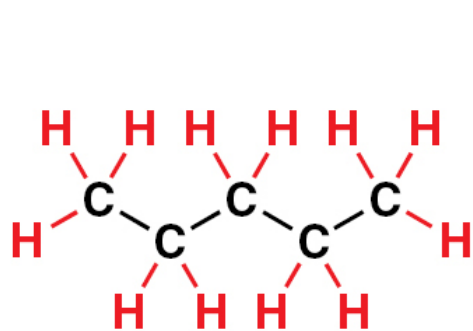
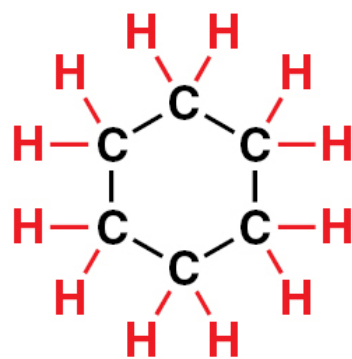


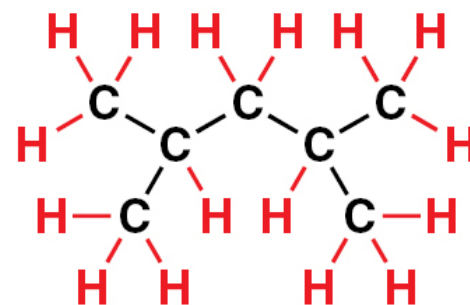
note sulle strutture organiche



a chain

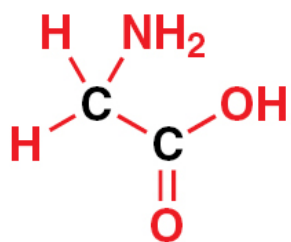


a ring

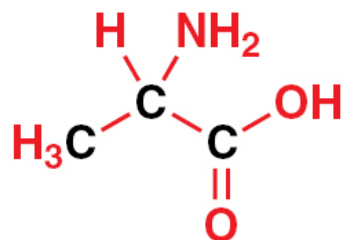


a branched chain

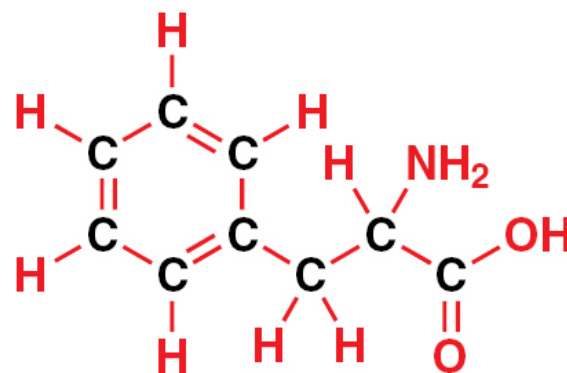
esempi di scheletro carbonioso di molecole organiche



glycine

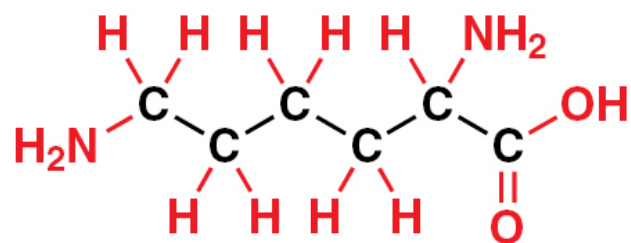


alanine

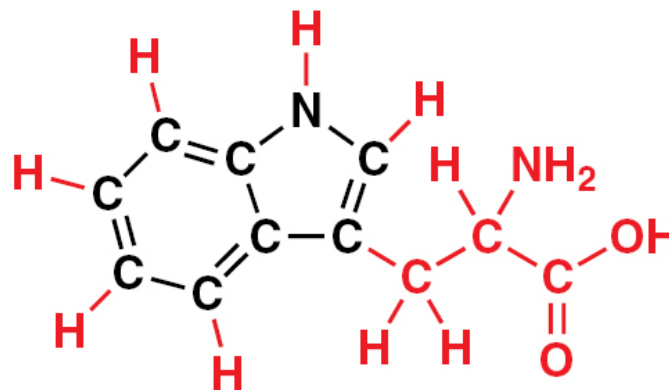


phenylalanine

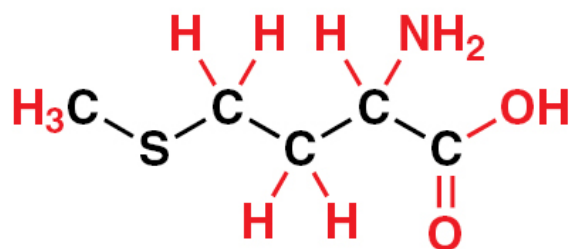
**lo scheletro di atomi di C è il supporto dei gruppi funzionali
(amminico e carbossilico per gli amminoacidi rappresentati)**



