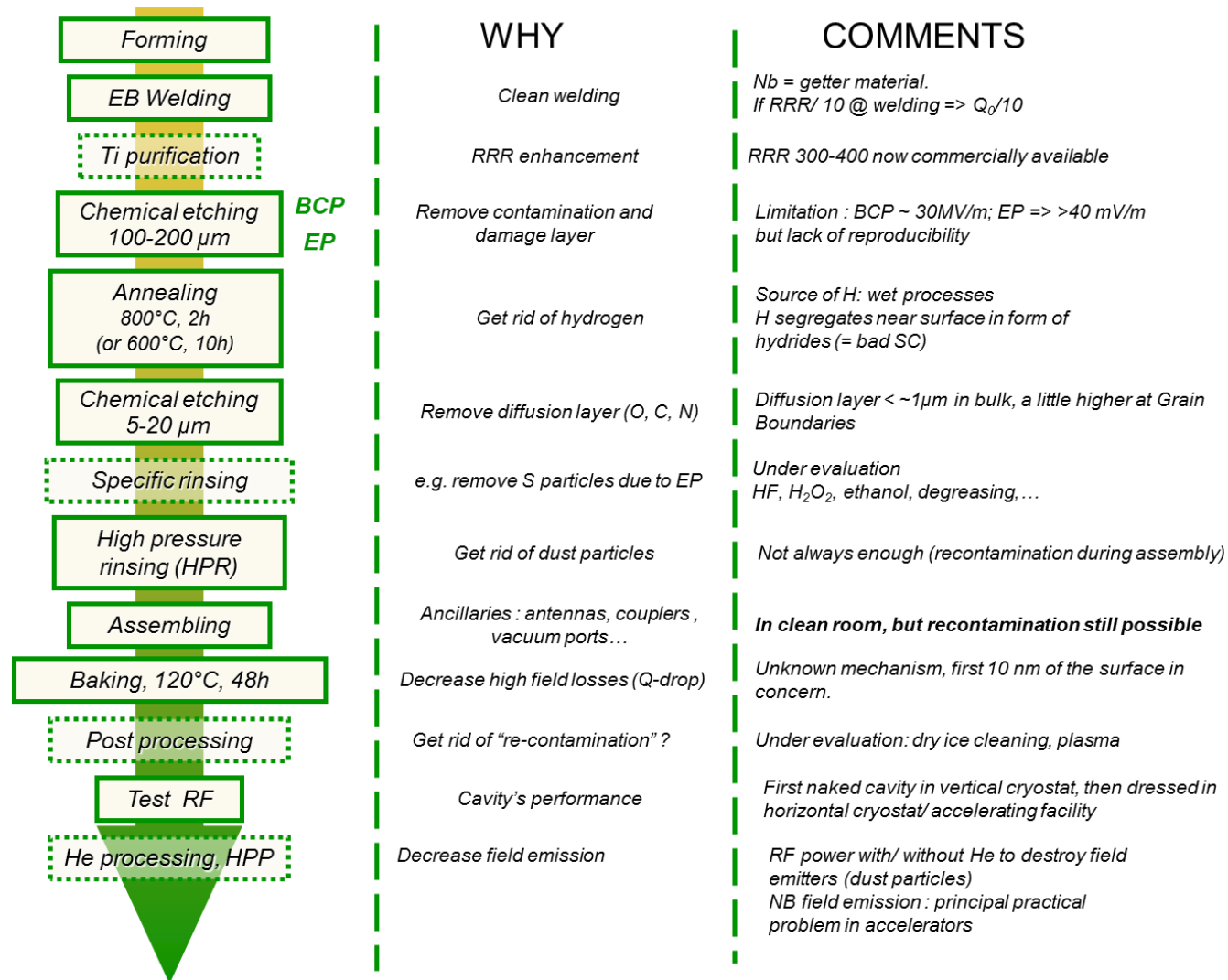


The team that realized the first seamless 400 MHz elliptical cavity

# Nb bulk cavities: cavity treatment



# Cavities' general treatment scheme



# A general consideration

**Do not make Nb surface worst than before with the next treatment!**

- **Do not apply treatments** that affect the Nb surface and **could not be “accepted” by the next step**
- If a **mistake is done, go back in the procedure** until the step where contaminant can be removed without contaminating the system
- **Chemical reactions** in many cases can **not be stopped** simply removing acid (residuals, no cooling, ...). **Rinsing is needed!**
- **Do not contaminate US bathes with material that can not be diluted**, as for **silicone grease**, oil, etc. Moreover take care of **contaminants** that can **float** over the liquid surface!
- **Wet components are more “sensitive” for collecting particles.**
- **Duration limit for a final treated cavity is about 24 hours (XFEL)**  
**Do not leave open cavities for longer time**

# Cavity preparation for SRF qualification

- Degreasing surfaces to remove contaminates
- Chemical removal of exterior films incurred from welding
- Removal of damage layer of niobium from fabrication ( $\approx 150\div 200\ \mu\text{m}$ )
- Removal of hydrogen from bulk Nb
- Mechanical tuning
- Chemical removal of internal surface for clean assembly ( $10\text{-}20\ \mu\text{m}$ )
  - – Additional “cleaning” steps if Electropolishing (EP) is used
- High Pressure Rinsing (HPR) to remove particulates from interior surfaces (incurred during chemistry and handling)
- Drying of cavity for assembly in cleanroom (reduce risk of particulate adhesion and reduce wear on vacuum systems)
- Clean assembly
- Clean evacuation
- Low-temperature baking



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# Degreasing and surface preparation @INFN LASA

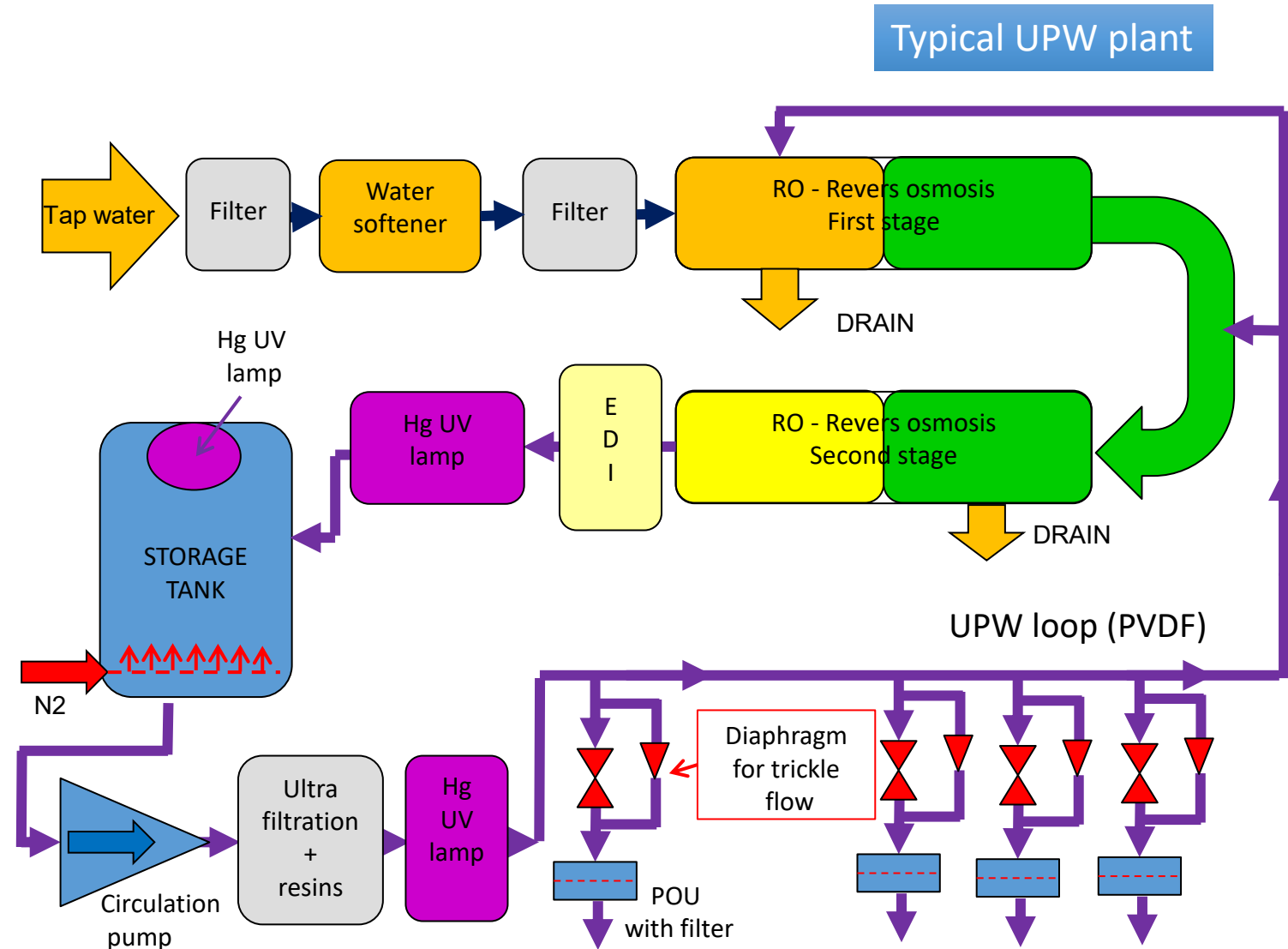
- After mechanical fabrication, **all contaminants** (fingerprints, oil, residuals from machining and QC) **must be removed**, similar to preparation of Ultra High Vacuum components.
- Typical process:
  - **Water rinsing** with specific detergent (Tikopur TR33, Micro -90, Liqui-Nox) usually 1-3%
    - For “dirty” component, alcohol and acetone could be used before
  - Water is usually **Ultra Pure Water** with 18 M $\Omega$  cm and filtered below 200 nm
  - Often in **HEPA filter environment**
- For entering **ISO7 clean room**, dishwashers are used for small components and car-wash for large components
- For entering **ISO4 clean room**, UltraSound is mandatory

# Ultra Pure Water

- UPW Specifications

- **Resistivity:**  
**18.2 MΩ cm**
- Total organic carbon (TOC):  
< 5 ppb
- Particulate counts (> 0.3 μm/l):  
< 10
- **Bacteria counts:**  
< 0.1 CFU/100 ml

The water quality is as for the **semiconductor industry:**  
**ASTM- D 5127-07 E-1.2**



# Industrial and laboratory plant



Large system:

- Production: **3000 l/h**
- **Storage: 9000 l**
- Typ. TOC: 3 ppb

Small lab production plant:

- Production: **170 l/h**
  - **Storage: 6000 l**
  - Typ. TOC: 3 ppb



# Ultrasonic Cleaning

- Immersion of components in DI water and detergent medium
- **Wave energy forms microscopic bubbles** on component surfaces
- **Bubbles collapse (cavitation)** on surface loosening particulate matter
- **Transducer** provides **high intensity ultrasonic fields** that set up standing waves. Higher frequencies lowers the distance between nodes which produce less dead zones with no cavitation
- Ultrasonic transducers are available in many different wave frequencies from 18 kHz to 120 kHz, **the higher the frequency the lower the wave intensity**

**Cavities and all hardware components (Flanges, nuts & bolts...)  
have to be degreased with ultrasonic cleaning**

# Water Break Test on Nb Sample

- It's a standard test for testing cleaning procedure with UPW and US



Not efficient cleaning



After good US cleaning procedure

ASTM F22 - 02(2007) Standard Test Method for Hydrophobic Surface Films by the Water-Break Test.

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- Clean evacuation
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# Removal of damage layer

After all the mechanical operations, a **thin layer of about 200  $\mu\text{m}$**  must be removed

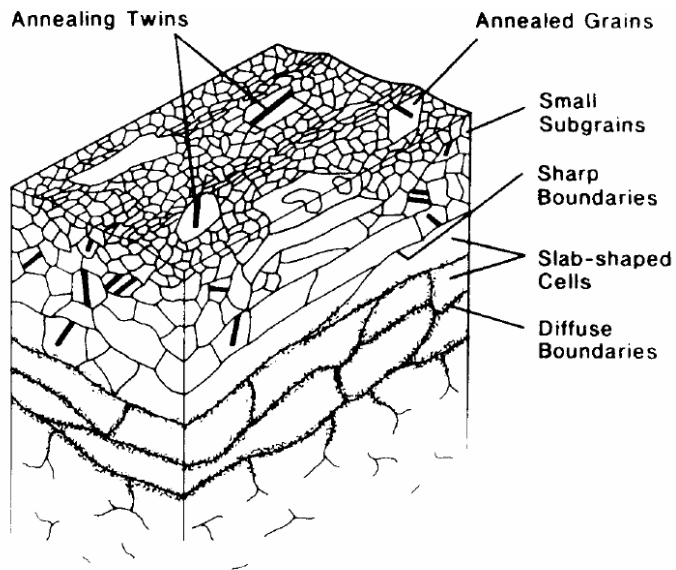
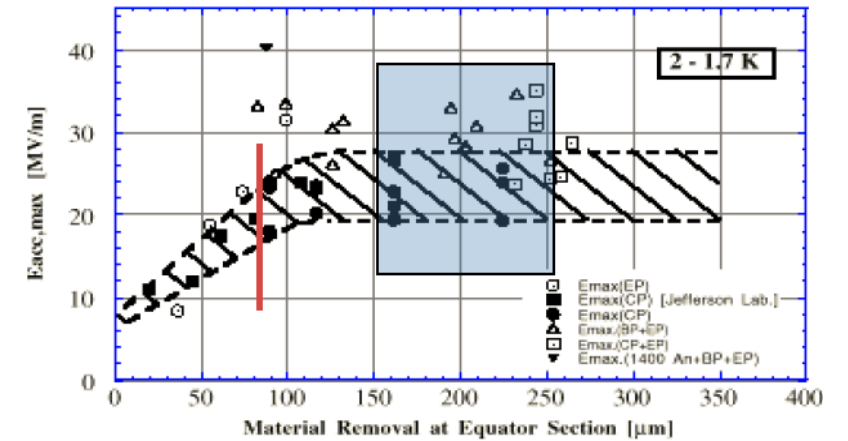
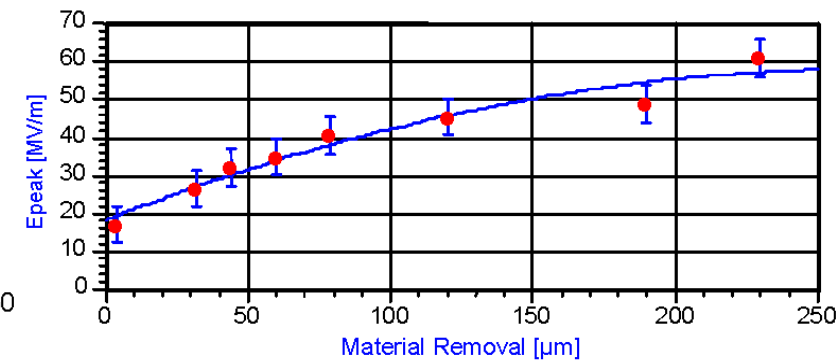
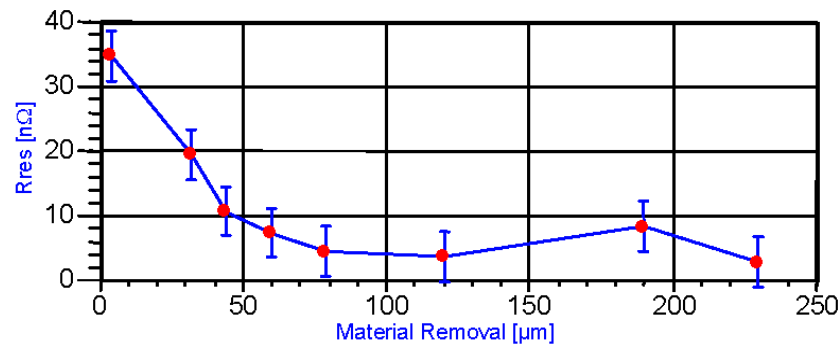


Fig. 1 - Schematic sketch of the structure observed for an abraded OFHC Copper surface.

V. Palmieri



K. Saito

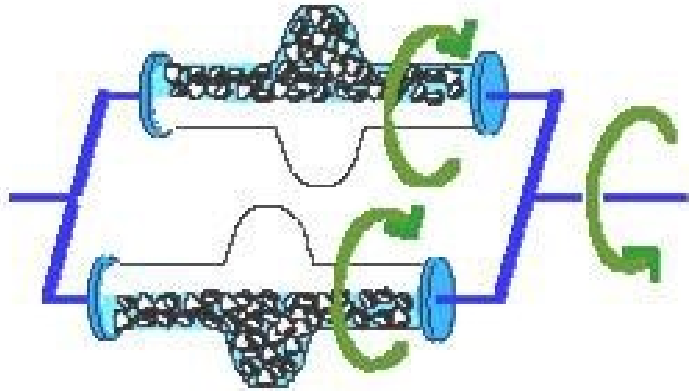


P. Kneisel



# Centrifugal Barrel Polishing (CBP)

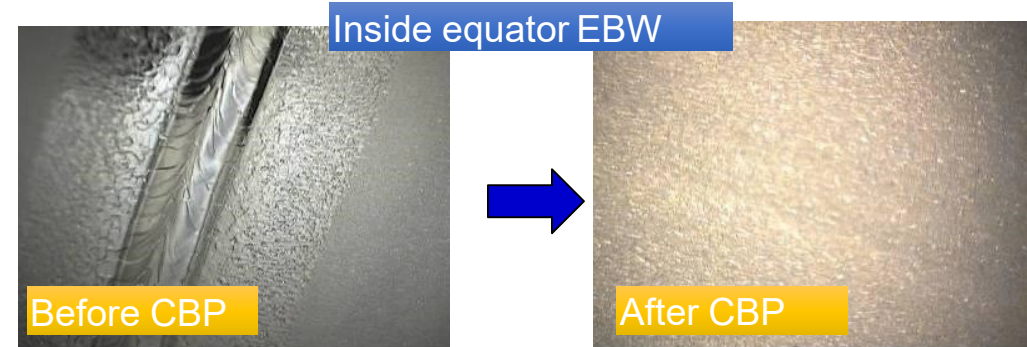
## Centrifugal Barrel Polishing (CBP)



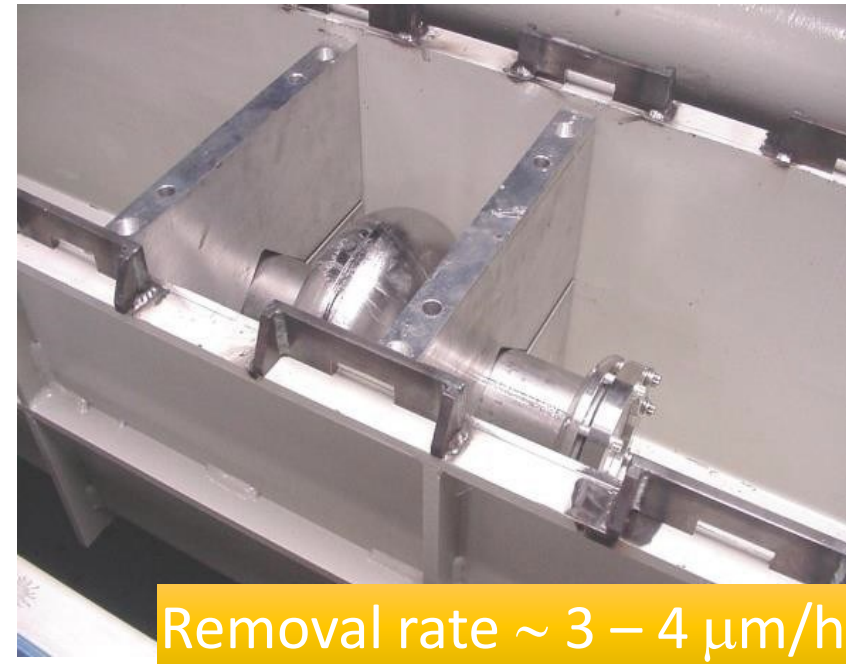
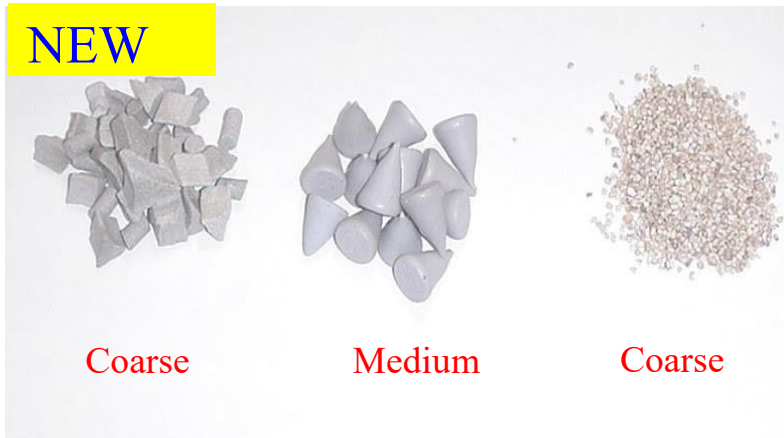
KEK 2001

## Implementation:

- **Plastic stones and liquid abrasive** added inside cavity and rotated
- **Stones rubbing on surface removes material** thus smoothing the surfaces (including weld areas)
- Benefit is **less overall chemistry** needed (80  $\mu\text{m}$ ) and smooth weld areas
- Removal of material **2x on equators** then irises. Average removal rate  $\approx 5 \mu\text{m/h}$



# Barrel Polishing Machine @ JLAB



Removal rate ~ 3 – 4  $\mu\text{m/h}$

# (Electro-)Chemical Nb removal

## Nb is resistant to chemical attack

- $\text{HNO}_3$ : oxidation of Nb surface and **passivation**, i.e. no more corrosion of the metal.
- $\text{HF}$ : **dissolve only Nb oxides**, but doesn't attack Nb itself
- $\text{HCl}$ : no attack
- $\text{H}_2\text{SO}_4$ : no attack
- Strong alkaline solution ( $\text{NaOK}$ ,  $\text{KOH}$ ,  $\text{NH}_4\text{OH}$ ): no attack

Two effects have to be coupled: **Nb oxidation** (e.g.  $\text{HNO}_3$ ) and **Nb oxides dissolution** ( $\text{HF}$ ).

# Buffer Chemical Polishing and Electro Polishing

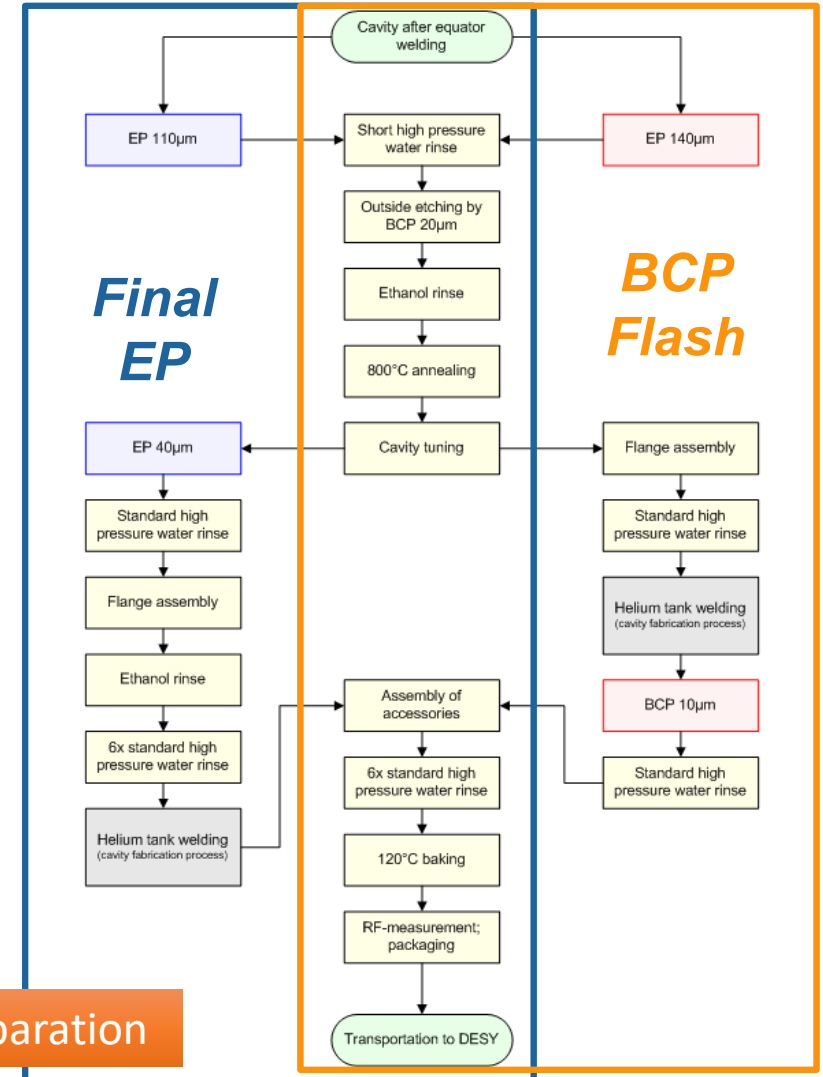
## Buffer Chemical Polishing (BCP)

- A mixture of Hydrofluoric (**HF**), Nitric Acid (**HNO<sub>3</sub>**) and Phosphoric acid (**H<sub>3</sub>PO<sub>4</sub>**), usually in 1:1:1 or 1:1:2 ratio in volume

## Electro Polishing (EP)

- A mixture of Hydrofluoric Acid (**HF**) and Sulfuric Acid (**H<sub>2</sub>SO<sub>4</sub>**) + electric current

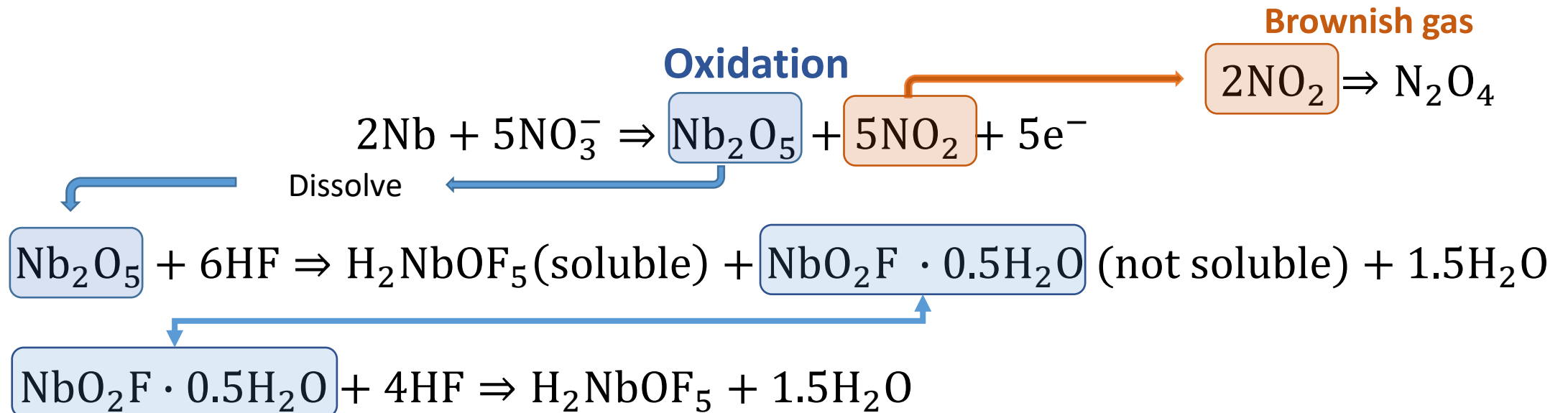
Sometimes, the **two processes** are used **together** to achieve better surface polishing (see EXFEL final steps)



EXFEL Cavity preparation

# Buffered Chemical Process (BCP)

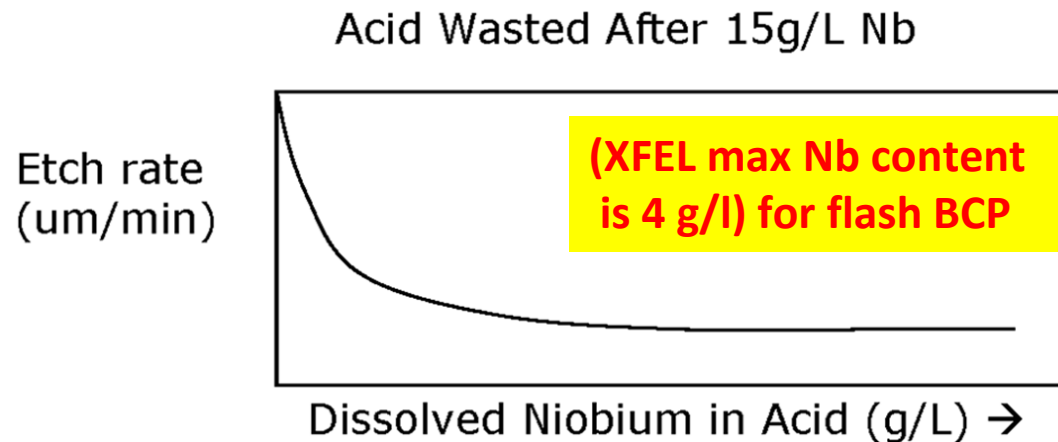
- Mixture of concentrated Hydrofluoric Acid (**HF, 40%**), Nitric acid (**HNO<sub>3</sub>, 70%**) and Phosphoric Acid (**H<sub>3</sub>PO<sub>4</sub>, 85%**)
- H<sub>3</sub>PO<sub>4</sub> doesn't participate the reaction: it act like a buffer slowing down the speed of the **exothermic reaction (self exiting!)**.
- **1:1:2**, generally used, **1 μm/min @ 20 °C**



