

Thin film deposition on Cu for SRF cavity production via DC magnetron sputtering at INFN-LNL

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INFN Legnaro National Laboratories

Outline

**DC
Magnetron
Sputtering**

1

**Nb₃Sn for
accelerating
cavities**

2

**NbTi for
haloscopes**

3

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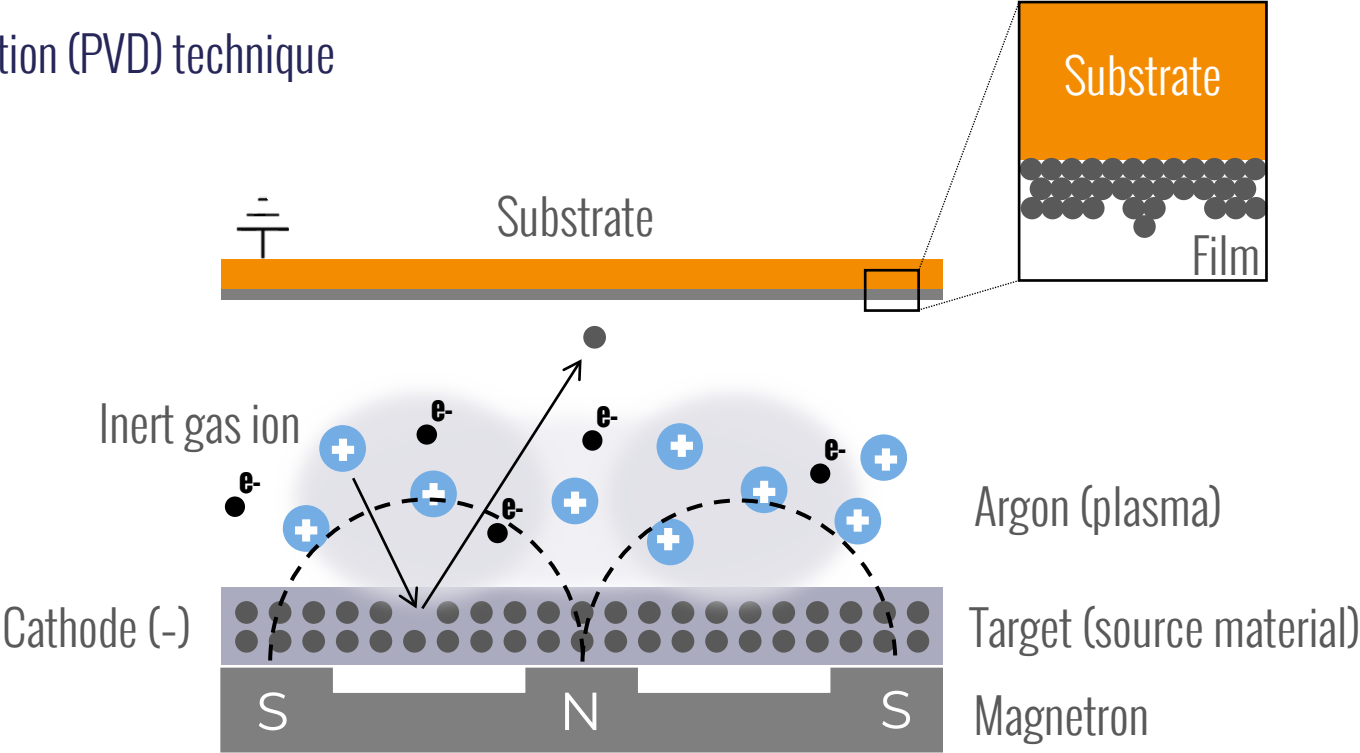
**NbTi for
haloscopes**

3

DC Magnetron Sputtering (DCMS)

Physical Vapour Deposition (PVD) technique

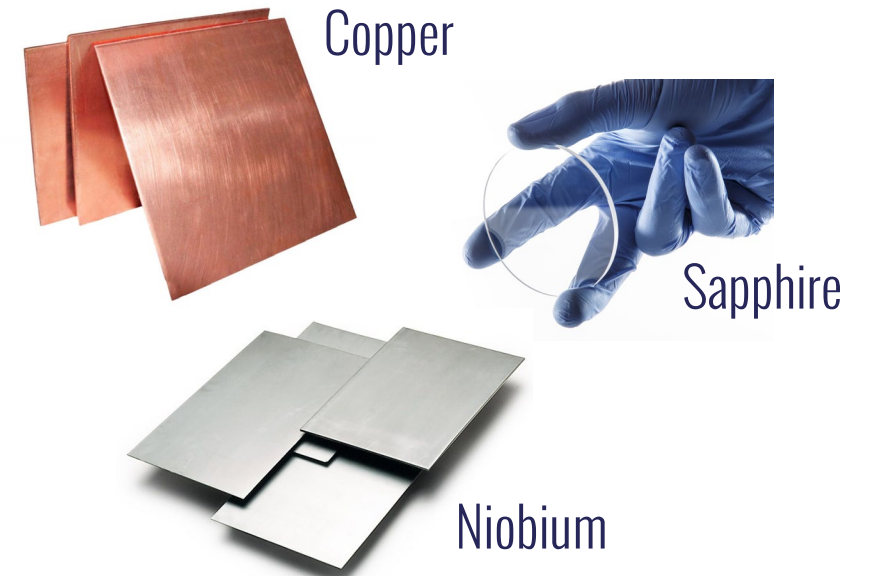
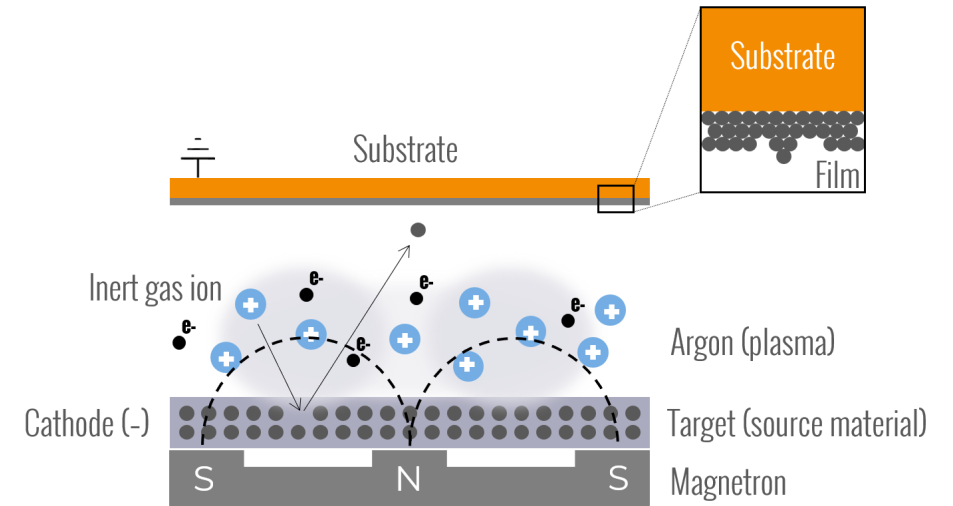
- Substrate temperature
- Gas pressure
- Cathode power



DC Magnetron Sputtering

ADVANTAGES

- Different target materials
- Different target/magnetron geometries
- Different substrate materials



SC thin films for SRF cavities



Accelerating cavities



Nb₃Sn



Copper



Haloscopes

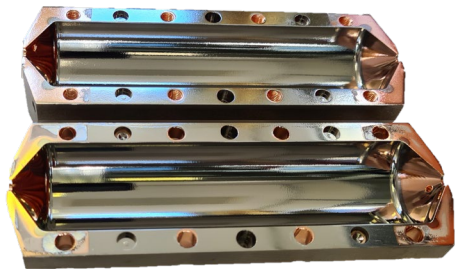
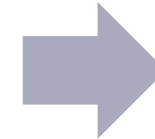


NbTi

Material choice

Advantages

Challenges



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Nb₃Sn for accelerating cavities: why?

- Nb close to its ideal performance

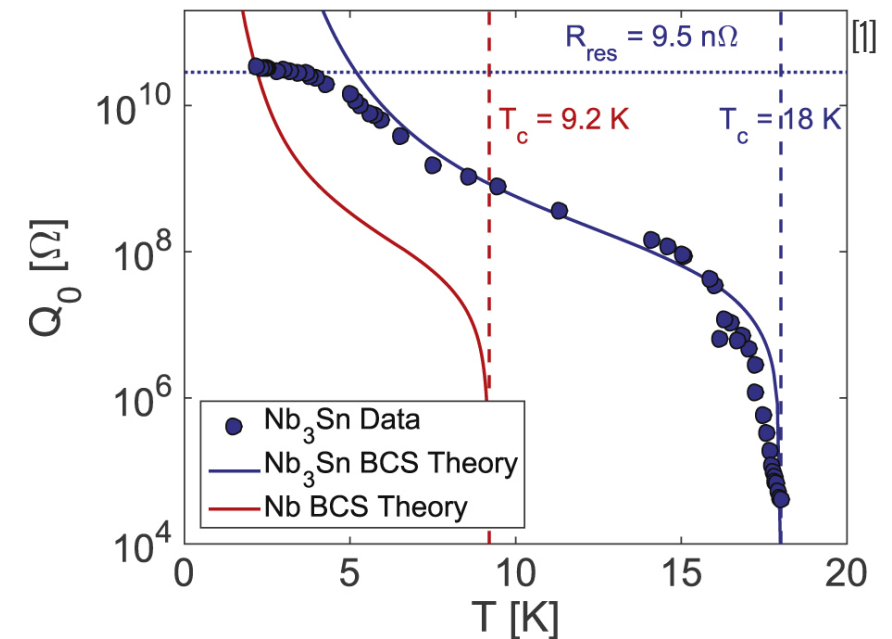
With Nb₃Sn:

- Lower R_{BCS} for a given temperature
- Operation at 4.5 K (instead of 2 K)
- Improve cryo plant efficiency by factor 3-4
- Potential demonstrated by vapour diffusion cavities:

$Q_0 \sim 2 \times 10^{10}$ at 20 MV/m (4.4 K, 1.3 GHz) [2]

*predicted

Material	T_c	H_{sh}
Niobium	9.2 K	240 mT
Nb ₃ Sn	18.3 K	420 mT*

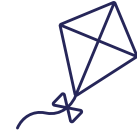


[1] S Posen and D L Hall 2017 *Supercond. Sci. Technol.* 30 033004

[2] S Posen et al 2021 *Supercond. Sci. Technol.* 34 025007

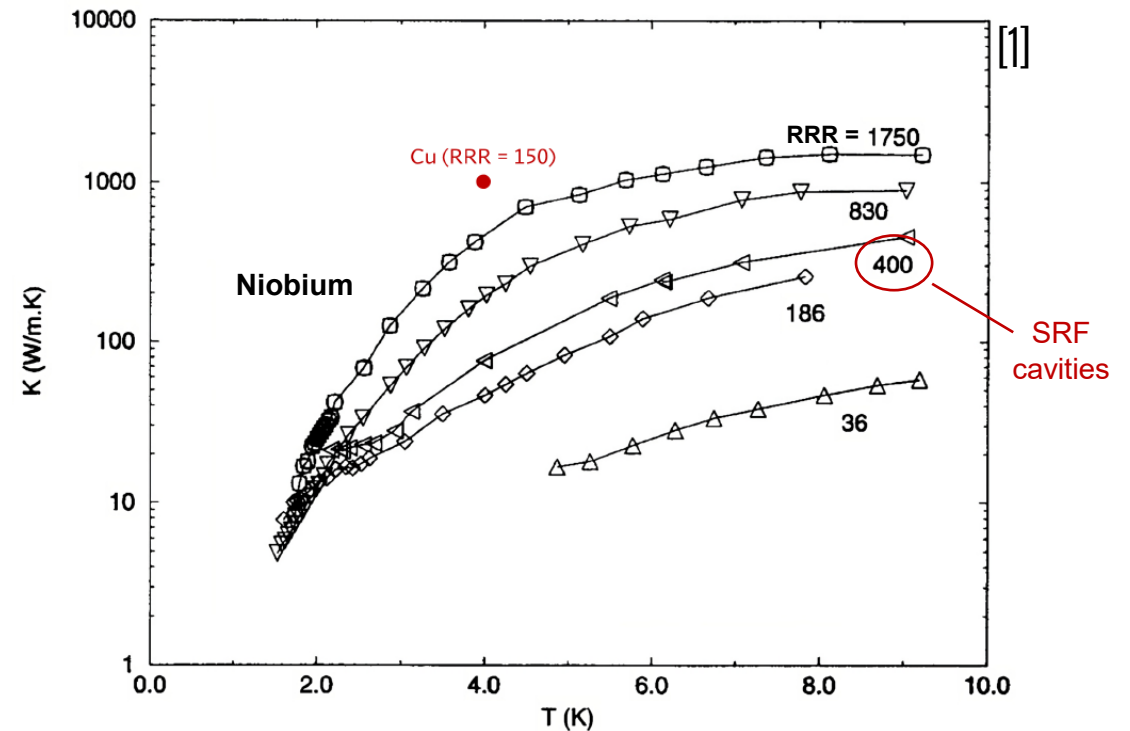
Nb₃Sn on copper: why?

DCMS Nb/Cu successful technology (LEP-II, LHC, ALPI, ISOLDE, FCC-ee)



Compared to bulk Nb:

- significant **thermal stability**
- much **cheaper** than bulk Nb
→ large scale production
- cryocooler based **conduction cooling**



[1] DOI: 10.5170/CERN-1996-003.191

Nb₃Sn on copper via DCMS: goals

Long term: scalable process to coat Nb₃Sn/Cu SRF accelerating cavities



Short term: produce Nb₃Sn films on Cu (small samples) which exhibit

⊗ bulk like T_c

⊗ correct stoichiometry (Nb-Sn ratio)

⊗ correct A15 phase

⊗ dense and crack-free morphology

Conditions must be satisfied
before further development
toward SRF application

Nb₃Sn on copper: challenges

❧ Complicated phase diagram:

~ 18 - 26 Sn At%

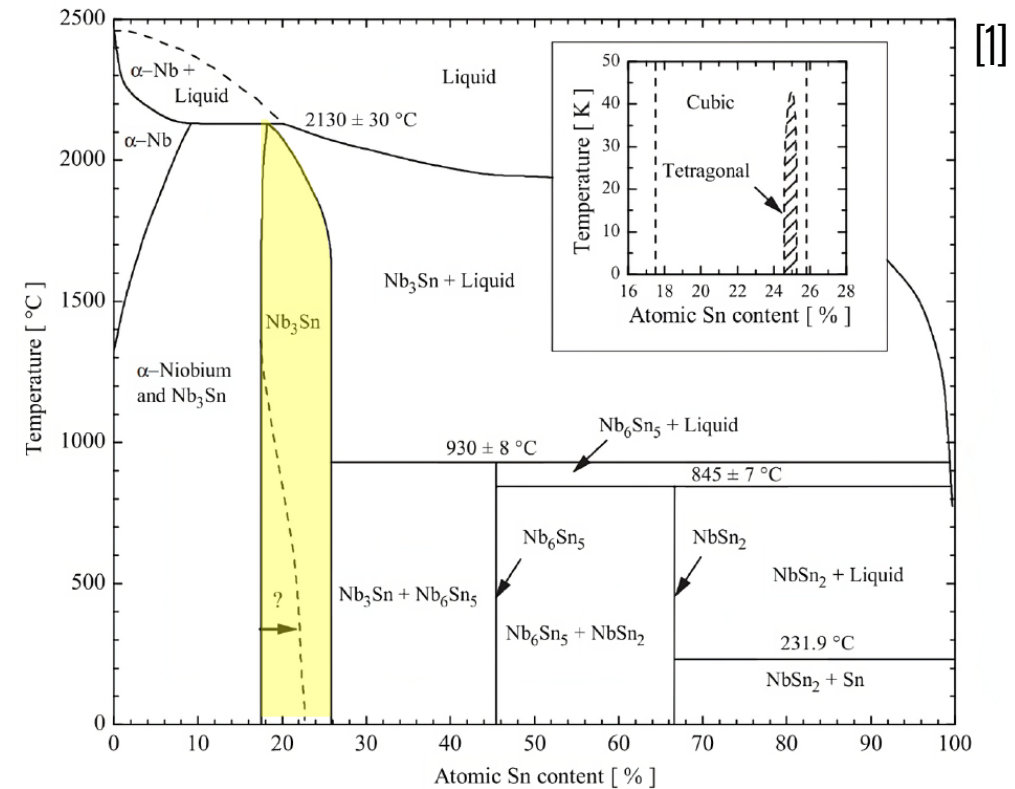
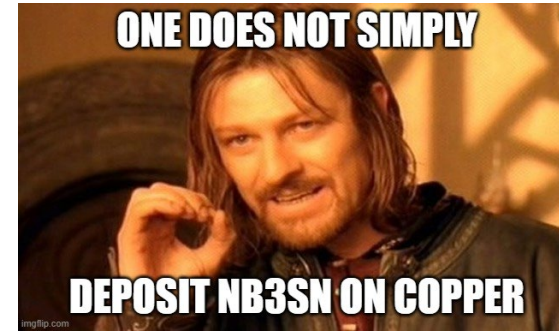
> 930 °C to form only A15 Nb-Sn