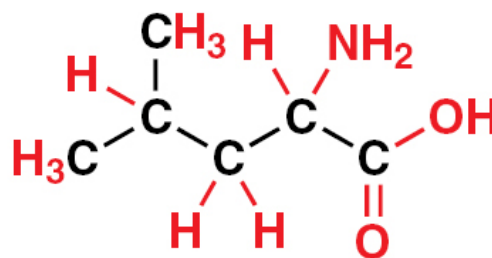
lysine



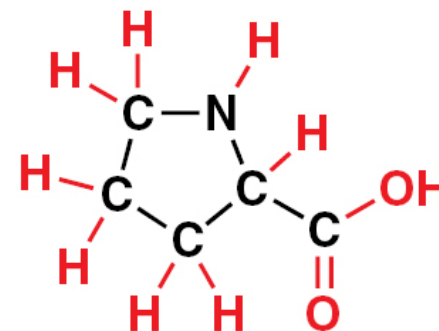
tryptophan



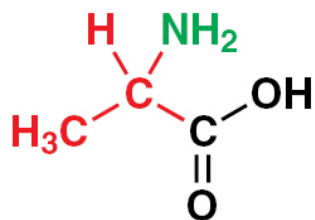
methionine



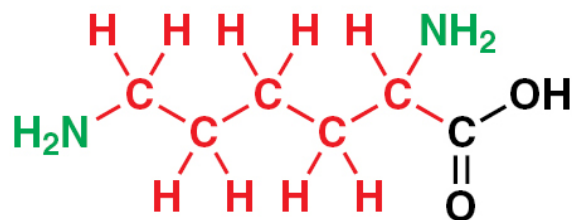
leucine



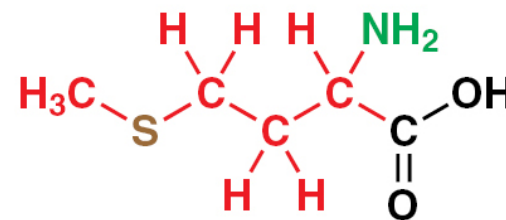
proline



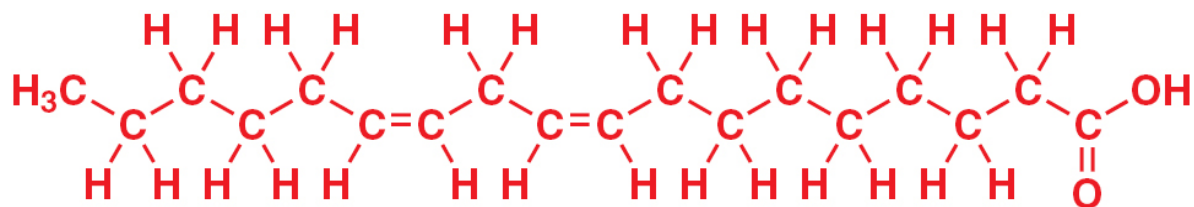
alanine
contains just the amino
and carboxylic acid
functional groups



lysine
has an additional
amino group



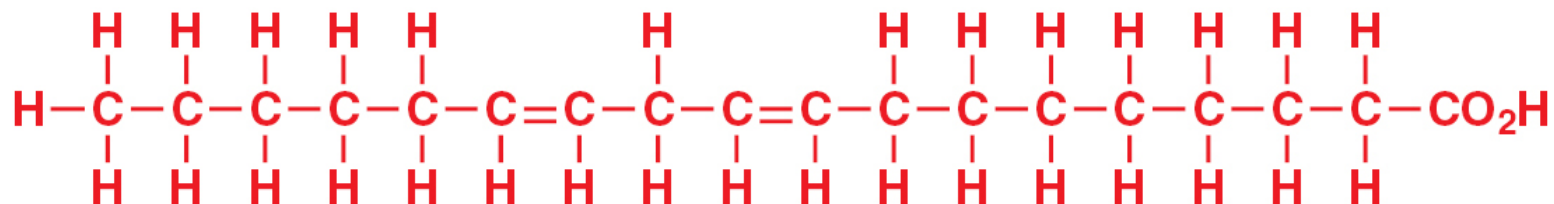
methionine
also has a sulfide
functional group



linoleic acid

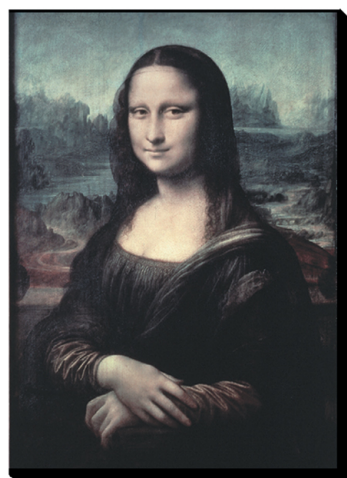


linoleic acid

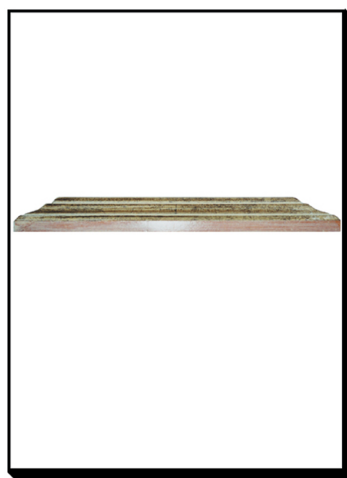


linoleic acid

tre modi per disegnare la stessa struttura molecolare



1

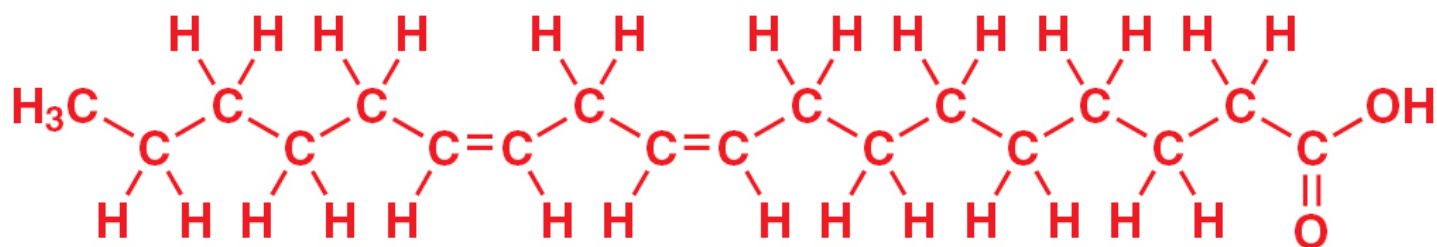


2

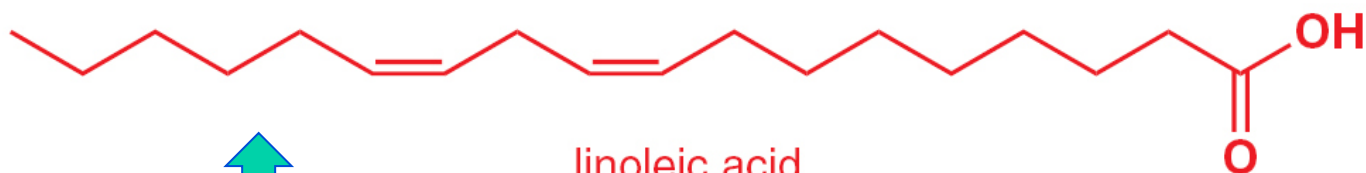


3

nel disegnare una struttura molecolare ricorda di seguire criteri di essenzialità e chiarezza