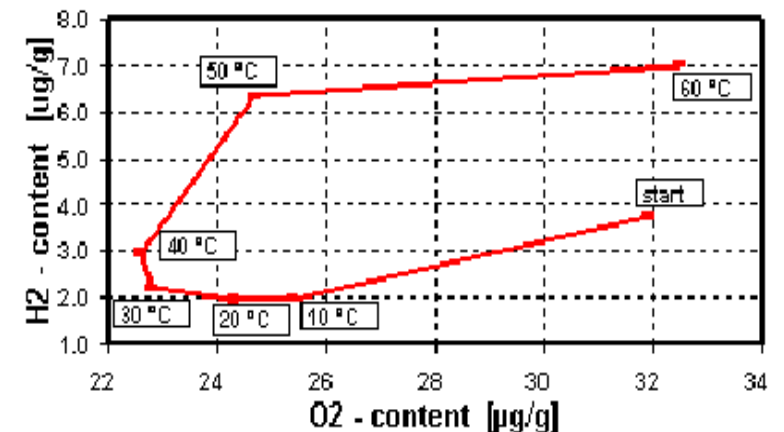
# Use of BCP Process

- **1:1:1** still used for subcomponents due to high etching rate ( $\sim 8 \mu\text{m}/\text{min}$ )
- **1:1:2** used for cavity treatment ( $\sim 1 \mu\text{m}/\text{min}$ )
- BCP must **mixed before used** because it stratifies
- BCP is usually **cool down before and during etching** to mitigate temperature increase and hydrogen content (starts at 3-5 °C and ends around 20 °C)

Etching rate versus dissolved Nb



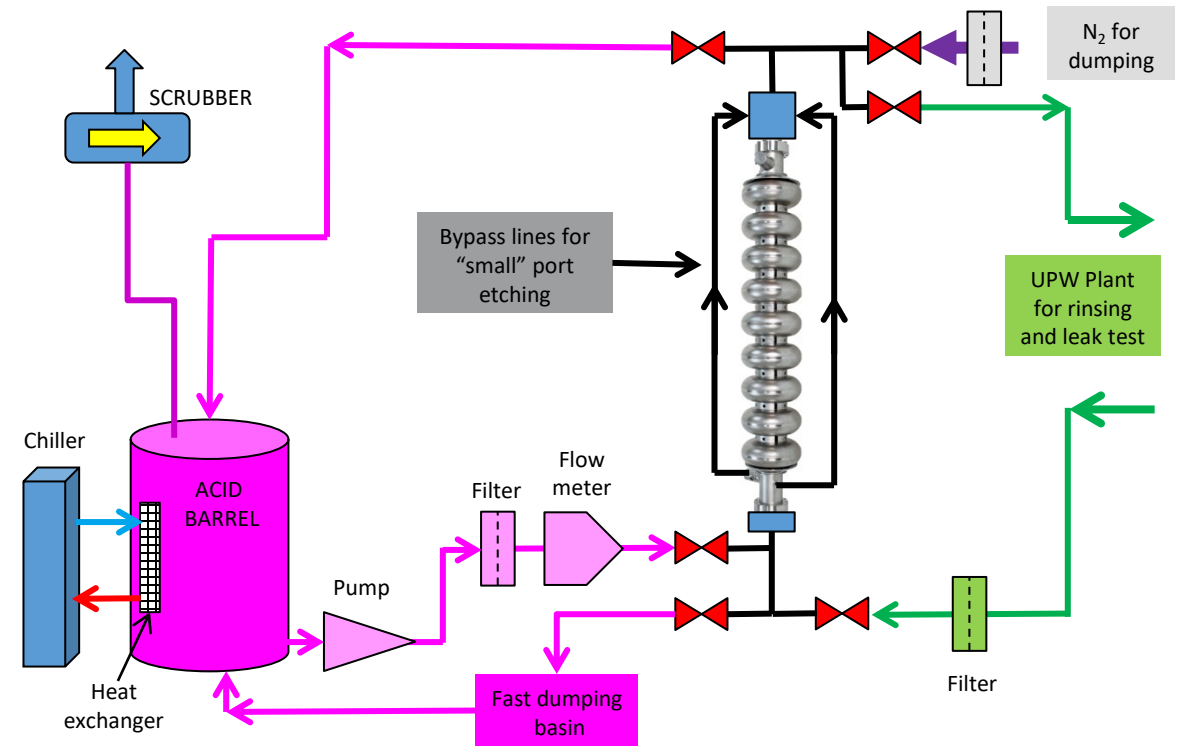
Hydrogen content vs BCP Temperature



Gas content of Nb at different etching temperatures (Schölz).  
Etching time: 20 min, BCP 1:1:1

# BCP Plant Layout

- **All components** in the acid mixture circuit **MUST be resistant** to acid attack
- Operative **temperature: below 20 °C**, to **reduce hydrogen diffusion in Nb**. Usually treatment starts at about **5 °C ÷ 6 °C**
- **Exothermic reaction**: heat exchanger or cooled barrel is needed
- **Cavity held in vertical position**, acid flow from the bottom part
- **Temperature gradient** causes **increased etching** from one end to the other
- **Usually etching rate on iris is 2 x the equator one**
- Used both for **bulk removal and final etching**: for **EXFEL** only for **final etching of half of cavities**



# BCP Plant in Operation in Labs



BCP Cabinet JLAB



DESY BCP



# BCP Plant in Operation at Qualified Vendors



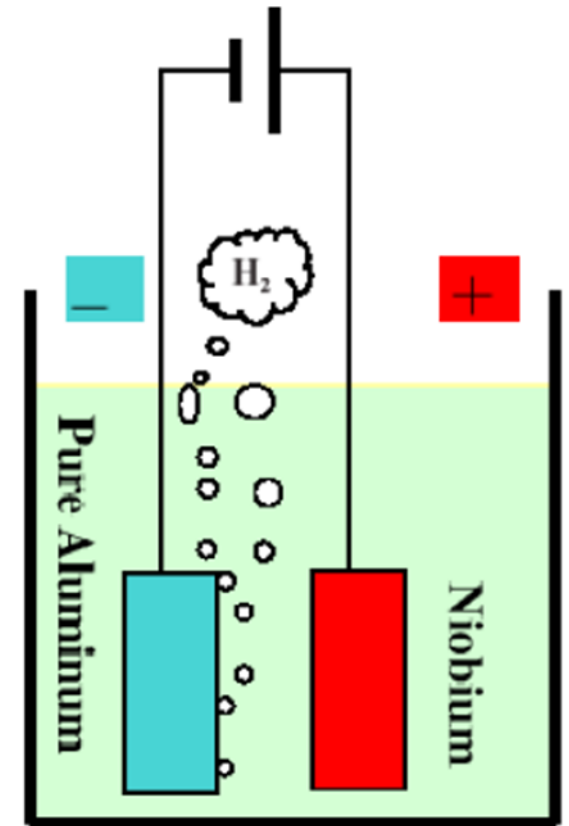
Automatic BCP system for subcomponents @ Ettore Zanon for EXFEL (etching + rinsing)



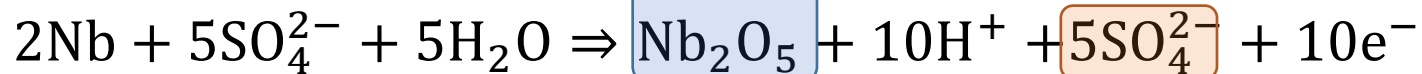
# Electropolishing (EP)

A **constant voltage** is kept between an Aluminum electrodes and the cavity immerse in a mixture of Hydrofluoric Acid (**HF**, 49%) and Sulfuric Acid (**H<sub>2</sub>SO<sub>4</sub>**, 96%) in a ratio 1:9 (typical) in volume.

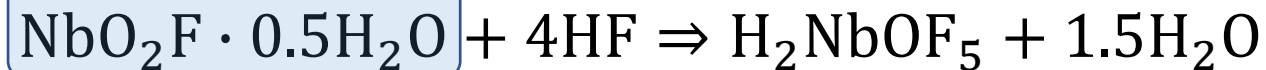
**Reaction is not self sustained:** no current - no reaction



## Oxidation

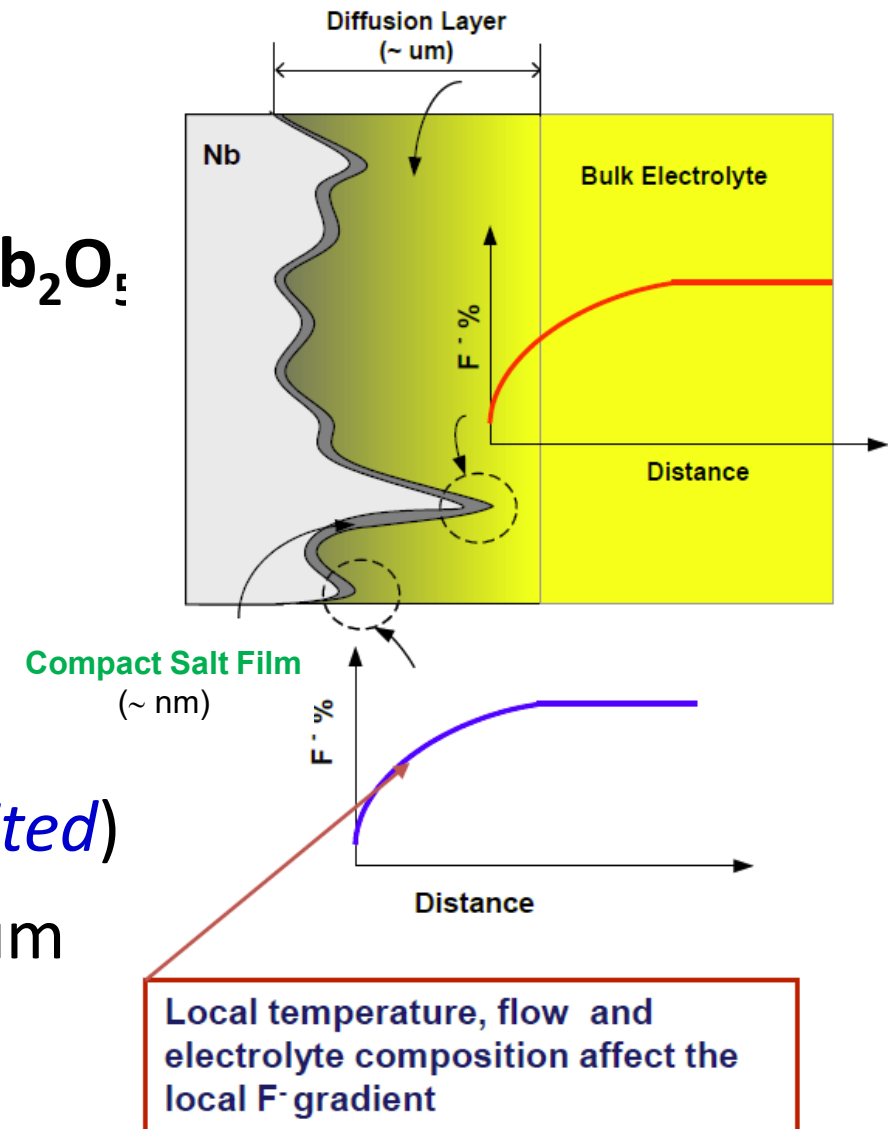


Dissolve



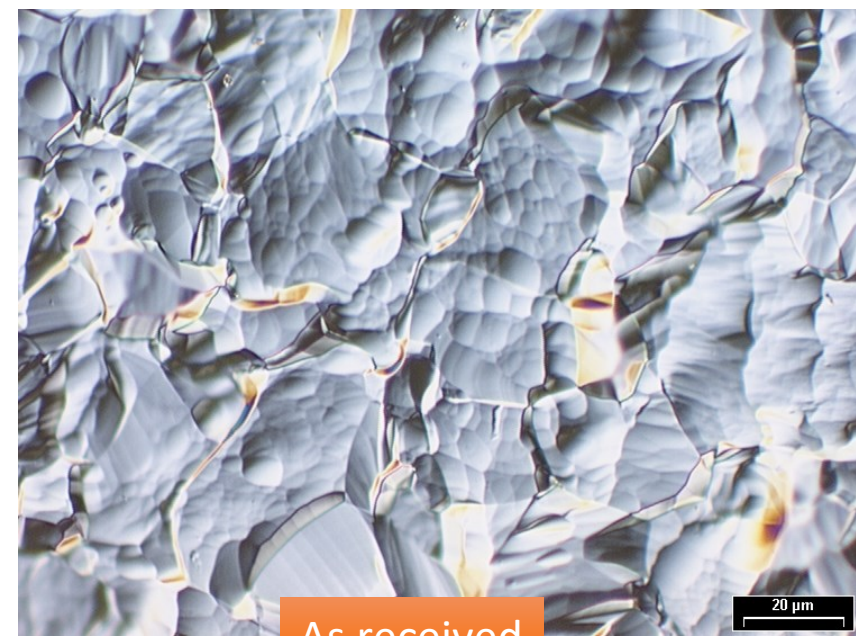
# EP Mechanism

- Anodization of **Nb** in  $\text{H}_2\text{SO}_4$  forces growth of  $\text{Nb}_2\text{O}_5$
- $\text{F}^-$  dissolves  $\text{Nb}_2\text{O}_5$
- These competing processes result in current flow and material removal
- Above a certain anodization potential, the **reaction rate plateaus**, limited by how fast fresh  $\text{F}^-$  can arrive at the surface (*diffusion-limited*)
- The diffusion coefficient sets a scale for optimum leveling effects



# Surface Polishing BCP vs EP

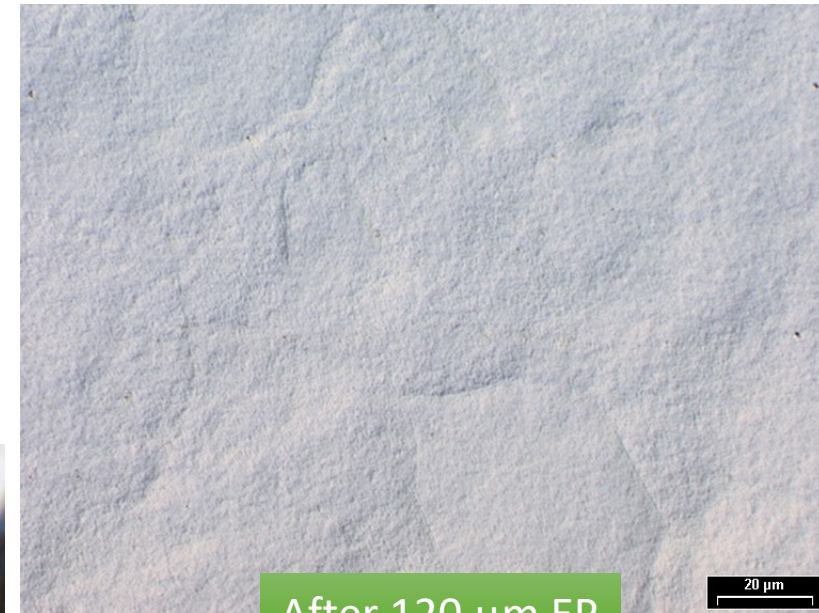
The main difference between **BCP** and **EP** is **smoothing of grain boundaries**



As received

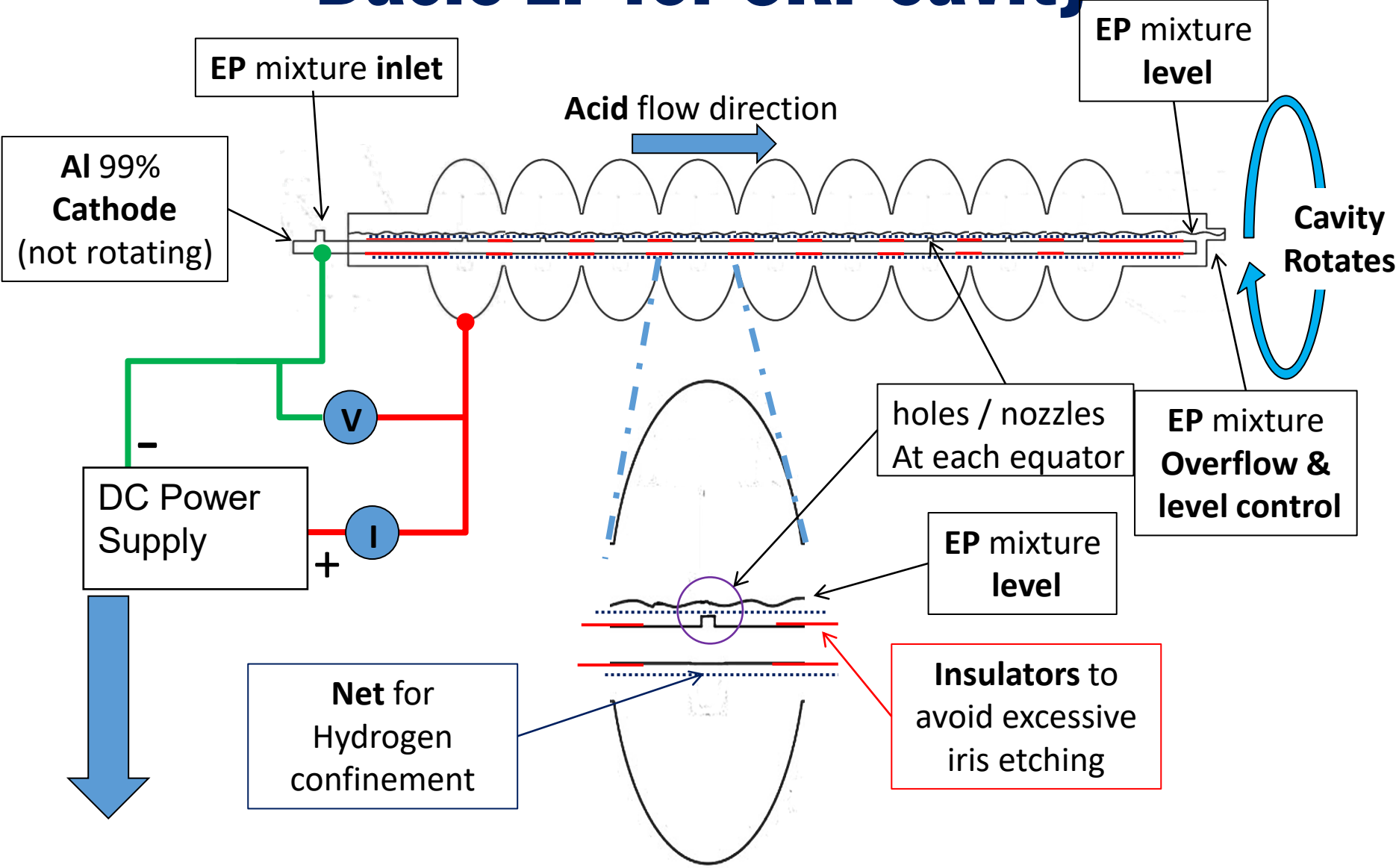


After 120  $\mu\text{m}$  BCP



After 120  $\mu\text{m}$  EP

# Basic EP for SRF cavity



# Electropolishing of SRF cavities

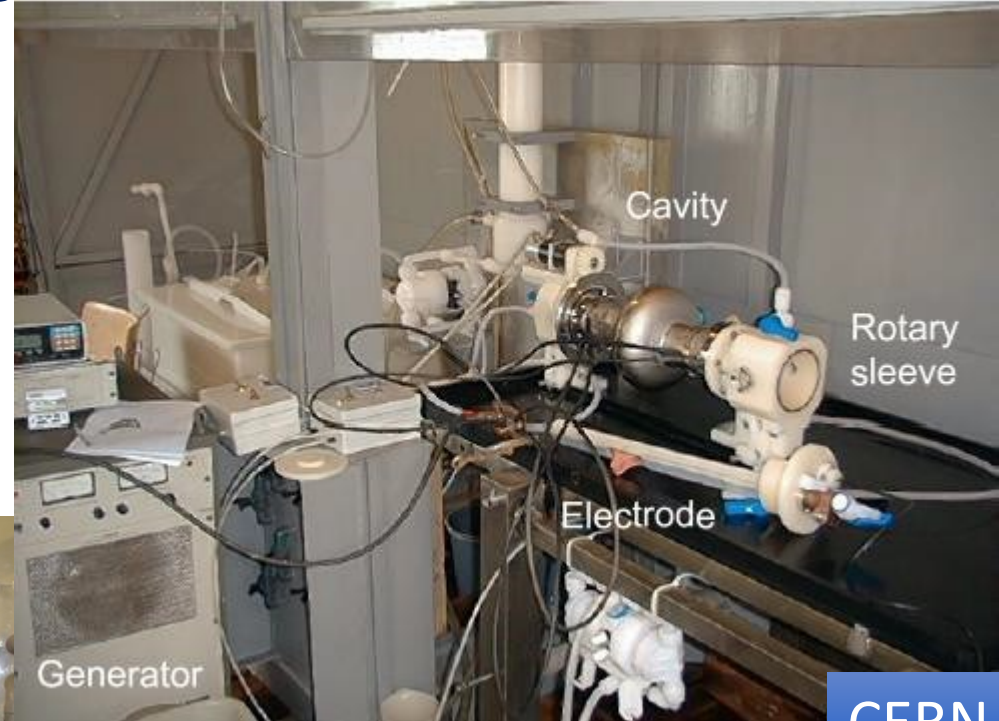
- Etching rate typical **0.3 ÷ 0.4  $\mu\text{m}/\text{min}$**
- Cavity (or electrode) is **rotating**
- It requires **ethanol rinsing** to remove Sulphur
- The current density (30-100 mA/cm<sup>2</sup>) in the plateau region:
  - decreases linearly with lower HF/H<sub>2</sub>SO<sub>4</sub> ratio
  - increases with increasing temperature
- **Temperature** during the process is maintained **between 25 – 35 °C**
- **Current oscillations often observed during polishing** (dynamic balance between oxide formation and dissolution). It's not a necessary condition for good surface finishing but indication of good processing parameters (temperature, voltage, agitation, HF concentration)

**Finding the right balance among the processing parameters becomes complicated when polishing multi-cell cavities!**

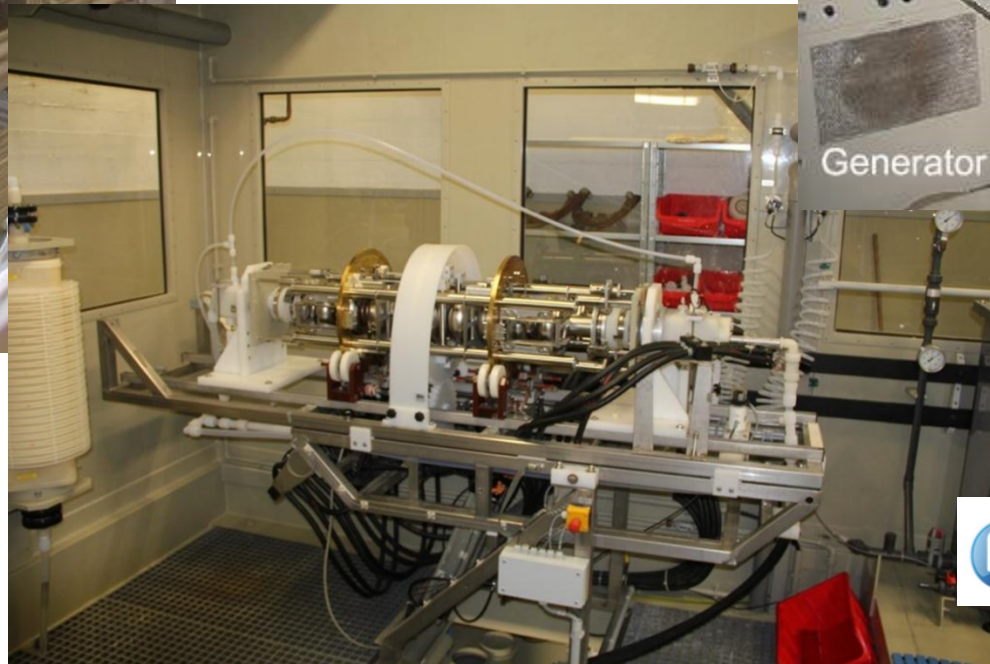
# EP Systems



DESY



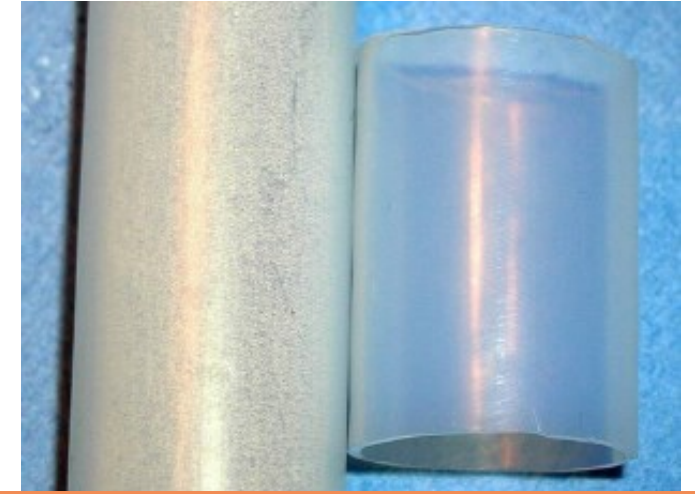
CERN



research instruments

# EP Ethanol Rinse

- **Motivation:** during EP process sulfur is produced and can cause **field emission**
- Sulfur segregates out of the acid as a **reaction with the Al electrode**, and is **deposited all over the system**, and also on the **Nb surface**
- Risk of **reaction with Nb during 800 °C** heat treatment: S must be removed before this step
- **Sulfur is insoluble in water**, but (slightly) soluble in ethanol
- Either **ethanol rinse or cleaning with detergent + US necessary**



PVDF tube before and after ethanol cleaning



Sulfur removed from a PVDF tube



# BCP vs EP

## BCP

- 2 Volumes of  $\text{H}_3\text{PO}_4$  (**buffer**, very viscous)
- 1 Volume of  $\text{HNO}_3$  (**oxidant**, transforms Nb into  $\text{Nb}^{5+}$ )
- 1 Volume of **HF** (**complexant** of  $\text{Nb}^{5+}$ , dissolves the oxide layer formed by  $\text{HNO}_3$  into  $\text{NbF}_5$ )

## Pros

- **Easy to handle**, middle stirring necessary
- **Fast etching rate**
- **Very reproducible**

## Cons

- **This is not “polishing” but “etching”**: all crystalline defects are preferentially attacked (etching pits, etching figures)
- Grains with various orientations are not etched at the same rate, which **induced roughness!**
- Except for a few cases  $E_{\text{acc}}^{\text{max}} \sim 25\text{-}30 \text{ MV/}$

## Caution!

- Do not process at temperatures higher than  $25 \text{ }^\circ\text{C}$
- Risk of runaway

# BCP vs EP

## EP

- 9 Volumes of  $\text{H}_2\text{SO}_4$  (**buffer**, very viscous)
- 1 Volume of **HF** (**complexant** of  $\text{Nb}^{5+}$ , dissolves the oxide layer formed due to the high potential applied to Nb)

## Pros (Ideal condition, i.e. viscous layer present)

- **This is really “polishing”**, not sensitive to crystallographic defects – it produces a smooth surface
- Should not be sensitive to the cathode-anode distance - the **same etching rate everywhere**
- It gives (but not always) **the best ever  $E_{\text{acc}}^{\text{max}} \sim 45 \text{ MV/m}$**  (TESLA shape  $\rightarrow \sim 180 \text{ mT}$ )

## Cons

- It is **not possible to reach an ideal state** in most of our processing conditions
- **Very sensitive** to stirring condition, temperature, and aging of the mixture
- **Not very reproducible**
- **Safety issues** (acid mixture sensitive to water,  $\text{H}_2$  evolution, etc.)

## Caution!

- If T increases: the etching rate increases but there is also a risk of pitting, H loading and HF evolution
- If V increases: the etching rate increases but there is also a risk of pitting, the generation of Sulphur particles and sensitivity to the cathode-anode distance

# BCP vs EP

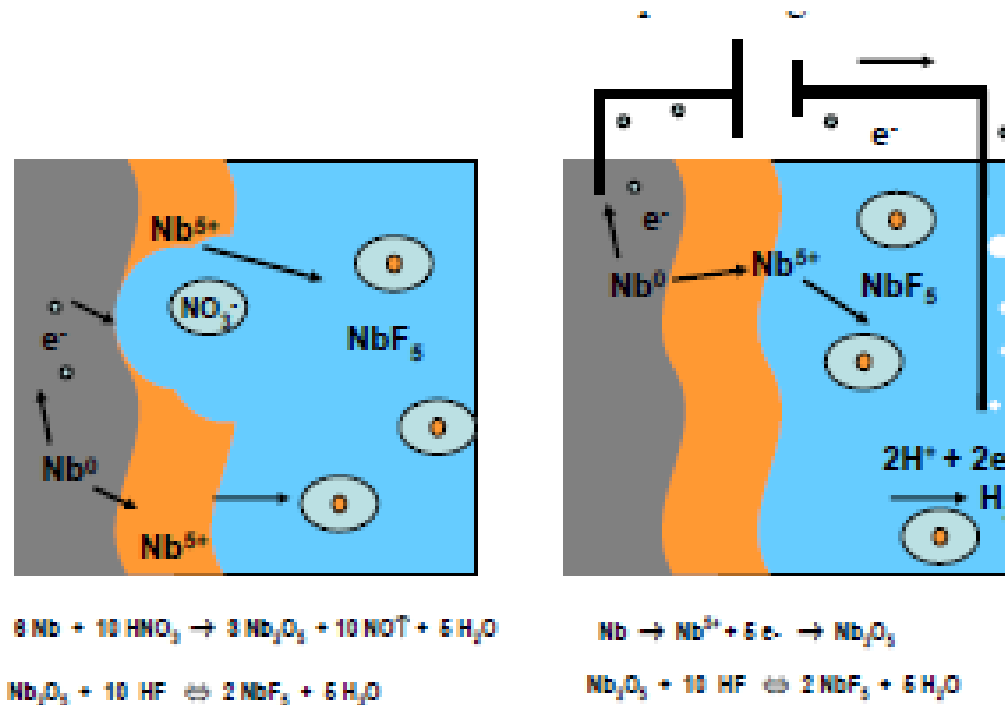


Fig. 6: A comparison between chemical polishing (left) and electropolishing (right). In both cases, niobium is oxidized into  $\text{Nb}^{5+}$ . In the case of chemical polishing, oxidation occurs because of the presence of a strong oxidant ( $\text{NO}_3^-$ ) in the solution, while in electropolishing oxidation occurs because of the bias applied to the anode. Because of the presence of water, the stable form of Nb is  $\text{Nb}_2\text{O}_5$ ; but HF decomposes the oxides into  $\text{NbF}_5$ , which is soluble in the solution.