❧ Temperature limit = **650 °C** (6 GHz)

→ weakening point of copper (400 °C)

❧ Sn and Cu are miscible

→ intermediate buffer layer needed



[1]

[1] A Godeke 2006 *Supercond. Sci. Technol.* 19 R68

Standard sample production process

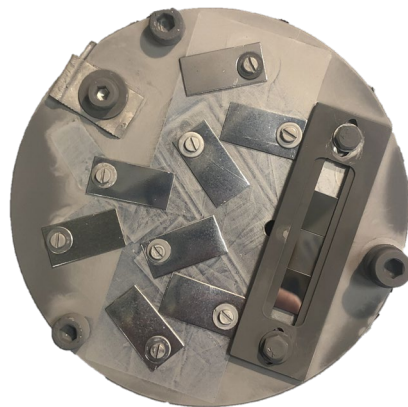
(1) Set parameters:

Cathode power P

Deposition pressure p_{dep}

Deposition temperature T_{dep}

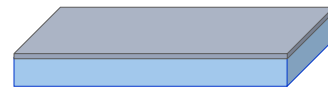
small samples
on holder plate



10 cm



(2) $1\ \mu\text{m}\ \text{Nb}_3\text{Sn}$



Sapphire



Cu + $1\ \mu\text{m}$ Nb buffer layer (BL)



Cu

+ 24 h annealing



(3) Characterisation:

T_c measurement

SEM + EDS (Sn composition)

XRD



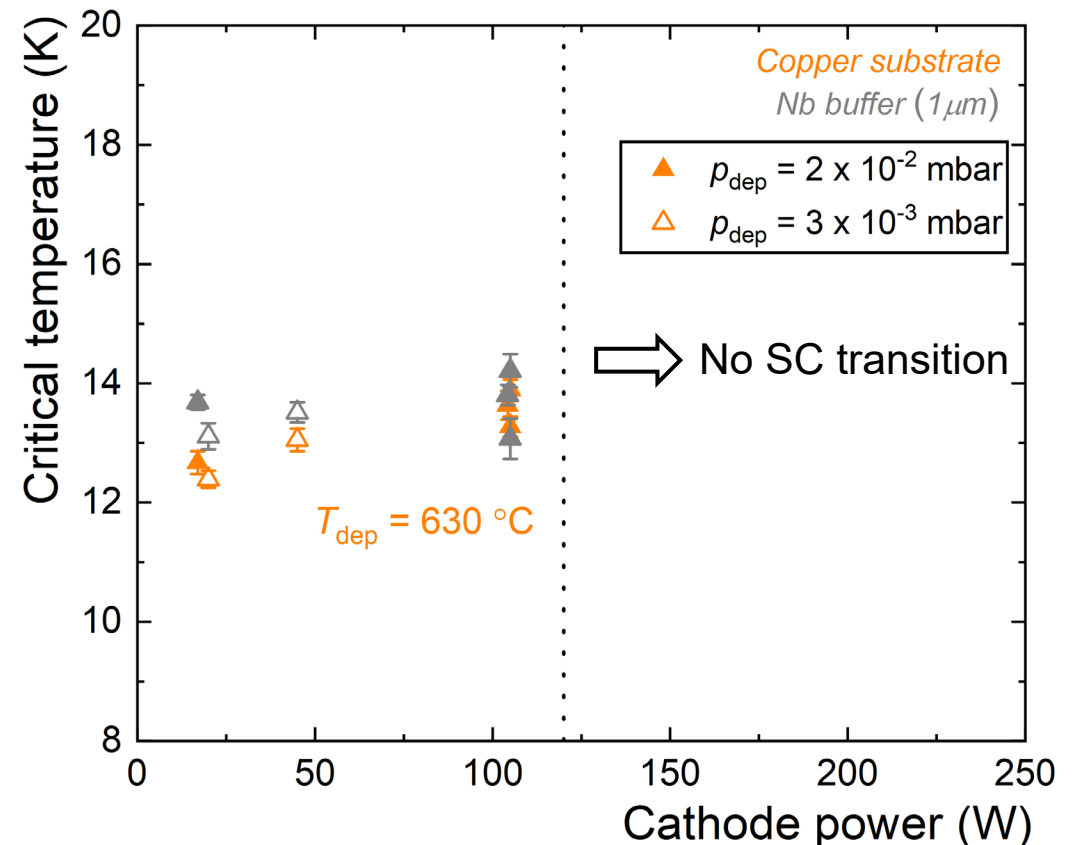
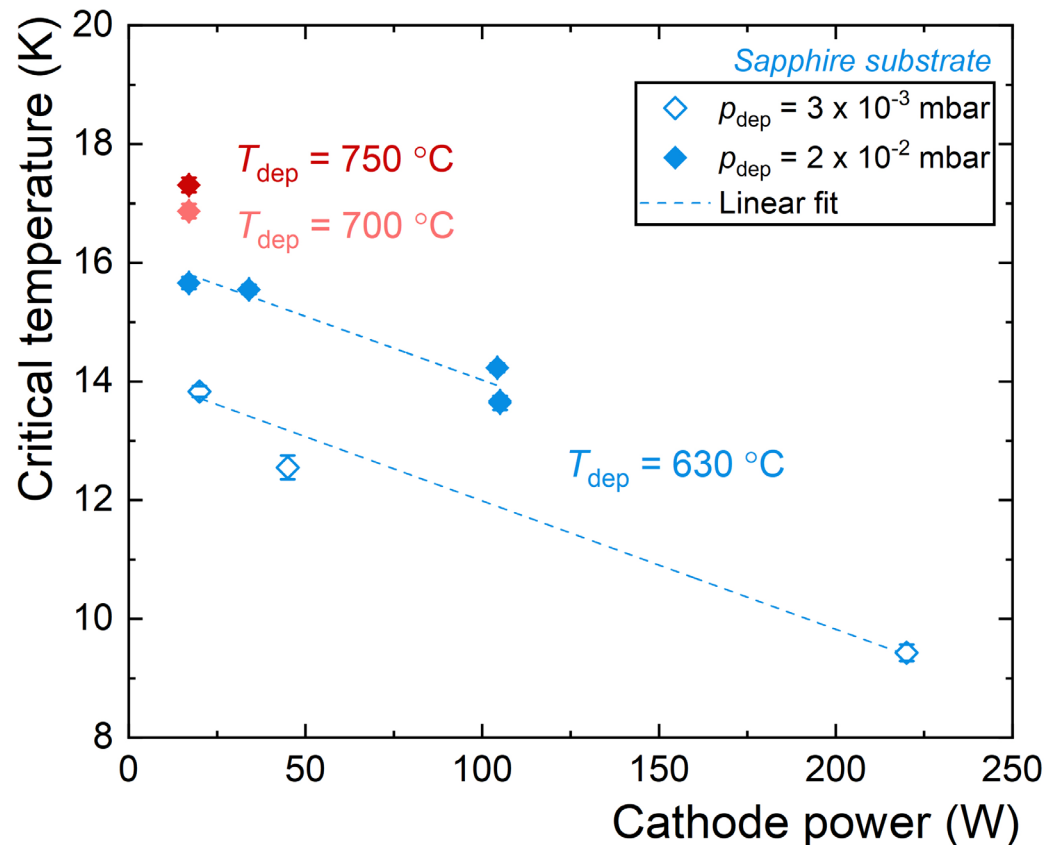
**Tune next parameters
based on trends (step 1)**

Nb₃Sn on copper: first trends



Sapphire as reference

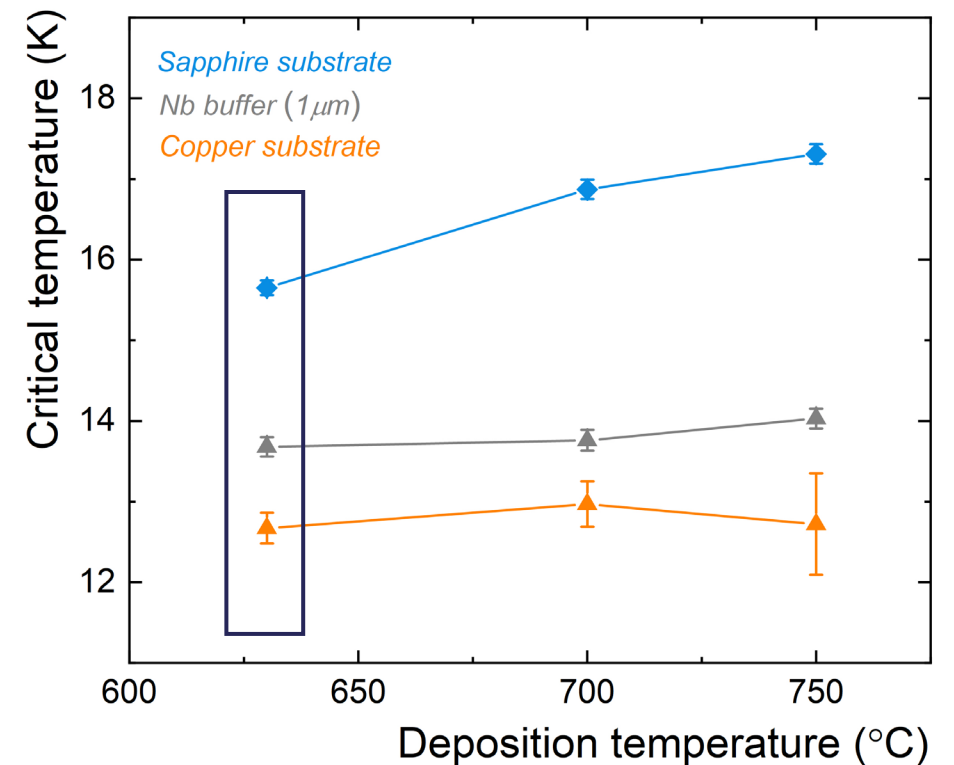
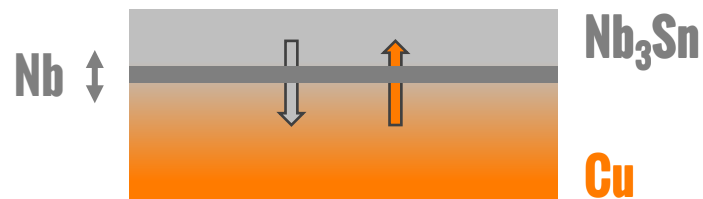
#1 - T_c -oriented study of dependencies from sputtering parameters



Nb₃Sn on copper: where to look

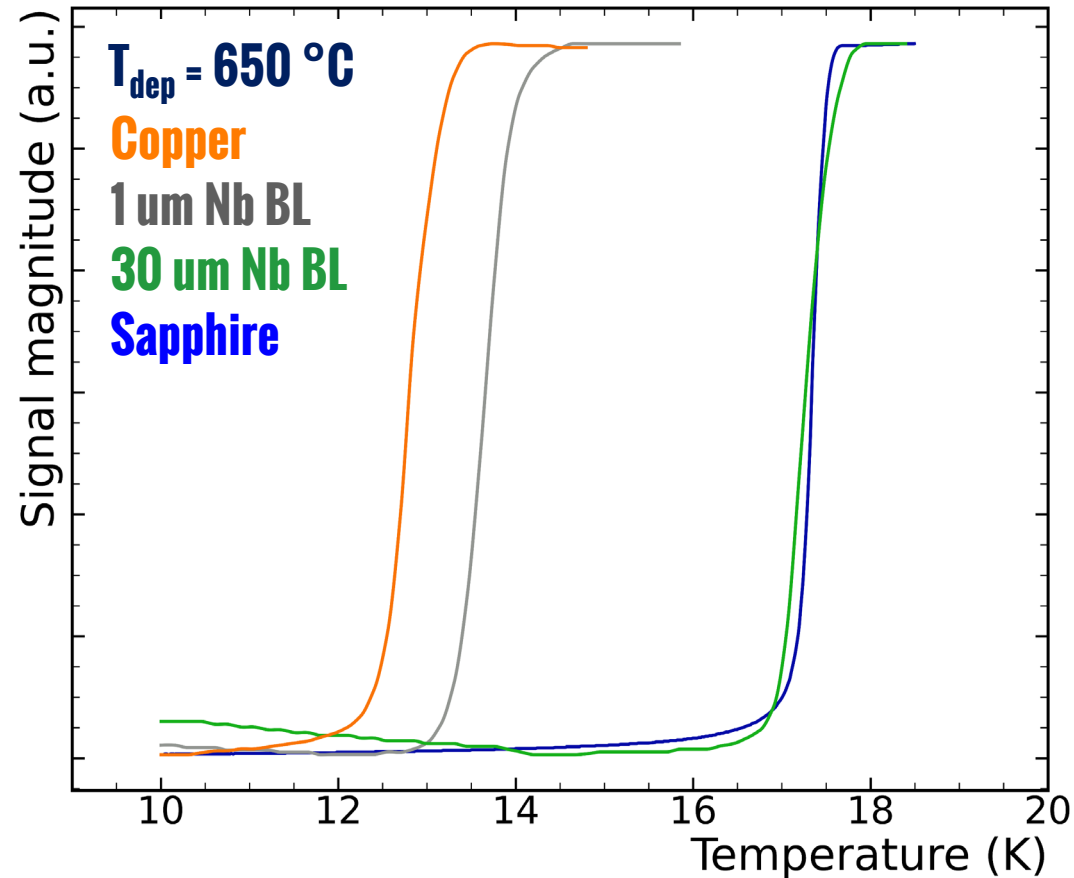
#2 - Fix cathode power and deposition pressure, keep temperature low

- Samples with best T_c on sapphire
- stick to lowest T_{dep} (≤ 630 °C)
- Investigate role of:
 1. annealing time \rightarrow decrease
 2. buffer layer thickness \rightarrow increase



Nb₃Sn on copper: SC transition curves

Recipe: 1 μm Nb₃Sn $T_{\text{dep}} = 650 \text{ }^\circ\text{C}$ $p_{\text{dep}} = 2 \times 10^{-2} \text{ mbar}$ $P_{\text{cathode}} = 15 \text{ W}$ **No annealing**



Buffer layer thickness

Nb₃Sn on copper: where we are

- shorter t_{ann} has no effect on Cu and Cu+Nb 1 μm BL
- shorter t_{ann} has positive effect on sapphire
- thicker Nb BL (1 μm \rightarrow 30 - 50 μm) comparable to sapphire

Role of buffer layer?