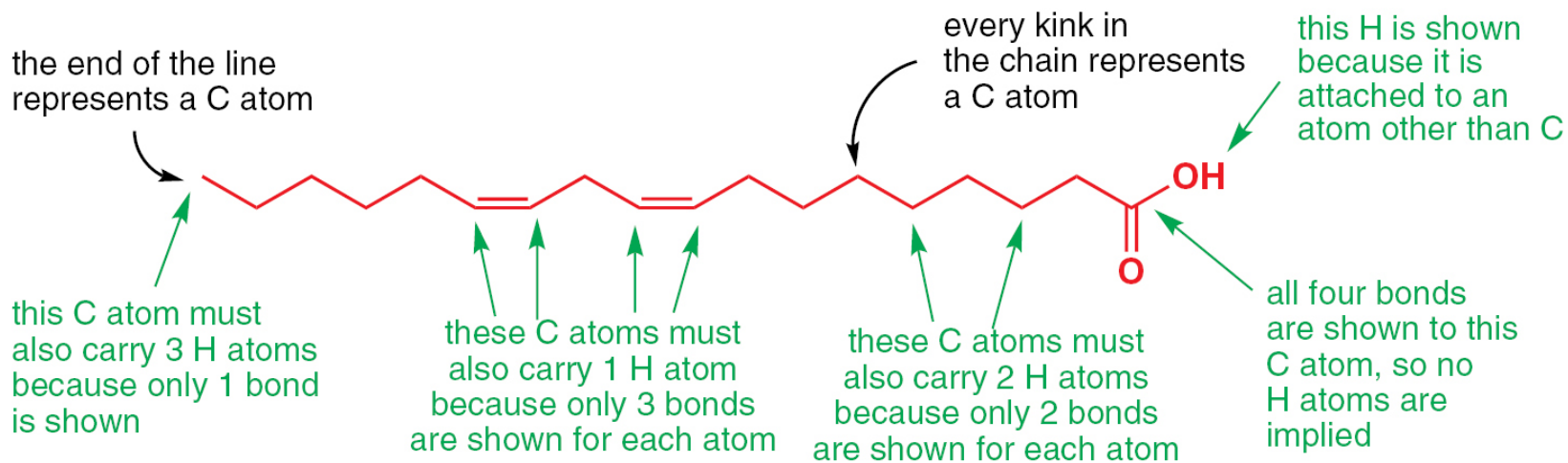
linoleic acid

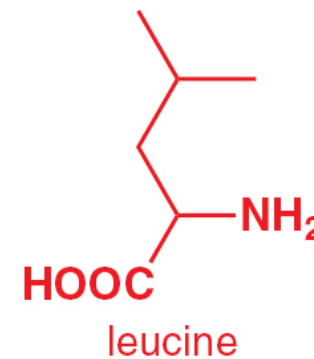
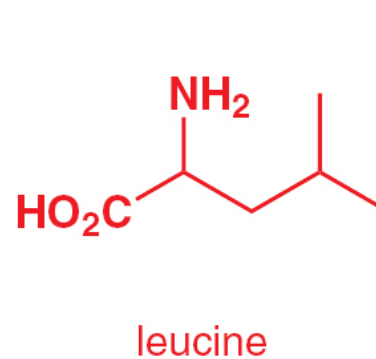
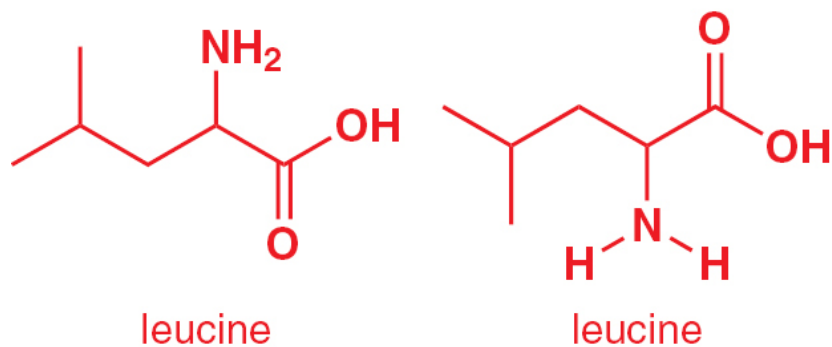


linoleic acid

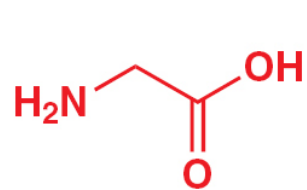


struttura chiara ed essenziale

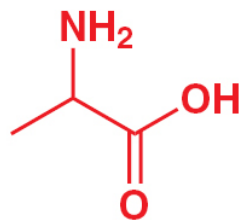




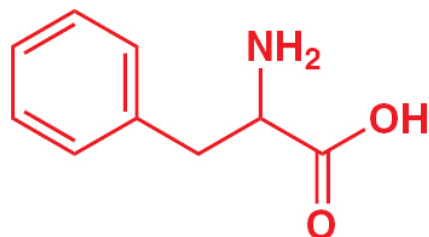
stesso amminoacido rappresentato da 4 strutture equivalenti (a parte l'orientazione e il modo con cui sono esplicitati i gruppi funzionali)