# Cavity preparation for SRF qualification

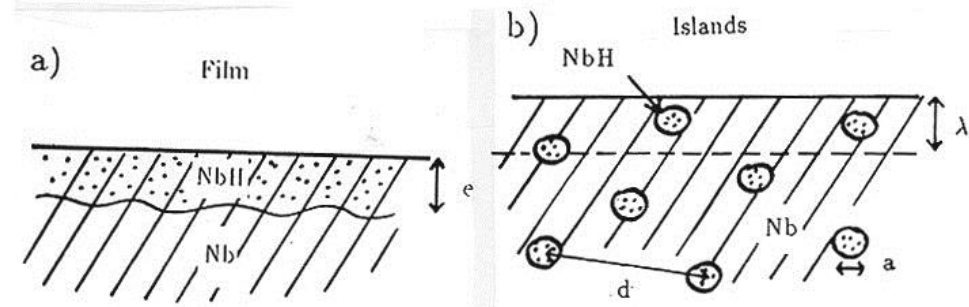
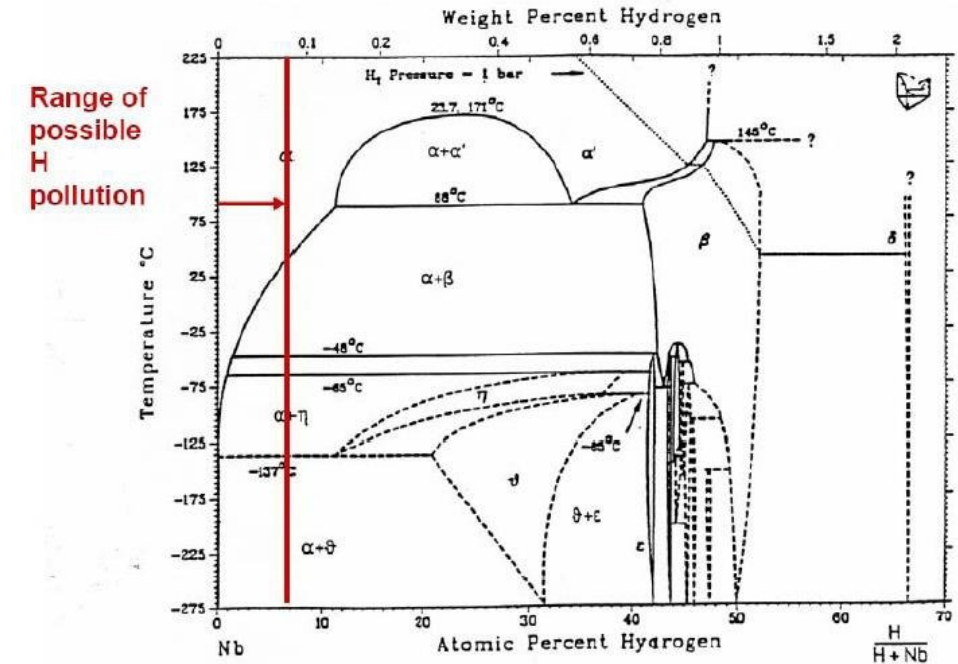
- Degreasing surfaces to remove contaminates
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- **Removal of hydrogen from bulk Nb**
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- Clean evacuation
- Low-temperature baking

# Annealing - Cavity firing at high temperature

- **H diffuses** in the bulk during the various etching treatments.  
See **R. E. Ricker and G. R. Myneni**, J. Res. Natl. Inst. Stand. Technol. **115**, 353-371 (2010), Evaluation of the Propensity of Niobium to Absorb Hydrogen During Fabrication of Superconducting Radio Frequency Cavities for Particle Accelerators.
- **Nb** is an **active metal** with respect to various gases: it acts like a getter.
- Hydrogen makes a **solid solution** in Nb,  $H_2$  equilibrium pressure is driven by **Sievert Law**
- Equilibrium pressure is **temperature dependent** and **increasing the temperature**, maintaining a **low  $H_2$  partial pressure**,  **$H_2$  is desorbed from the bulk (Nb)**

# Hydrogen in Niobium Q “disease”

- Cavities that remain at **70-150 K** for **several hours** (or slow cool-down,  $< 1$  K/min) experience a **sharp increase of residual resistance**
- **More severe** in cavities which have been heavily **chemically etched**
- **H is readily absorbed into Nb** where the oxide layer is removed (during chemical etching or mechanical grinding)
- **H has high diffusion rate in Nb**, even at low temperatures.
- **H precipitates** to form a hydride phase **with poor superconducting properties**:  $T_c=2.8$  K,  $H_c=60$  G
- At room temperature the required concentration to form a hydride is  $10^3$ -  $10^4$  wppm
- At 150 K it is  $< 10$  wppm



# Annealing

## Hydrogen outgassing

=> most efficient at 750°C – 800°C, **2h under good vacuum**

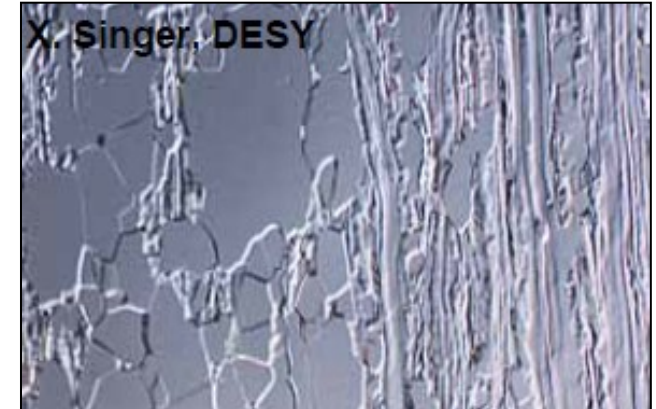
## Recrystallization (goal is close to 100% with highest RRR)

- Removing of defects and curing of dislocations
- Nucleation of new grains and growing of new crystals
- Grain growth (depending on temperature and purity)

**Nb becomes softer and this facilitate the cavity tuning process**

Different parameters at different labs:

- 600 °C/10 h at Jlab
- 800 °C/2 h at DESY
- 750 °C/3 h at KEK

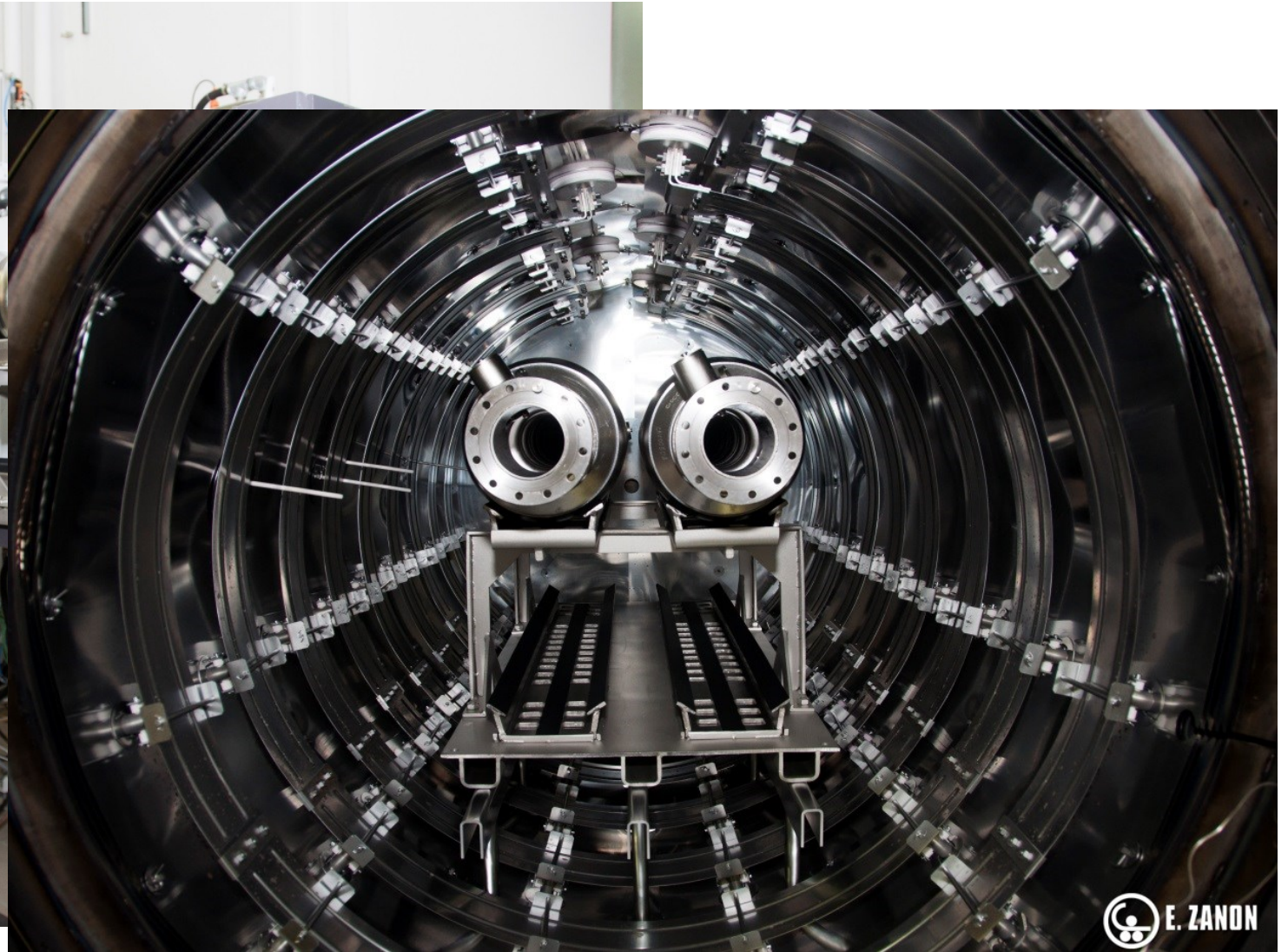
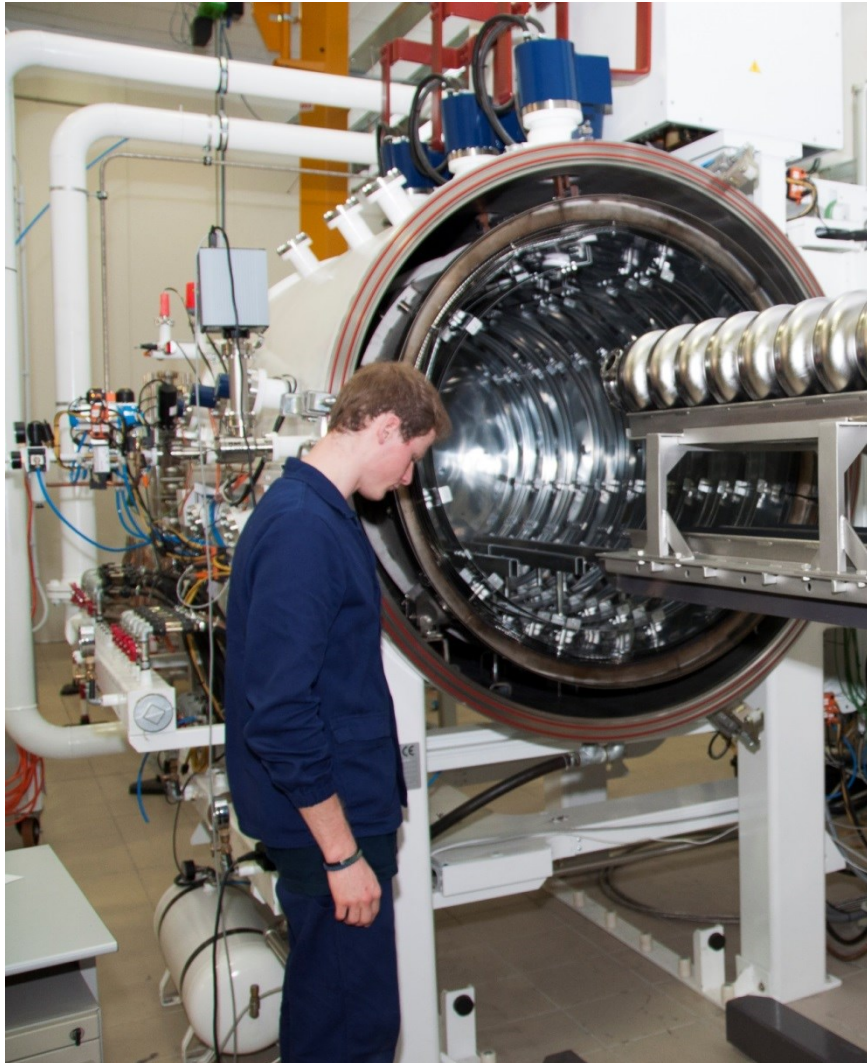


No completely recrystallized Nb



Completely recrystallized Nb

# EZ Furnace



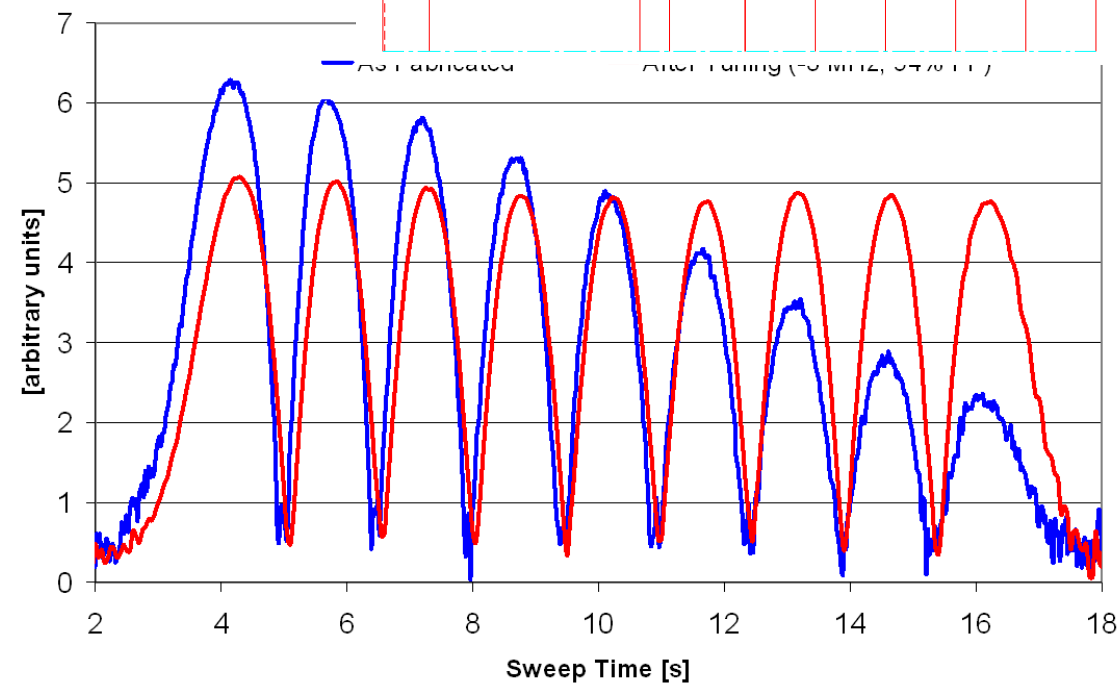
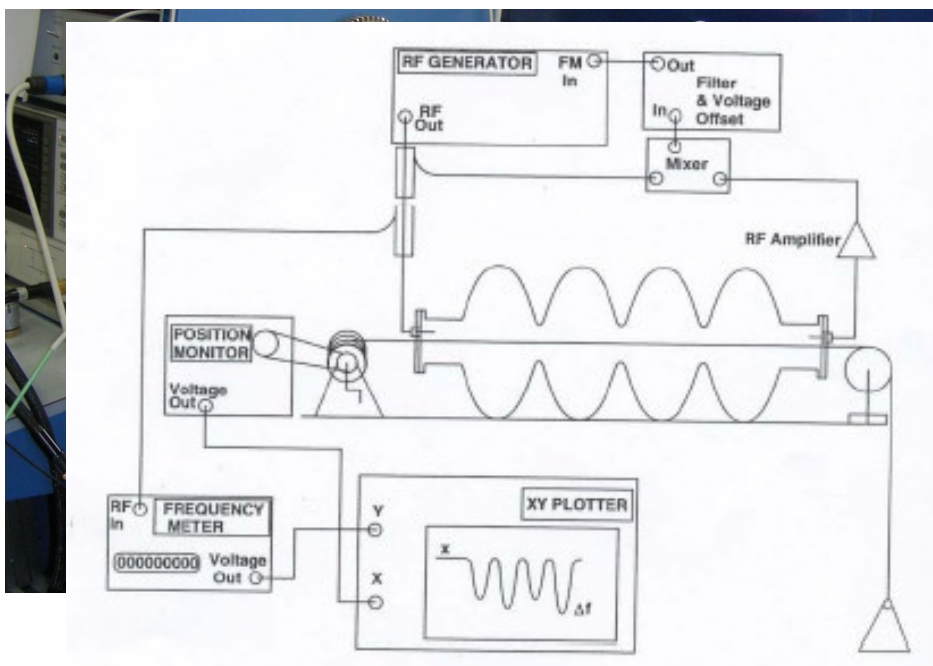
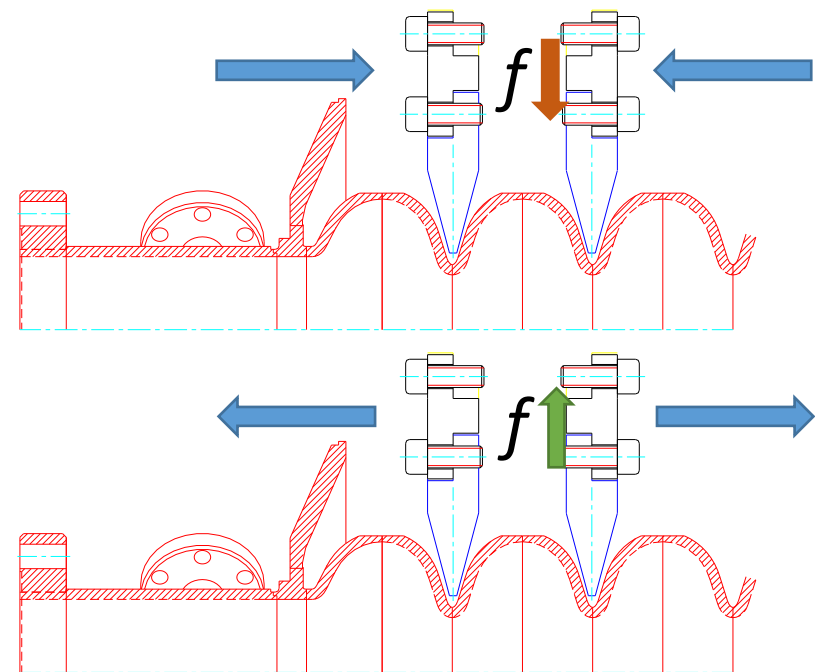


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- Clean assembly
- Clean evacuation
- Low-temperature baking

# Frequency tuning

- After the treatments, the cavity needs to be tuned to the **right frequency** and **field flatness**.
- This operation is done by **tuning each single cell** to achieve proper field distribution

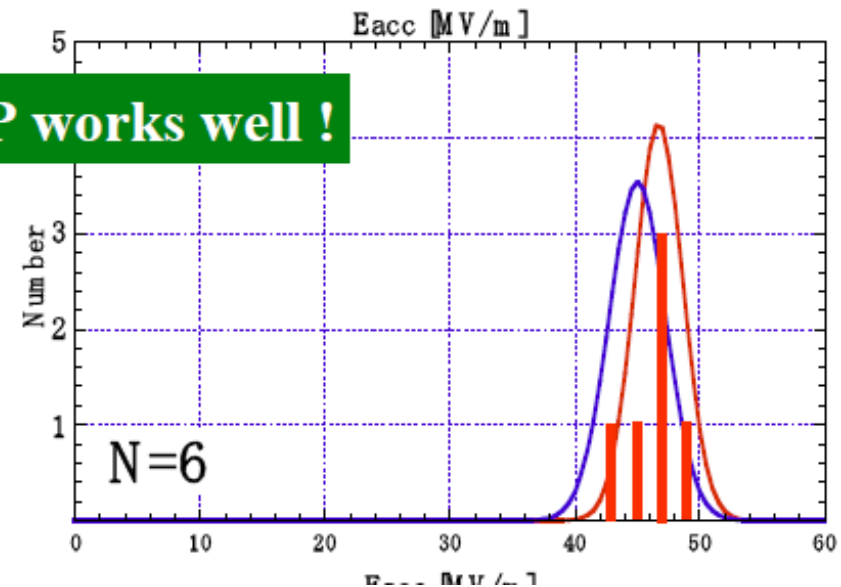
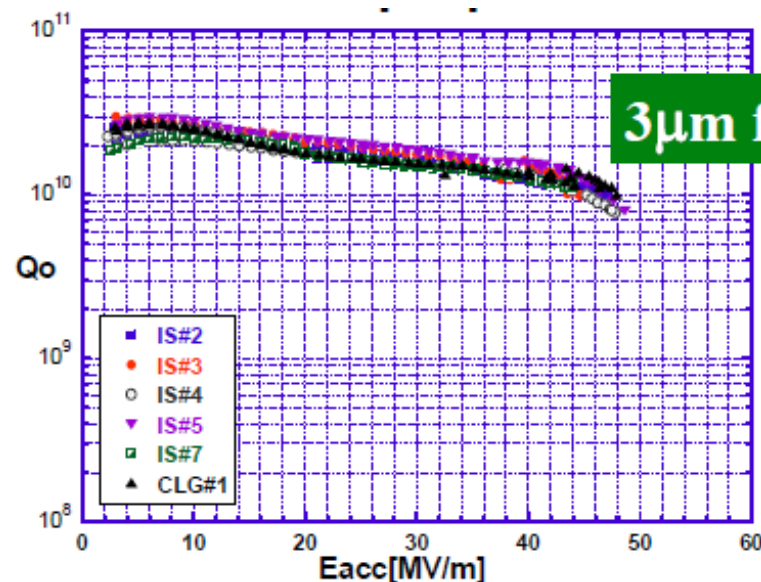


# Cavity preparation for SRF qualification

- Degreasing surfaces to remove contaminates
- Chemical removal of exterior films incurred from welding
- Removal of damage layer of niobium from fabrication ( $\approx 150\div 200\ \mu\text{m}$ )
- Removal of hydrogen from bulk Nb
- Mechanical tuning
- Chemical removal of internal surface for clean assembly ( $10\text{-}20\ \mu\text{m}$ )
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- High Pressure Rinsing (HPR) to remove particulates from interior surfaces (incurred during chemistry and handling)
- Drying of cavity for assembly in cleanroom (reduce risk of particulate adhesion and reduce wear on vacuum systems)
- Clean assembly
- Clean evacuation
- Low-temperature baking

# Post EP treatment

- Ethanol Rinse (DESY)
- “Flash” BCP (10  $\mu\text{m}$ ) (DESY)
- “Flash” EP (3  $\mu\text{m}$ , fresh acid, no re-circulation) (KEK)
- Ultrasonic Degreasing with Micro-90 and hot water (JLab)

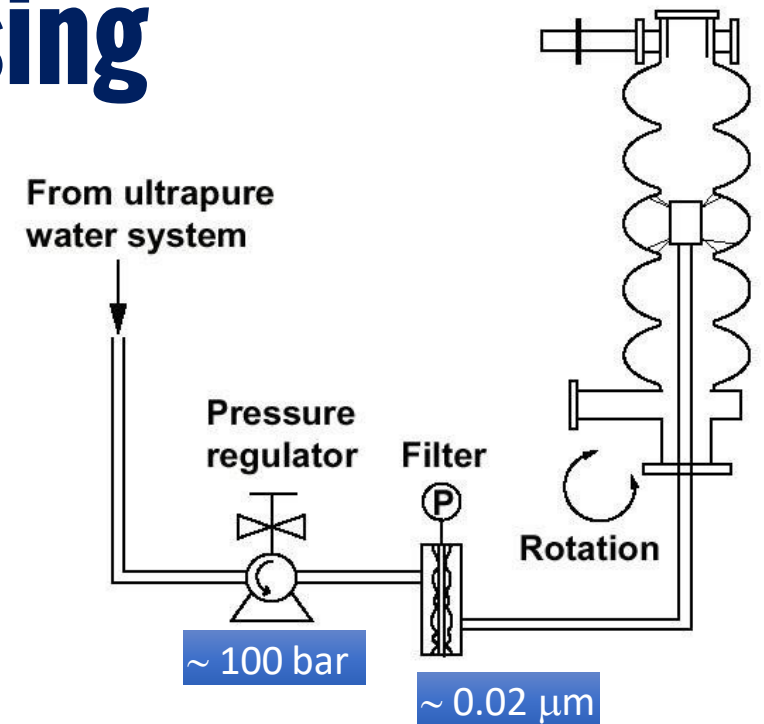


# Cavity preparation for SRF qualification

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# High Pressure Rinsing

- The **final step** in cavity assembly is the **Rinsing with a High Pressure UPW jet** to remove particulate from the handling and residual from chemical treatments
- Water jet **must be moved continuously**: if jet impacts stably in one-point Nb surface can be damaged
- Continuous **motion of the cavity respect jets** (drawing a spiral behavior that cover completely the Nb surface)
- Ultra pure (6.0) filtered (40 nm) **nitrogen protection gas injection coaxial** with water to reduce risk of particles entering
- **Cavity** must be **grounded** otherwise it will be electrically charged

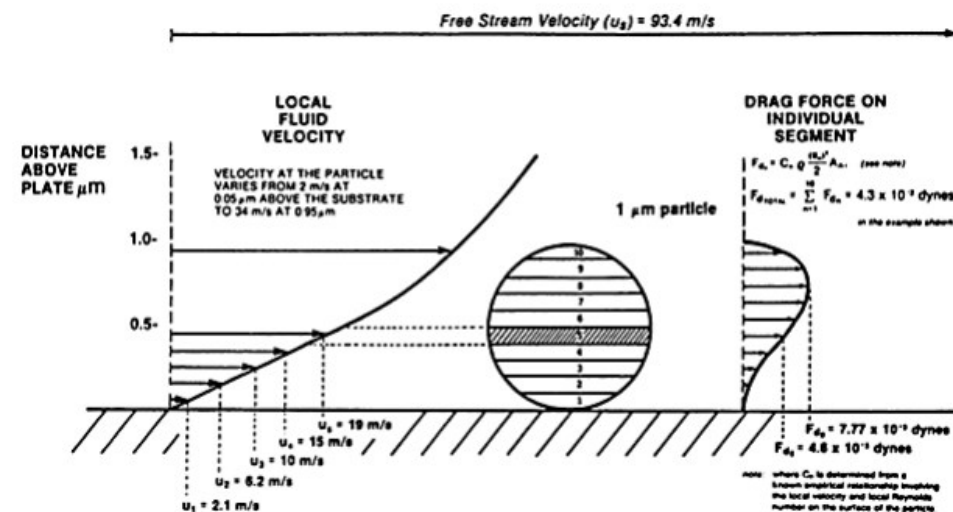
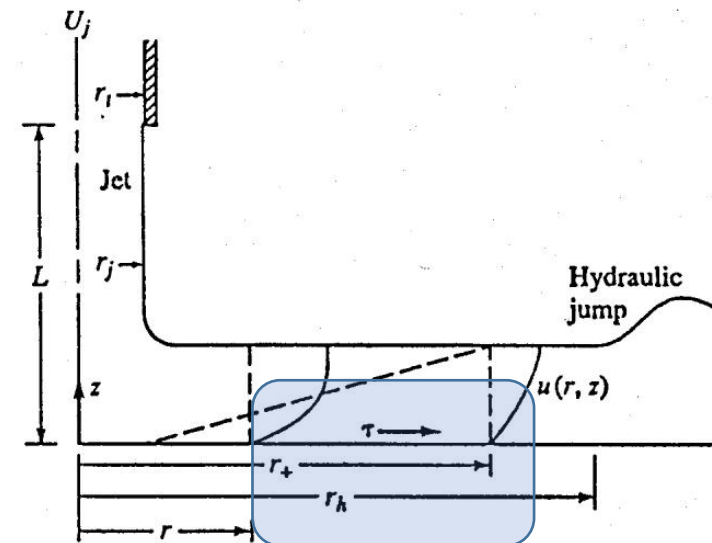
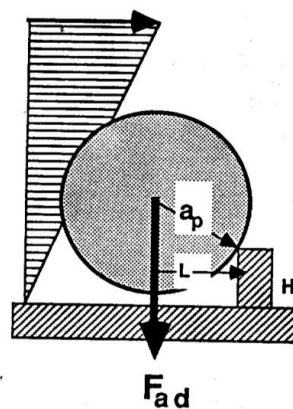


Oxidation induced by fixed HPR jets

# HPR Process

- Hydrodynamic model allows estimating the **shear stress  $\tau$  of the water jet**, which depends on flow rate and pressure
- Particle removal by rolling** if the water shear stress is greater than a critical shear stress  $\tau_0$ , related to the particle size, adhesion force and surface roughness

$$\tau_0 = \frac{F_{ad}}{44 a_p^2} \sqrt{2 \frac{H}{a_p} + \left(\frac{H}{a_p}\right)^2}$$

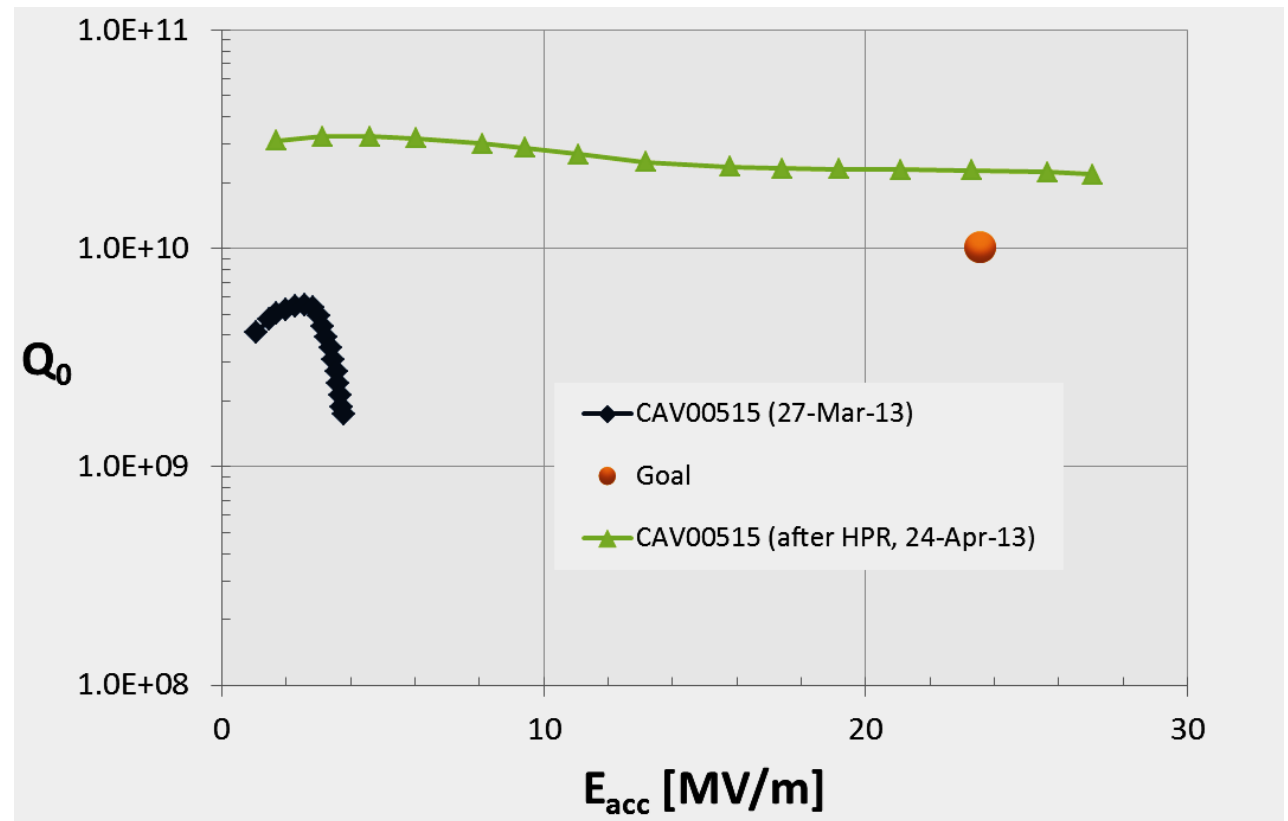


<https://doi.org/10.1063/1.494924>

# HPR Effect

Cavity had a **problem** in the 120°C treatment (vacuum system power failure) that **produced a rapid change of pressure (a bump) in the cavity during last pump-down**, with particle movement in the system. Consequences are clearly visible (**dark blue curve**).