Sn-Cu interdiffusion



Film accommodation



Temperature limit

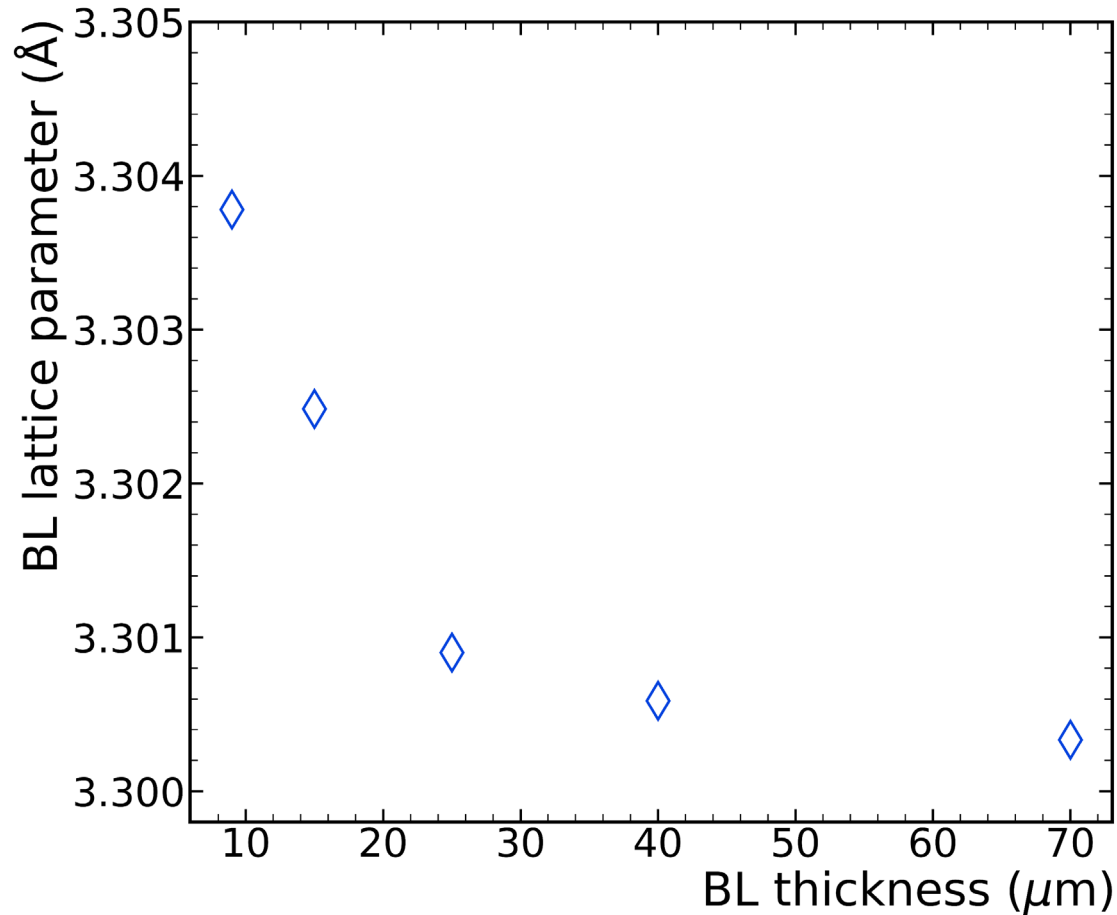


$T_c = 17.33 \pm 0.25$ K on Cu+50 μm BL at $T_{\text{dep}} = 600$ °C

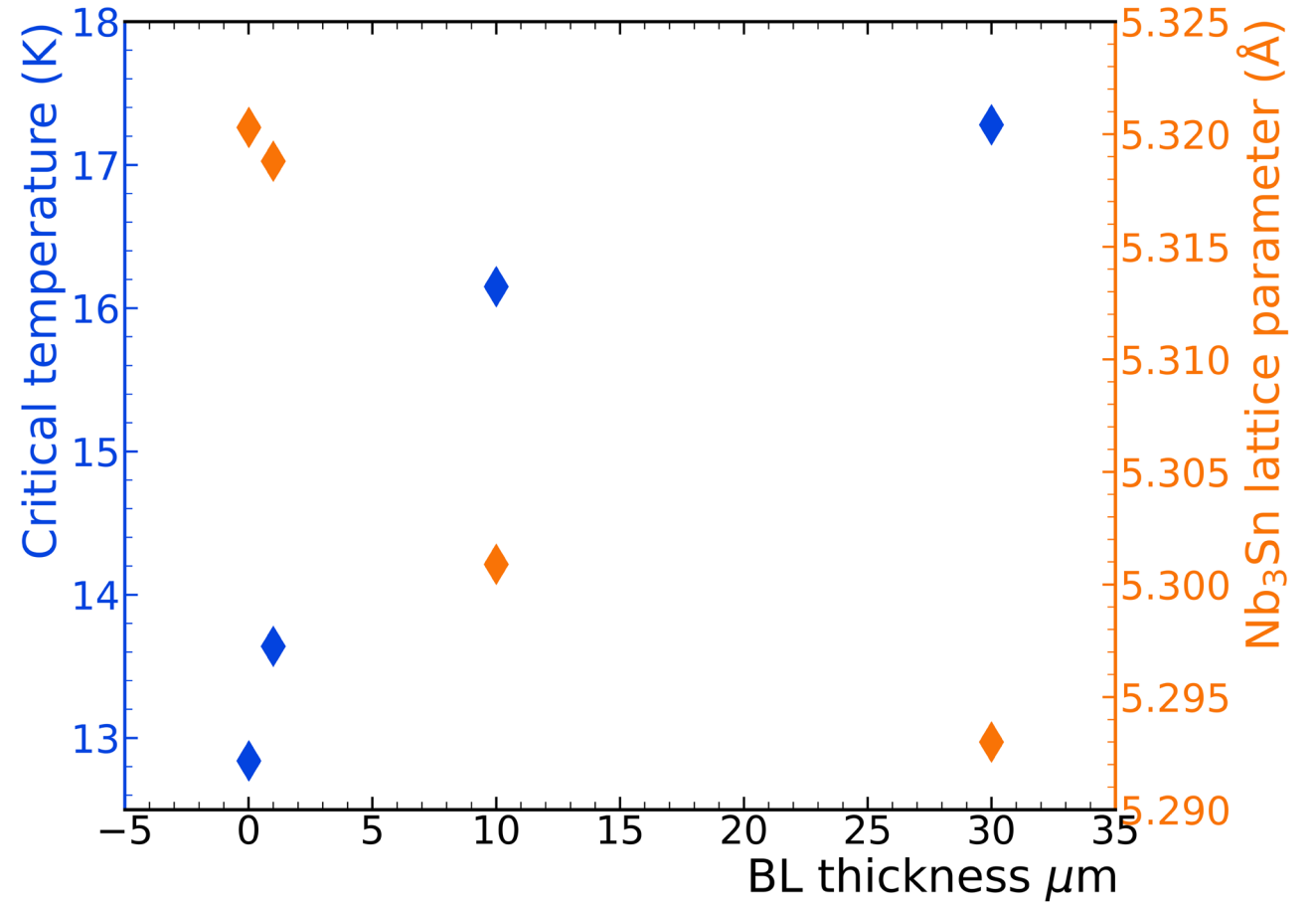


Accomodation effects

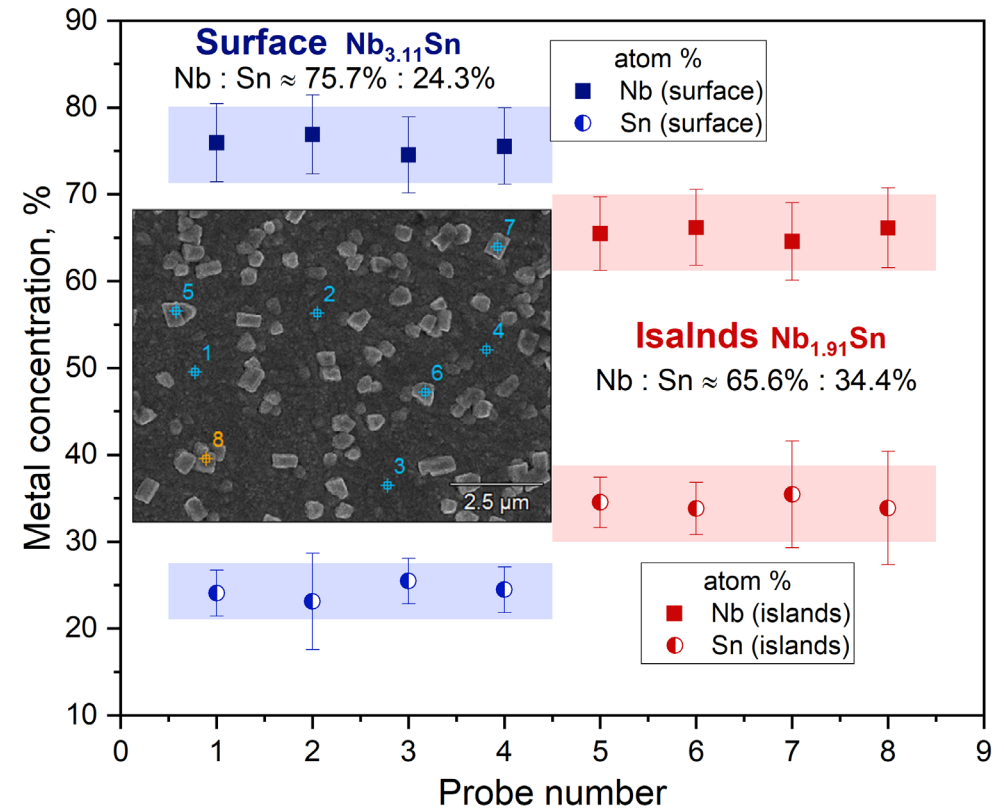
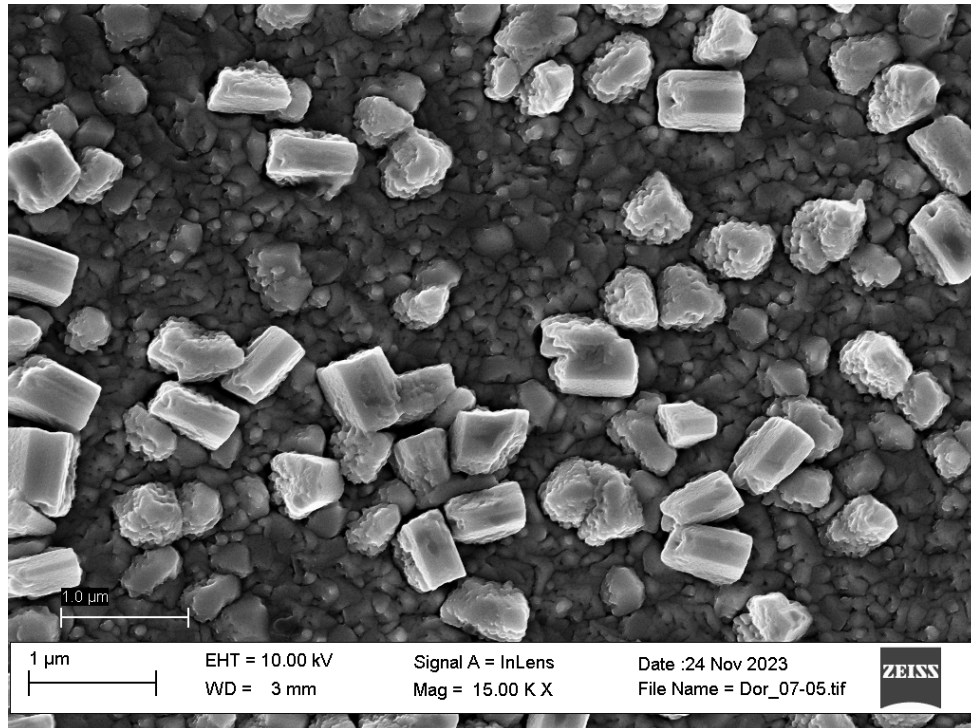
Nb buffer layer



Nb₃Sn/Nb/Cu 650 °C



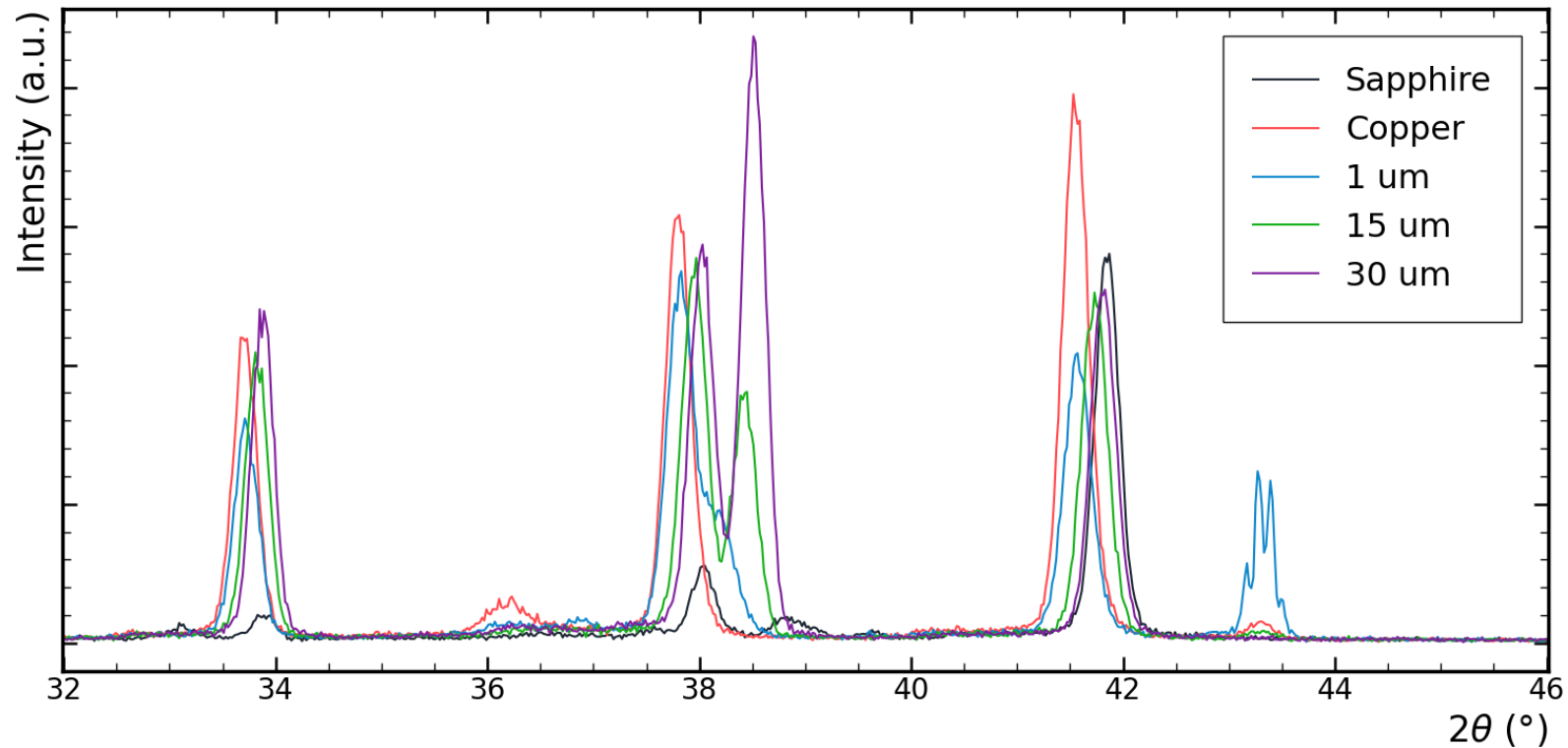
Morphology and composition



XRD patterns

Nb3Sn peaks								
(002)	(012)	(112)	(023)	(123)	(004)	(024)	(124)	(233)
33.870	38.012	41.801	63.352	66.041	71.263	81.284	83.736	86.176