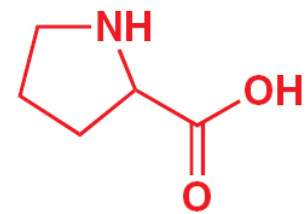
glycine



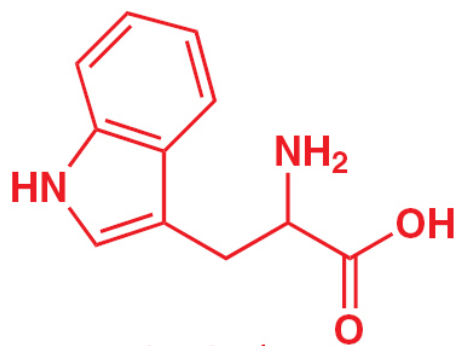
alanine



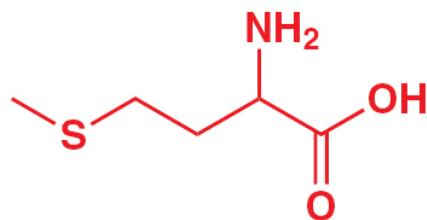
phenylalanine



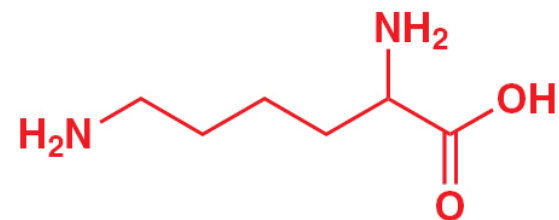
proline



tryptophan

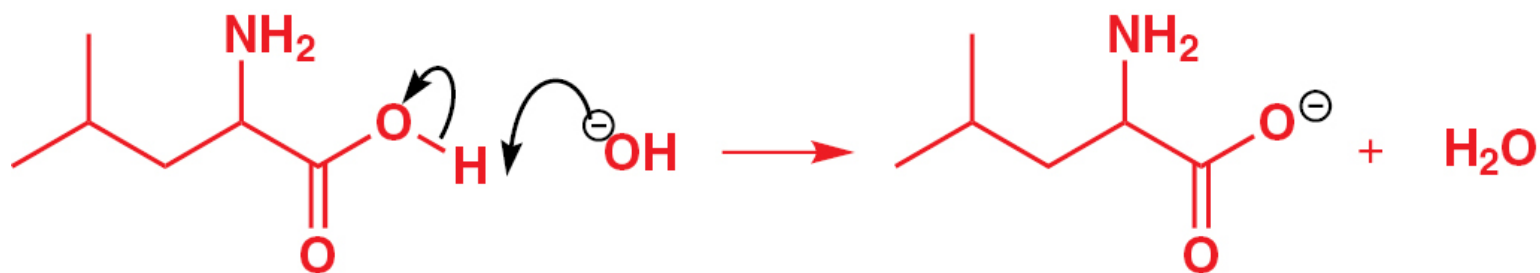


methionine

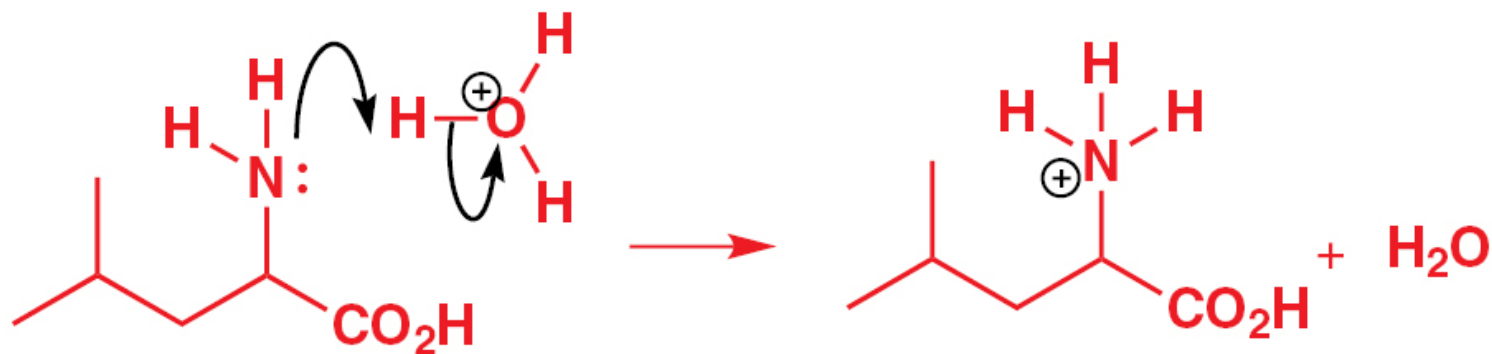


lysine

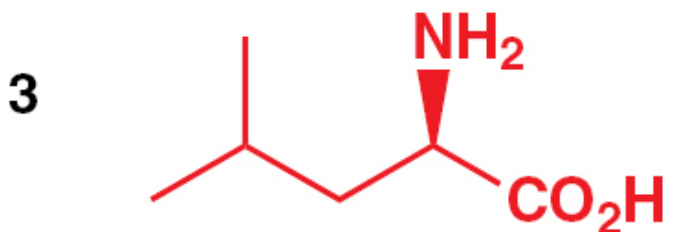
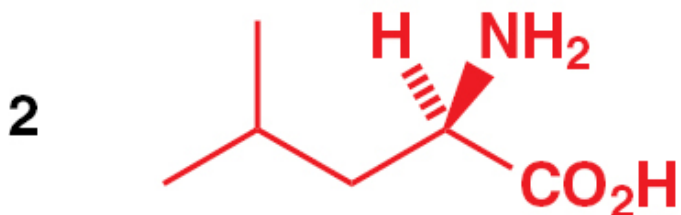
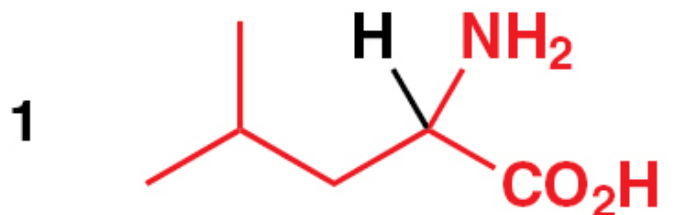
**amminoacidi disegnati in modo chiaro ed
essenziale con i gruppi funzionali evidenti**



una reazione acido-base tra il gruppo funzionale acido carbossilico e l'anione ossidrillico: il legame O-H viene evidenziato per chiarezza (la carica negativa viene inclusa in un cerchietto affinché sia maggiormente in evidenza)



una reazione acido-base tra il gruppo funzionale amminico e il catione idrossonio: il legame N-H viene evidenziato per chiarezza (la carica positiva viene inclusa in un cerchietto affinché sia maggiormente in evidenza)