**After HPR**, with **no further chemical etching**, cavity performances are completely recovered.





# HPR Systems



Rinsing cabinet of  
"old" DESY HPR system



Rinsing cabinet of "new" DESY  
HPR system with "plastic" cavity



CEA HPR system

# HPR QA

- **Examples of QC at HPR systems**  
(DESY, EXFEL cavities production@ companies)
- Check of **Point-of-use supply** water quality:
  - **UPW conductivity**
  - **Particles: online** particle counter
  - **Particles: off-line** sampling & identification (SEM optical microscope)
  - **TOC: online monitoring**
  - **TOC drain line: sampling, after maintenance**
  - **Bacteria (=> offline)**

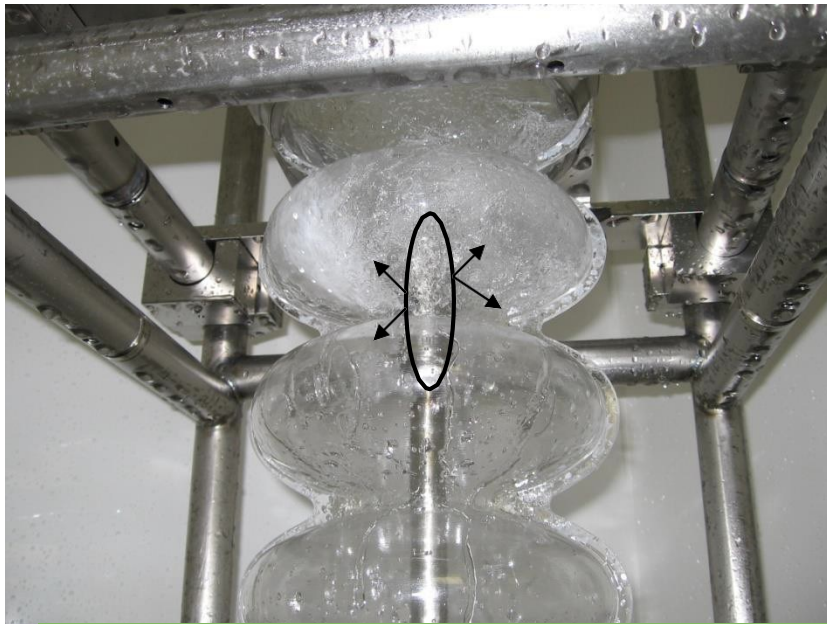
HPR Head QA



HPR Water Collector

# HPR Spray Head Optimization

- For a given pump displacement **the nozzle opening diameter and number of nozzles sets the system pressure and flow rate**
- The **HPR spray head needs to be optimized for each cell geometry!**



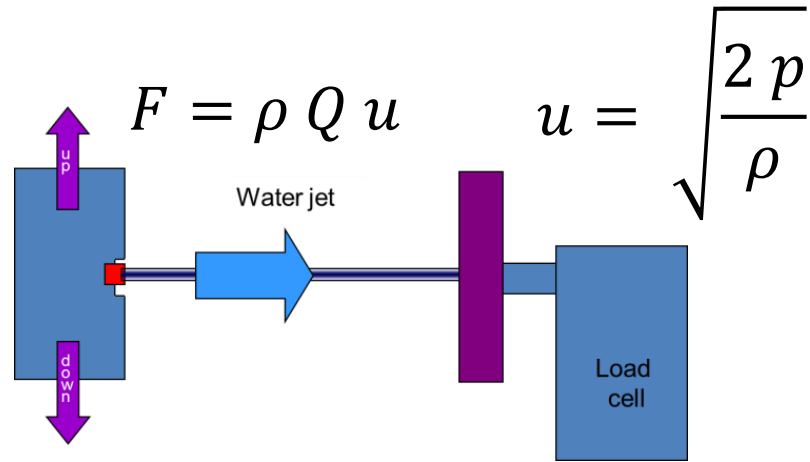
Very effective on irises



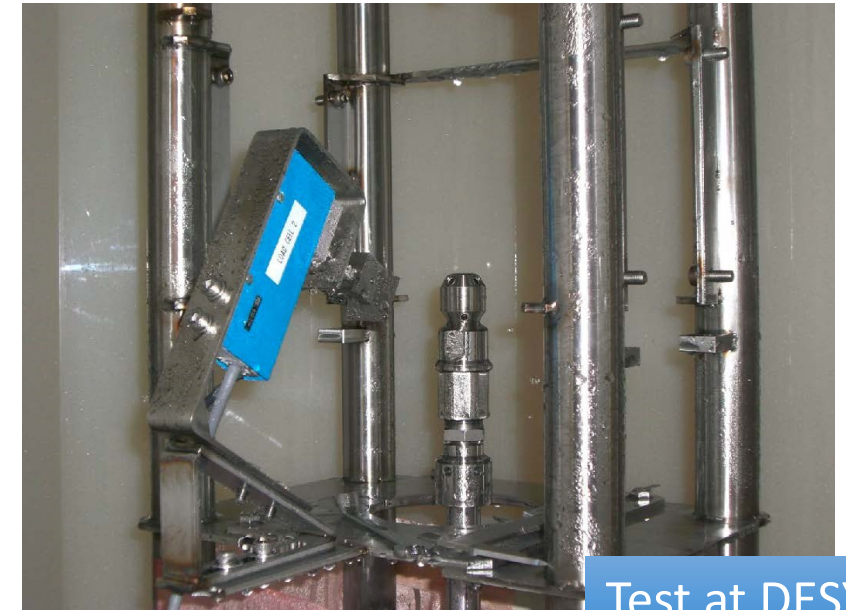
Equator fill with water → too high flow rate

# HPR Water Jet Characterization (INFN-LASA)

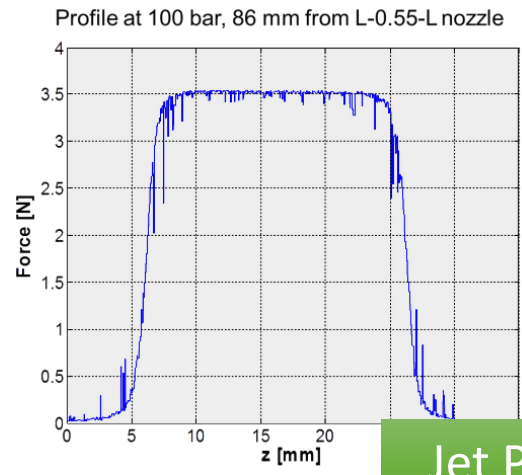
- Use a load-cell to measure the jet force



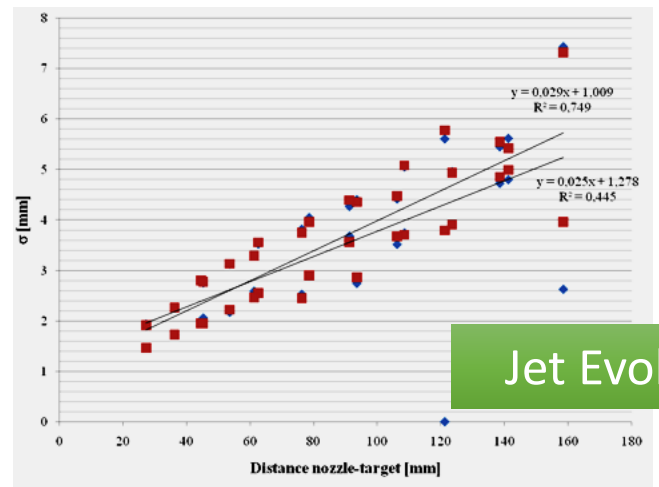
u = velocity  
 Q = flow  
 p = pressure  
 rho = density



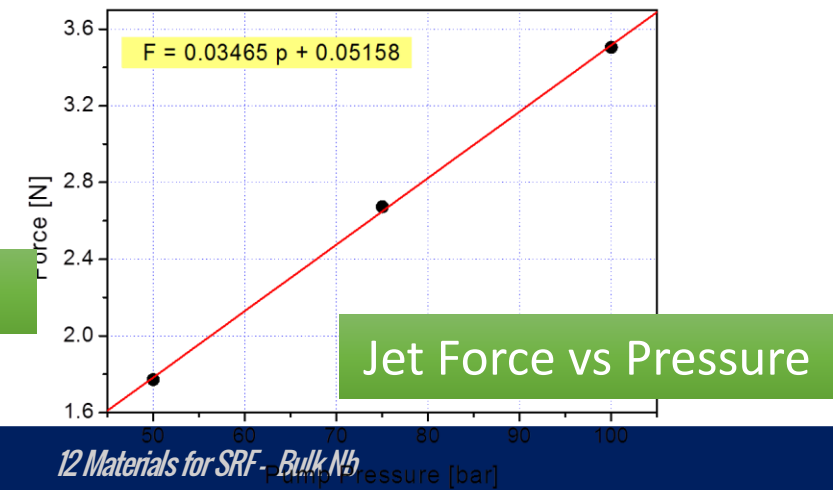
Test at DESY



Jet Profile



Jet Evolution



Jet Force vs Pressure

# Cavity preparation for SRF qualification

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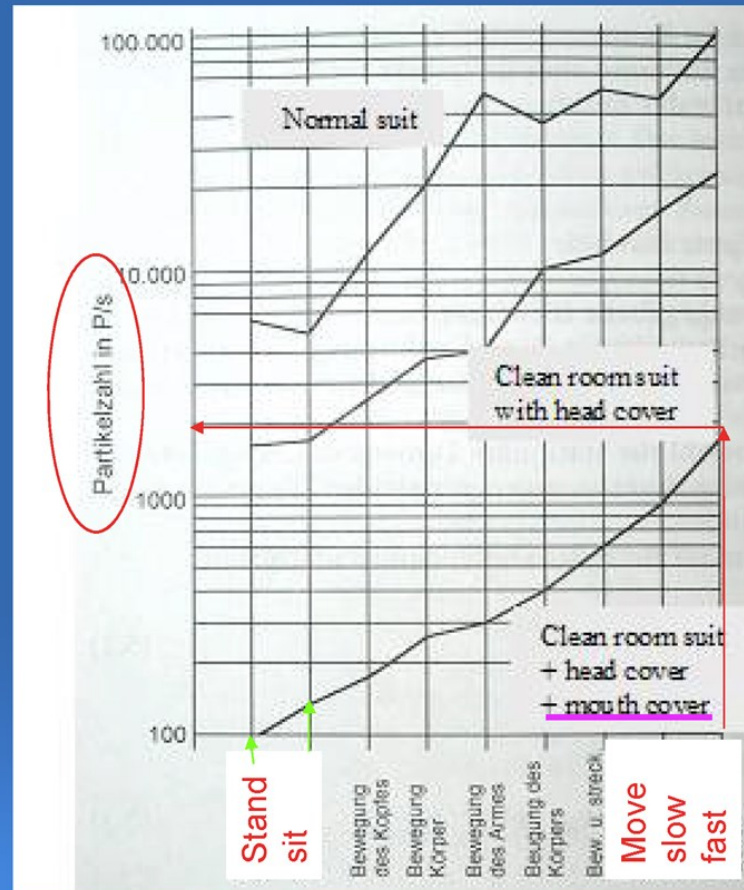
# What is a cleanroom?

## The ISO 44644 definition

- “A room in which the **concentration of airborne particles is controlled**, and which is constructed and used in a manner to **minimize the introduction, generation and retention of particles** inside the room and in which other relevant **particles** inside the room and in which other relevant parameters, e.g. **temperature, humidity and pressure**, are **controlled as necessary**.”
- **A cleanroom** is likely to have between **some tens of air changes per hour up to many hundreds of them**.
- A cleanroom uses filters that would normally be 99.97 % and more efficient in removing particles greater than  $0.3 \mu\text{m}$  from the room air supply. These filters are known as **High Efficiency Particle Air (HEPA) filters**, although **Ultra Low Particle Air (ULPA) filters**, which have a higher efficiency, are used in microelectronic fabrication areas.

# Human generated particle

One major part inside a cleanroom is PERSONAL



1st Dress code



A. Matheisen SRF workshop 2007  
Beijing China October 2007

# Clean room «dress» code

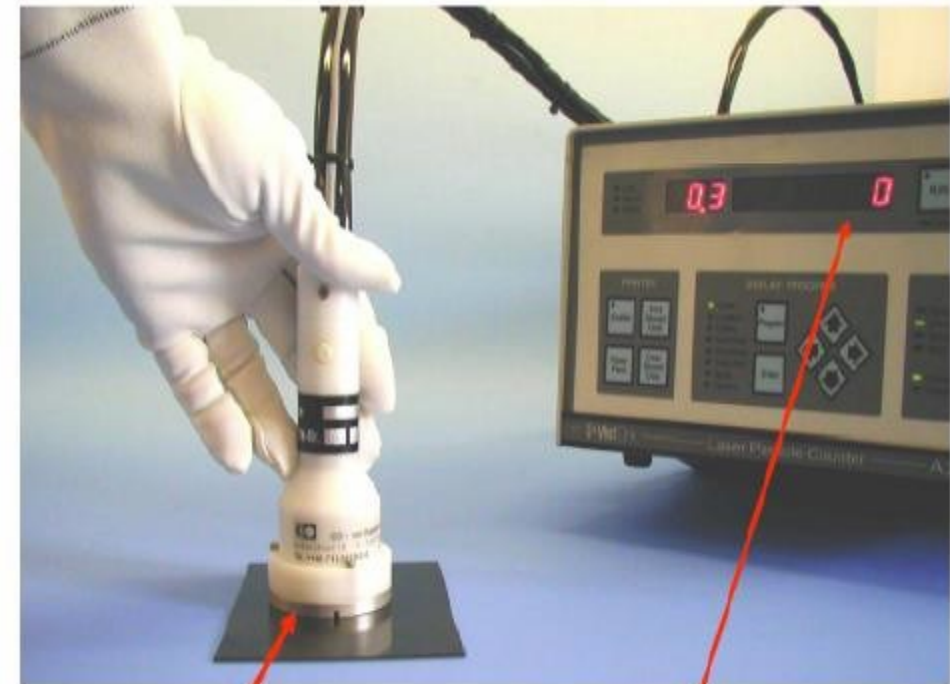
- **People are a major source of particulate contamination** inside a clean room through:
  - **Body Regenerative Processes** - Skin flakes, oils, perspiration and hair.
  - **Behavior** - Rate of movement, sneezing and coughing.
  - **Attitude** - Work habits and communication between workers.





# Particle counters

- To ensure the proper cleanness, **all components** before installation need to be **washed, rinsed and particle counted**



Samplehead

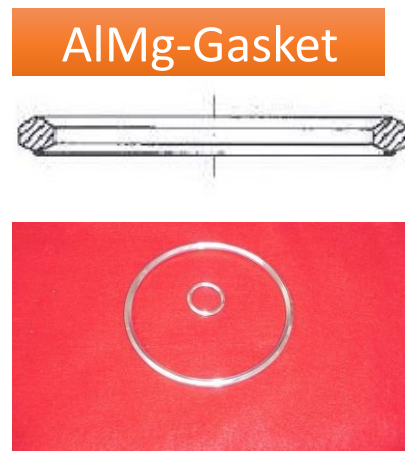
Particlecounter

# Cavity Assembly

- The cavity strings have to be **vacuum tight to a leak rate of  $< 1 \cdot 10^{-10}$  mbar l/sec**
- The **sealing gaskets and hardware** have to be **reliable and particulate-free**
- The clamping hardware should minimize the space needed for connecting the beamlines

- **UHV Gasket**

- Present choice for SRF cavities:  
diamond-shaped AlMg<sub>3</sub> –gaskets +  
NbTi flanges + bolts

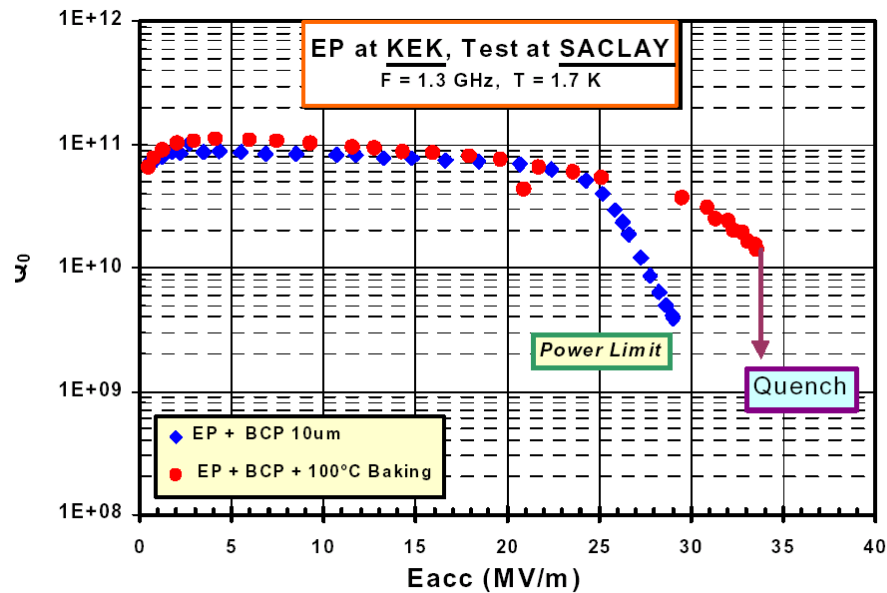


# Cavity preparation for SRF qualification

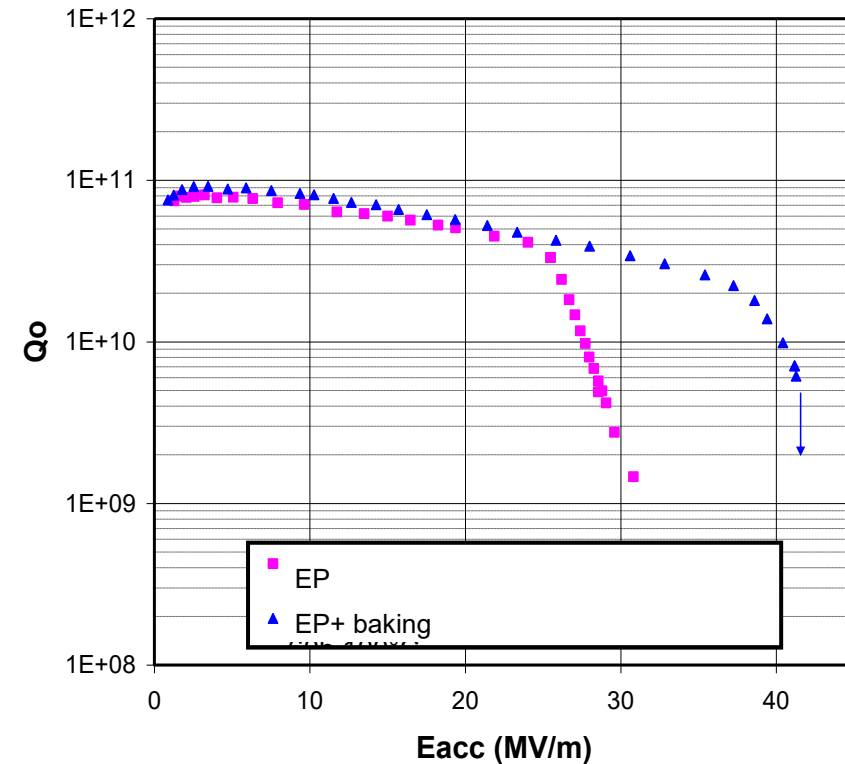
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- Clean assembly
- Clean evacuation
- **Low-temperature baking**

# Why low temperature baking (120 °C)?

- **Baking:** shifts high field dissipation to higher field
  - Discovered at Saclay in 1998 (B. Visentin)
  - Low temperature treatment : 110-120°C, 48 H : few changes expected
  - **Dramatic effect on performances**
  - Still resists full explanation



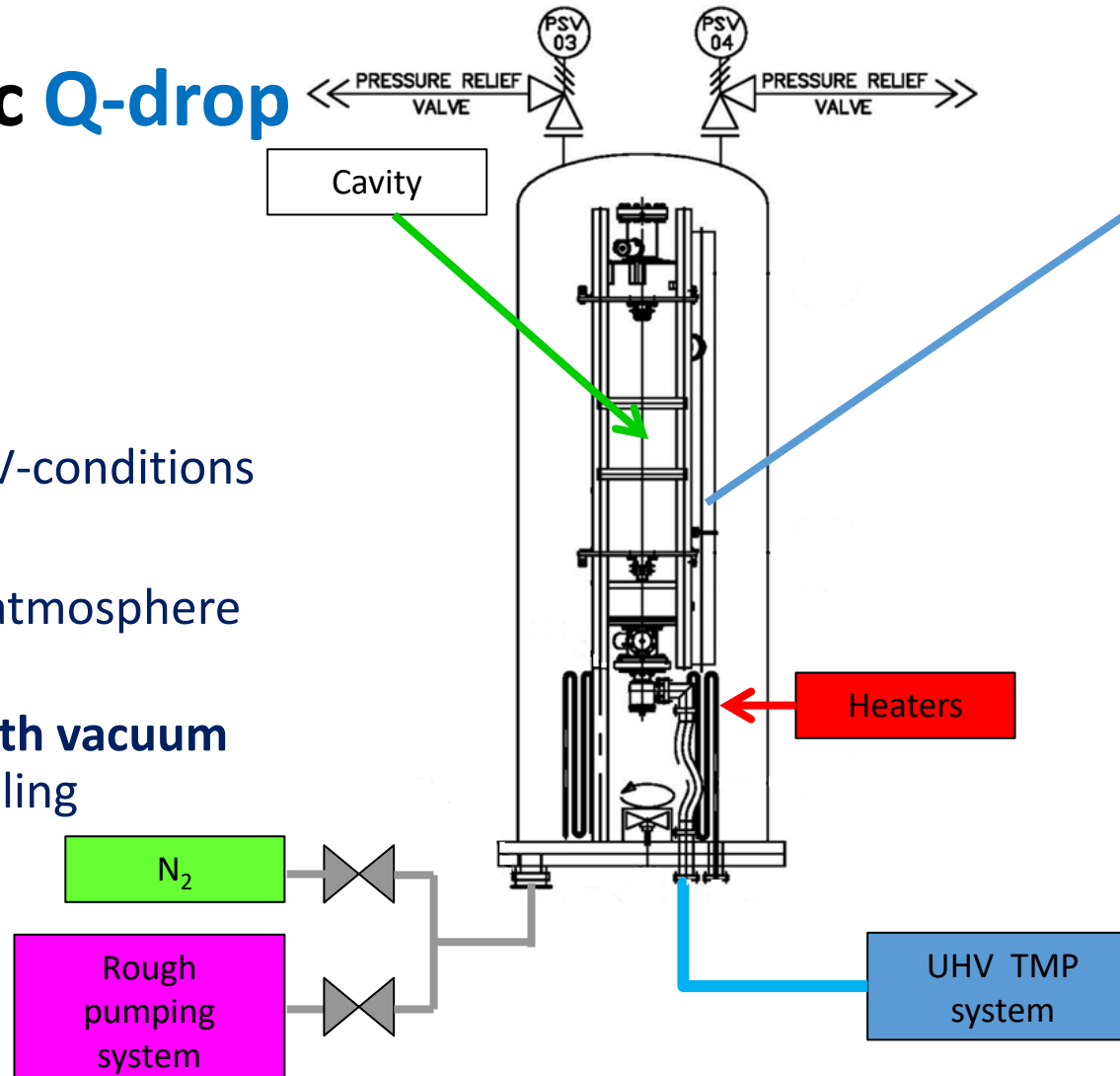
Collaboration KEK / CEA Saclay, August 1999



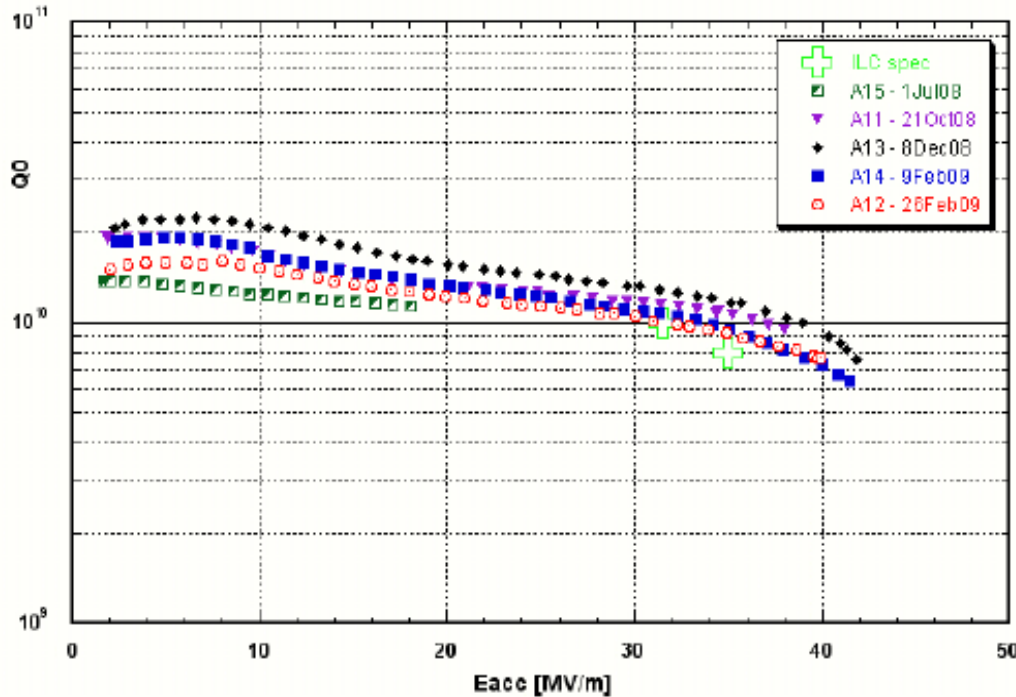
# 120 °C baking

## Cure of characteristic Q-drop

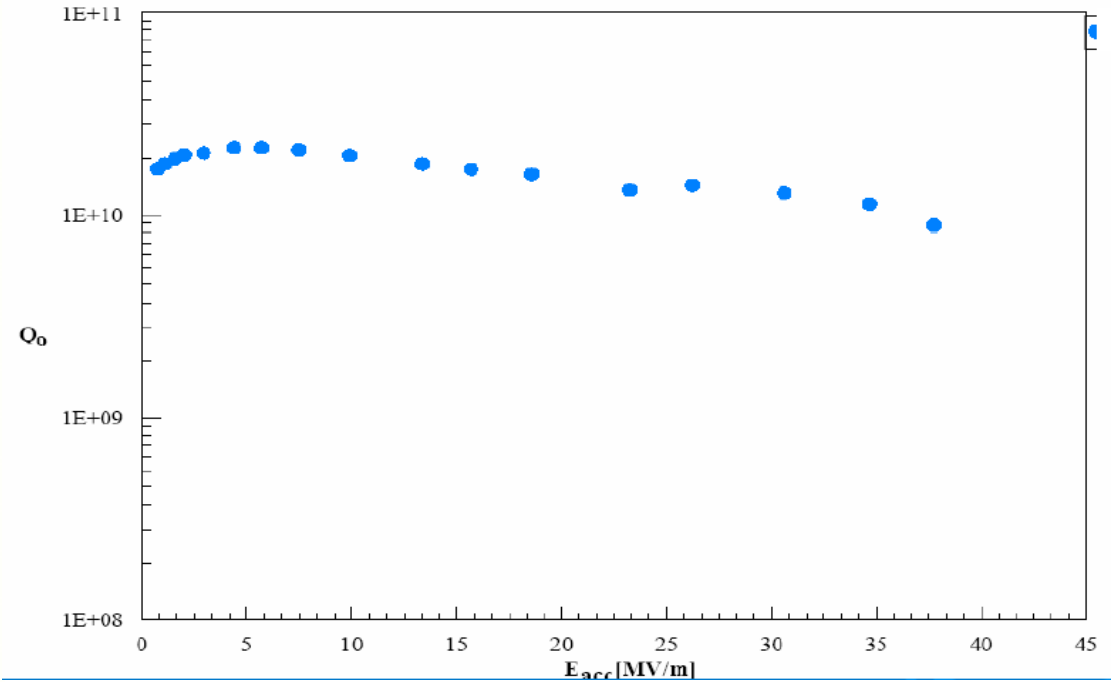
- **Standard recipe:**
  - **T = 110 - 125°C for 48h**
  - **Active pumping (TMP)**
  - Oil free vacuum system, UHV-conditions
  - Fully assembled cavity
  - **Nitrogen or argon** external atmosphere to avoid oxidation
  - External volume: **purging with vacuum before** inert gas (N<sub>2</sub> or Ar) filling