Nb Im-3m peaks				Cu peaks			
(011)	(002)	(112)	(022)	(111)	(002)	(022)	(113)
38.575	55.696	69.796	82.695	43.342	50.479	74.174	89.622



Nb₃Sn on copper: outlook

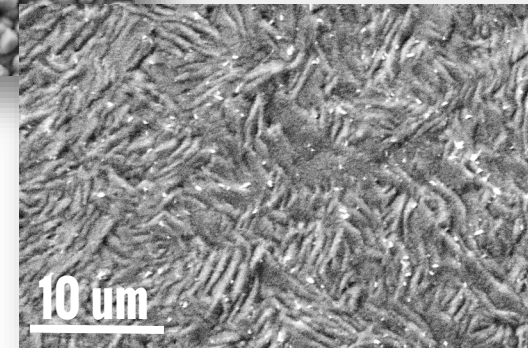
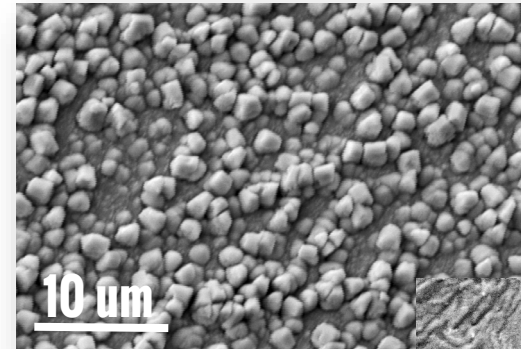


ONGOING (on our side):

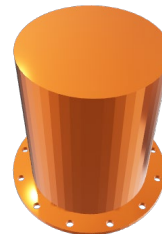
- $T_{\text{dep}} = 500 \text{ }^\circ\text{C} \rightarrow$ find lower limit
- buffer layer thickness dependencies
- EDS composition analysis
- XRD analysis

ONGOING (within I.FAST):

- SQUID magnetometry
- SEM surface and FIB cross-section imaging



Toward QPR sample for RF test



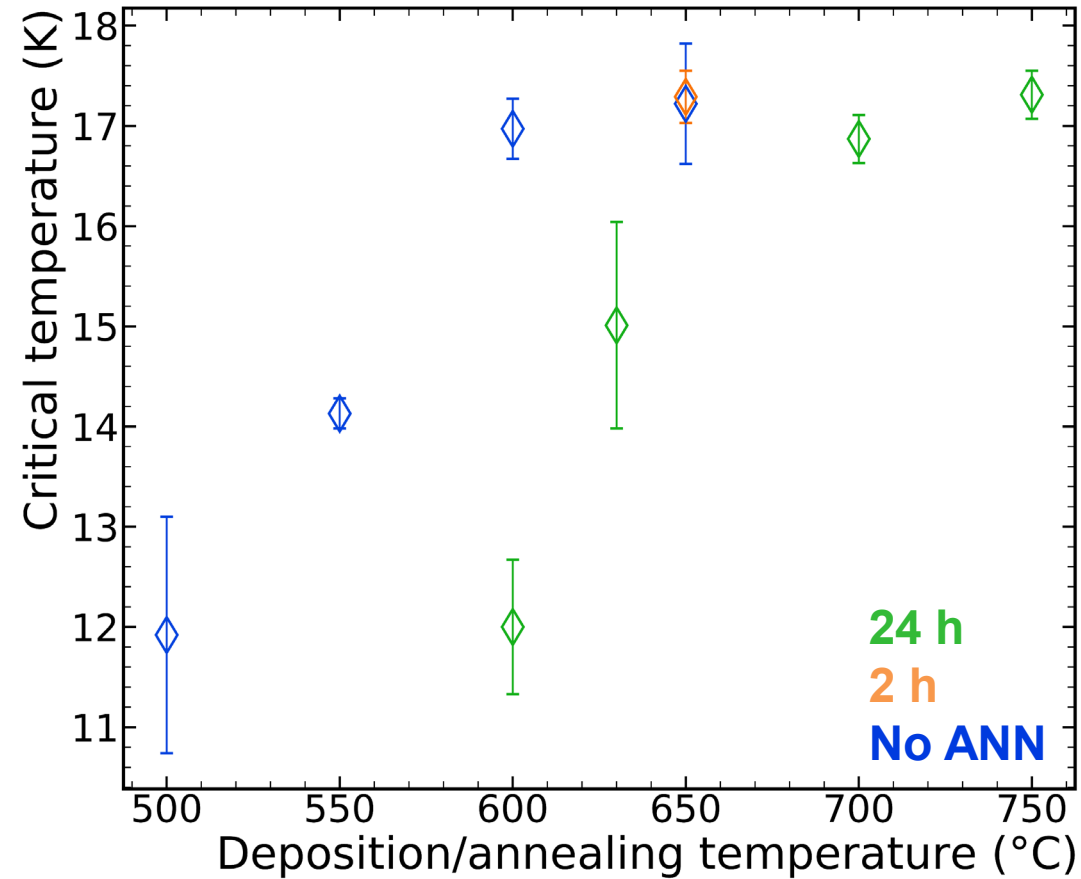
Thank you!



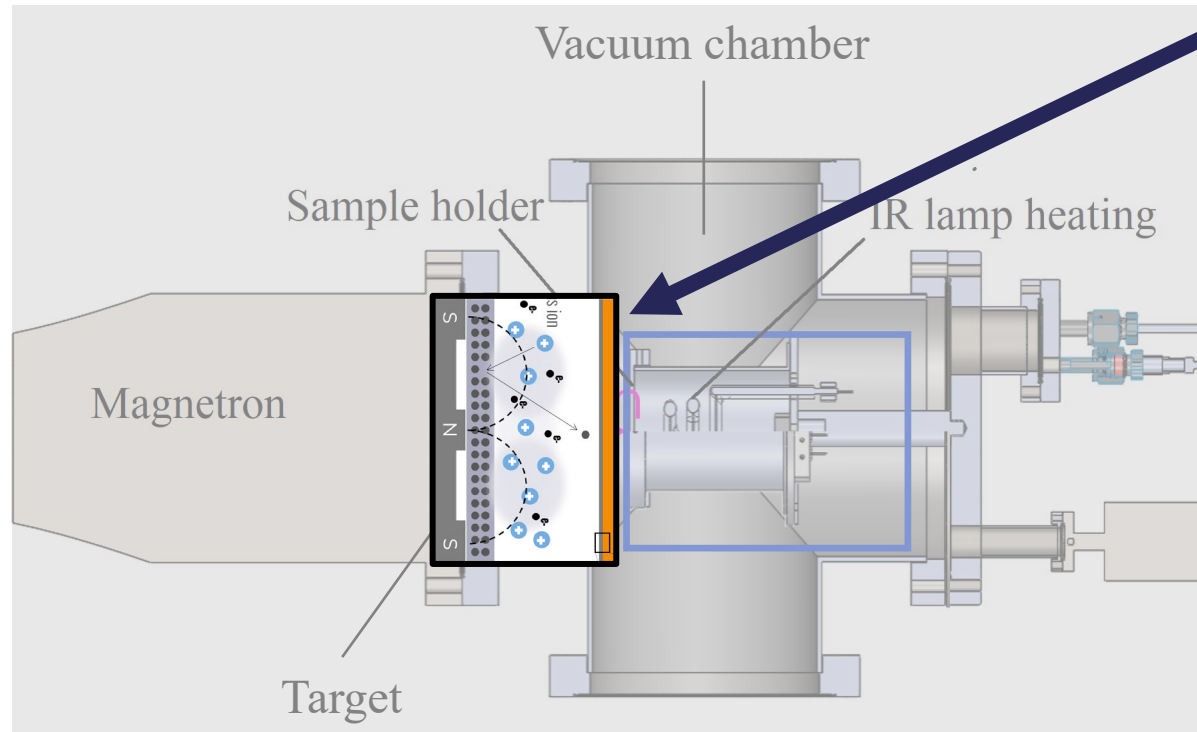
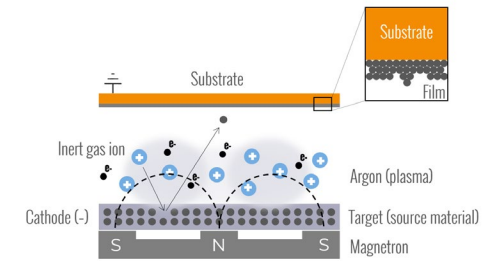
Backup

ANNEALING TIME

Sapphire substrate



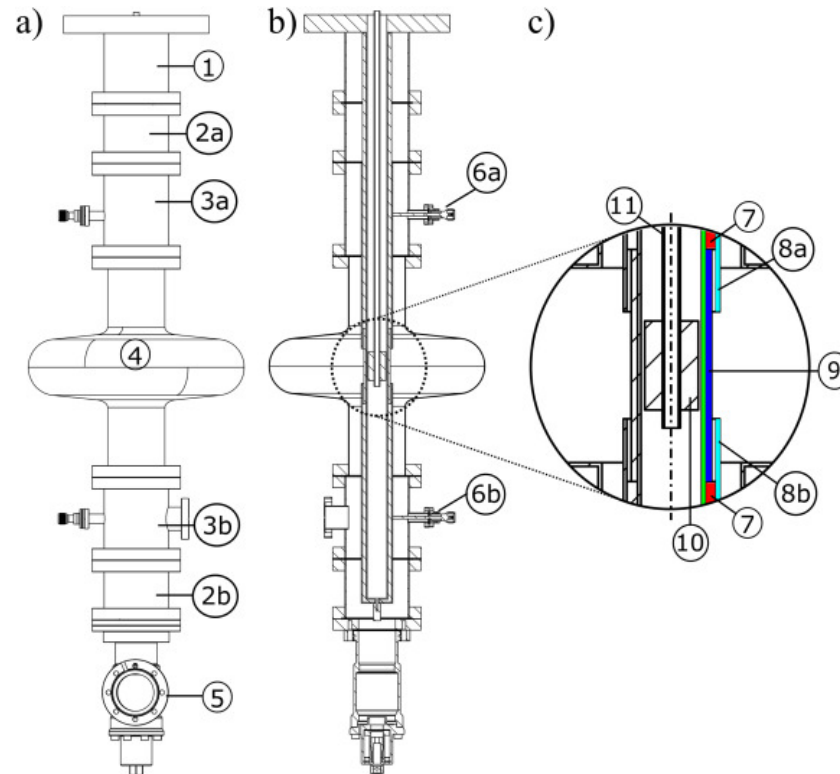
Sputtering chamber



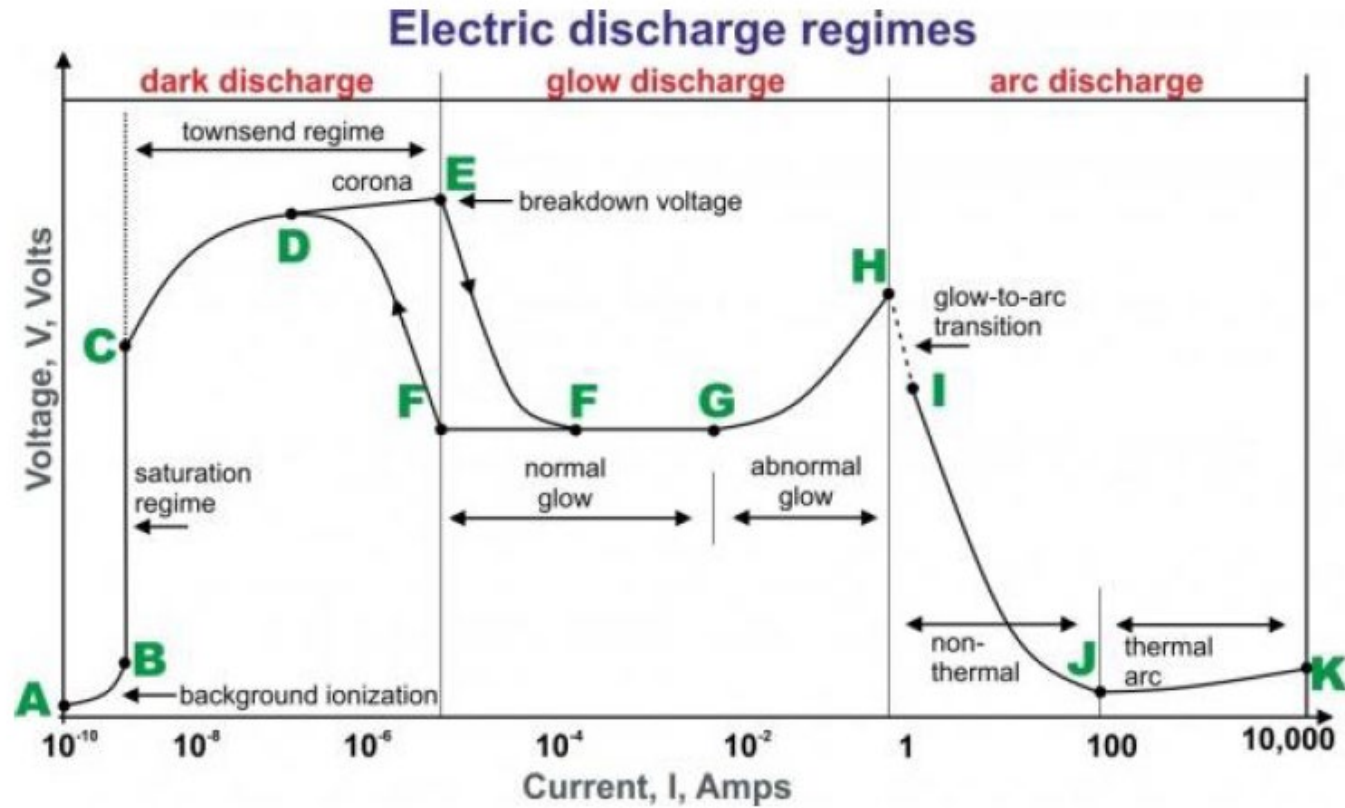
side view

- Planar DCMS on small flat samples
- Single Nb_3Sn stoichiometric target
- Argon atmosphere
- T substrate regulated via IR lamps
- Film thickness $1\ \mu\text{m}$