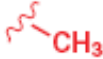

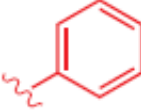

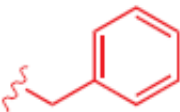




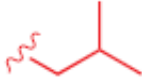


1. rappresentazione
“bidimensionale”

2) e 3) sono rappresentazioni che indicano la disposizione nello spazio dei sostituenti all'atomo di C che porta i gruppi funzionali amminico e carbossilico (triangolo pieno=verso di noi; tratteggio=dentro il piano del foglio; linea piena=sul piano del foglio)

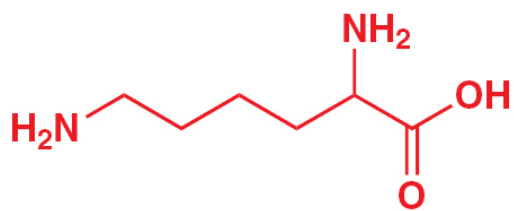


a section of the structure of polythene

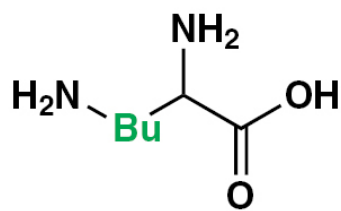
rappresentazione a zig-zag di una catena carboniosa

| | | | | | |
|------------------|-----------|-------------------------------------------------------------------------------------|-------------|------------|---------------------------------------------------------------------------------------|
| R | alkyl | | t-Bu | tert-butyl |  |
| Me | methyl |  | Ar | aryl | any aromatic ring |
| Et | ethyl |  | Ph | phenyl |  |
| Pr (n-Pr) | propyl |  | Bn | benzyl |  |
| Bu (n-Bu) | butyl |  | Ac | acetyl |  |
| i-Pr | isopropyl |  | | vinyl |  |
| i-Bu | isobutyl |  | | allyl |  |
| s-Bu | sec-butyl |  | X | halide | F, Cl, Br or I |

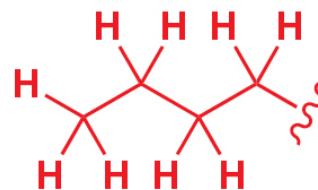
da ricordare



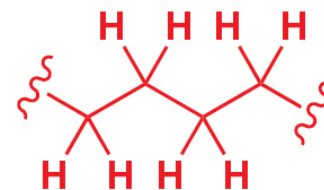
lysine



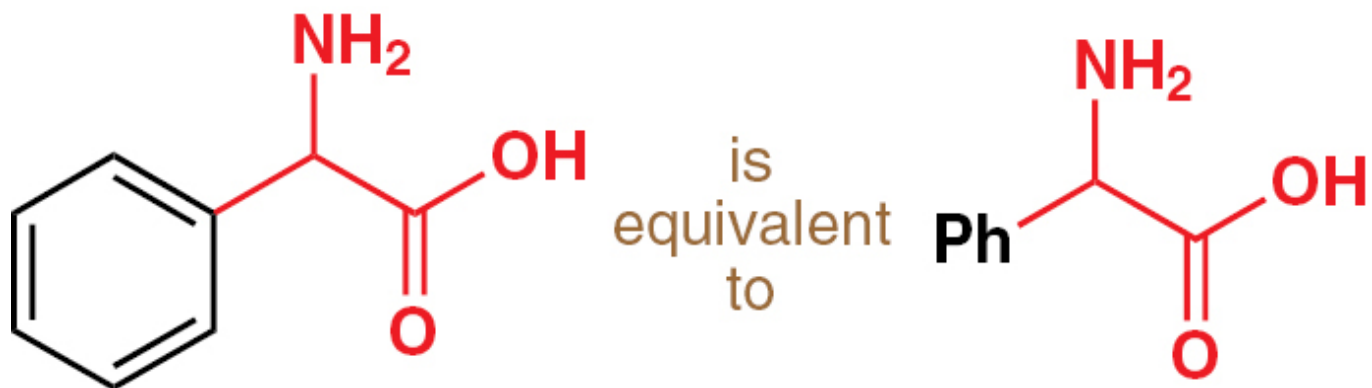
3 NOT CORRECT

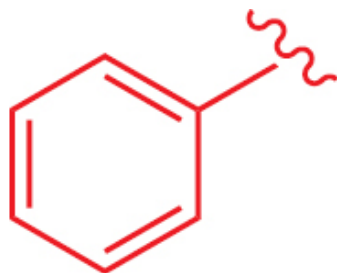


4 C₄H₉ = Bu

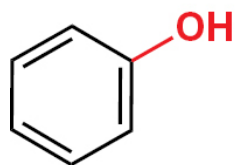


5 C₄H₈ NOT Bu

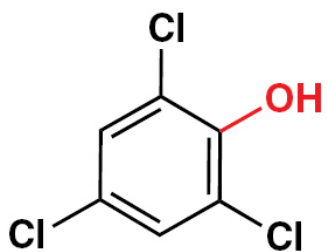




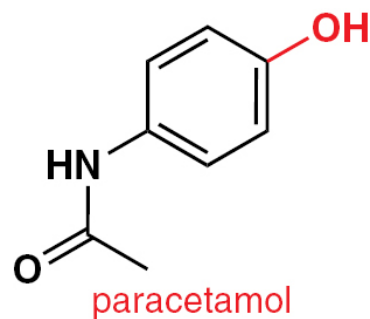
the phenyl group, Ph



PhOH =
phenol



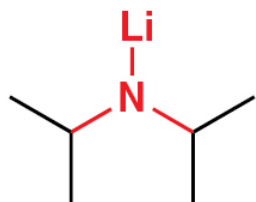
2,4,6-trichlorophenol



paracetamol

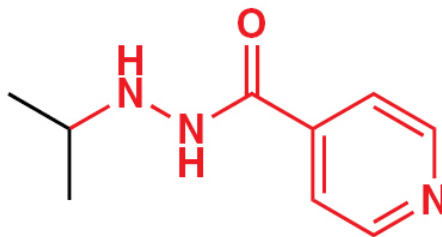
=



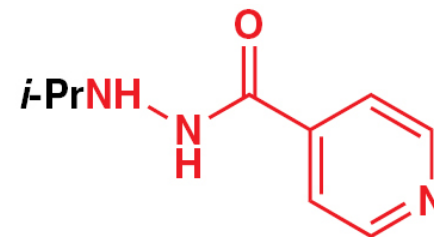


lithium diisopropylamide (LDA)