# If everything went well you will get ...



JLAB



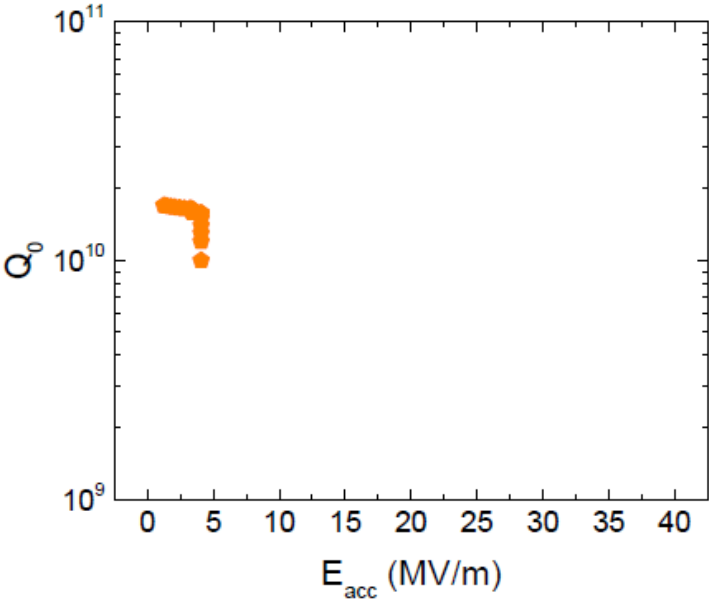
DESY

$E_p \cong 80$  MV/m,  $B_p \cong 170$  mT can be achieved in the vertical test of 9-cell ILC cavities ( $\sim 1$  m<sup>2</sup> of Nb surface)

# Nb bulk cavities: performance evolution



# SRF Performance Evolution



3-4 MV/m  
Multipacting



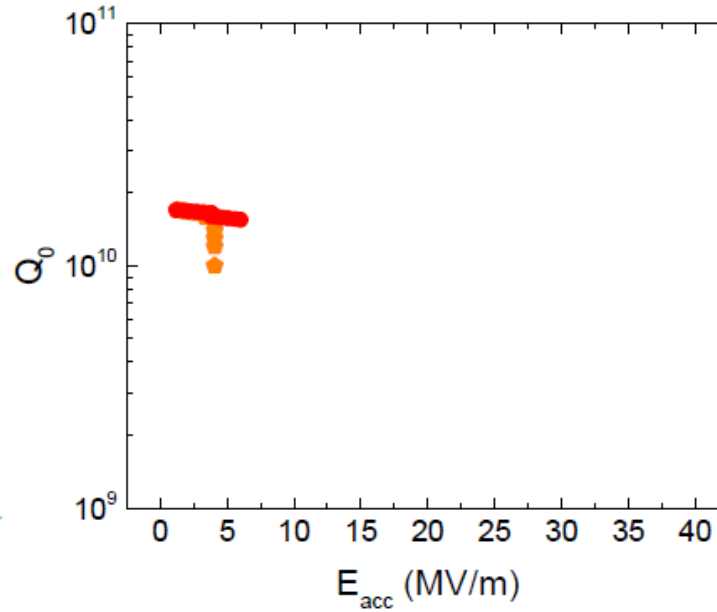
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# SRF Performance Evolution



1.3 GHz, 2K

3-4 MV/m  
Multipacting

5 MV/m  
Thermal Breakdown

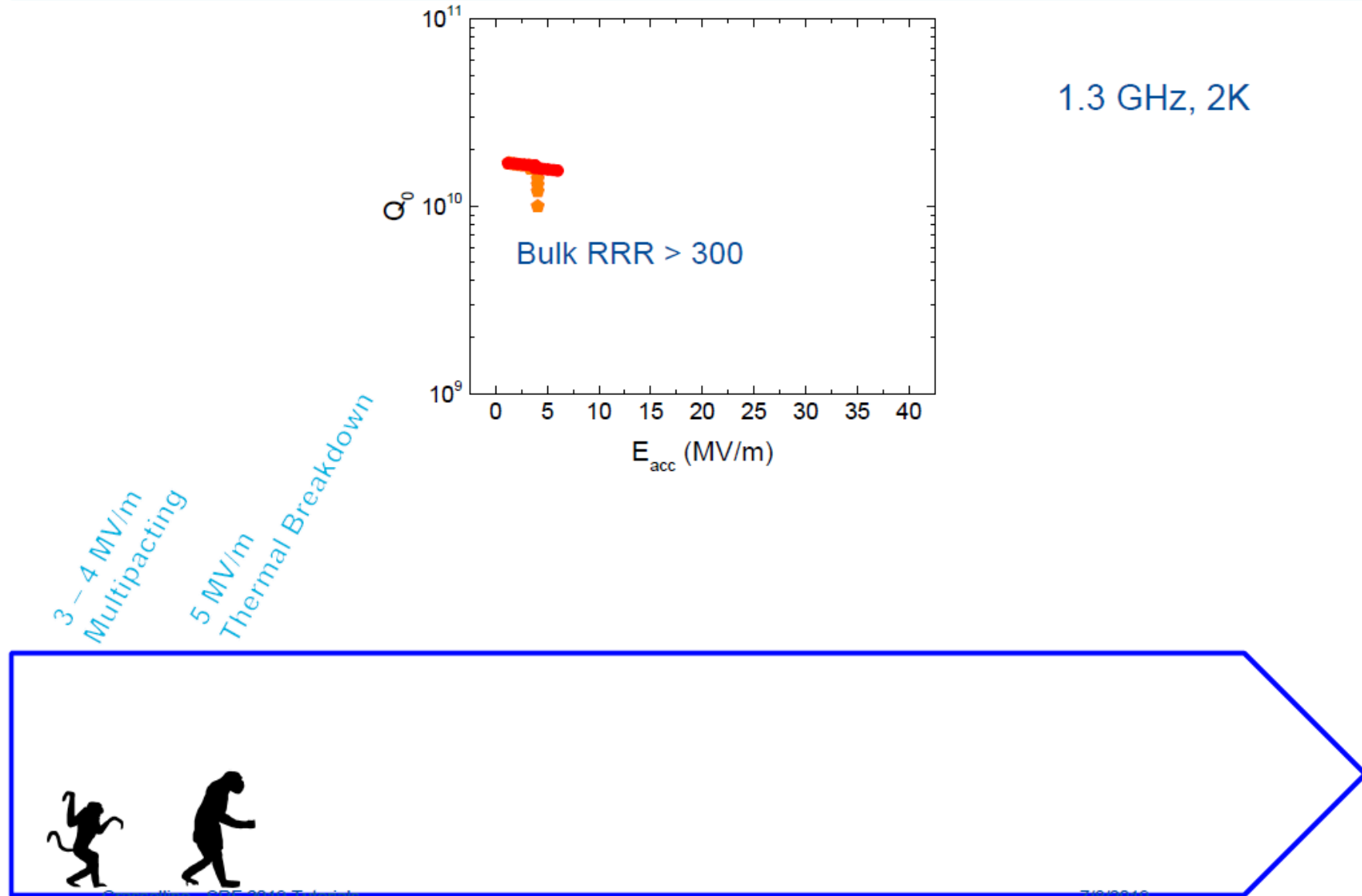


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# SRF Performance Evolution

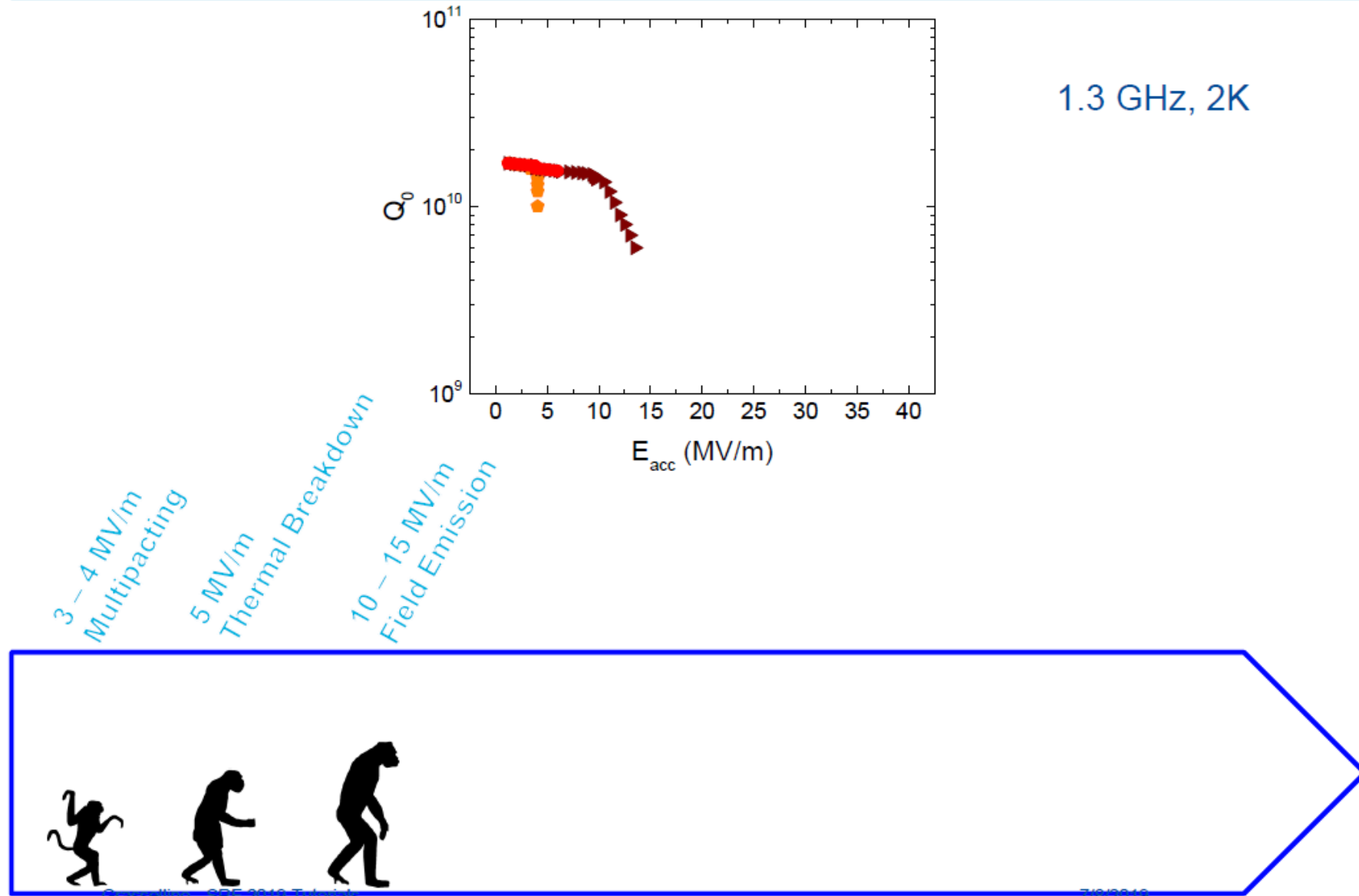


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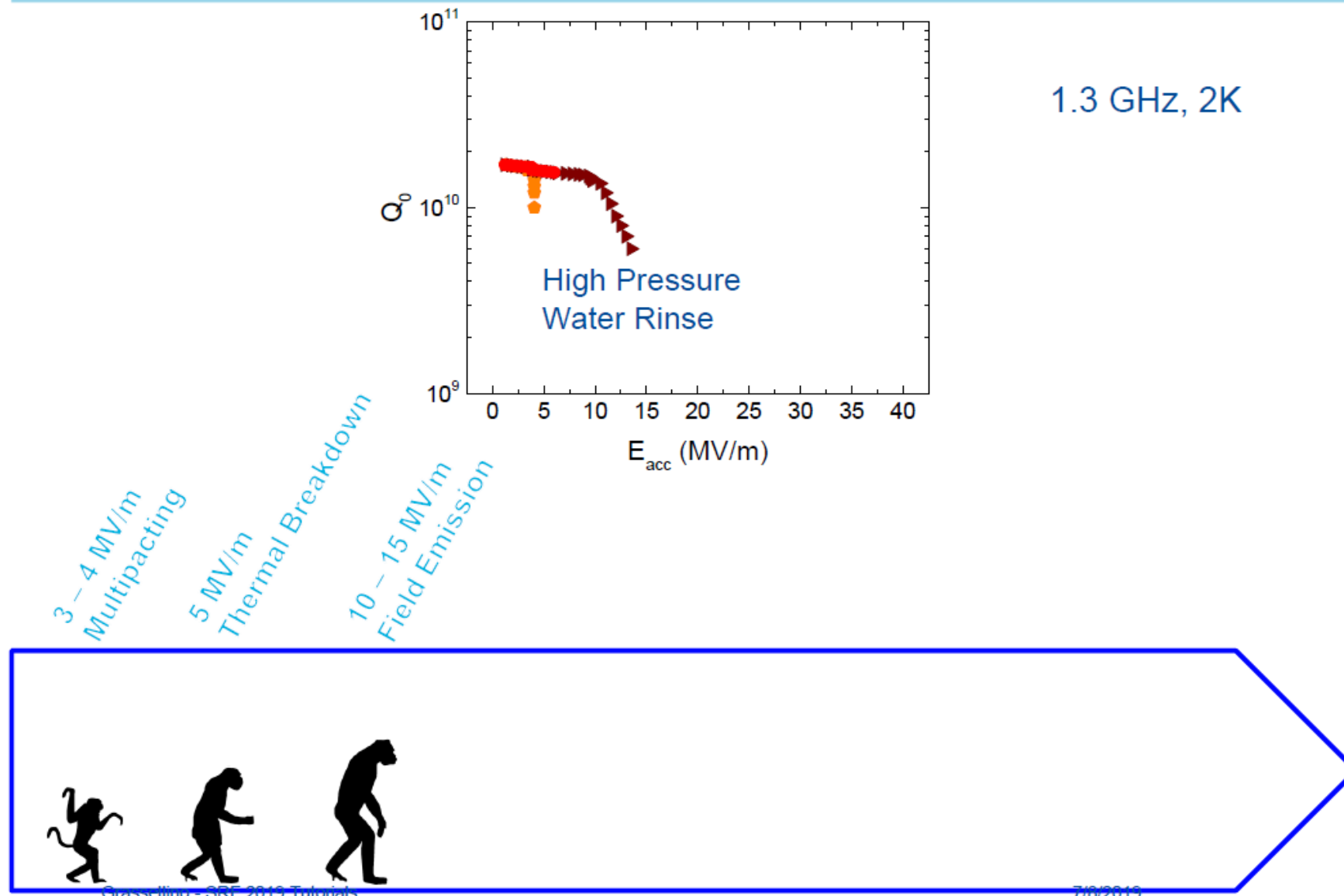
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# SRF Performance Evolution



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# SRF Performance Evolution

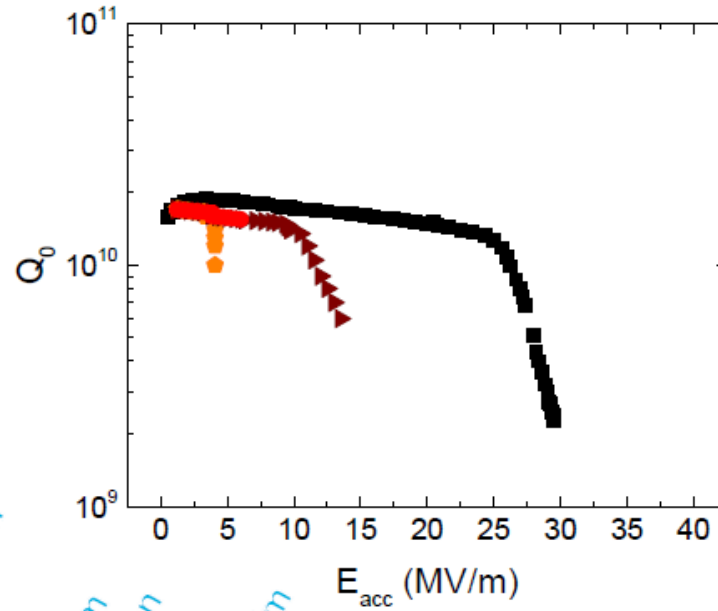


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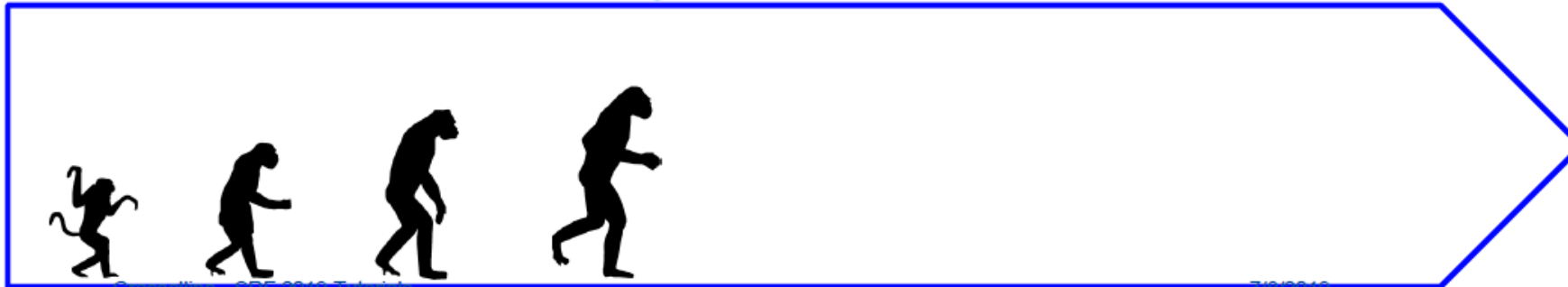


3 - 4 MV/m  
Multipacting

5 MV/m  
Thermal Breakdown

10 - 15 MV/m  
Field Emission

20 - 25 MV/m  
High field  
Q-SLOPE



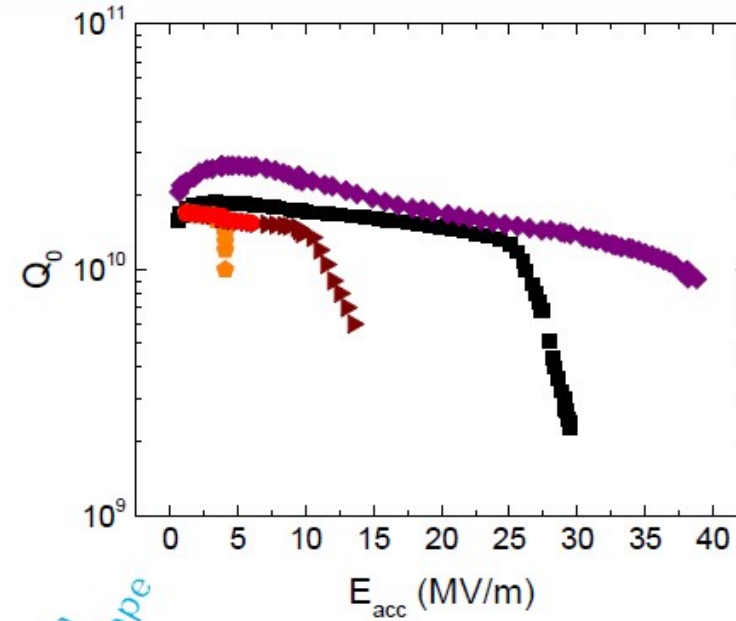
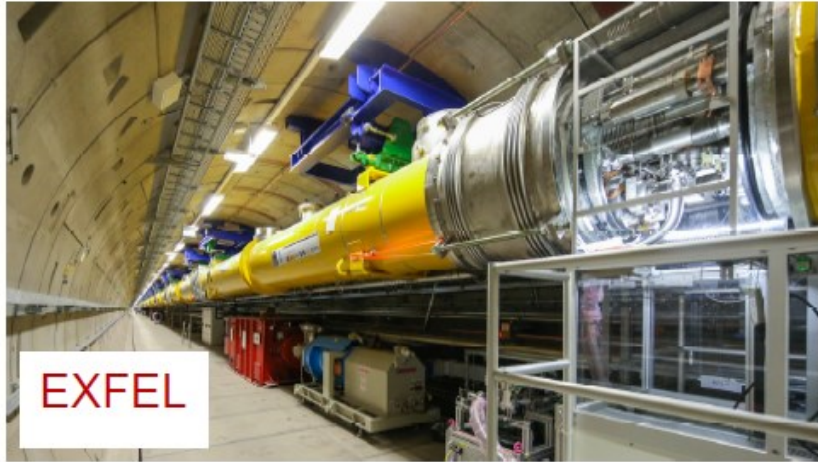
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# SRF Performance Evolution

1.3 GHz, 2K



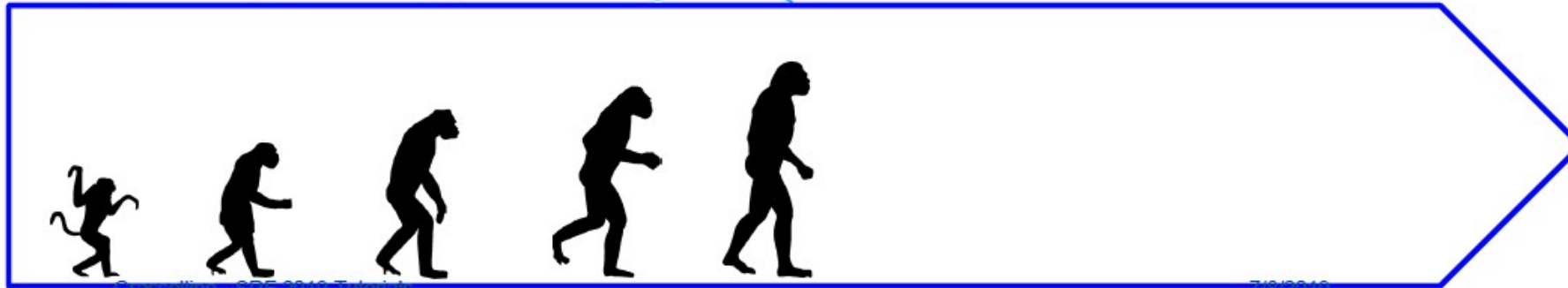
3 - 4 MV/m  
Multipacting

5 MV/m  
Thermal Breakdown

10 - 15 MV/m  
Field Emission

20 - 25 MV/m  
High field  
Q-SLOPE

35 - 40 MV/m  
mid field Q-slope



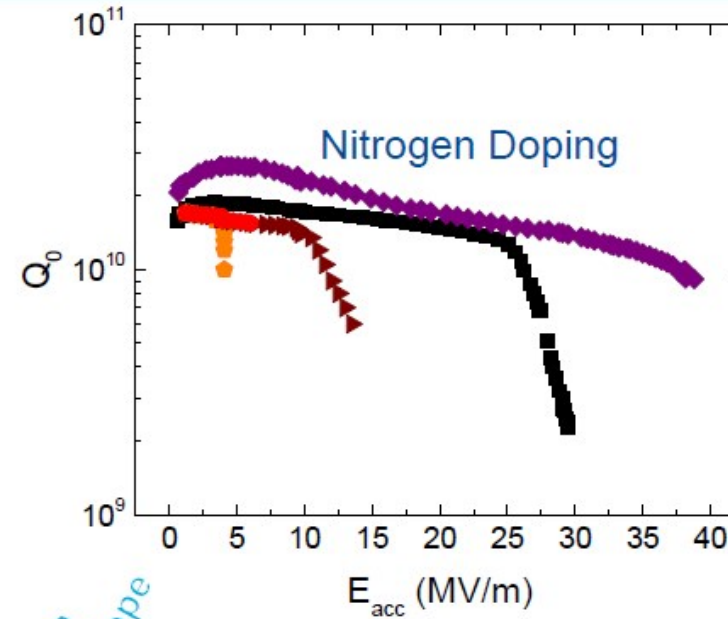
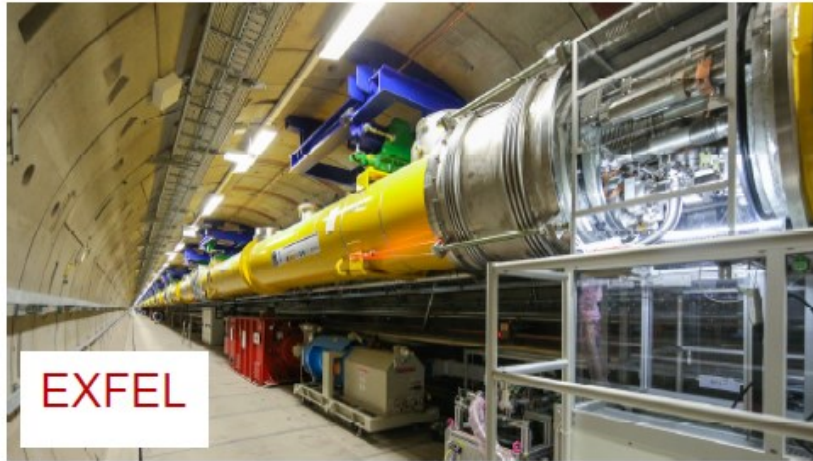
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# SRF Performance Evolution

1.3 GHz, 2K



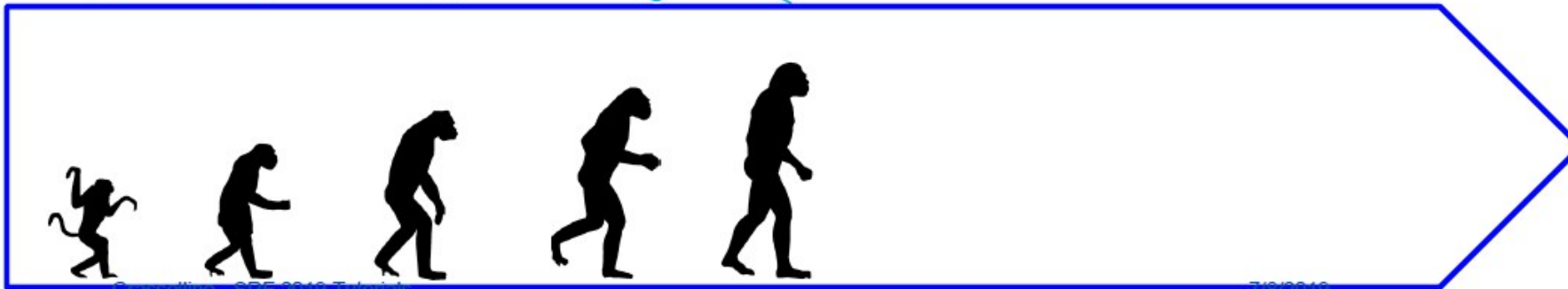
3 - 4 MV/m  
Multipacting

5 MV/m  
Thermal Breakdown

10 - 15 MV/m  
Field Emission

20 - 25 MV/m  
High field  
Q-SLOPE

35 - 40 MV/m  
mid field Q-slope

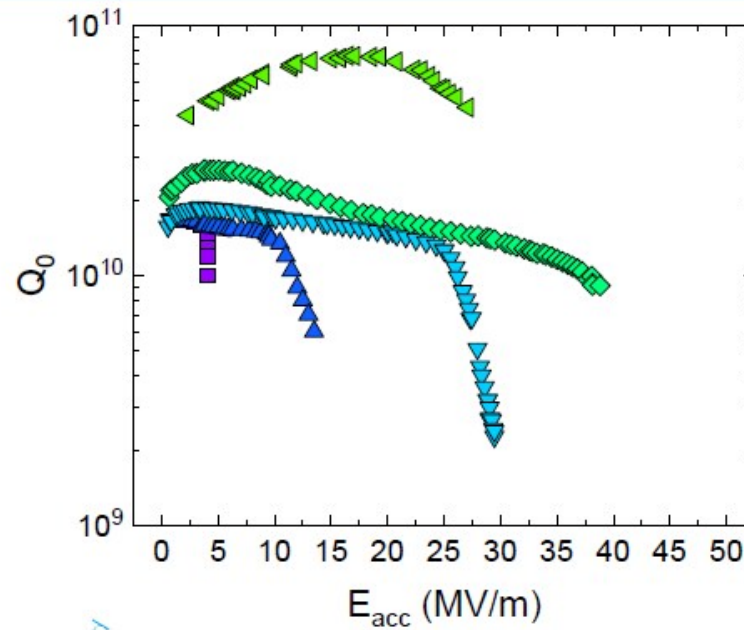


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# SRF Performance Evolution – 2013: N doping!