DCMS cylindrical configuration



Plasma

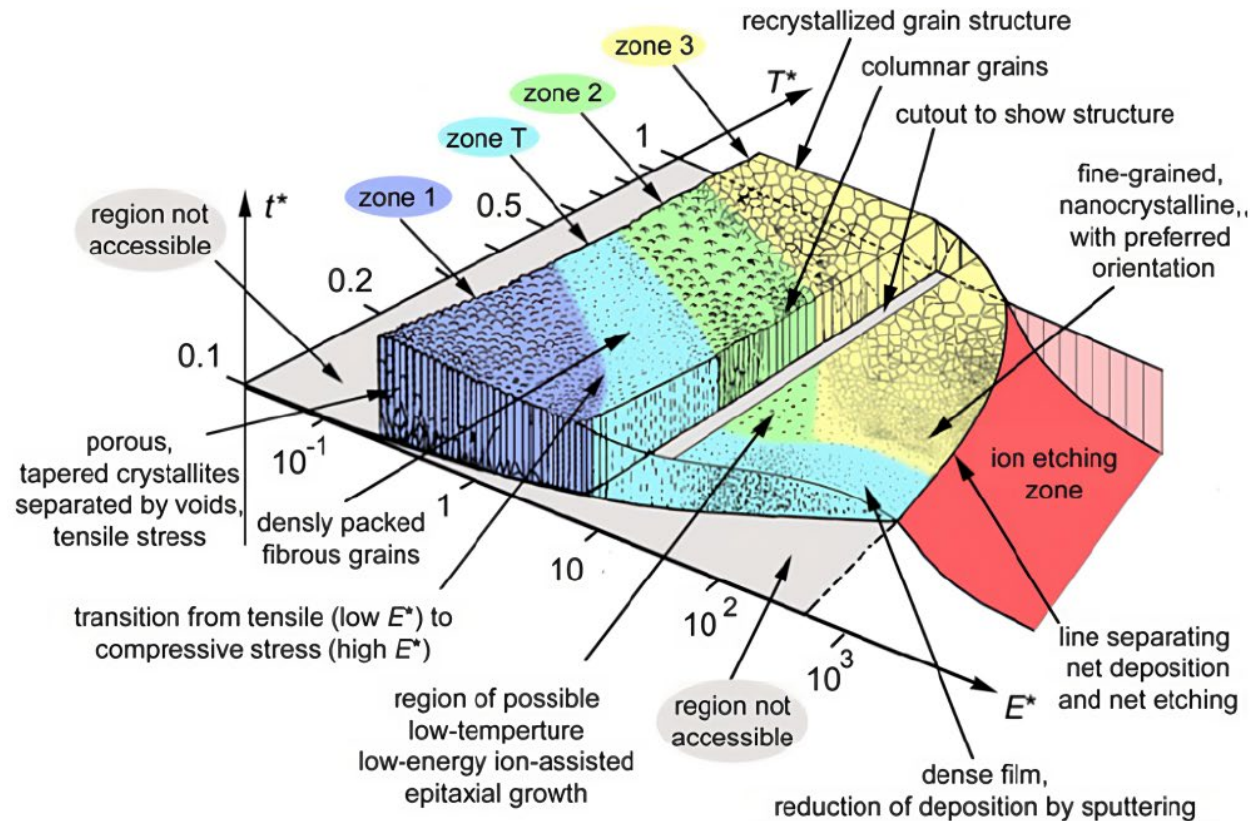


Sputtering

$$E_{\text{th}} = E_b \begin{cases} \frac{1+5.7(M_i/M_a)}{\Lambda} & \text{for } M_i \leq M_a \\ \frac{6.7}{\Lambda} & \text{for } M_i > M_a \end{cases} \quad \Lambda = \frac{4M_iM_a}{(M_i+M_a^2)}$$

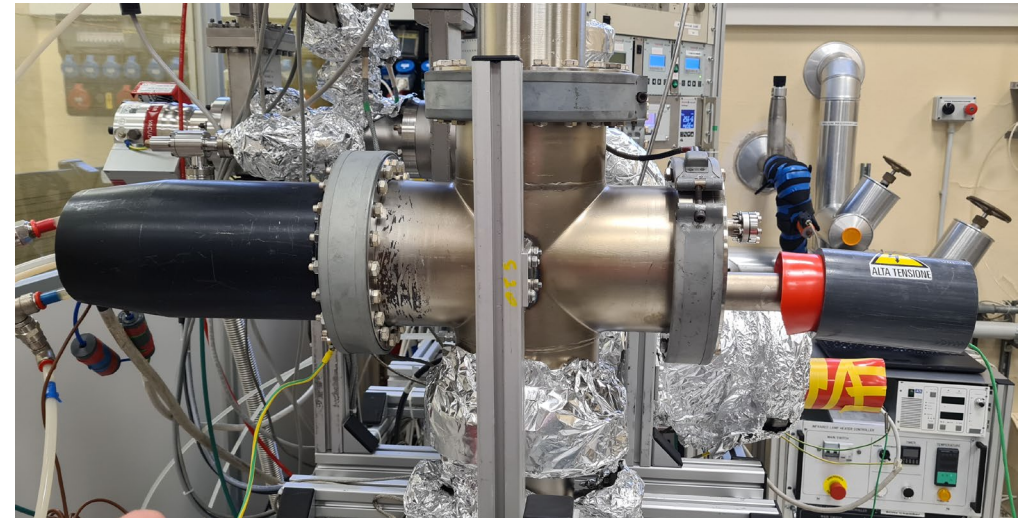
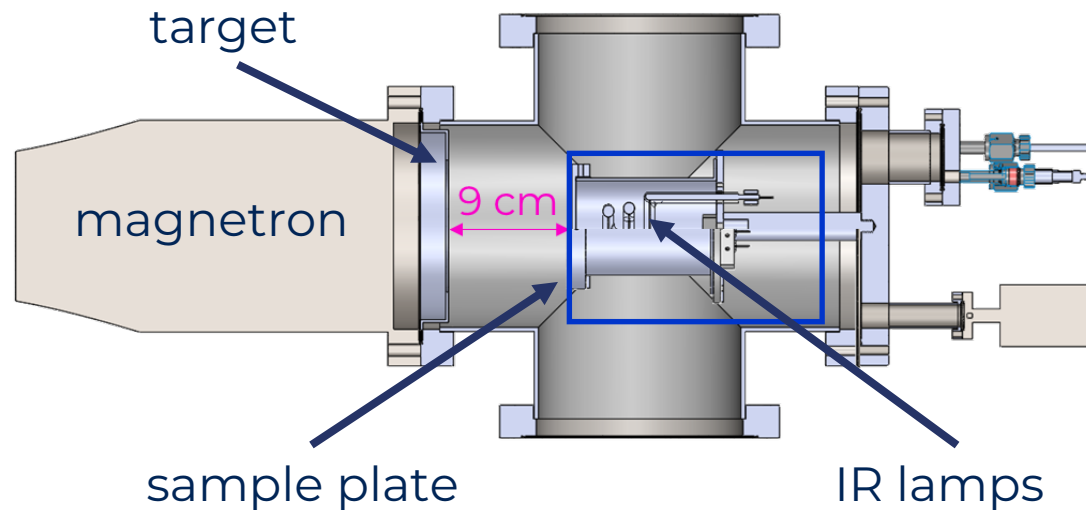
$$Y(E)dE \propto \frac{1 - \sqrt{(E_b + E)/\Lambda E_0}}{E^2(1 + E_b/E)^3} dE$$

Structure zone diagram

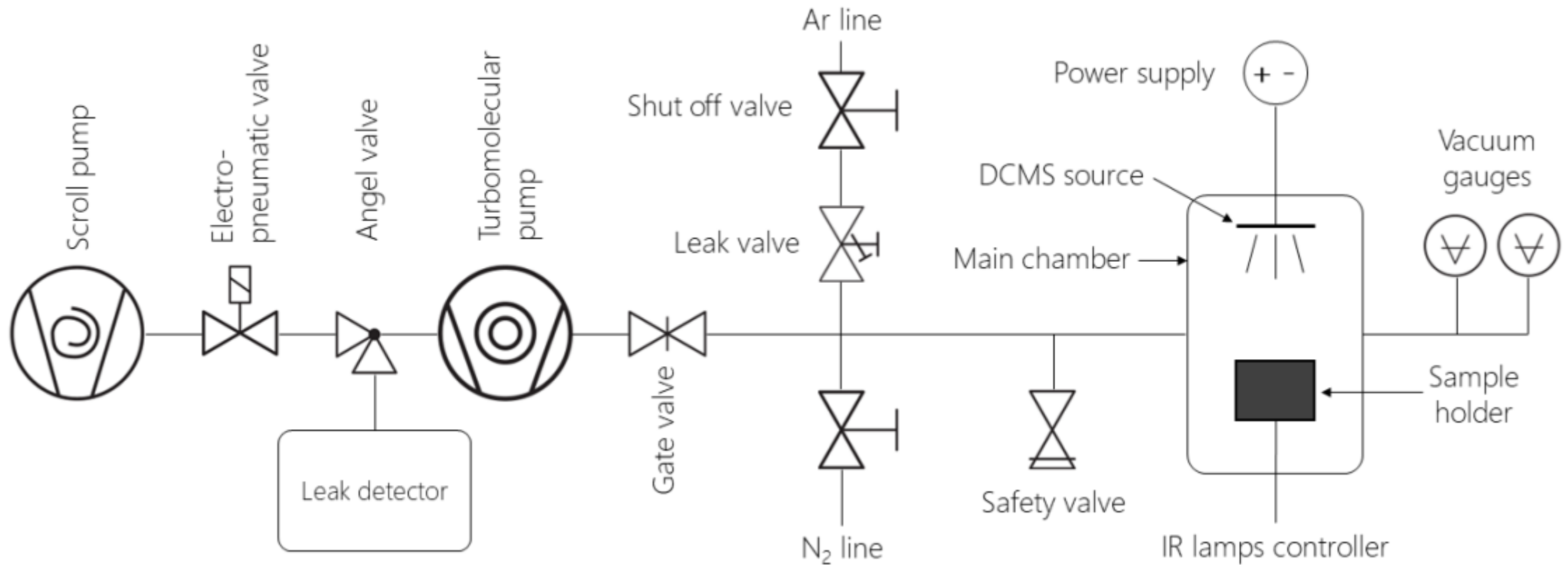


Nb₃Sn Experimental Set-up

- Commercial Nb₃Sn stoichiometric planar target (4" diameter)
- System base pressure: 5×10^{-9} mbar
- Heated sample holder: up to 950 °C via IR lamps



Nb₃Sn Experimental Set-up (PID)

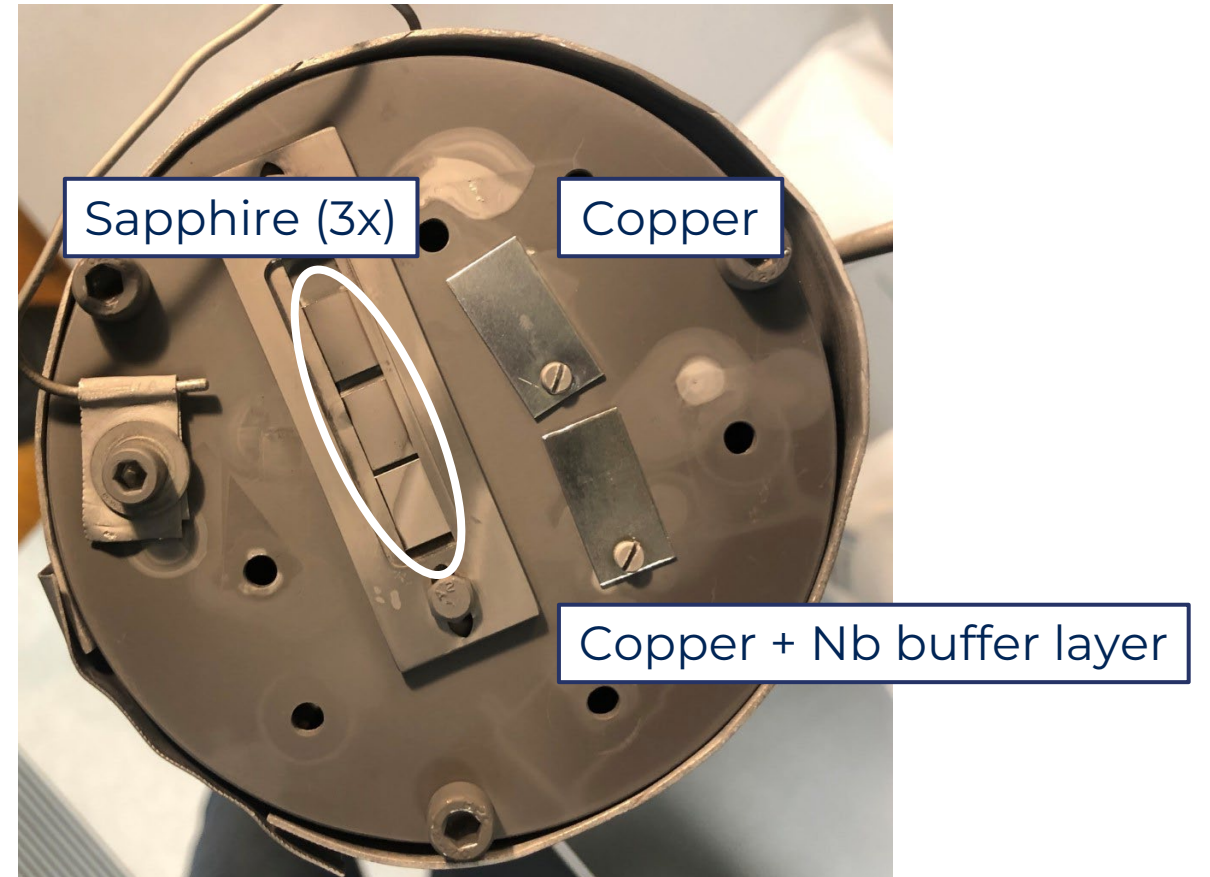


Experimental Set-up

Standardized procedure (1)

Fixed number and type of substrates

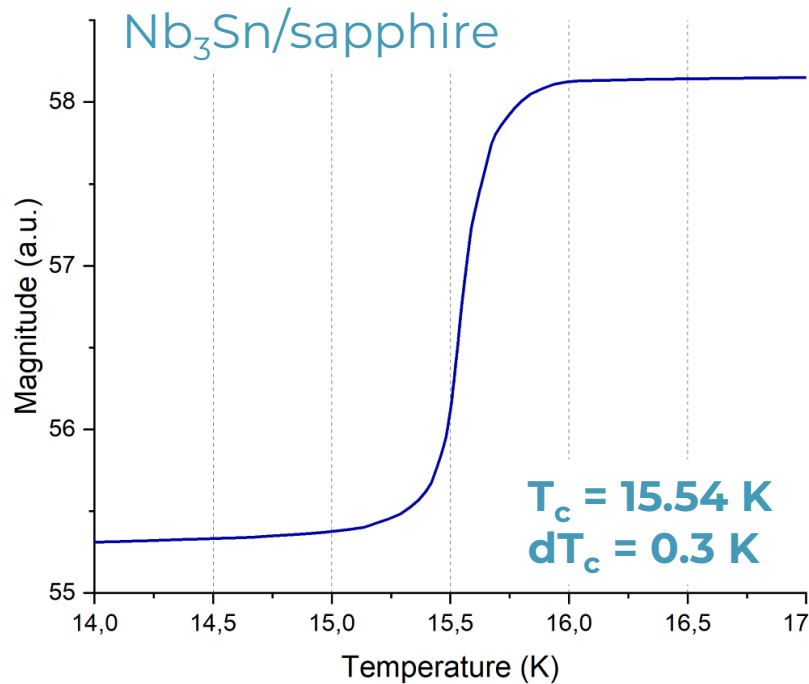
- 5 substrates per run:
 - 3x sapphire
 - 1x copper
 - 1x copper + Nb buffer layer