is equivalent to $\text{Li}i\text{-Pr}_2$

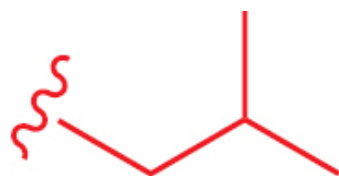


is equivalent to

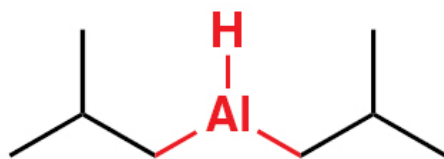


iproniazid

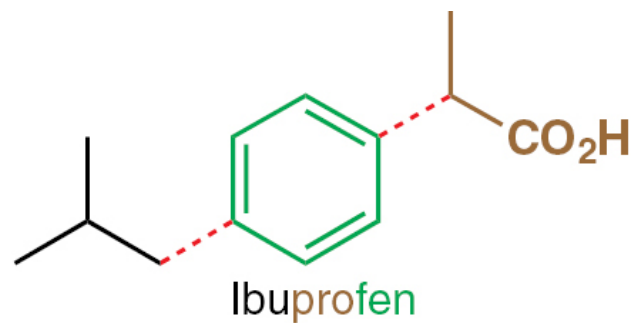
isopropile

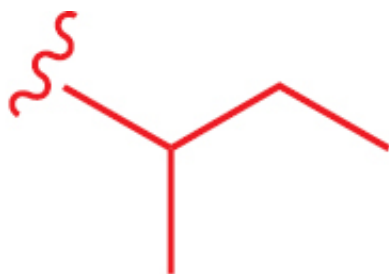


the isobutyl group
i-Bu

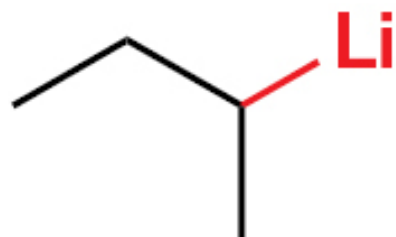


diisobutyl aluminium hydride (DIBAL)
is equivalent to **HAl*i*-Bu₂**





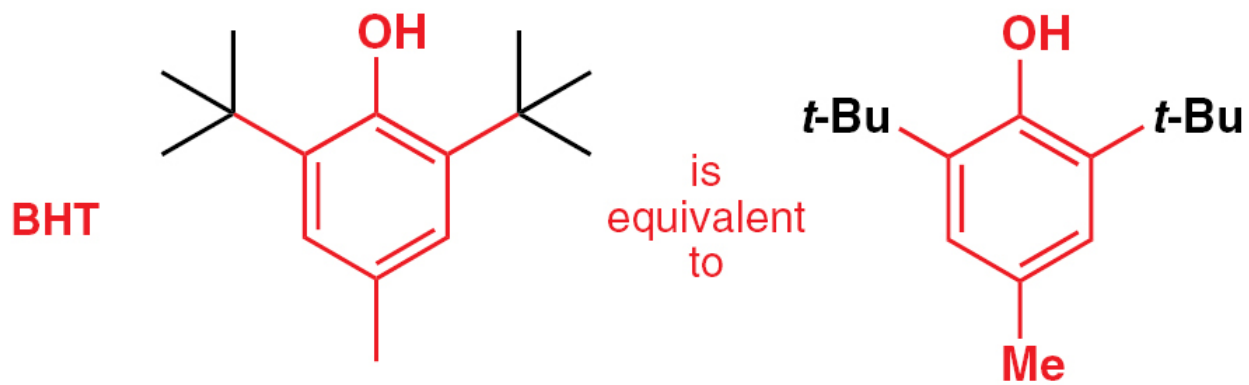
the *sec*-butyl group, *s*-Bu

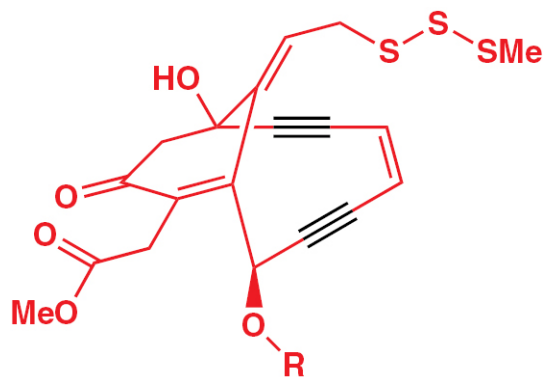


is equivalent to **s-BuLi**



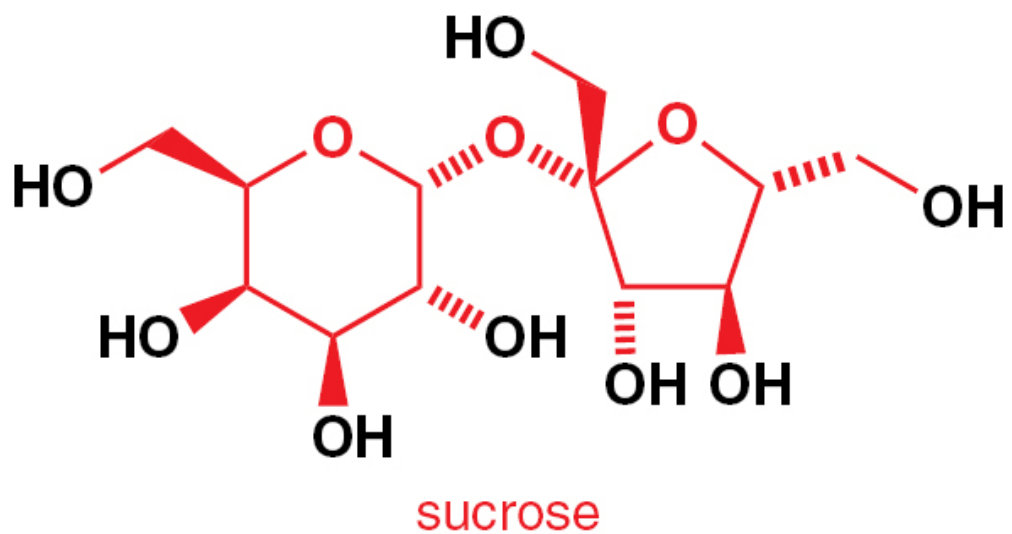