3 – 4 MV/m  
Multipacting

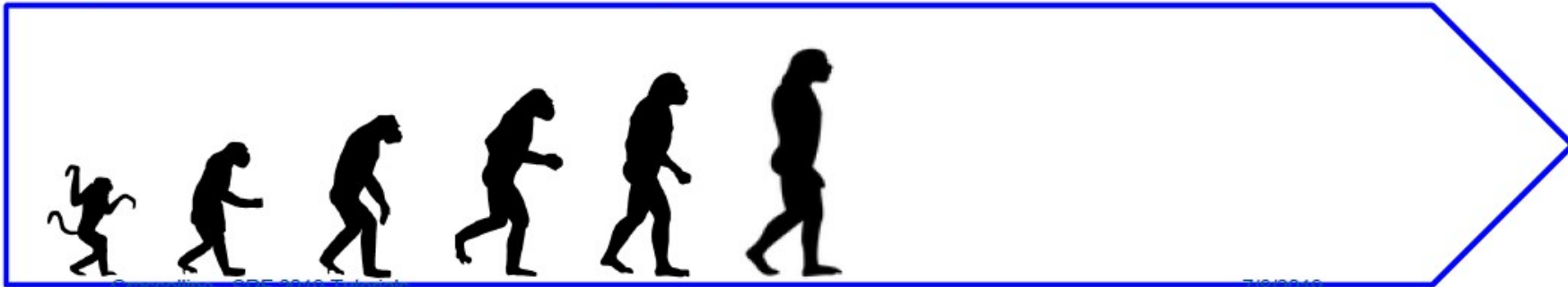
5 MV/m  
Thermal Breakdown

10 – 15 MV/m  
Field Emission

20 - 25 MV/m  
Q-SLOPE

35 – 40 MV/m

$Q > 5 \times 10^{10}$   
At medium field



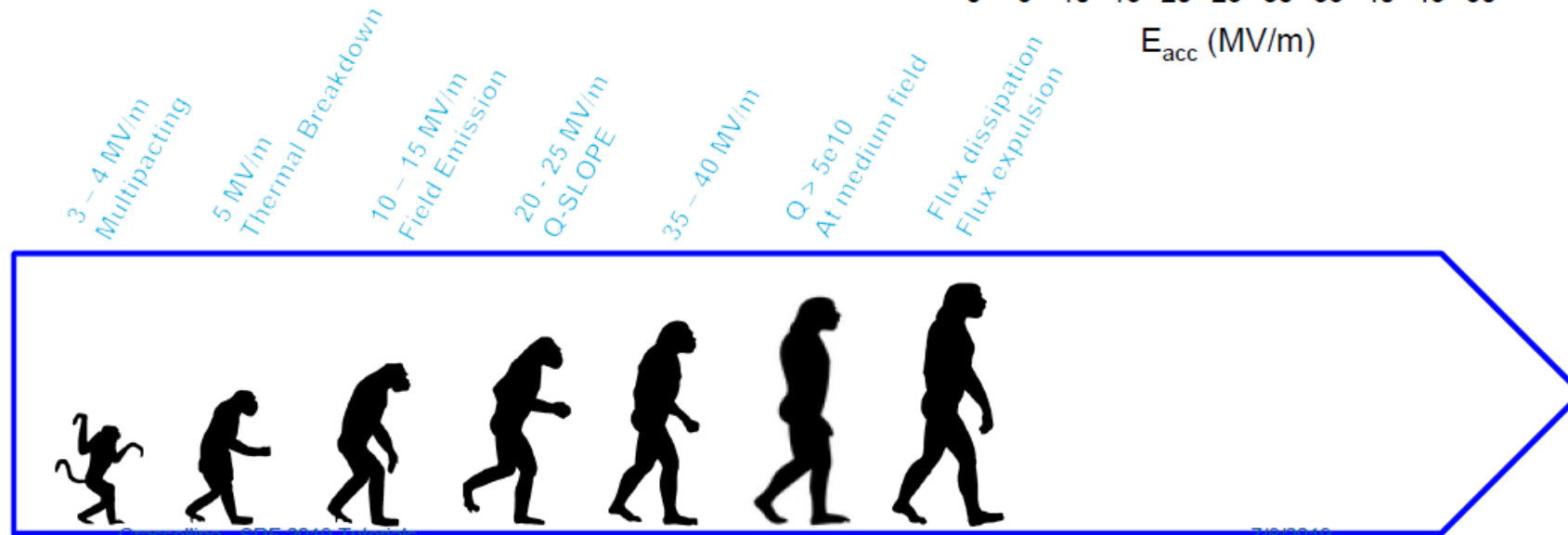
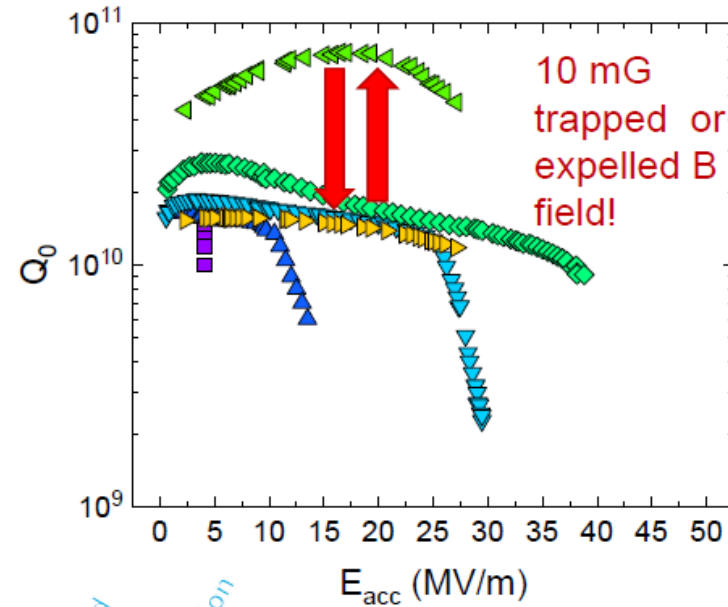
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# SRF Performance Evolution – 2014: Magnetic flux trapping with slow cooldown/ efficient expulsion with fast



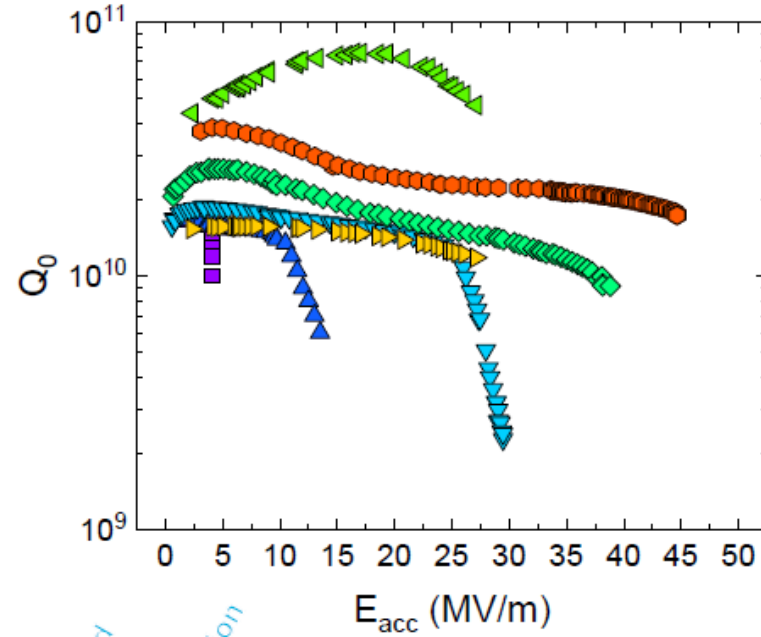
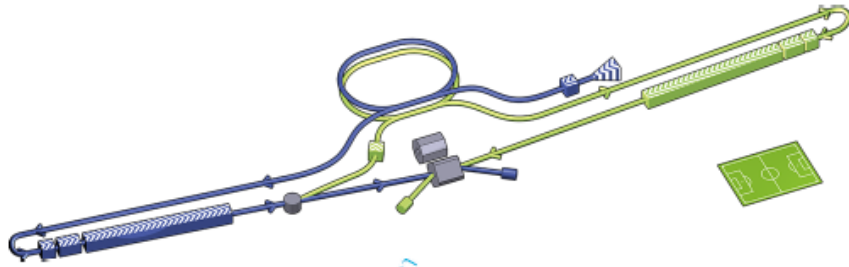
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# SRF Performance Evolution – 2017: N infusion

ILC cost reduction??



3 – 4 MV/m  
Multipacting

5 MV/m  
Thermal Breakdown

10 – 15 MV/m  
Field Emission

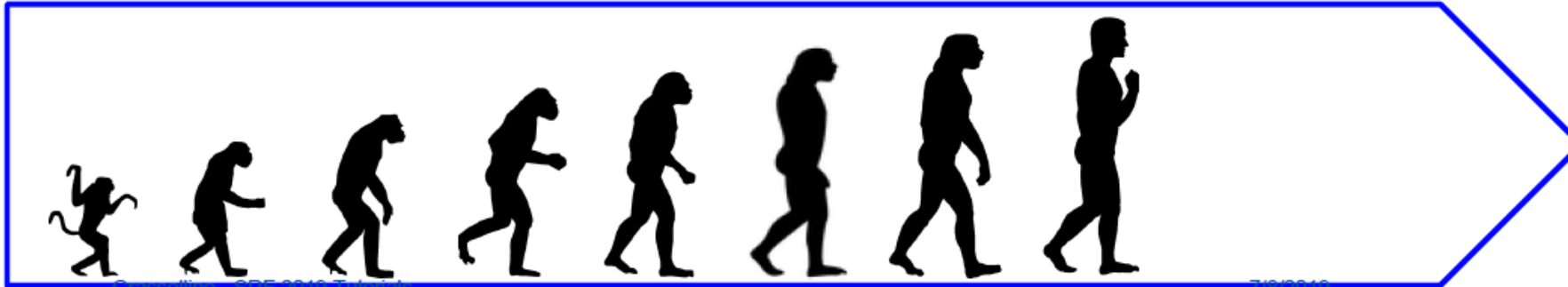
20 – 25 MV/m  
Q-SLOPE

35 – 40 MV/m

$Q > 5e10$   
At medium field

Flux dissipation  
Flux expulsion

$Q > 2e10$   
At high field



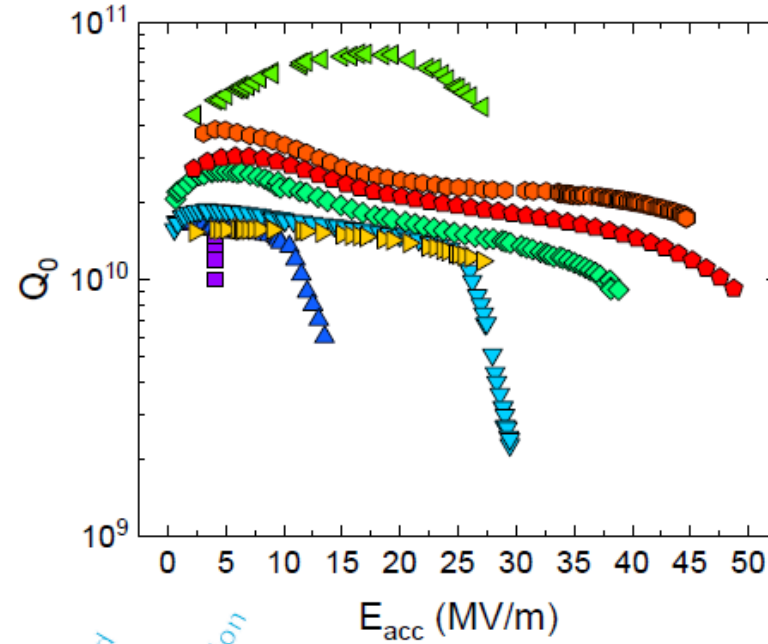
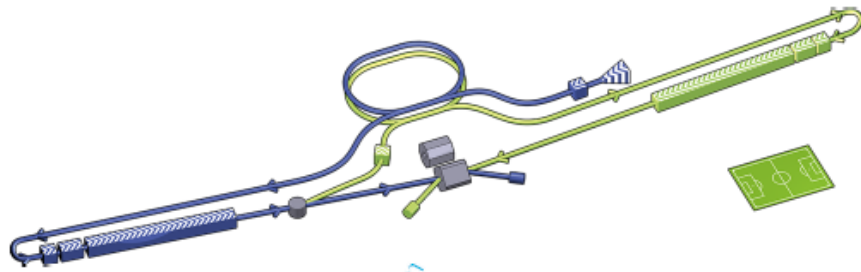
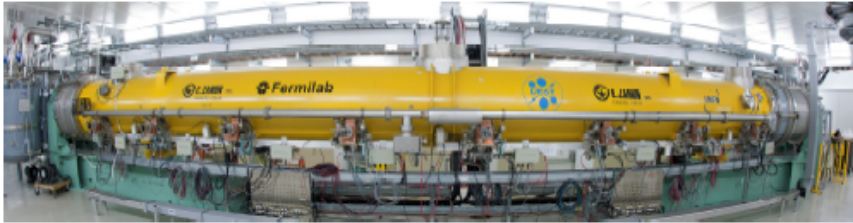
Grassellino - SRF 2019 Tutorials

7/6/2019

Grassellino, SRF2019 Tutorials

# SRF Performance Evolution – 2018: the 75C bake

ILC cost reduction??



3-4 MV/m  
Multipacting

5 MV/m  
Thermal Breakdown

10-15 MV/m  
Field Emission

20-25 MV/m  
Q-SLOPE

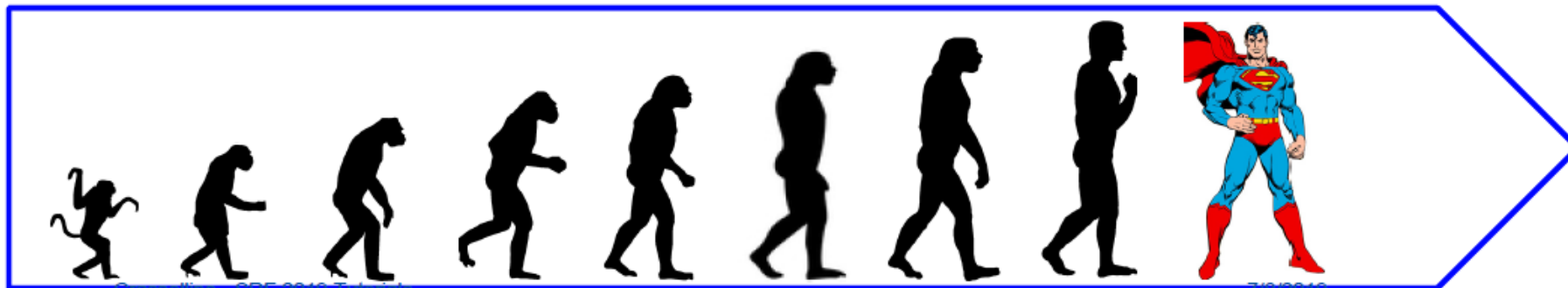
35-40 MV/m

$Q > 5e10$   
At medium field

Flux dissipation  
Flux expulsion

$Q > 2e10$   
At high field

50 MV/m !!!

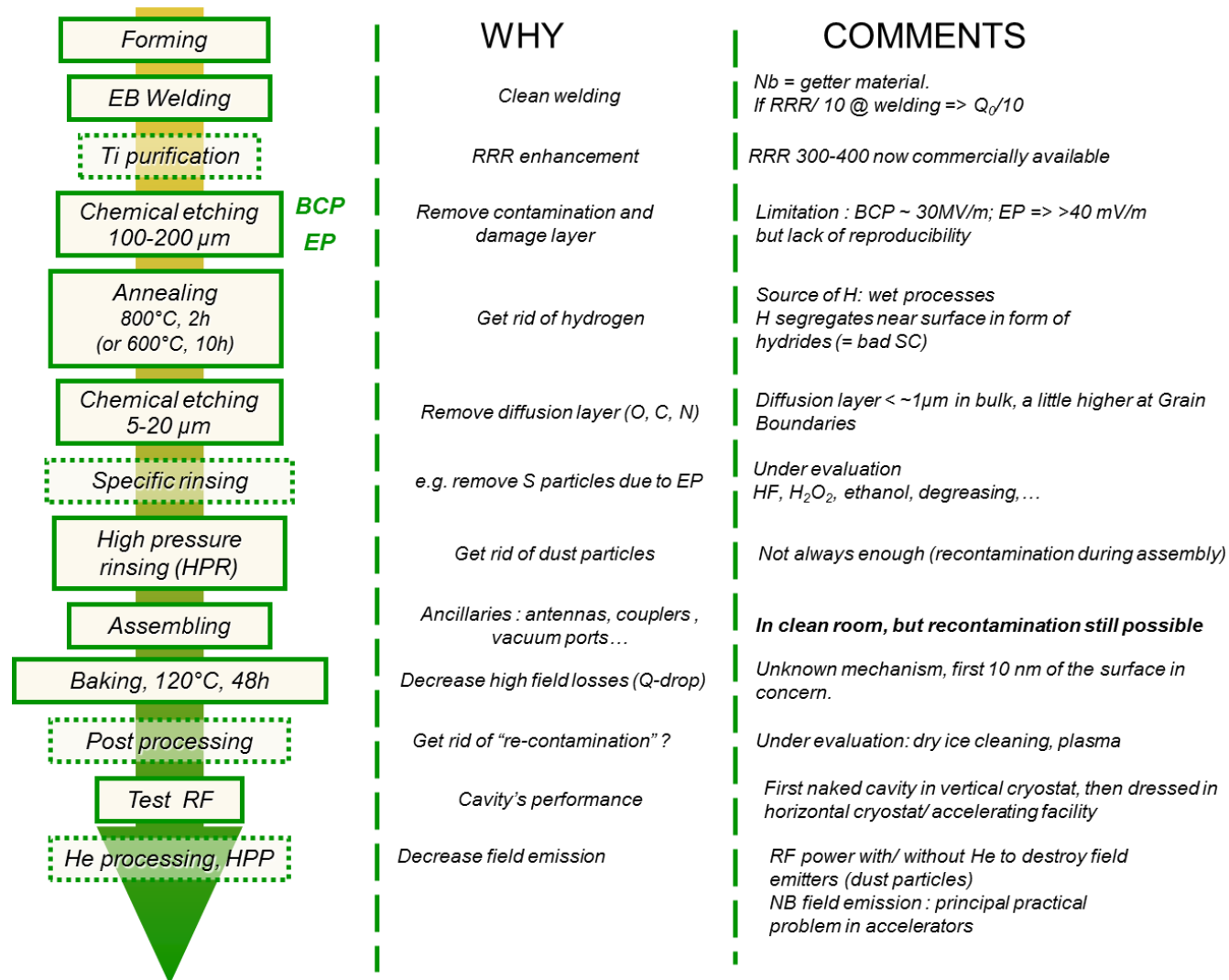


Grassellino - SRF 2019 Tutorials

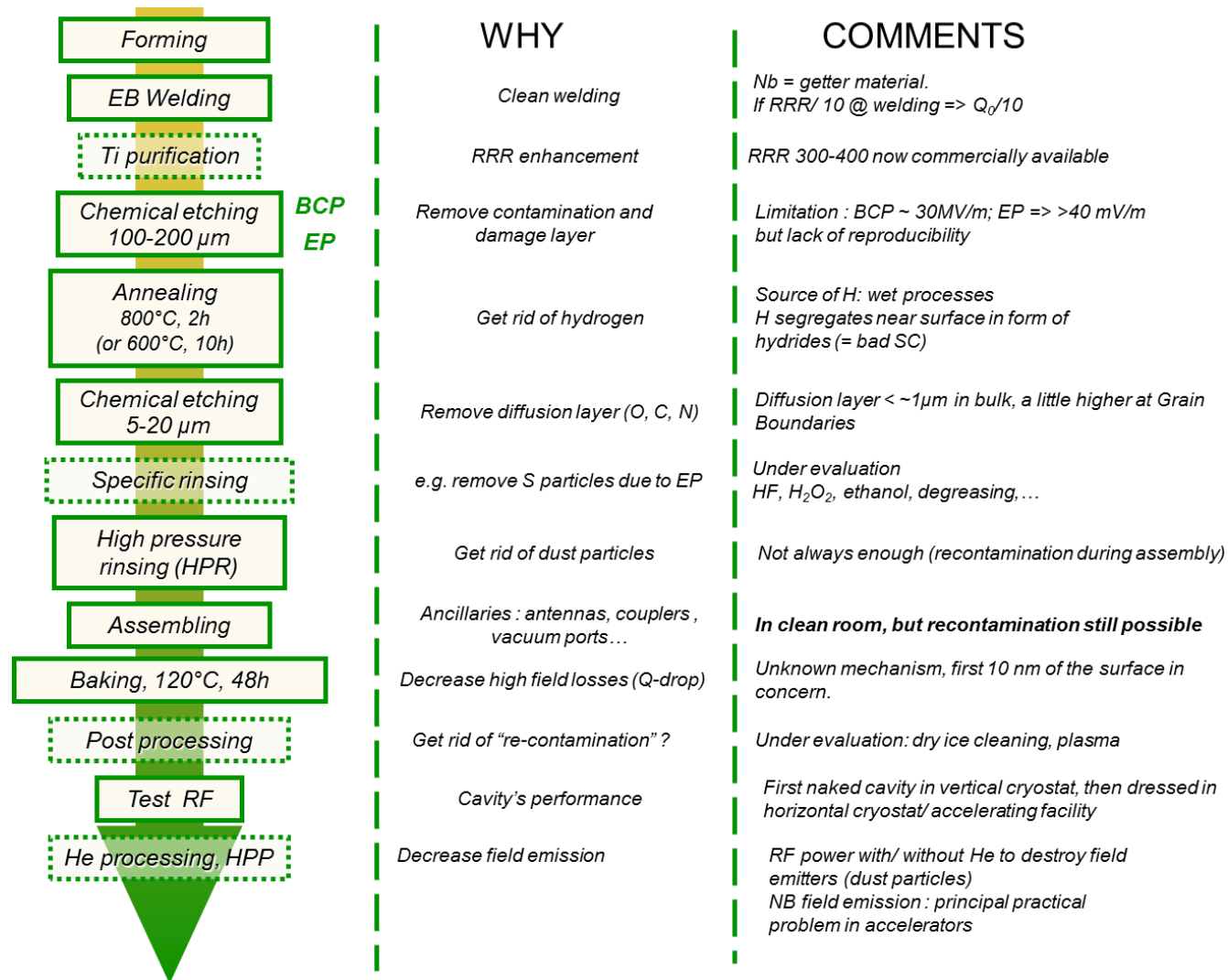
7/6/2013

Grassellino, SRF2019 Tutorials

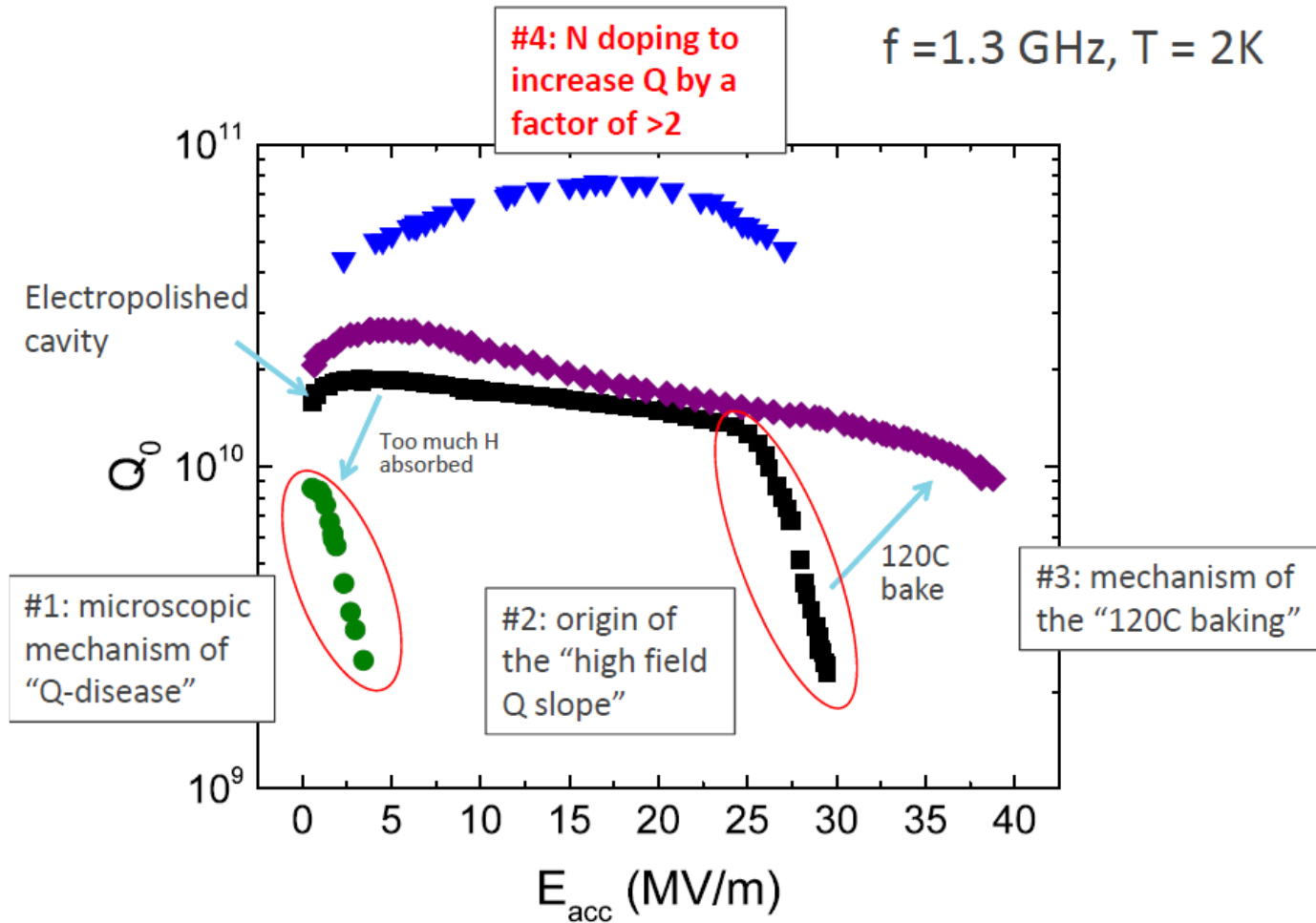
# Cavities' general fabrication scheme



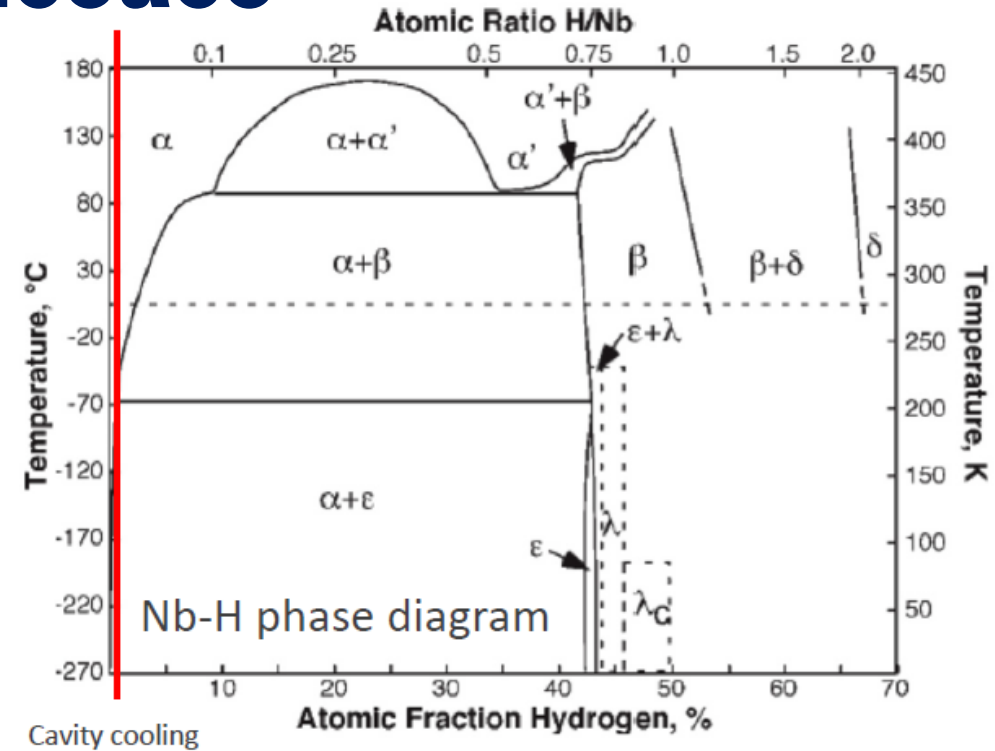
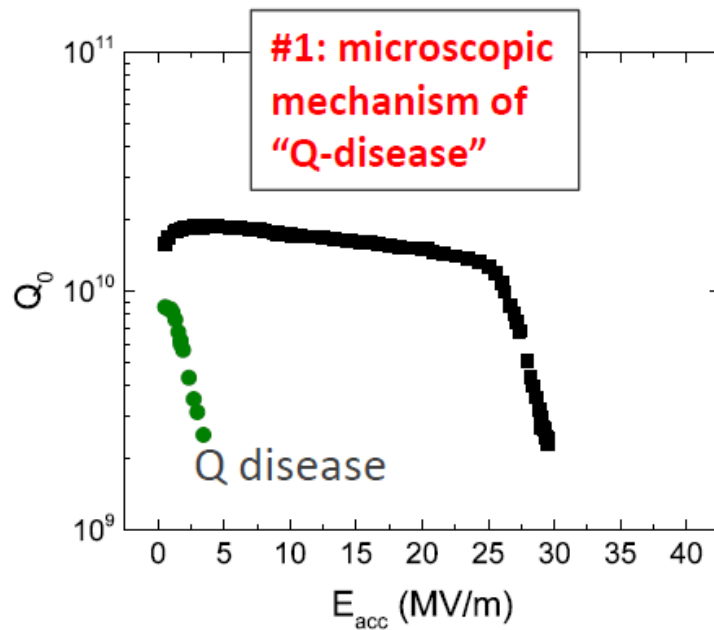
# Cavities' general fabrication scheme



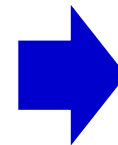
# Possible explanations to performance limitations and solutions



# Q-disease



Q-disease is related to the **excess of hydrogen**, which forms non-SC Niobium hydrides upon cooldown



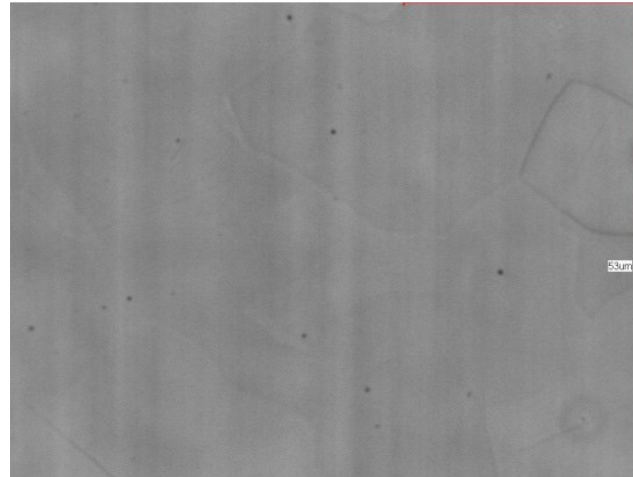
**600-800 °C vacuum anneal to degas hydrogen**