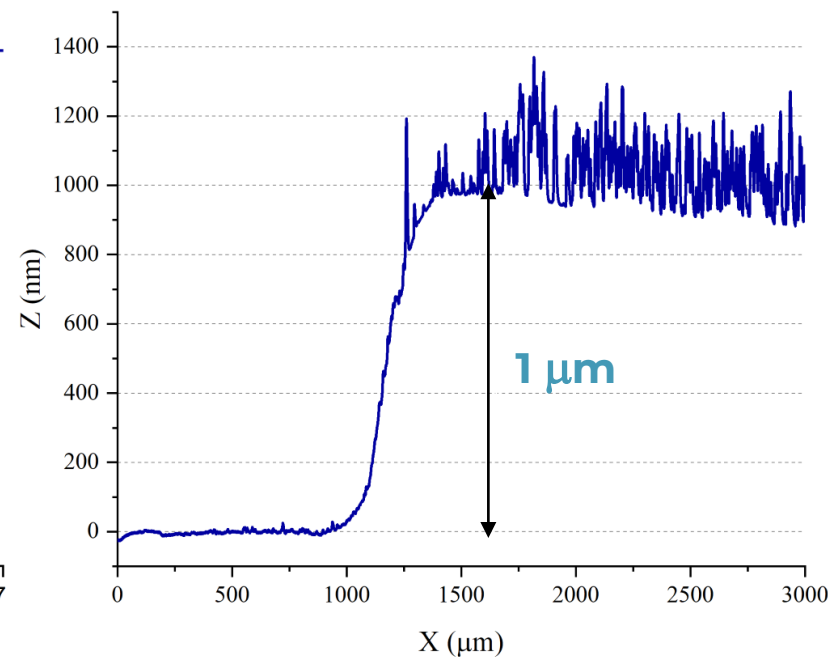
Sample characterization

Standardized procedure (2): fixed measurement routine and T_c extraction method

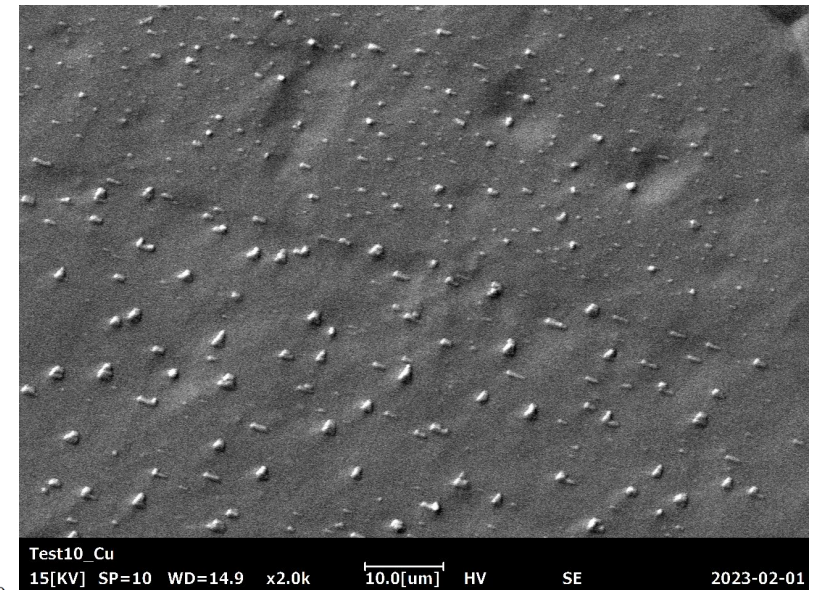
T_c by induction coil



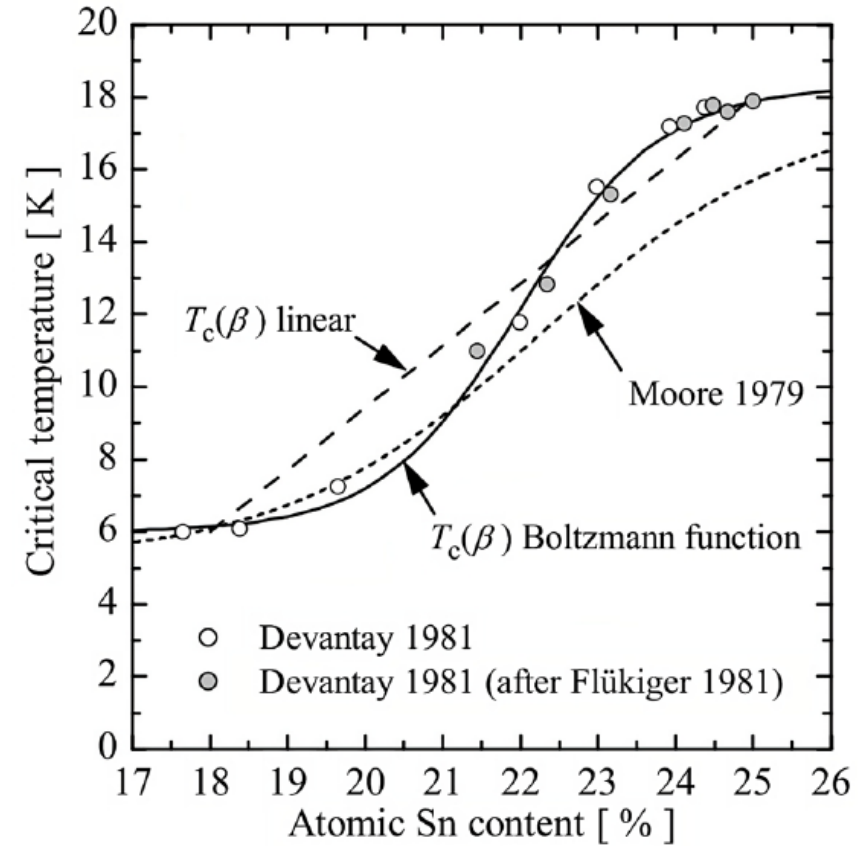
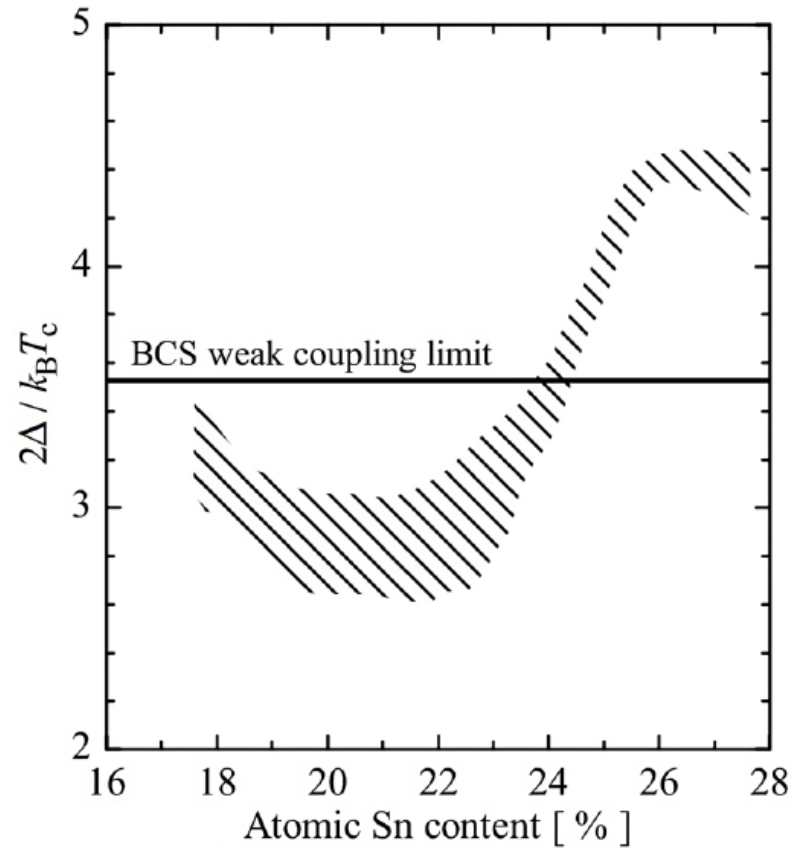
Thickness by profilometer
and SEM cross-section



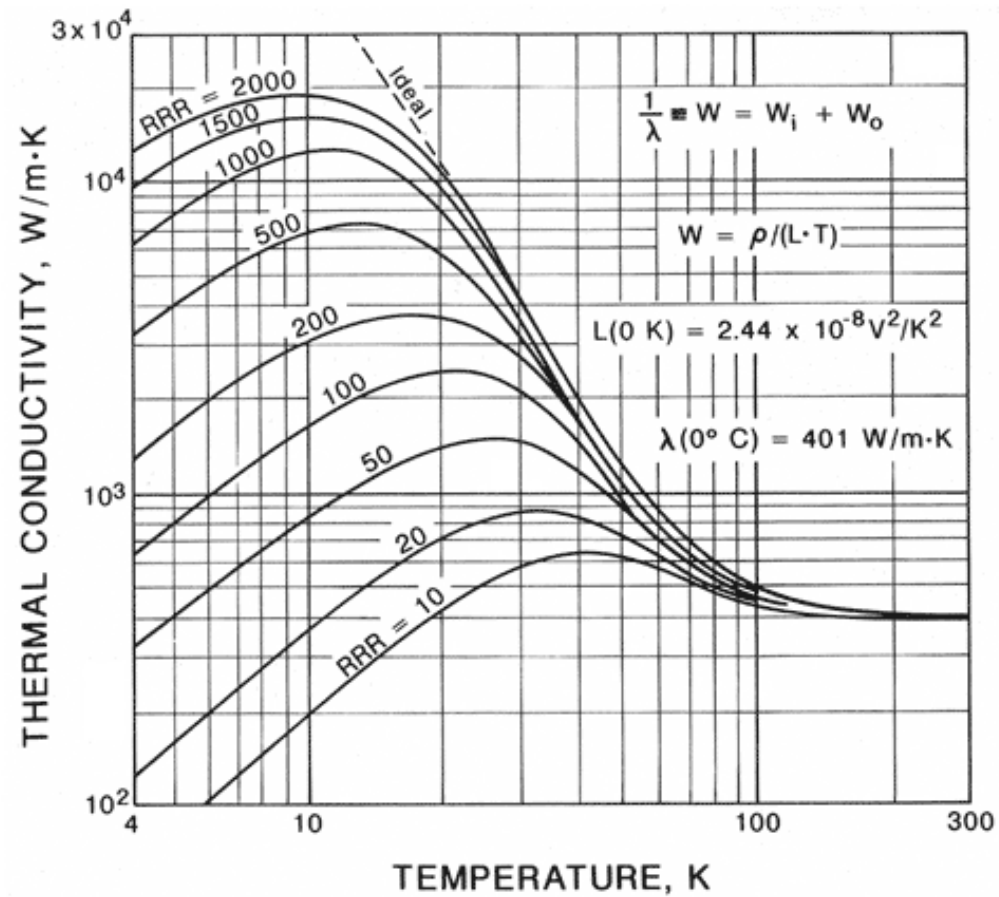
Morphology and composition
by SEM + EDS



Nb₃Sn coupling



Copper



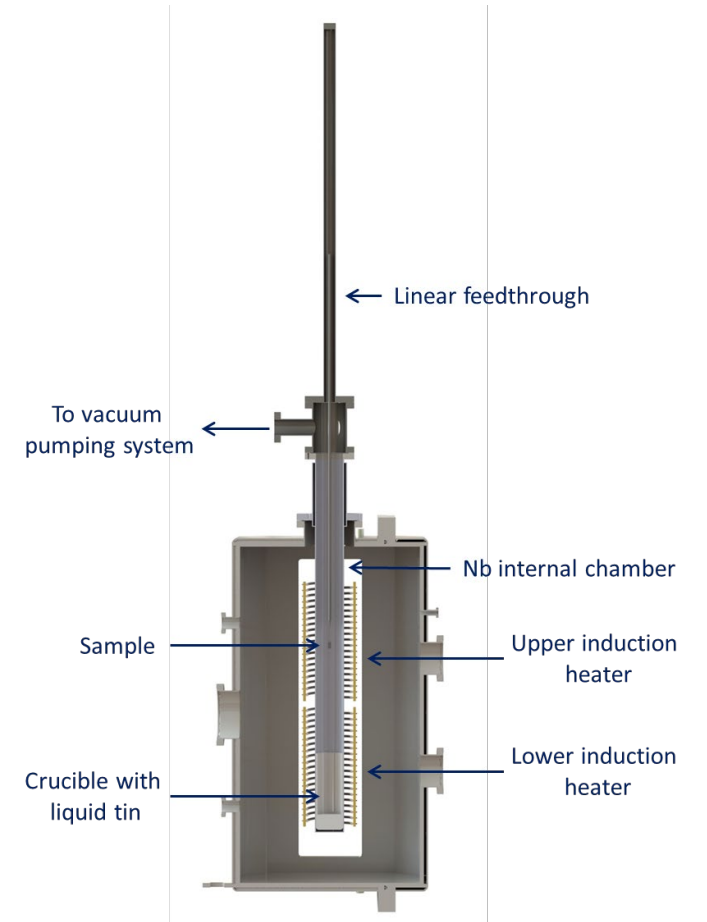
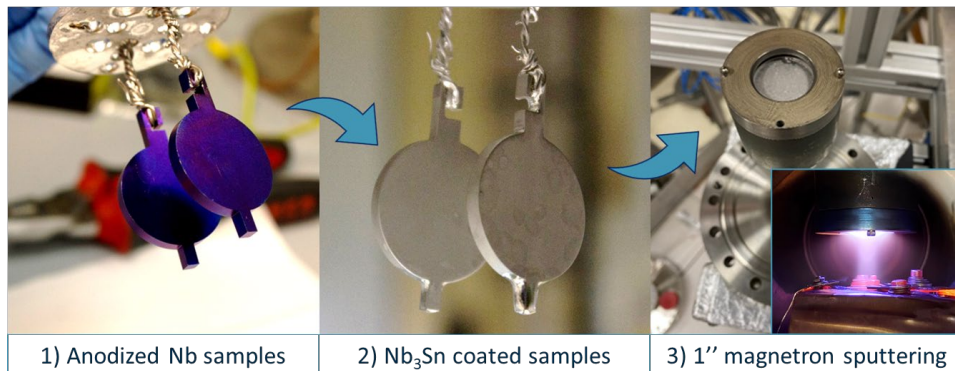
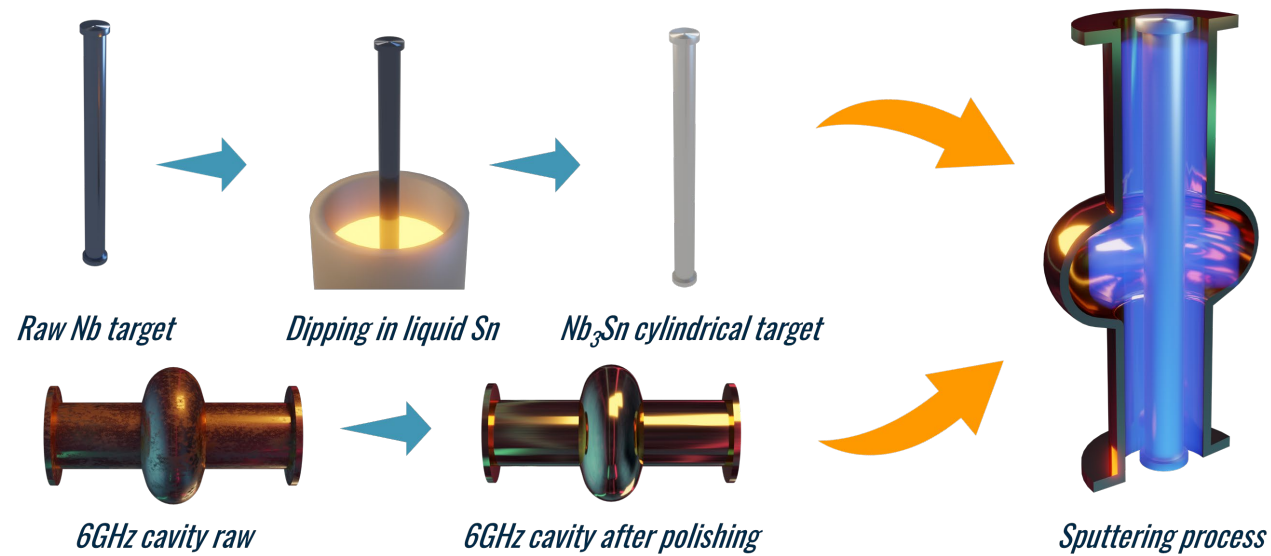
Copper vs. Nb costs