the *tert*-butyl
group *t*-Bu



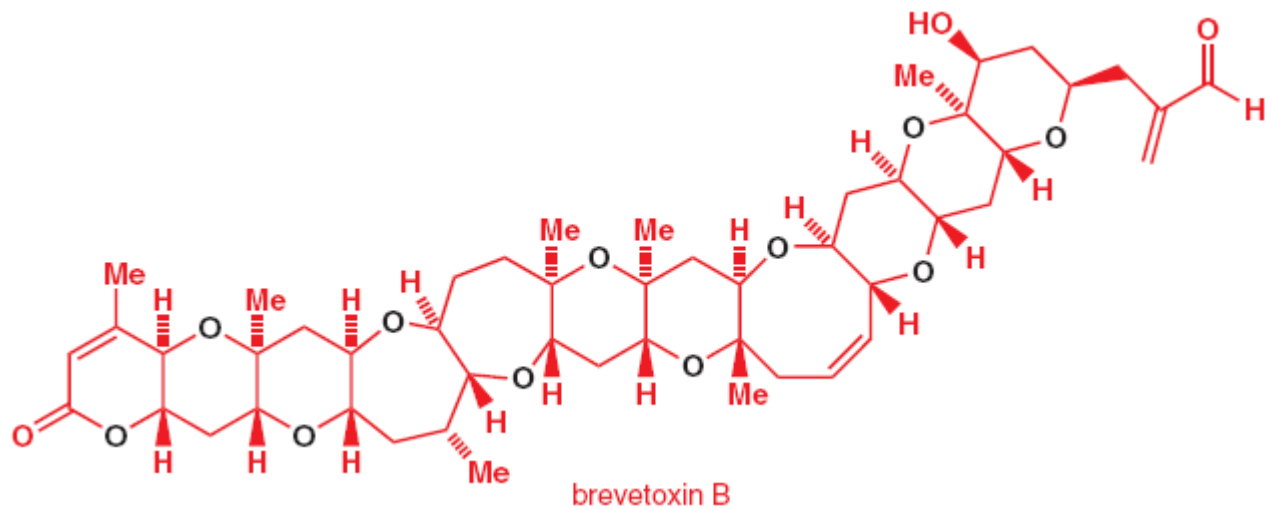


calicheamicin
(R = a string of sugar molecules)

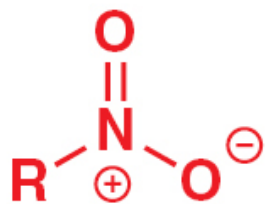
esercizio: riconosci i gruppi funzionali nella molecola



esercizio: riconosci i gruppi funzionali nella molecola

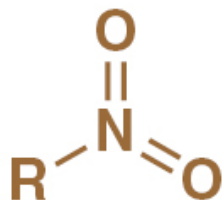


esercizio: riconosci i gruppi funzionali nella molecola

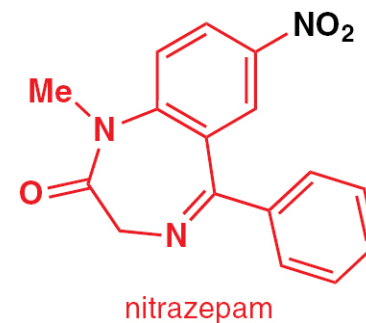
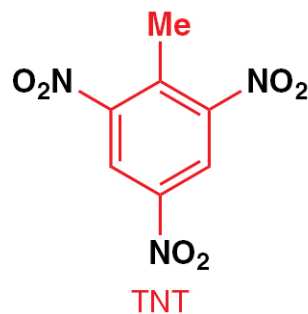


the nitro
group

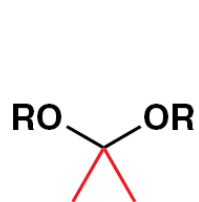
nitrogen cannot
have five bonds!