# Microscopic Mechanism of Q-disease

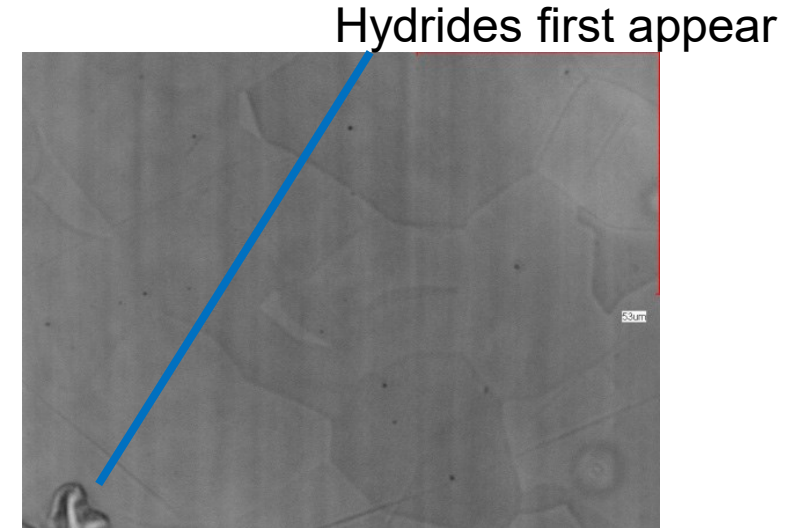


cryostage in the laser confocal scanning microscope

## COOLDOWN



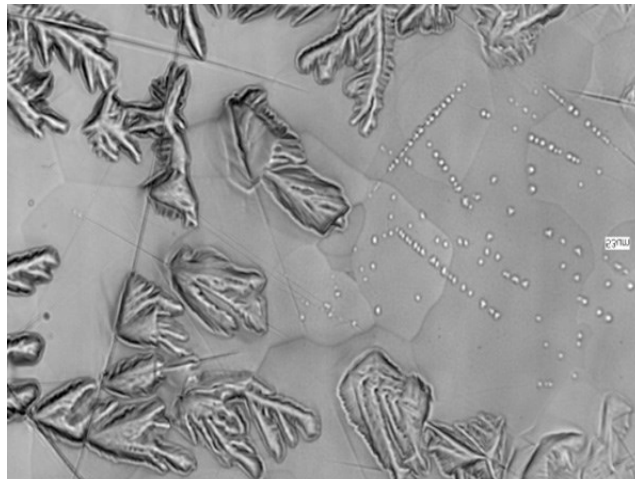
T = 300 K



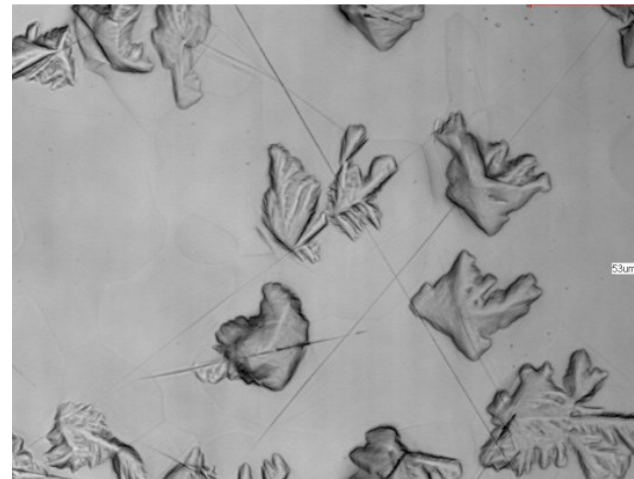
Hydrides first appear

T = 160 K

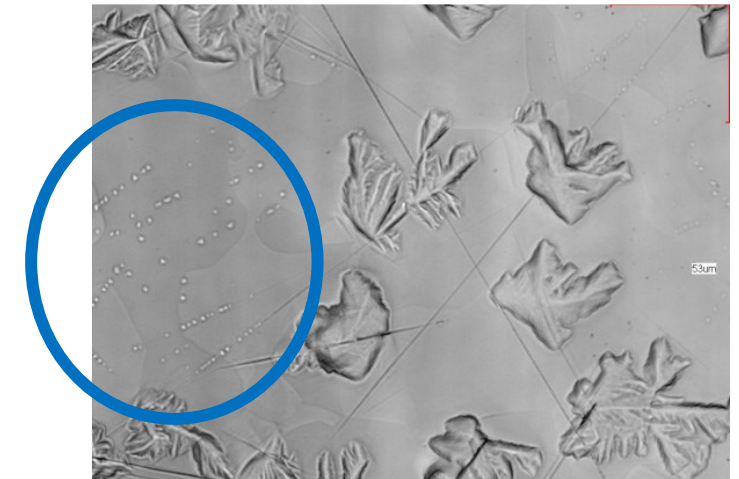
Second (smaller) phase of hydride forms



T = 6 K



T = 100 K



T = 140 K



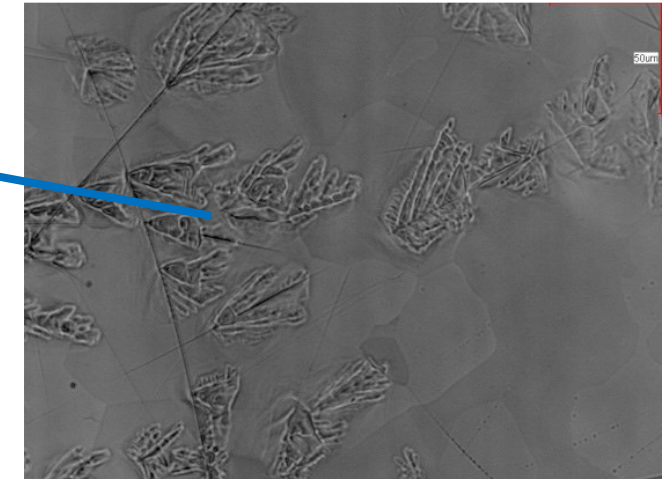
# Microscopic Mechanism of Q-disease



cryostage in the laser confocal scanning microscope

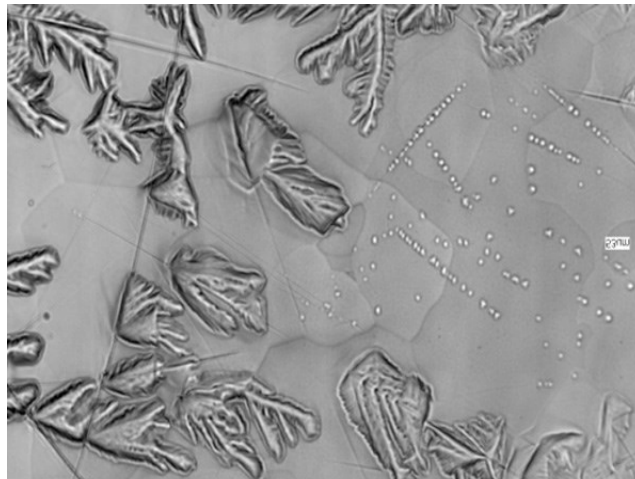
## WARM-UP

Hydrides gone, dislocation skeleton (deformation) remains on the surface



T = 260 K

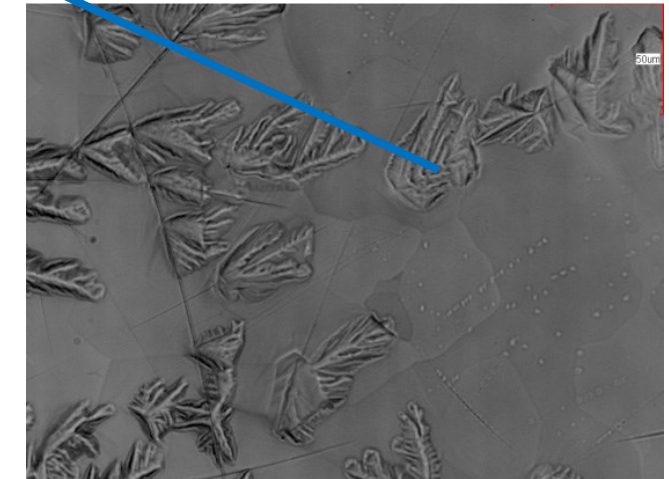
Large phase starts to dissolve



T = 6 K

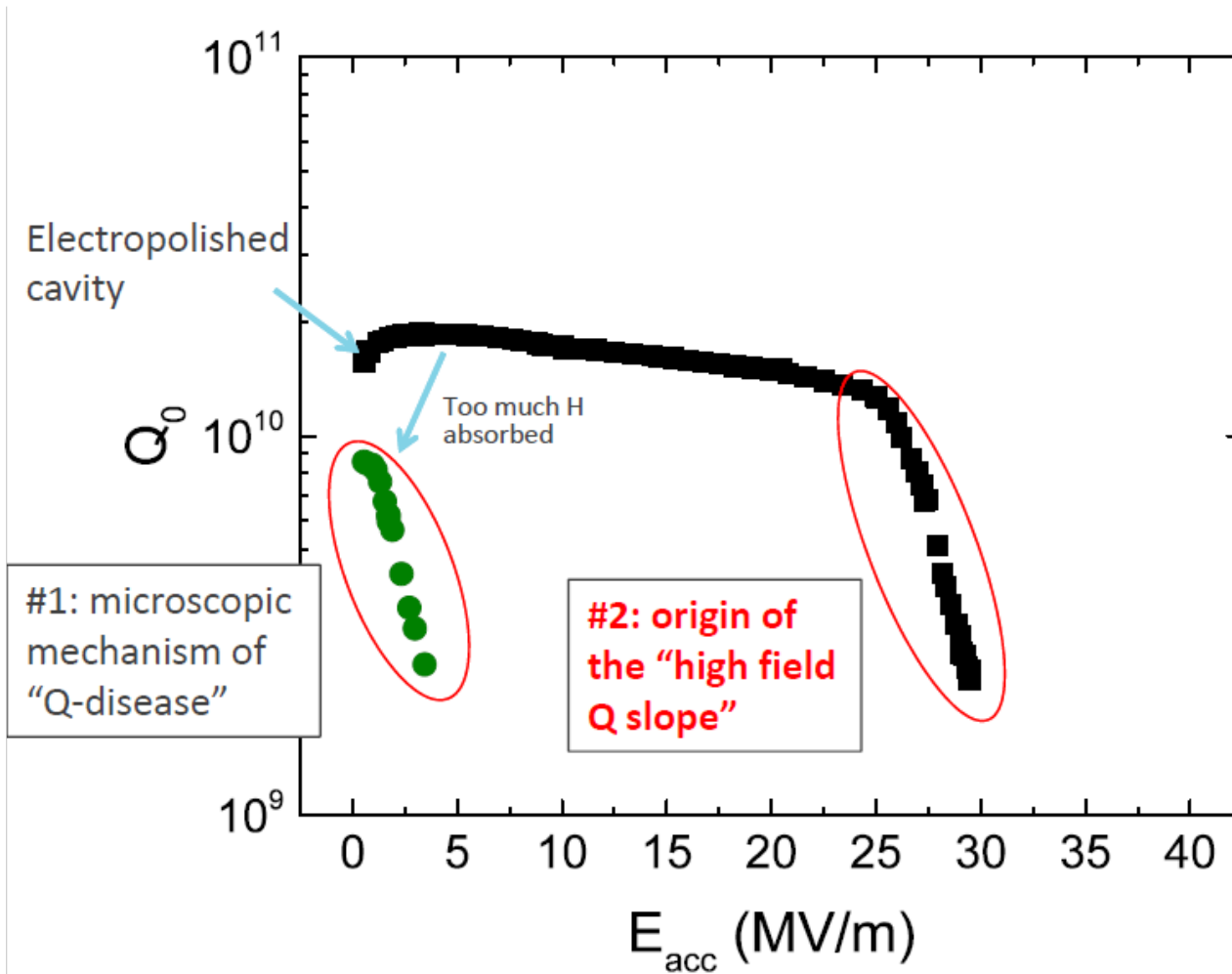


T = 180 K



T = 210 K

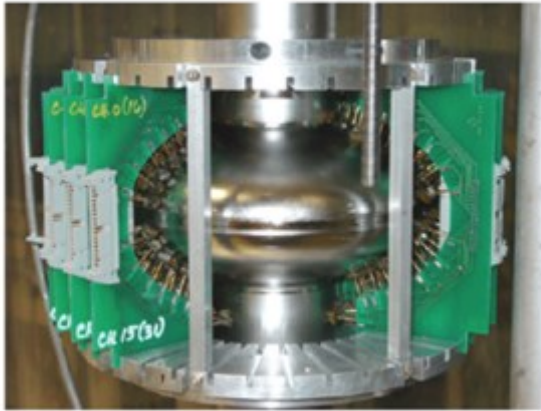
# Possible explanation of the high field Q slope



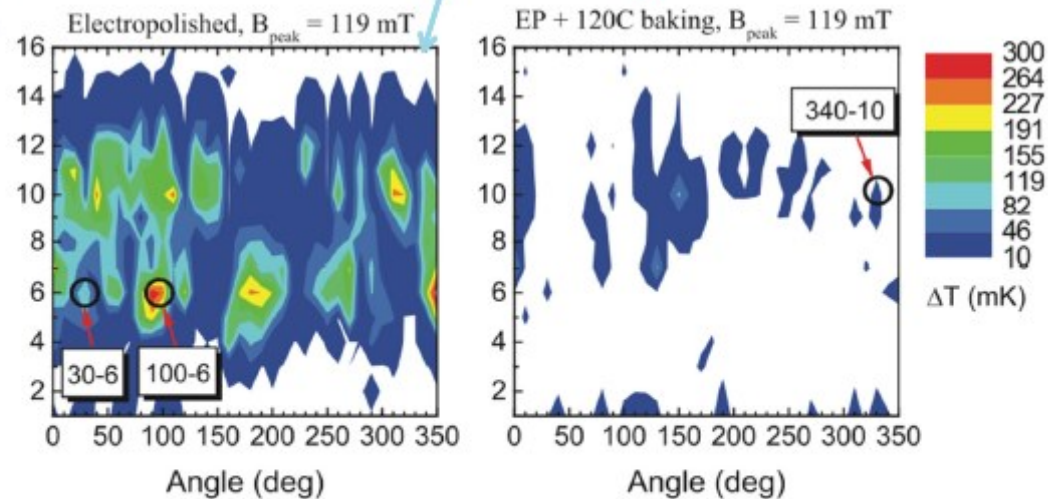
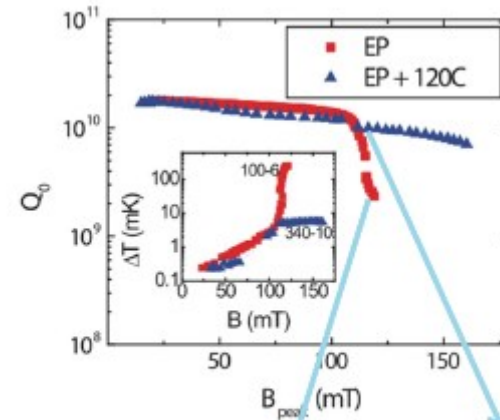
Could be the same mechanism?

But nanohydrides instead of micron-sized one?

# First: find the piece of cavity to characterize



Array of 576 thermometers attached to the outside cavity walls allows mapping wall dissipation



# Cut the cavity

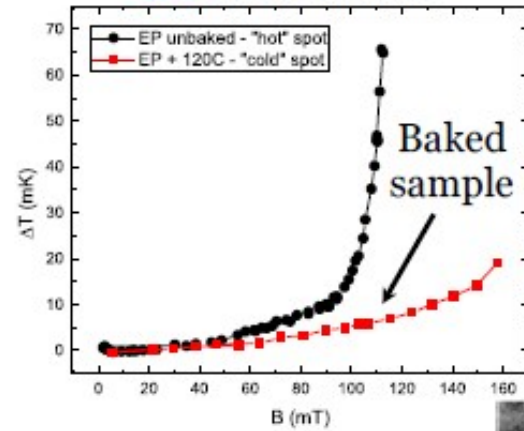
Extract samples from cavity walls locations identified by temperature mapping -> direct correlation of RF losses with material structure



# TEM characterization

Cold: 120C in situ bake for 48hours

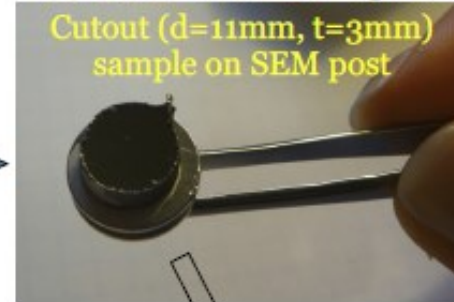
Hot: no such bake



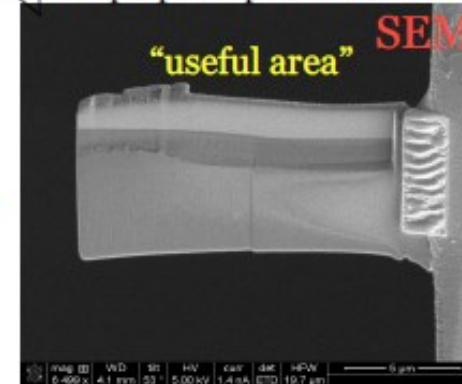
Identify samples based on T-map

Heating: comparison of "cold" and "hot" spots

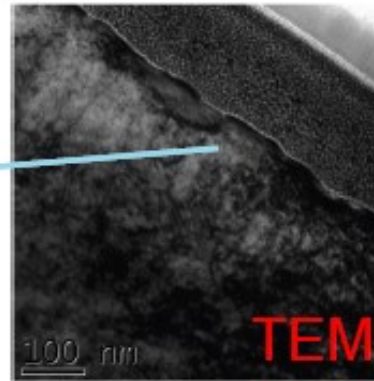
Cutout (d=11mm, t=3mm) sample on SEM post



FIB prep sample for TEM

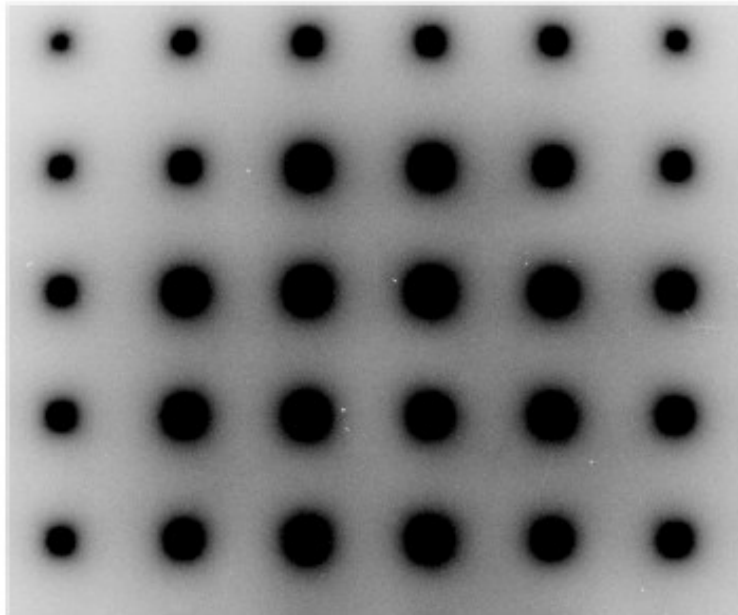


Look at this area with subnanometer resolution in TEM at room AND  $T < 100K$  temperatures

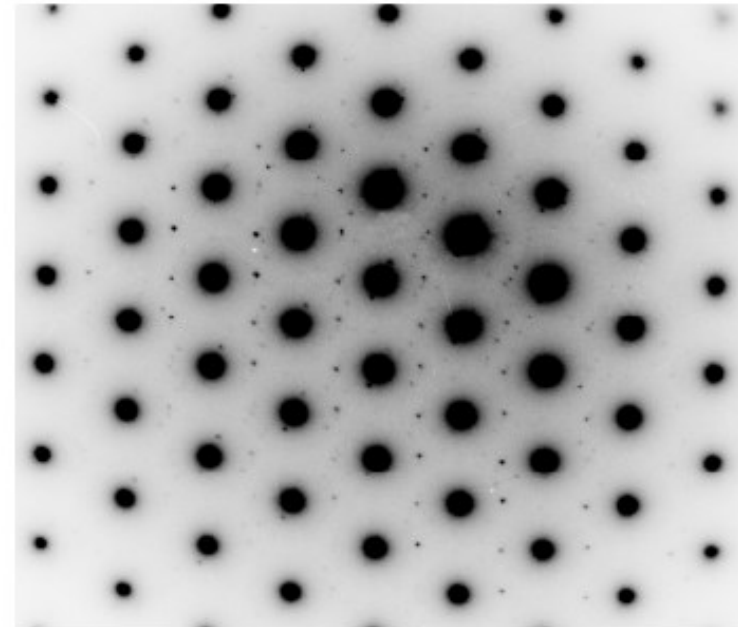


# Confirmation of the existence of nanohydrides

TEM diffraction on cavity cutouts confirms the existence of nanohydrides



Room T: BCC Nb patterns, NO additional phases



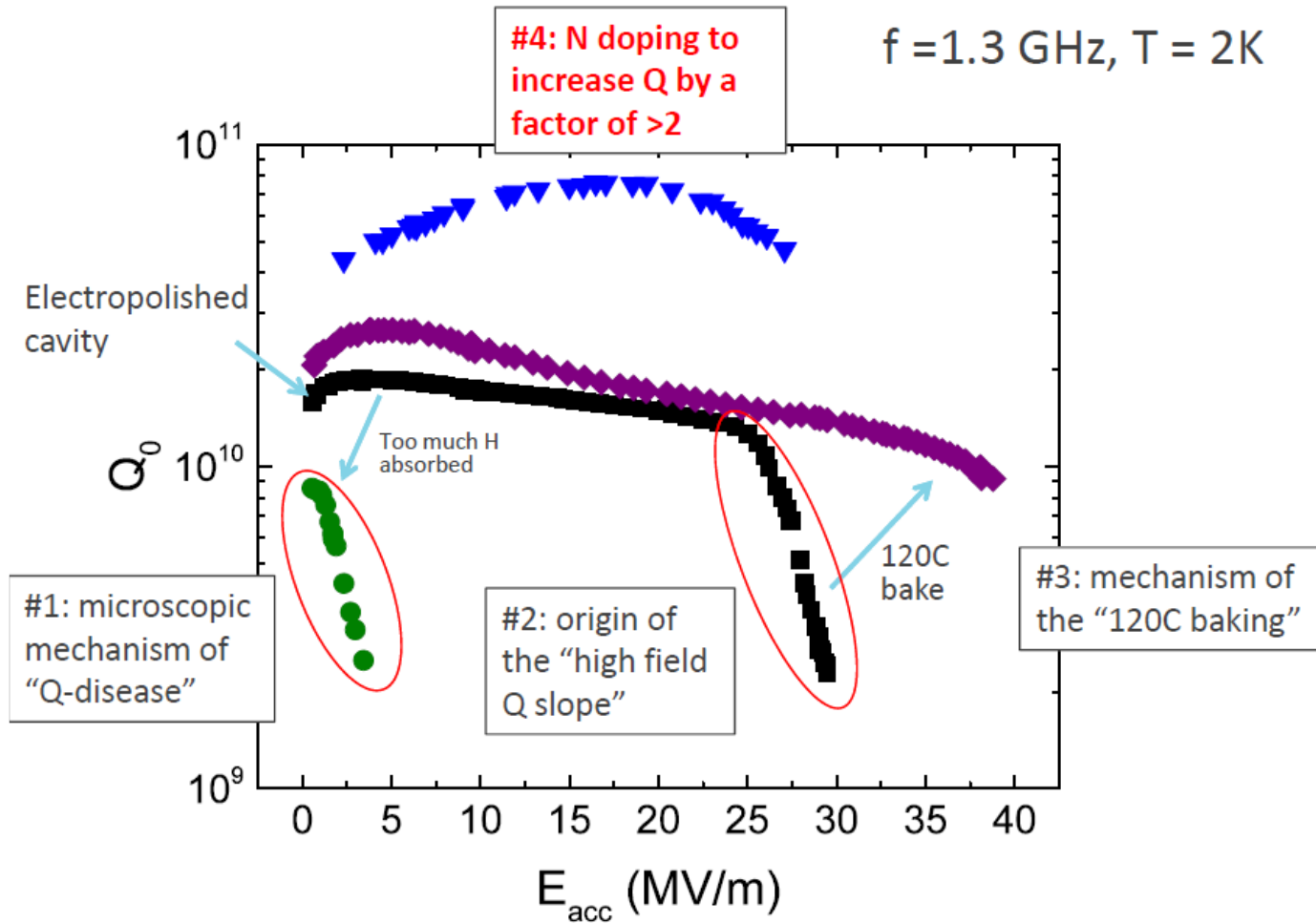
94K: stoichiometric Nb hydride phases!

Y. Trenikhina, A. Romanenko, J. Zasadzinski, Proceedings of SRF'2013, TUP04

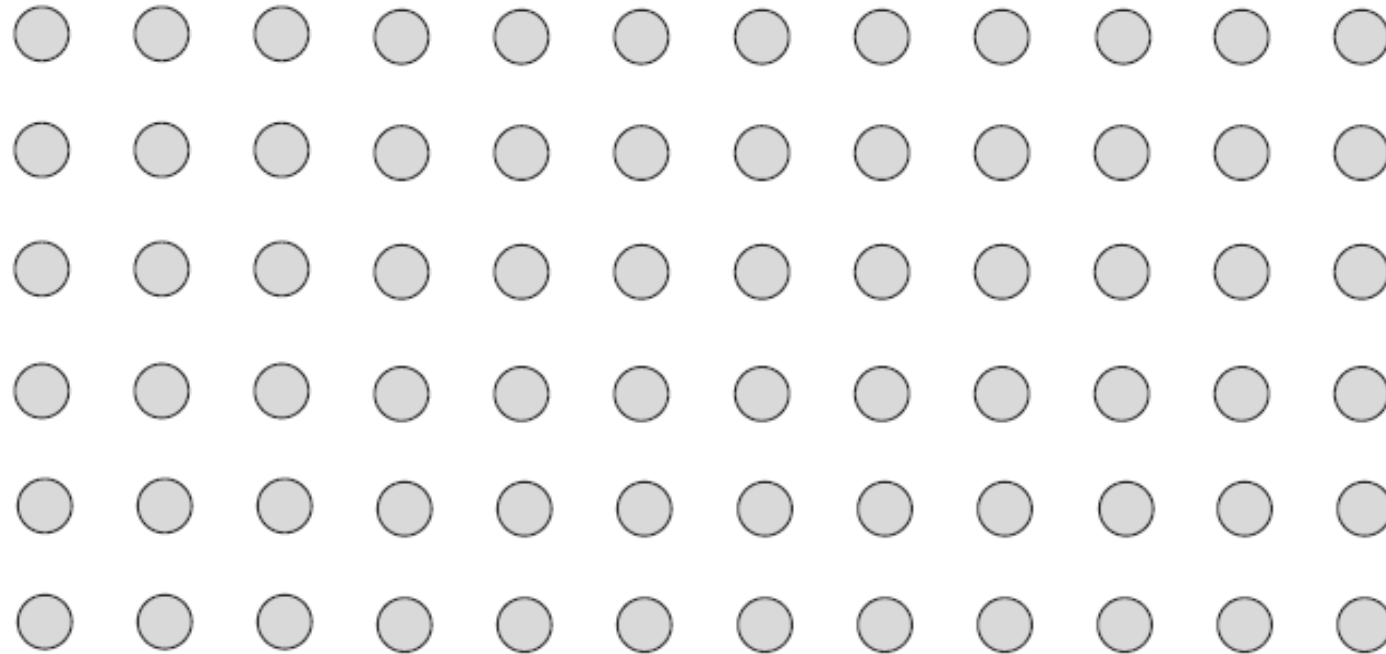
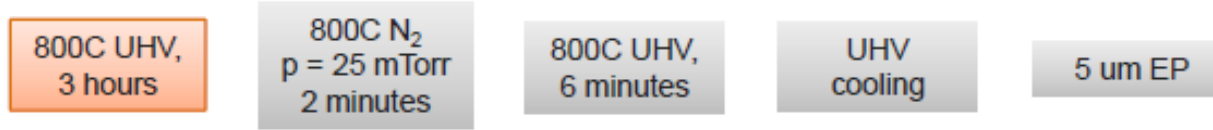


Grassellino, SRF2019 Tutorials

# N - doping, a breakthrough for Q



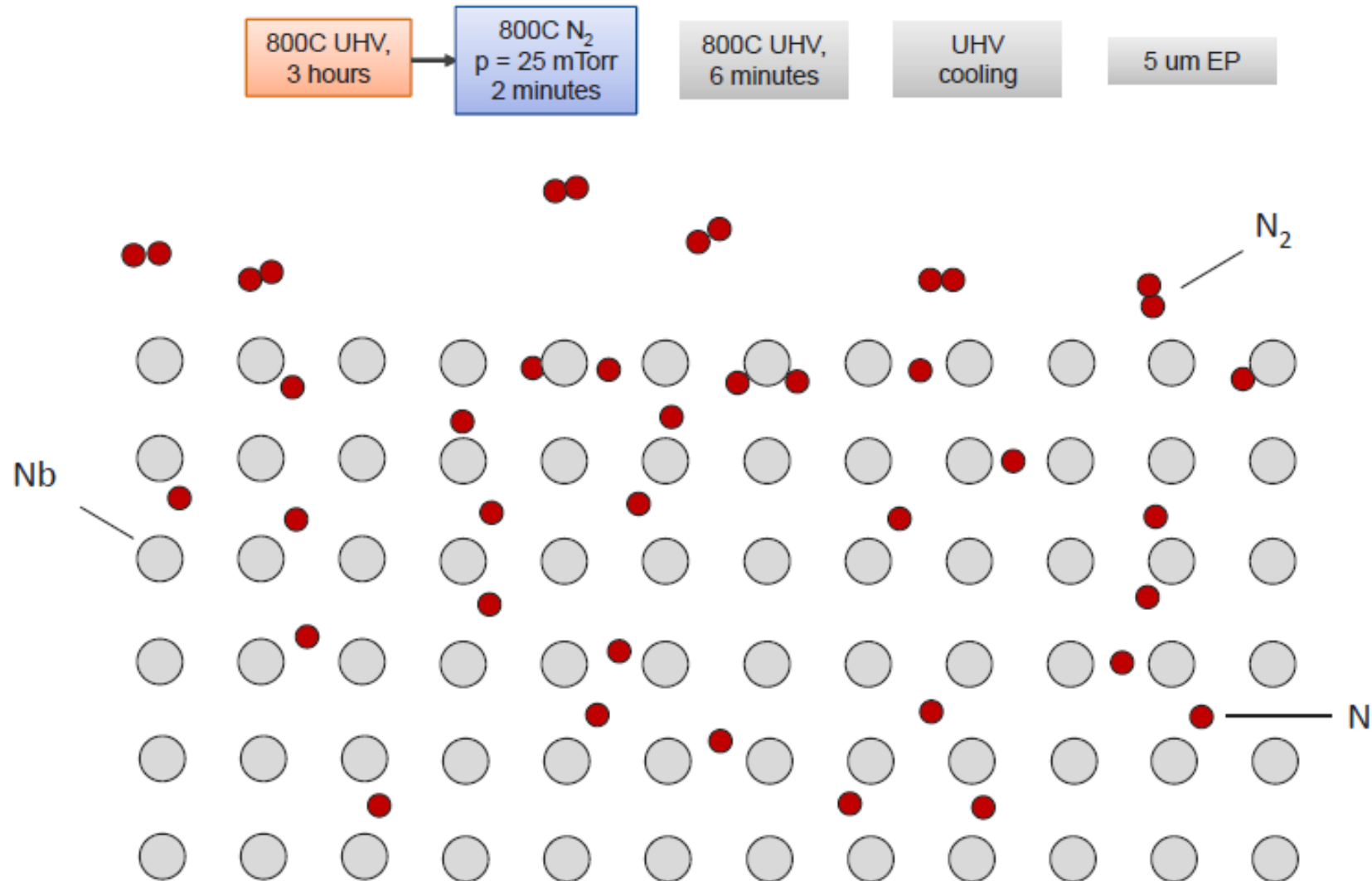
# N-doping how it works?