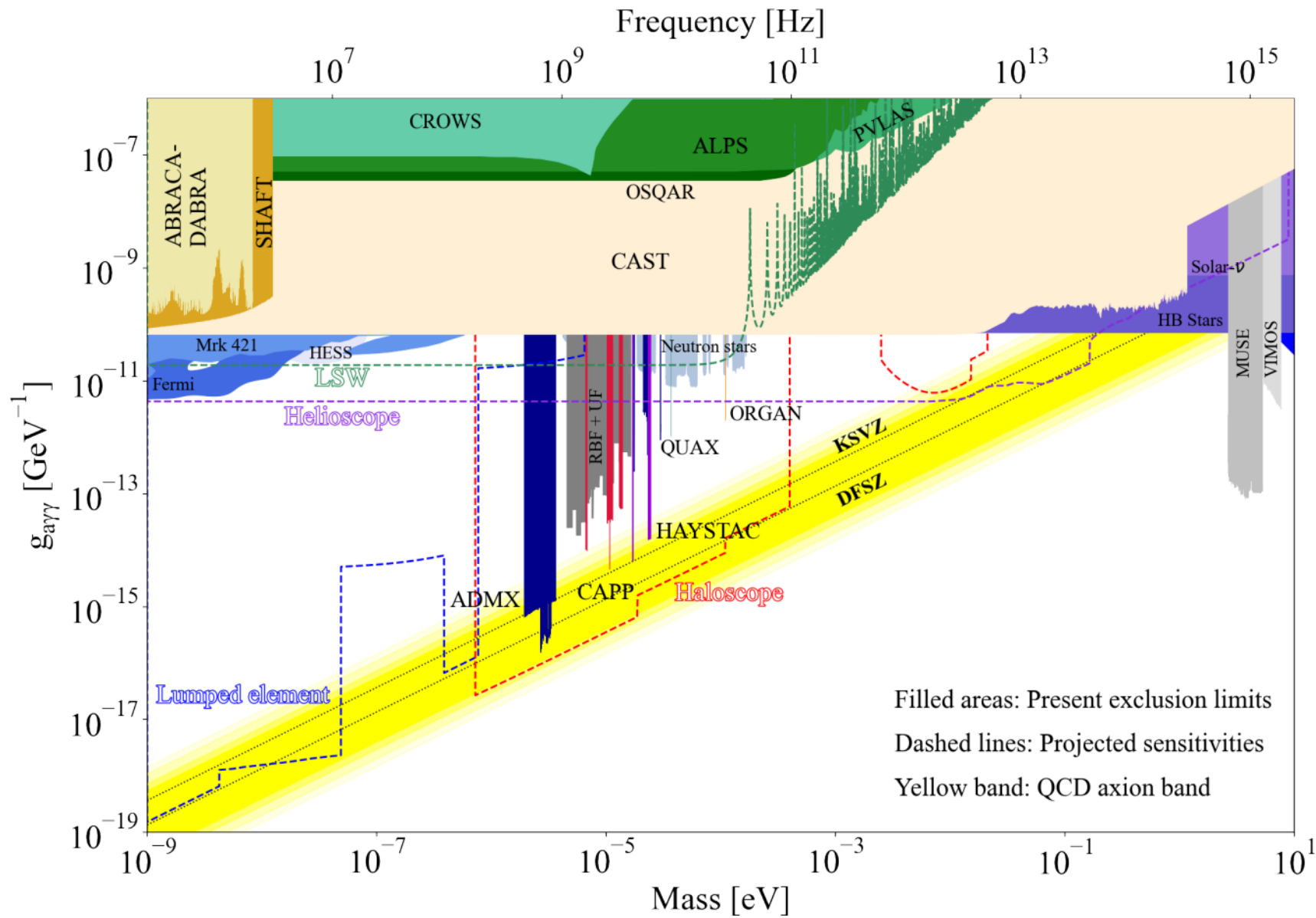
Type I Nb ~ 420 \$/Kg (year 2014)

Cu ~ 6 \$/kg (year 2014)

~ factor 70 difference

Dipping (Nb_3Sn target production via LTD)





Fluxons

$$\eta = \frac{\phi_o B_{c2}}{\rho_n}$$

$$k = \frac{2\pi J_c \phi_o}{d}$$

A. Gittleman and Rosenblum (GR) model

In this seminal work⁵⁷ no thermal and Hall terms were considered: $\mathbf{F}_{\text{thermal}}=0$, $\alpha_H=0$ in Eq. (1). Thus:

$$\rho_{vm,GR} = \frac{\Phi_0 B}{\eta} \frac{1}{1 - i \frac{\omega_p}{\omega}}. \quad (2)$$

In this model η and ω_p can be directly calculated from the data by simple inversion. In the high-frequency limit ($\omega \gg \omega_p$) $\rho_{vm,GR} \rightarrow \rho_{ff}$, being $\rho_{ff} = \Phi_0 B / \eta$ the free flux flow resistivity. Equation (2) gave for many years the theoretical

First SC haloscope

→ Hybrid Geometry

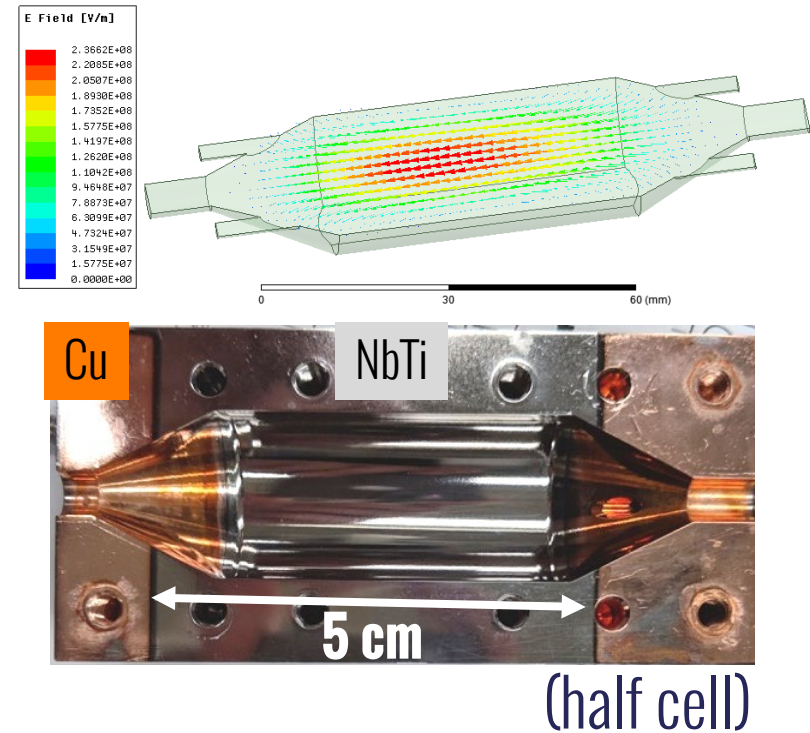
Cavity designed at LNF

Cavity coated at LNL with 4 μm NbTi layer

Cu endcaps to reduce vortex motion dissipation

Freq. (TM010)	9.1 GHz
G_{cyl}	6270 W
G_{cones}	482 W
$R_{S^{Cu}}$	4.9 mW
Q_0^{Max}	1.3×10^6

$$\frac{1}{Q_0} = \frac{R_s^{cyl}}{G_{cyl}} + \frac{R_s^{cones}}{G_{cones}}$$



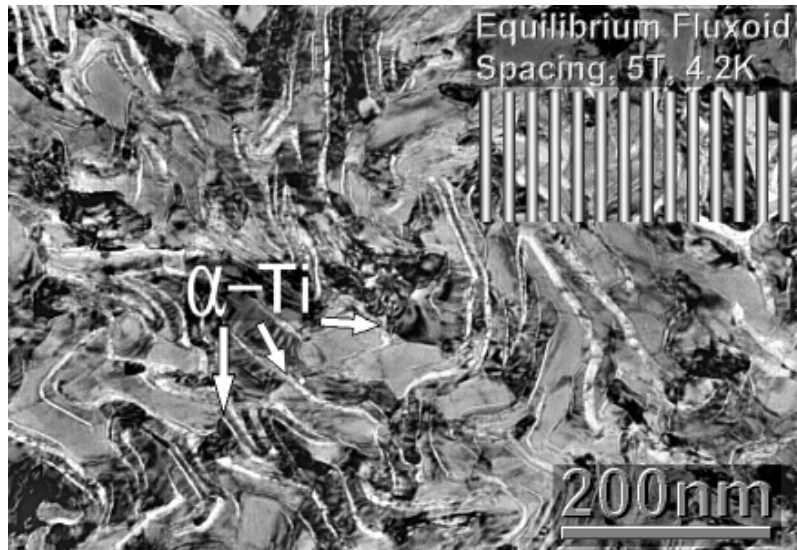
Deposition via DCMS:

- allows easy deposition (NbTi \neq Nb₃Sn)
- allows easy exploration of Ti concentration
- allows application of mask for copper cones

[1] D. Alesini, Phys. Rev. D 99, 101101(R) (2019)

NbTi pinning

α -Ti precipitates act as pinning centers in NbTi alloys



D. C. Larbalestier and P. J. Lee, "Proceedings Particle Accelerator Conference, Dallas, TX, USA, 1995

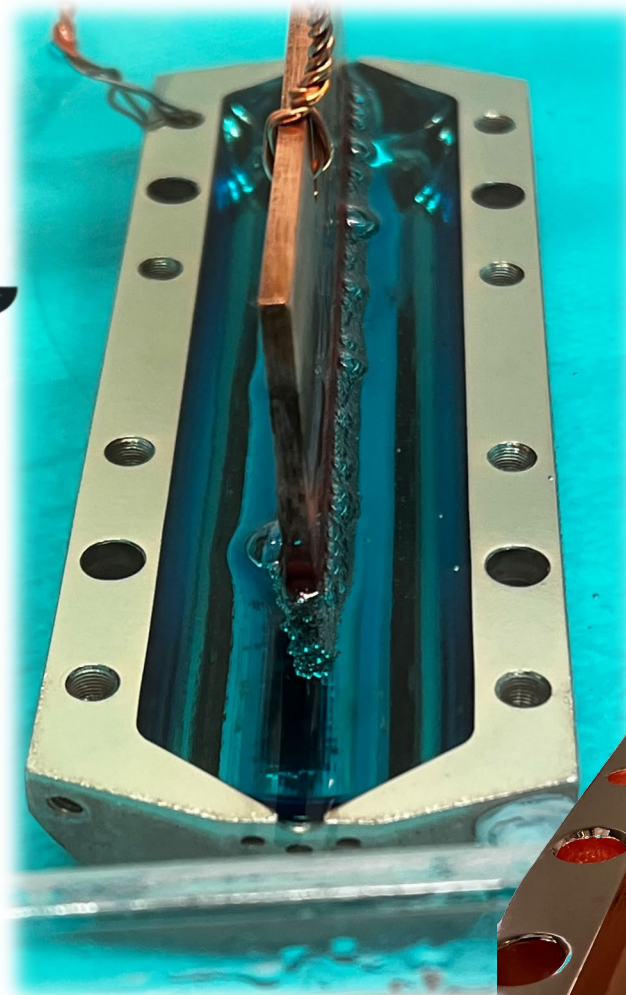
- ▶ **Pinning Force has a maximum with Ti content**
(we expect similar pinning for $\text{Nb}_{0,31}\text{Ti}_{0,69}$ and $\text{Nb}_{0,38}\text{Ti}_{0,62}$)
- ▶ α -Ti precipitates density and dimension depends on thermal treatments
- ▶ Data only for wires → **no data for thin films**

Surface Preparation

1. Ultrasonic degreasing in Rodatel-30 soap
2. Ultrasonic in deionized water
- 3. Electropolishing**
in H_3PO_4 : Butanol at 3:2 volume ratio
4. Chemical polishing in SUBU-5 solution
5. Surface passivation with sulfamic acid
6. Ultrasonic, ethanol rinsing and drying
7. 100 bar High Pressure Water Rinsing