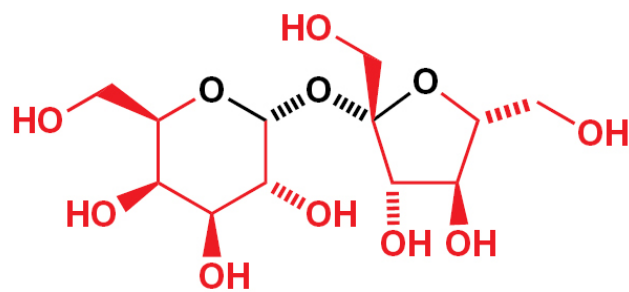
incorrect structure
for the nitro group



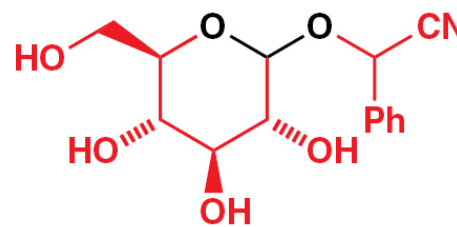
il gruppo nitro



an acetal

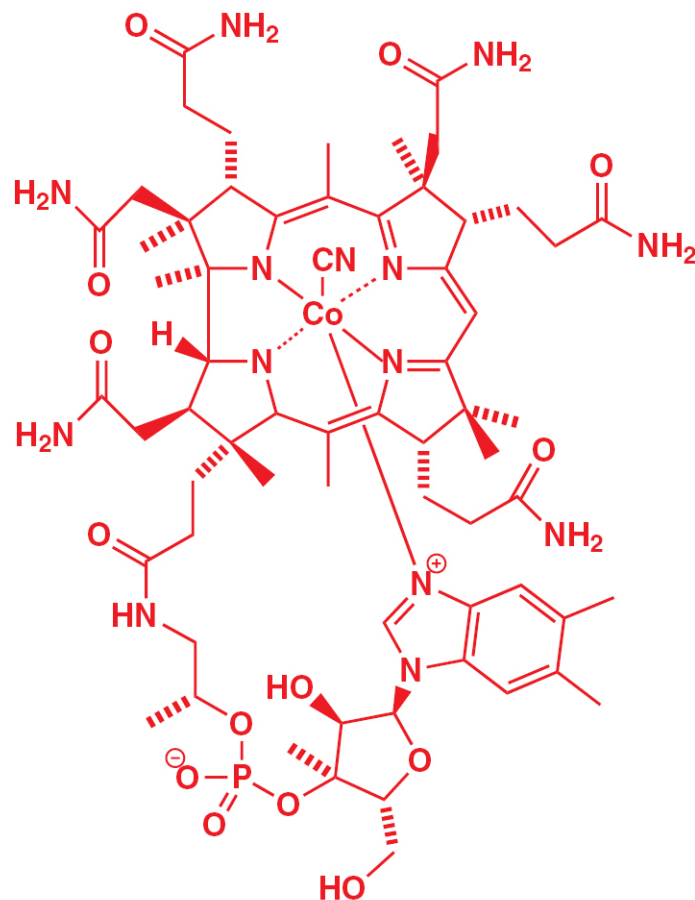


sucrose



laetrile

| Zero bonds to heteroatoms: alkane oxidation level | One bond to heteroatoms: alcohol oxidation level | Two bonds to heteroatoms: aldehyde oxidation level | Three bonds to heteroatoms: carboxylic acid oxidation level | Four bonds to heteroatoms: carbon dioxide oxidation level |
|---------------------------------------------------------------------------------------------------------------------------|-----------------------------------------------------|-------------------------------------------------------|----------------------------------------------------------------|---------------------------------------------------------------------------------------------------------------------------------|
| $\begin{array}{c} R^2 & R^3 \\ & \diagdown \quad / \\ & C \\ & / \quad \diagdown \\ R^1 & R^4 \end{array}$ <p>alkanes</p> | $R-CH_2-OH$ <p>alcohols</p> | $R-C(=O)H$ <p>aldehydes</p> | $R-C(=O)OH$ <p>carboxylic acids</p> | $O=C=O$ <p>carbon dioxide</p> |
| | $R-CH_2-OR^2$ <p>ethers</p> | $R^1-C(=O)R^2$ <p>ketones</p> | $R^1-C(=O)OR^2$ <p>esters</p> | $RO-C(=O)-OR$ <p>carbonates</p> |
| | $R-CH_2-NH_2$ <p>amines</p> | $R^3O-C(OR^3)R^2$ <p>acetals</p> | $R-C(=O)NH_2$ <p>amides</p> | $\begin{array}{c} F & F \\ & \diagdown \quad / \\ & C \\ & / \quad \diagdown \\ Cl & Cl \end{array}$ <p>tetrahalo compounds</p> |
| | $R-CH_2-Cl$ <p>alkyl halides</p> | $R-C\equiv C$ <p>alkynes</p> | $R-C\equiv N$ <p>nitriles</p> | |
| | $R-CH_2-Br$ <p>alkyl halides</p> | | $R-C(=O)Cl$ <p>acyl chlorides</p> | |
| | $R-CH_2-I$ <p>alkyl halides</p> | | | |
| | $R-CH=CH_2$ <p>alkenes</p> | | | |

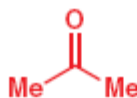


vitamin B₁₂, or...

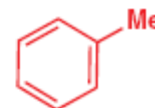
esercizio: riconosci i gruppi funzionali nella molecola

Important structures to learn

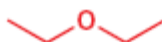
acetone



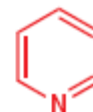
toluene



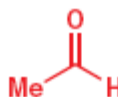
ether or diethyl-ether



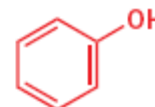
pyridine



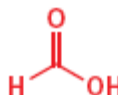
acetaldehyde



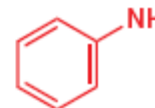
phenol



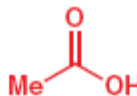
formic acid



aniline



acetic acid or AcOH



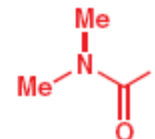
THF or tetrahydrofuran



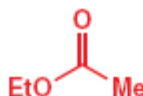
benzene



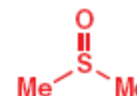
DMF, Me₂NCHO, or dimethylformamide



ethyl acetate or EtOAc

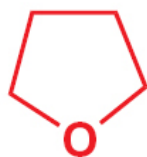


DMSO

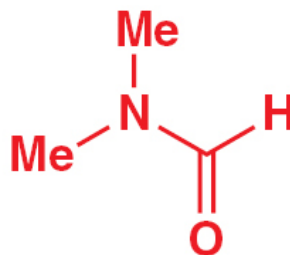


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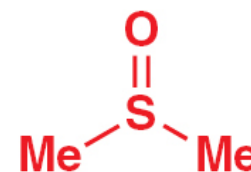
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THF
(TetraHydroFuran)



DMF
(DiMethylFormamide)



DMSO
(DiMethylSulfOxide)

solventi comuni