Martinello, Fermilab

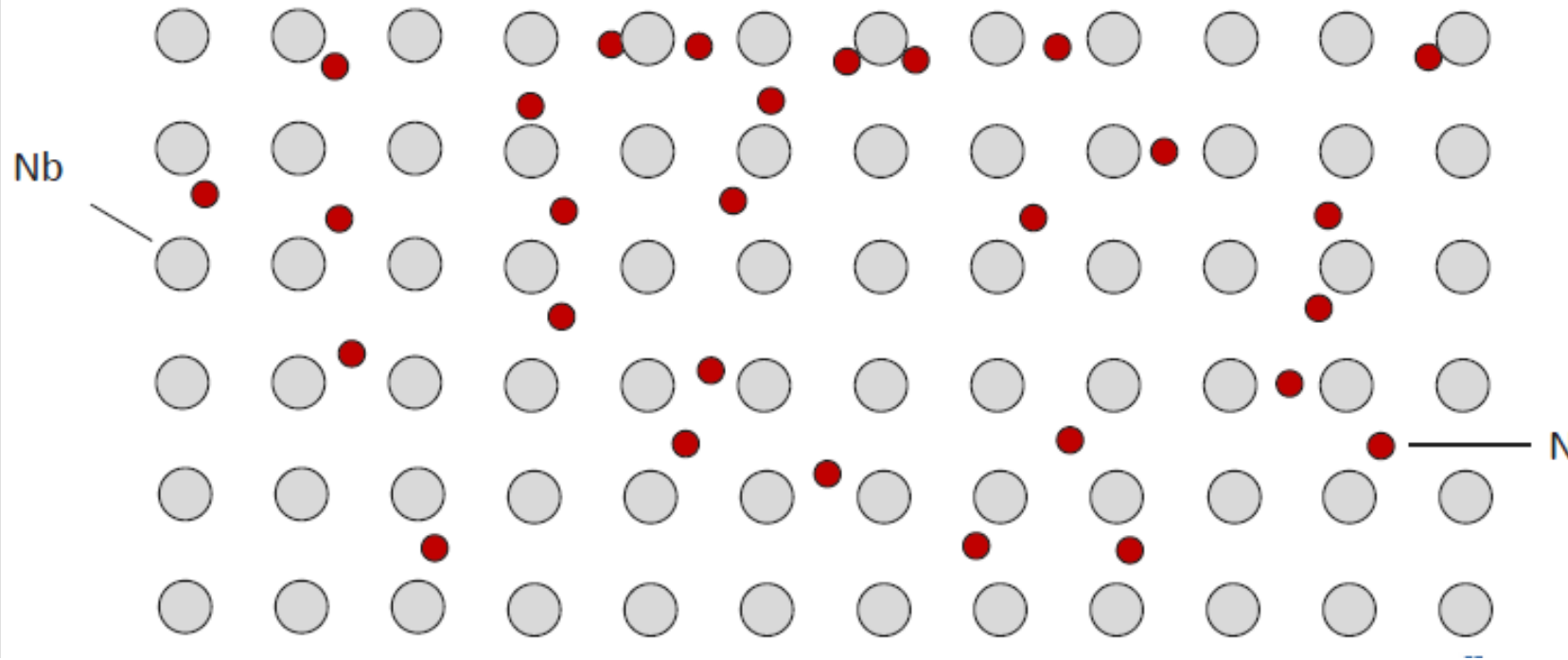
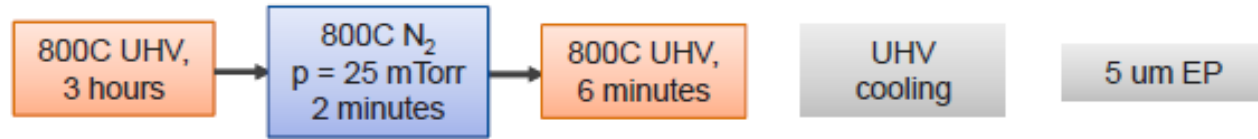


# N-doping how it works?



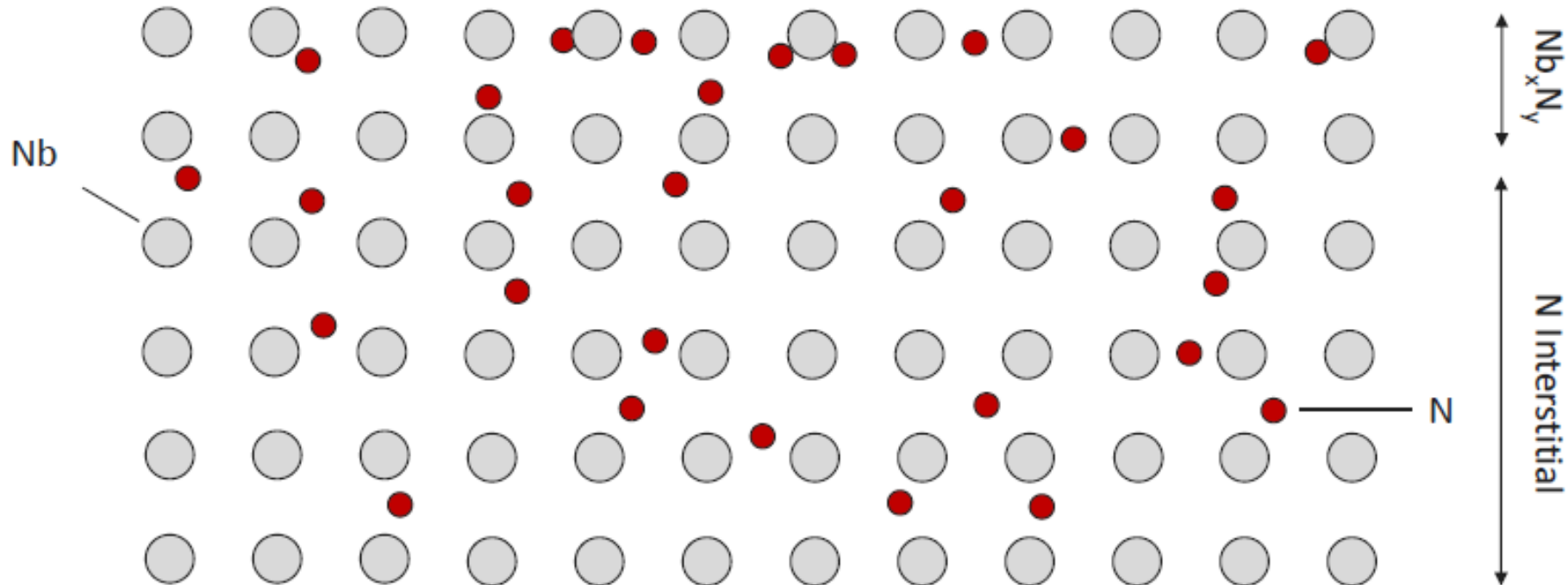
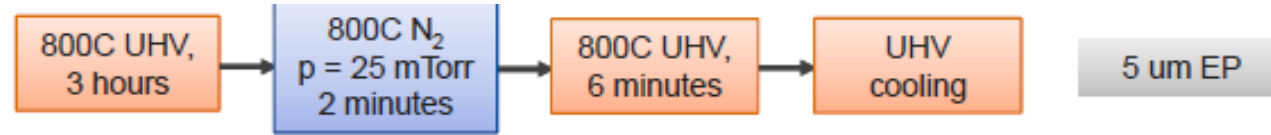
Martinello, Fermilab

# N-doping how it works?



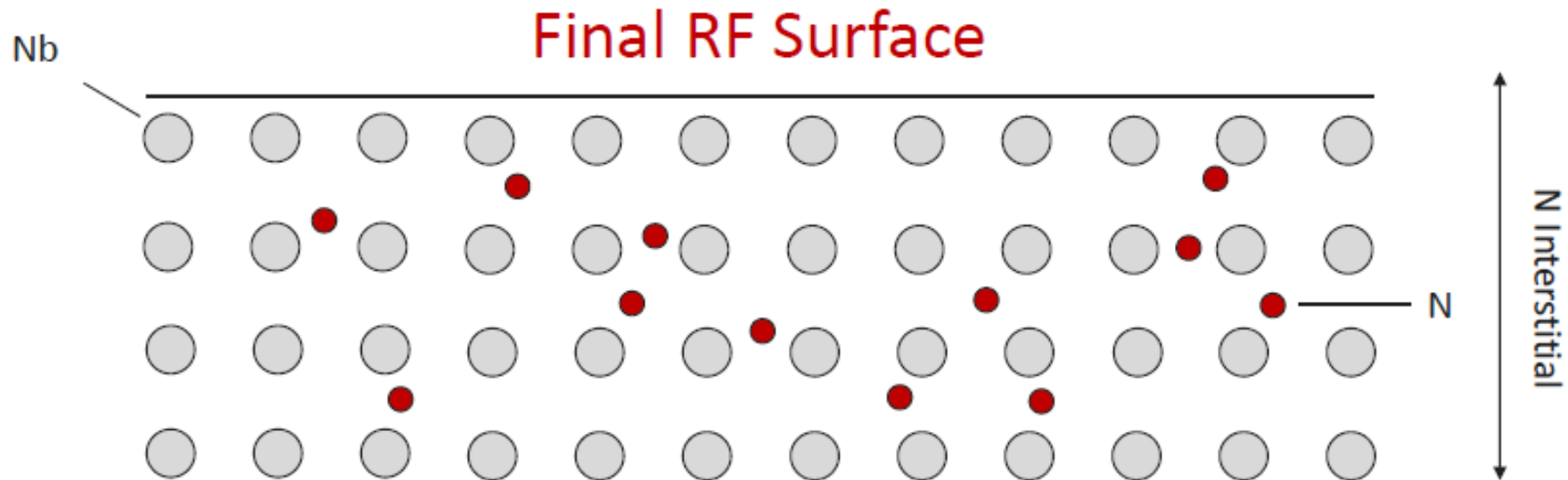
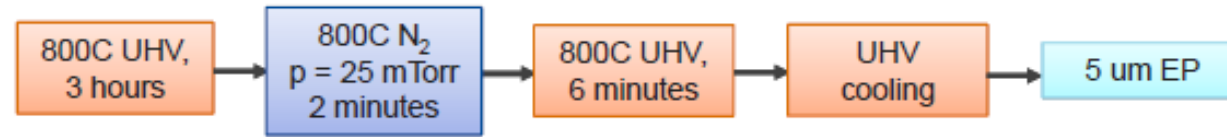
Martinello, Fermilab

# N-doping how it works?



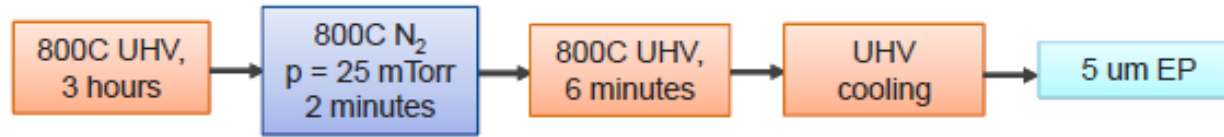
Martinello, Fermilab

# N-doping how it works?

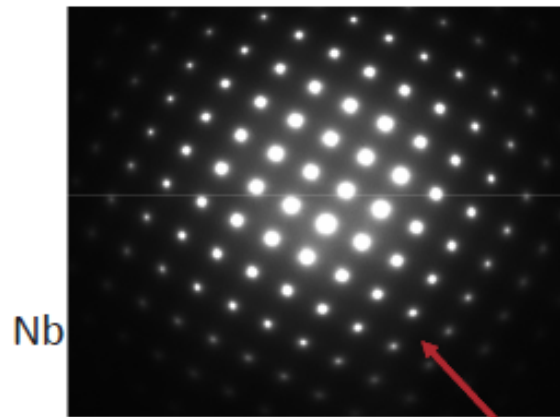


Martinello, Fermilab

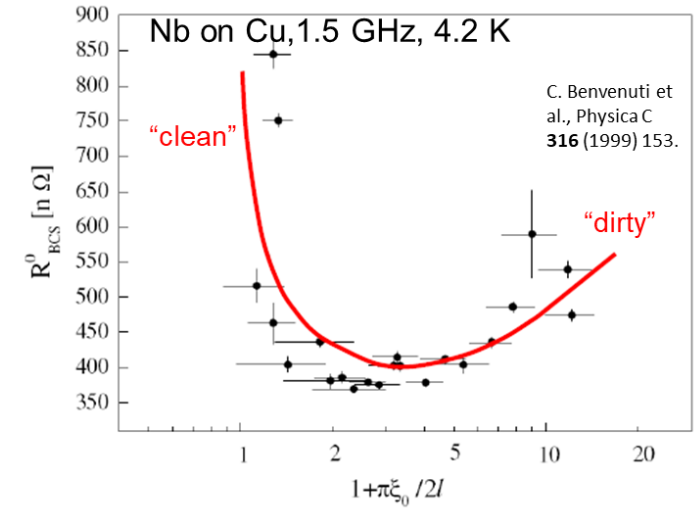
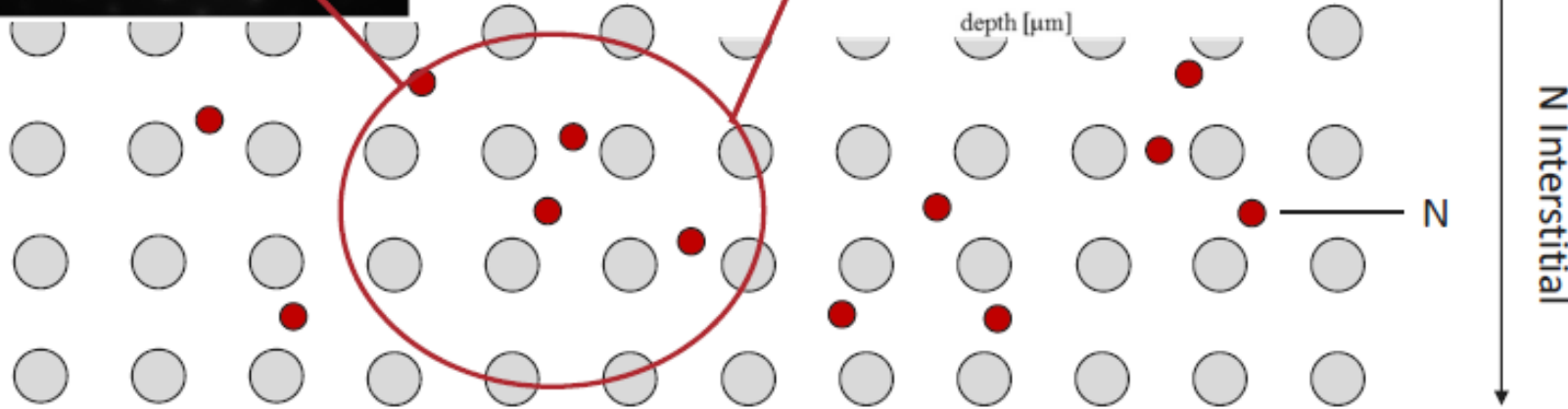
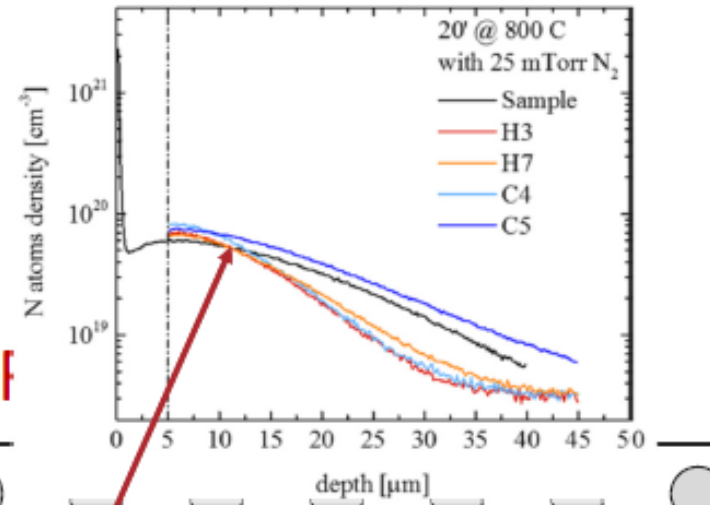
# N-doping how it works?



Y. Trenikhina et al, Proc. of SRF 2015



Final RF

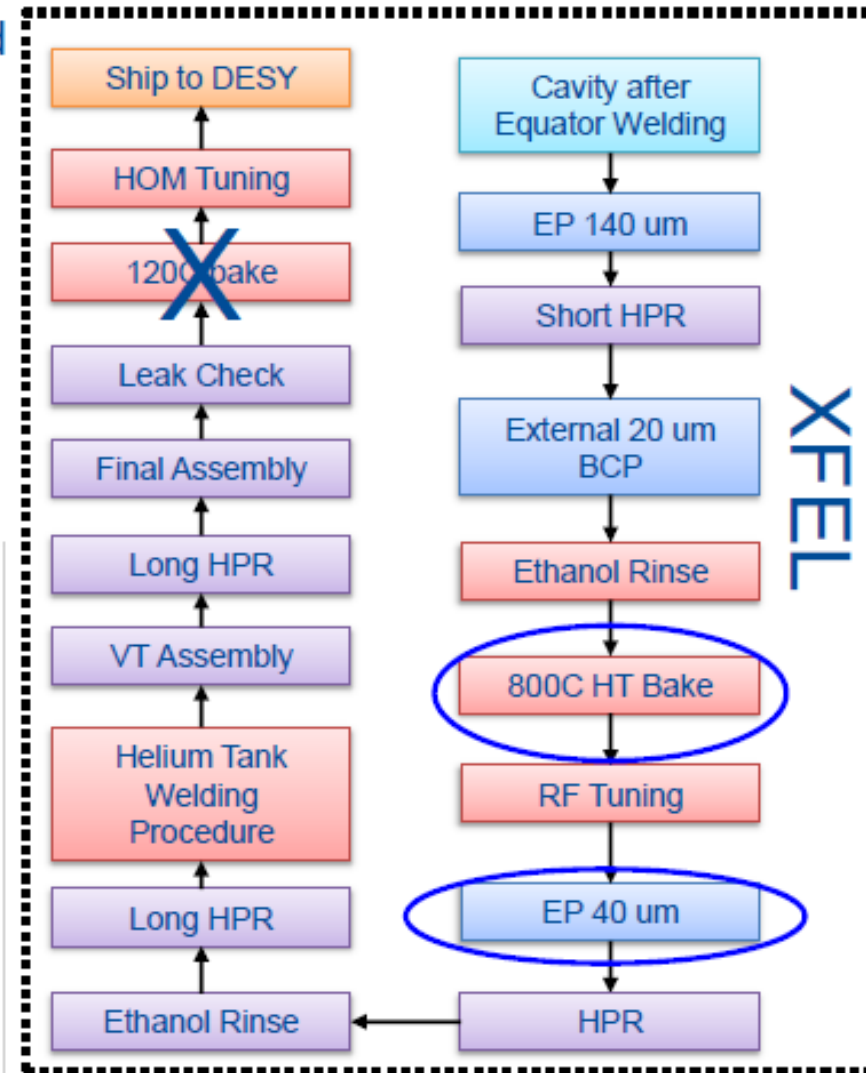
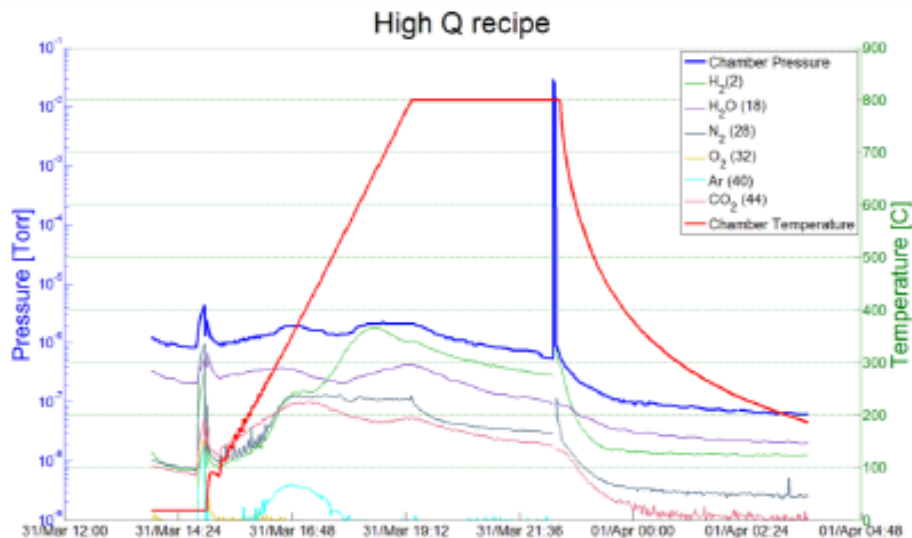


$R_s$  has a minimum for  $\ell = \pi \xi_0 / 4$

# New protocol immediately implement in a real accelerator

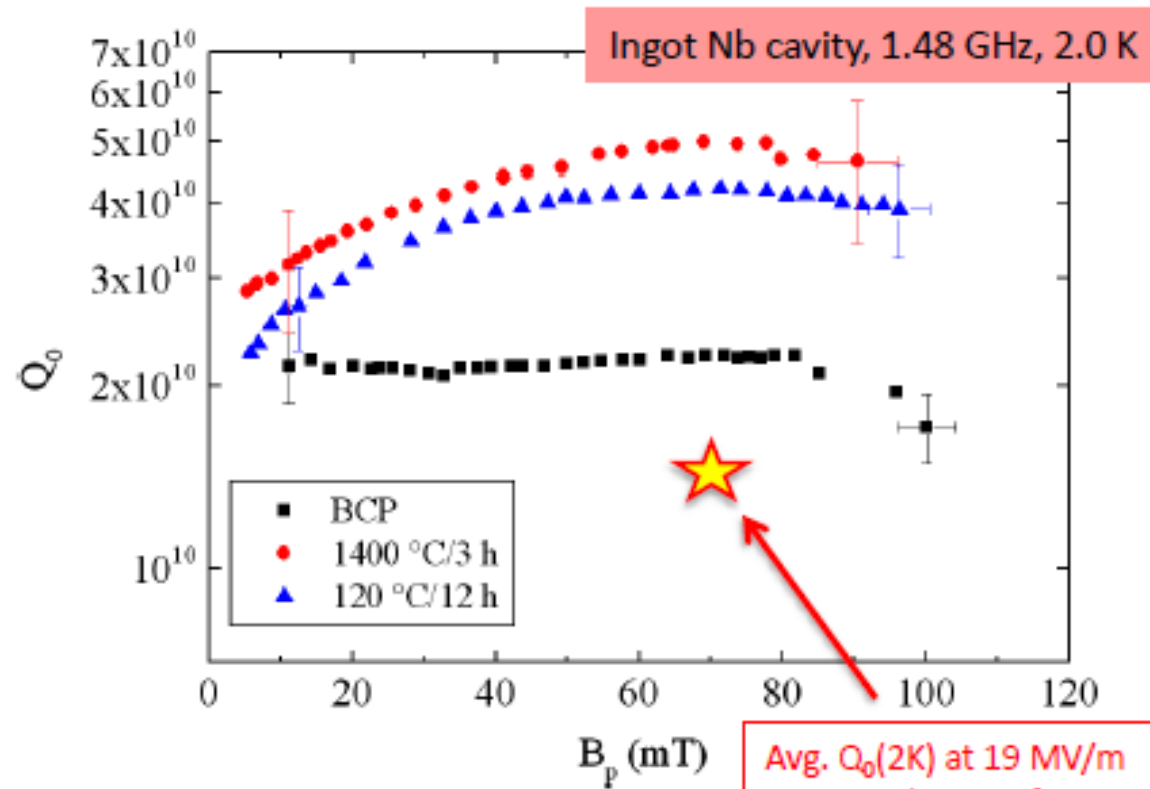
Example from a doping process developed for LCLS-2:

- Bulk EP
- 800 C anneal for 3 hours in vacuum
- 2 minutes @ 800C nitrogen diffusion
- 800 C for 6 minutes in vacuum
- Vacuum cooling
- 5 microns EP



Martinello, Fermilab

# Doping with Ti works as well

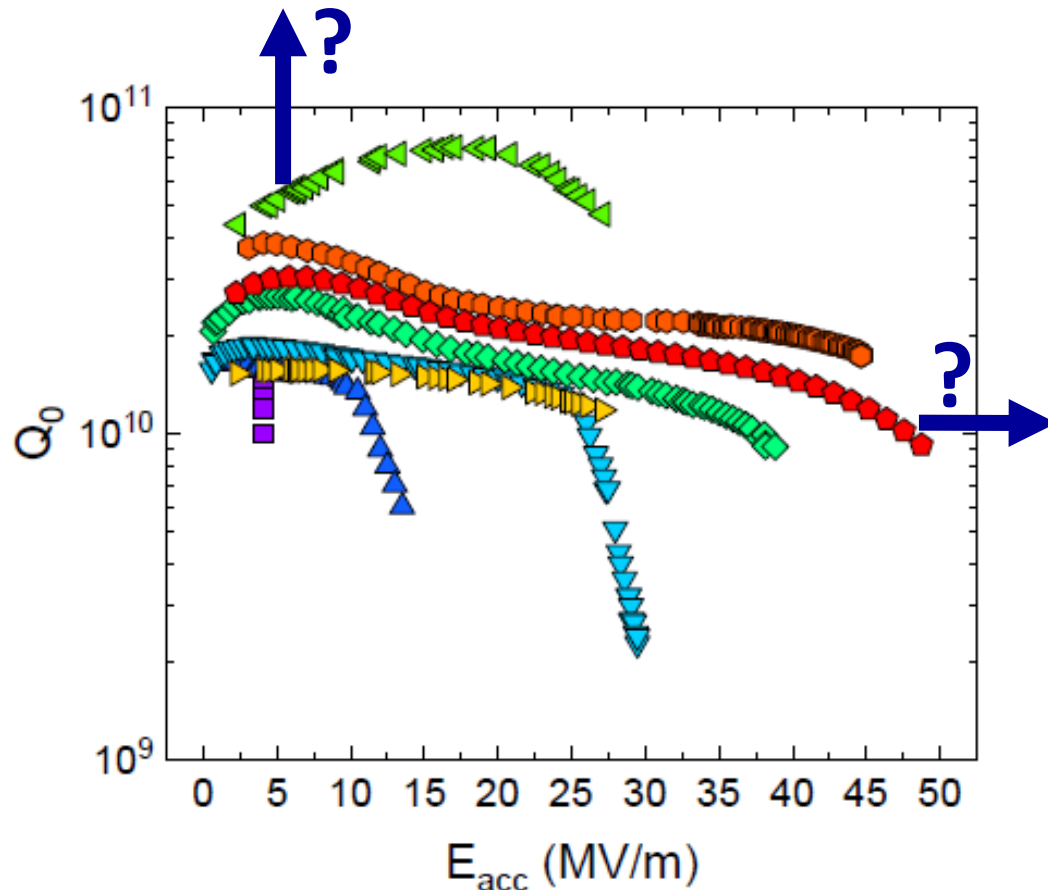


P. Dhakal, Rev. Sci. Inst. 83, 065105 (2012)  
P. Dhakal et al., Phys. Rev. ST Accel. Beams 16, 042001 (2013)  
P. Dhakal et al., IPAC'14, p. 2651

Ciovati

# Why looking beyond bulk Nb?

Nb has the highest critical temperature  $T_c$  ( $=9.25\text{k}$ ) and the highest lower critical magnetic field  $H_{c1}$  ( $\approx 180\text{ mT}$ ) of any elemental SC



Breakdown fields close to the de-pairing limit of 50 MV/m for Nb have been achieved

**Best Nb cavities approaching their intrinsic limit at  $H_{max} = H_C$**

**For further improvement, innovation needed**

**Possibilities to use higher performance superconductors other than bulk Nb?**

*A-M Valente, SRF2017 Tutorials*



# Recommended Literature

- R. Padamsee, J. Knobloch and T. Hays – « RF Superconductivity for Accelerators », Wiley-VCH, 2008
- J. P. Turneaure, J. Halbritter, and H. A. Schwettman. « The surface impedance of superconductors and normal conductors: The Mattis-Bardeen theory. » Journal of Superconductivity 4.5 (1991): 341-355
- A. Gurevich « Theory of RF superconductivity for resonant cavities. » Superconductor Science and Technology, 30(3), 034004 (2017).
- SRF Tutorials (<https://jacow.org/Main/Proceedings?sel=SRF> and websites of the SRF conferences)

*This slides are mainly based on: D. Sertore, SRF Cavity Fabrication, EASISchool 3 Genoa, 2020*