Surface preparation is a key process for SRF cavities

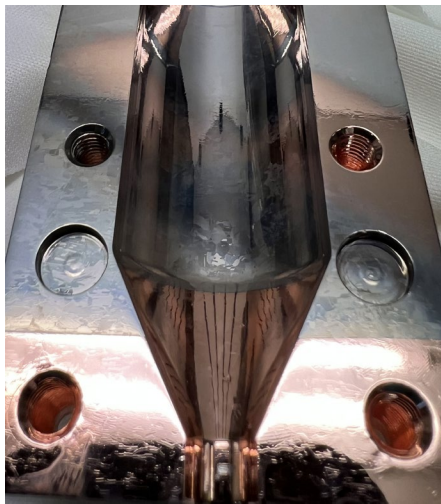
Coating mimate substrate surface



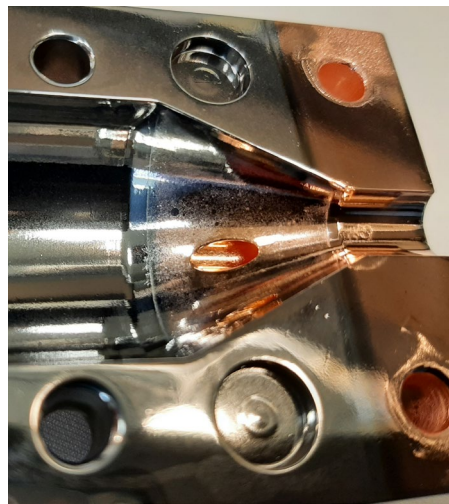
$R_a < 100 \text{ nm}$



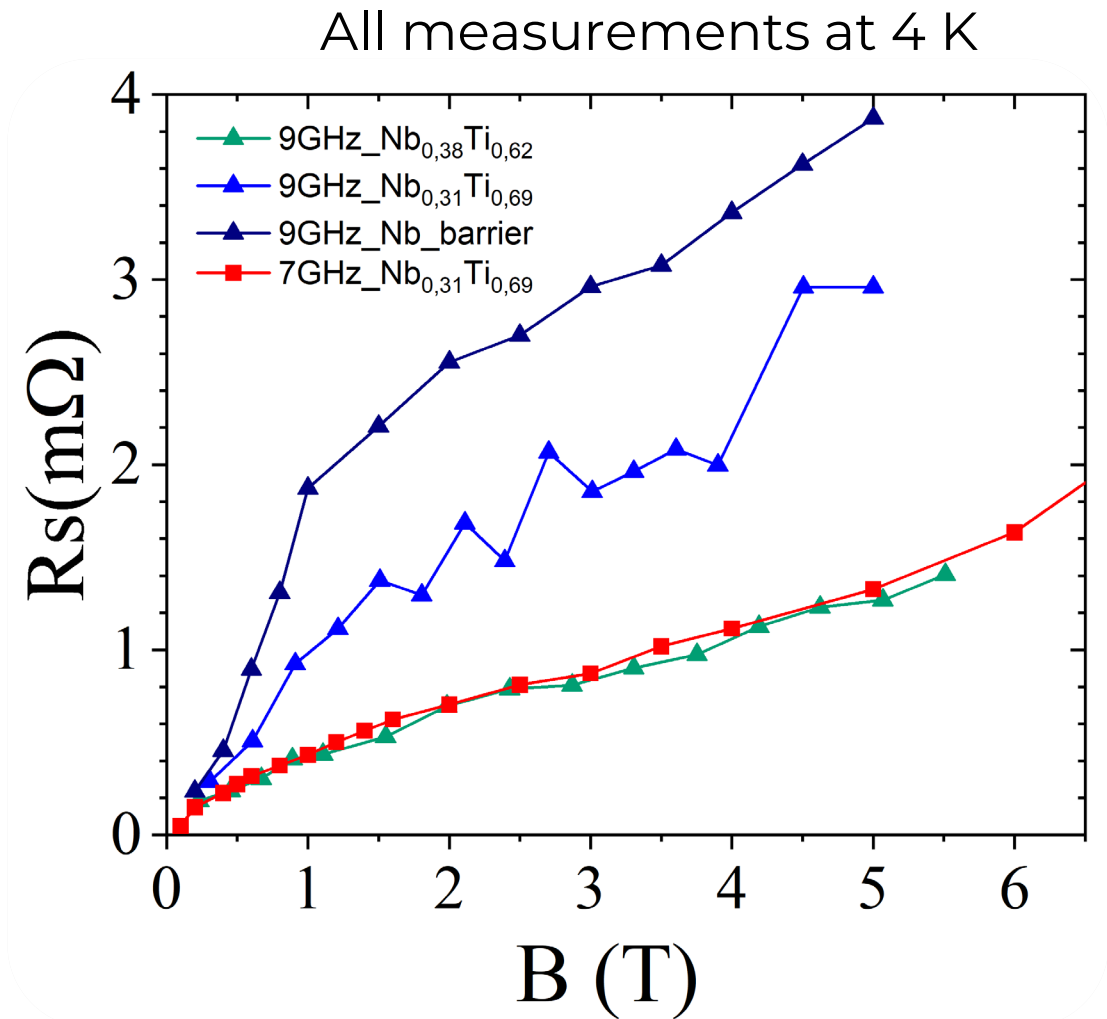
Defects on the cavity surface



Large grain boundaries

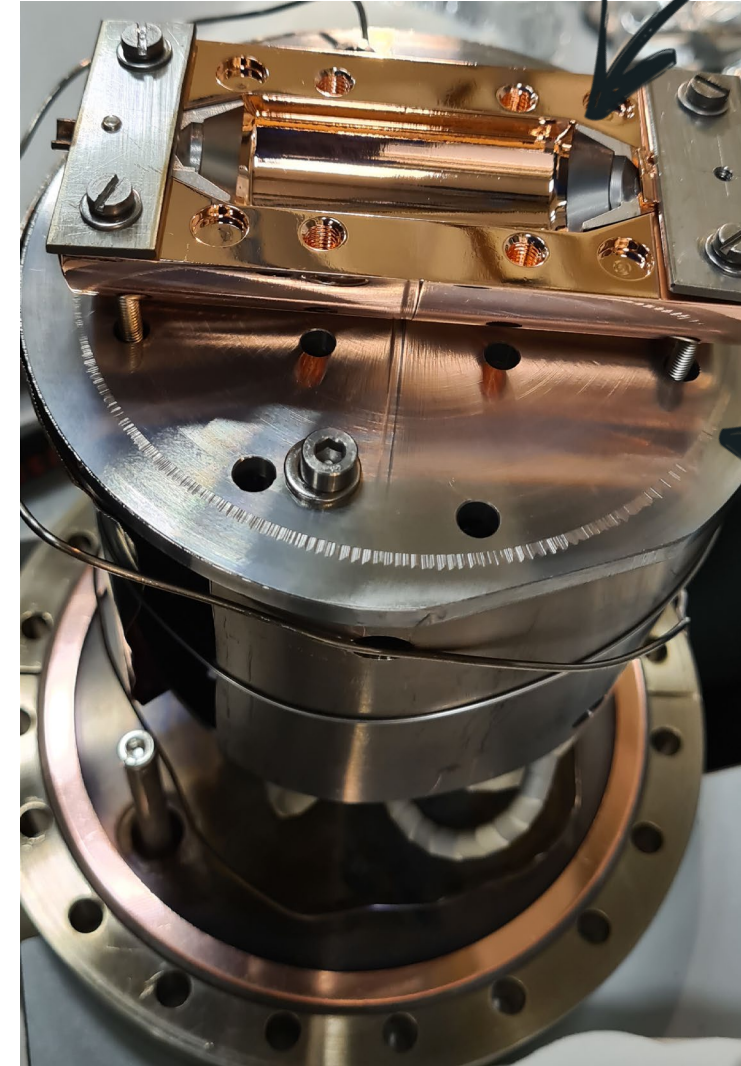


Pitting + NbTi coating on Cu cones



NbTi Coating Set-up

- ▶ DC Magnetron Sputtering
- ▶ Single NbTi target
- ▶ Ar pressure $6 \cdot 10^{-3}$ mbar
- ▶ T substrate $500 \text{ }^{\circ}\text{C}$
- ▶ Film thickness $2,5 - 3,5 \text{ } \mu\text{m}$
- ▶ Base pressure: $< 9 \cdot 10^{-9}$ mbar @roomT

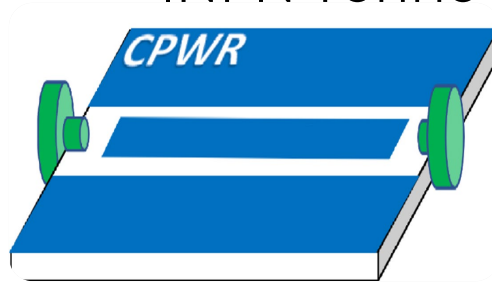


cone
masks

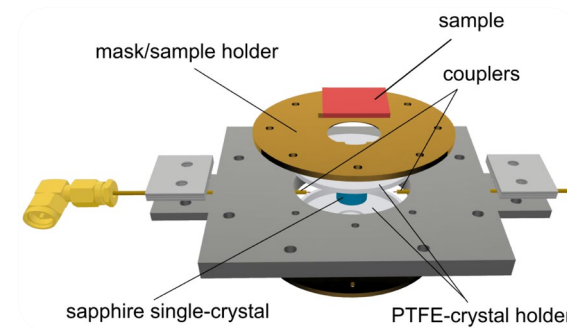
IR heater

NbTi characterisation

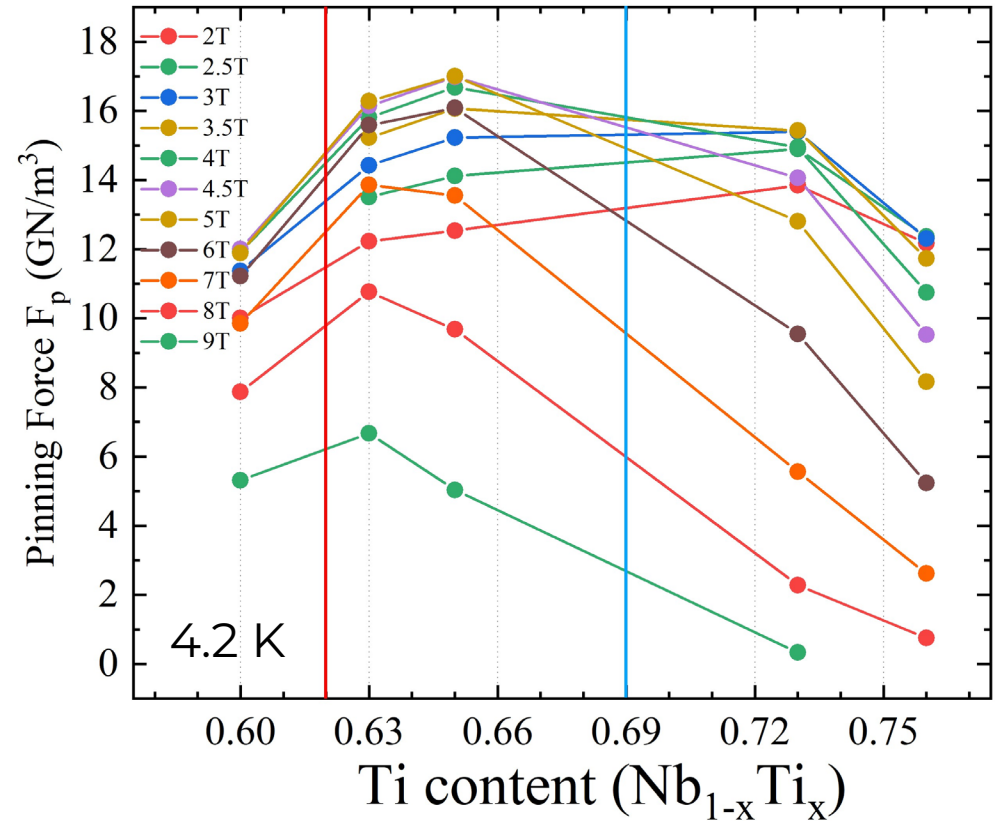
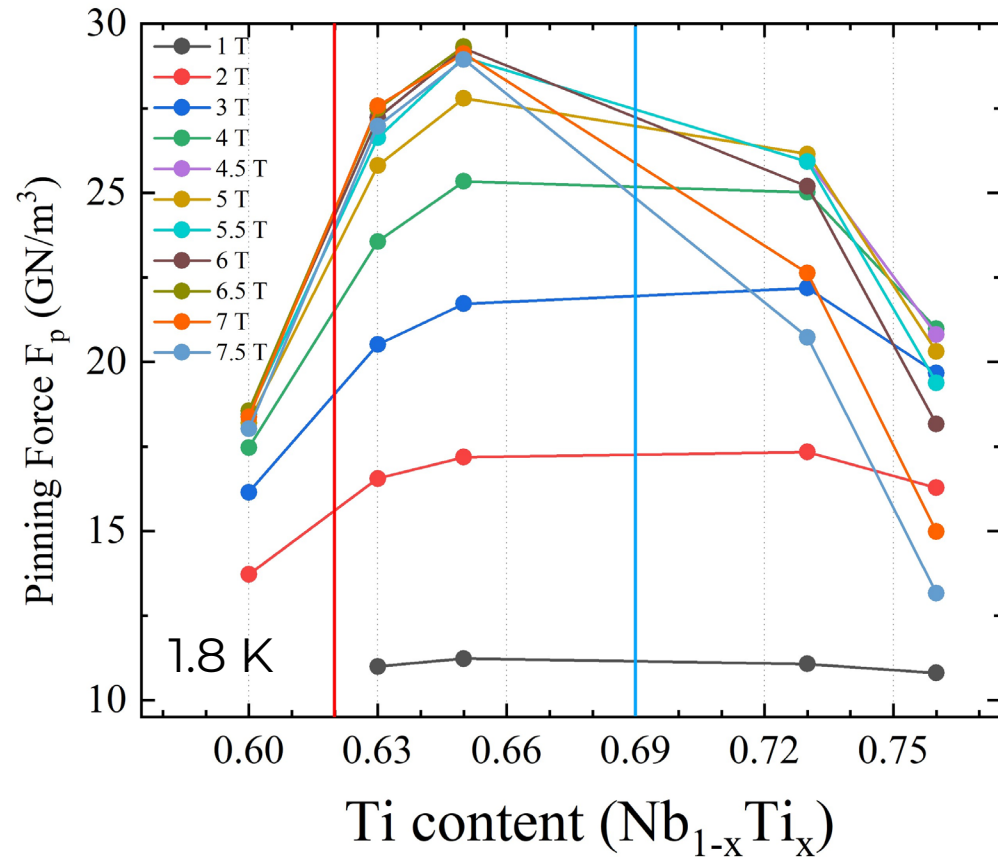
INFN Torino



Roma 3 university



NbTi pinning force

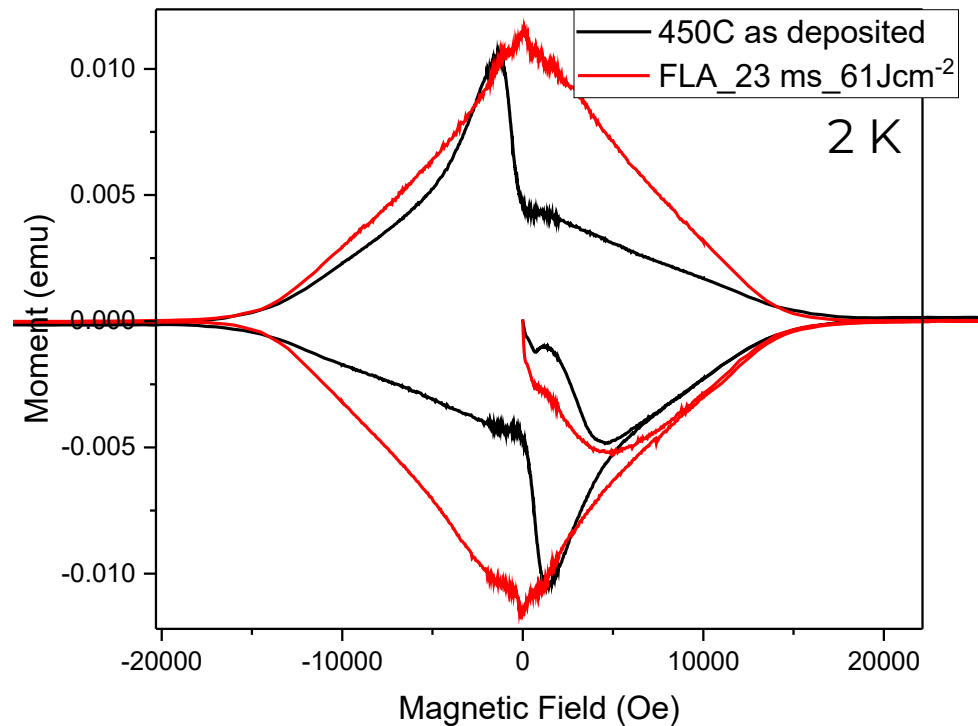
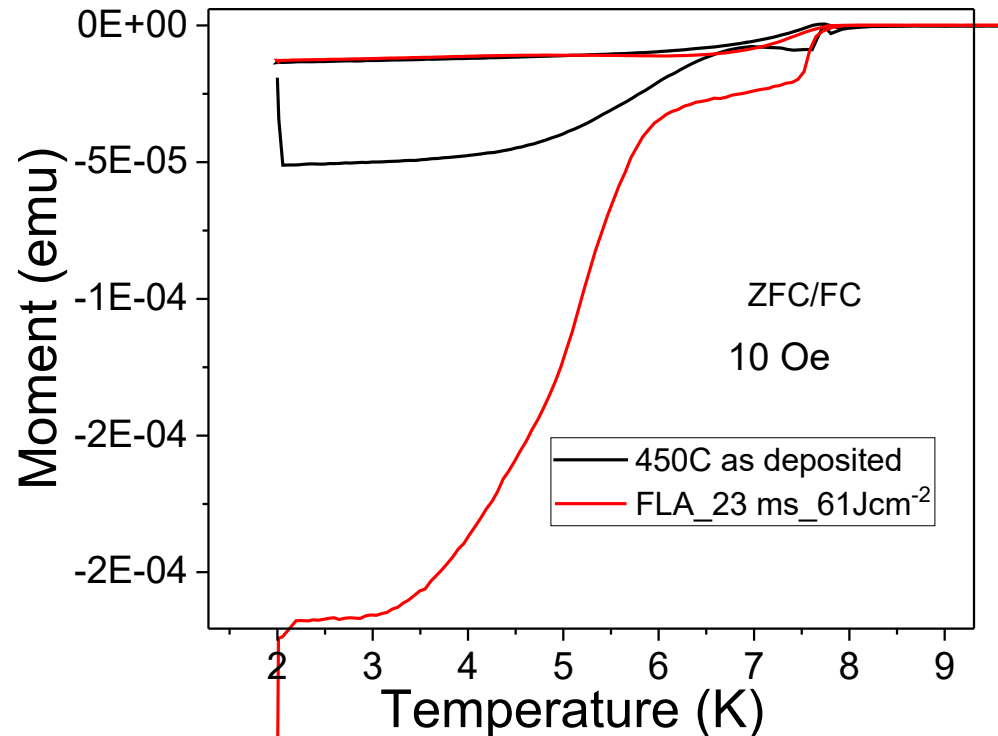


Flash annealing tests (HZDR)

Original idea: can a “bad sample” improve with FLA? (seems not...)

- 1 μm Nb_3Sn on sapphire, Cu, Cu+ 1 μm Nb buff
- coating parameters: $T_{\text{coat}} = 450 \text{ C}$, $p_{\text{coat}} = 2 \times 10^{-2} \text{ mbar}$, $P = 20 \text{ W}$
- two sets:
 - 1 set annealed, $t_{\text{ann}} = 24 \text{ h}$ (shown here)
 - 1 set not annealed

NEXT:
test effect on
“good sample”



Technical issues

Two headaches:

1. Leaks

- runs to be repeated several times due to CF copper gaskets not standing the temperature gradient (both time and temperature-wise)
- OFHC copper gaskets by Pfeiffer are granted for $-196 - 200$ °C T range
- other companies offer up to 400 °C
- OFE copper might be an option

2. IR lamps

- not designed to stand high temp for long times, fail each 3 - 4 runs



**What's your experience?
Any advice?**