

MICHELE MAGGINI
DIPARTIMENTO DI SCIENZE CHIMICHE
VIA MARZOLO, 1

Ricevimento: MERCOLEDI' ore 16-17

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www.chimica.unipd.it/michele.maggini

**Informazioni sul corso, appunti di lezione:
sito Moodle STEM UniPD:**

<https://stem.elearning.unipd.it/course/index.php>

Esami: UNIWEB (<https://uniweb.unipd.it>)

Chimica Organica 1

SCIENZA DEI MATERIALI

a.a. 2023-2024

27 settembre 2023 - 19 gennaio 2024

64 ore di lezione

Mercoledì: 9.30-11.15

Giovedì: 10.30-11.15

Venerdì: 11.30-13.15

Aula L2, DiSC

2023

settembre						
lu	ma	me	gi	ve	sa	do
28	29	30	31	1	2	3
4	5	6	7	8	9	10
11	12	13	14	15	16	17
18	19	20	21	22	23	24
25	26	27	28	29	30	1
2	3	4	5	6	7	8

ottobre						
lu	ma	me	gi	ve	sa	do
25	26	27	28	29	30	1
2	3	4	5	6	7	8
9	10	11	12	13	14	15
16	17	18	19	20	21	22
23	24	25	26	27	28	29
30	31	1	2	3	4	5

novembre						
lu	ma	me	gi	ve	sa	do
30	31	1	2	3	4	5
6	7	8	9	10	11	12
13	14	15	16	17	18	19
20	21	22	23	24	25	26
27	28	29	30	1	2	3
4	5	6	7	8	9	10

dicembre						
lu	ma	me	gi	ve	sa	do
27	28	29	30	1	2	3
4	5	6	7	8	9	10
11	12	13	14	15	16	17
18	19	20	21	22	23	24
25	26	27	28	29	30	31
1	2	3	4	5	6	7

2024

gennaio						
lu	ma	me	gi	ve	sa	do
1	2	3	4	5	6	7
8	9	10	11	12	13	14
15	16	17	18	19	20	21
22	23	24	25	26	27	28
29	30	31	1	2	3	4
5	6	7	8	9	10	11

Giallo: vacanza
Arancione: lezione

L' ESAME DI CHIMICA ORGANICA 1

Sessione INVERNALE:

Sessione ESTIVA:

Sessione DI RECUPERO:

L'esame di CO1 consiste in una prova scritta e un colloquio orale. La prova scritta è costituita da 30 domande a scelta multipla e 5 domande aperte. Per ogni risposta giusta saranno corrisposti 3 punti, zero punti per ogni risposta non data e sarà tolto un punto per ogni risposta sbagliata. Il punteggio ottenuto sarà normalizzato a 30.

I miei corsi



Esami on-line >>



Portale video Mediaspace >>



ZOOM >>

Piattaforma Moodle della Macroarea STEM

Cerca corsi

I miei corsi



maggini_CO1_2023

SCQ2099499 - CHIMICA ORGANICA 1 2023-2024

Corso Impostazioni Partecipanti Valutazioni Report Altro ▾

▾ SCQ2099499 - CHIMICA ORGANICA 1 2023-2024 - PROF. MICHELE MAGGINI

Minimizza tutto

LEZIONI IN PRESENZA - AULA L2 (piano terra DiSC)

LT Scienza dei Materiali

Il anno, I semestre - a.a. 2023-2024

mercoledì dalle 9.30 alle 11.15

giovedì dalle 10.30 alle 11.15

venerdì dalle 11.30 alle 13.15

Ricevimento studenti: mercoledì dalle 16 alle 17 su appuntamento (secondo piano, stanza 7, edificio 170)

tel.: 049 8275662

email: michele.maggini@unipd.it




Annunci



Informazioni sull'insegnamento di CO1



Programma del corso 

Il corso di chimica organica_1 ha lo scopo di fornire i concetti fondamentali sulla struttura, nomenclatura e reazioni di alcuni composti organici monofunzionali. E' necessario aver seguito il corso di chimica generale e inorganica per affrontare e portare a termine proficuamente il corso di introduzione alla chimica organica.

Testi consigliati:

P. Yurkanis Bruice: Chimica Organica, Edises, III edizione - 2017

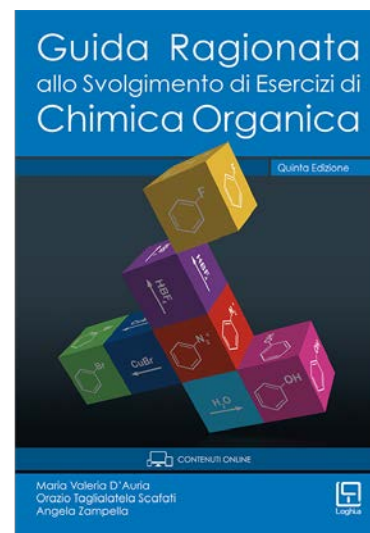
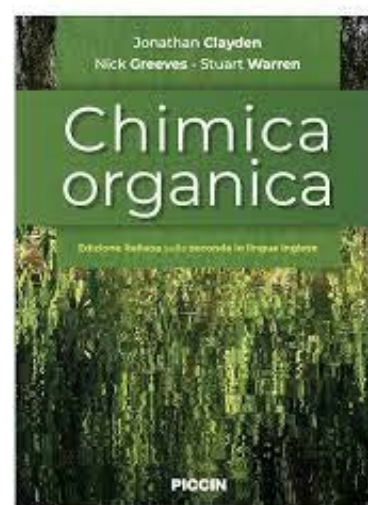
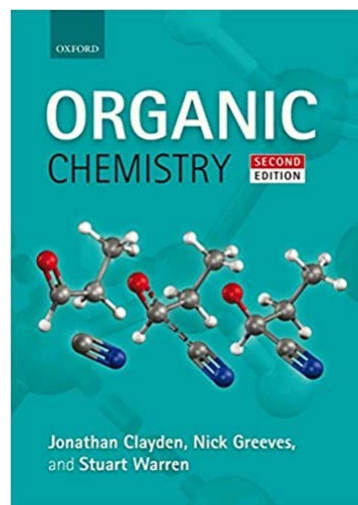
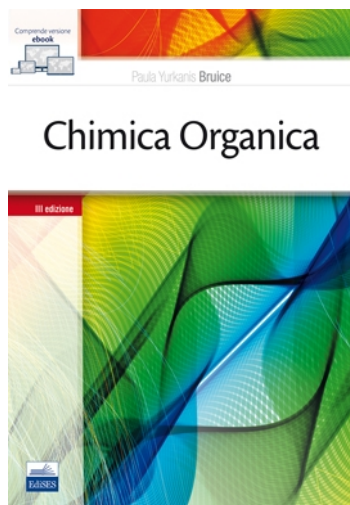
J. Clayden, N. Greeves, S. Warren: Organic Chemistry, 2nd Edition – March 2012
(traduzione italiana, Chimica Organica ed. Piccin, maggio 2023)

W.H. Brown, B.L. Iverson, E.V. Anslyn, C.S. Foote: Chimica Organica, Edises VI edizione – 2020.

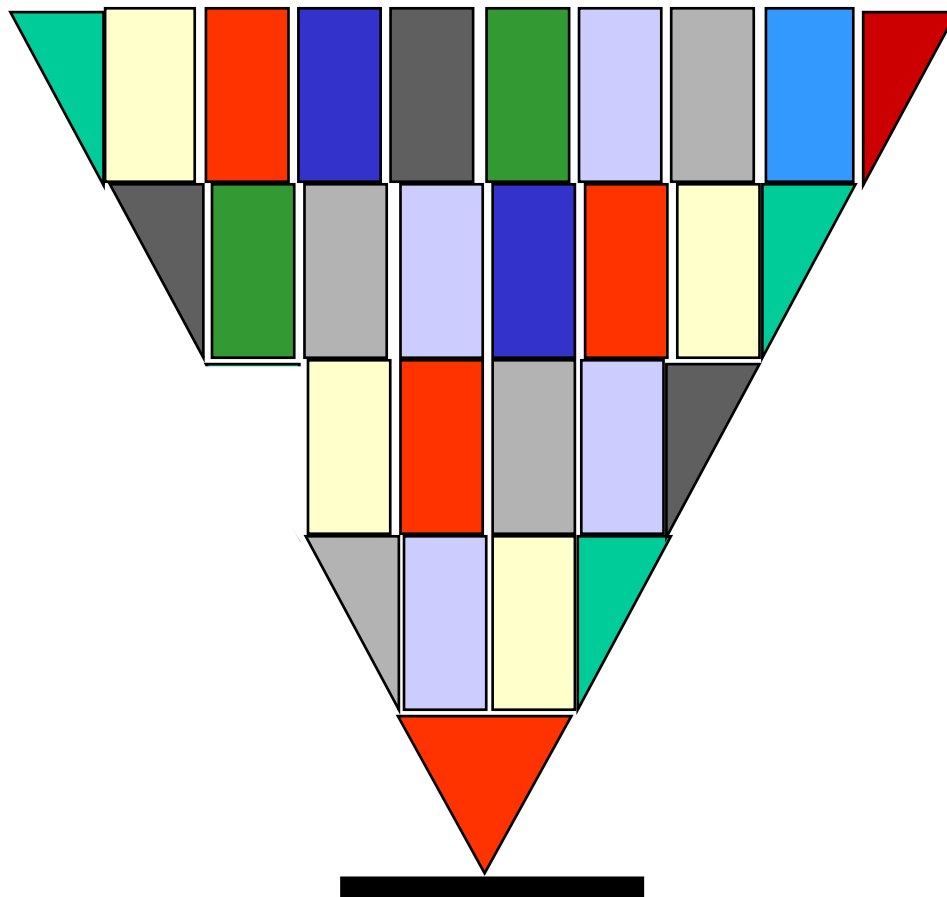
Esercizi

D'Auria, Tagliatela Scafati, Zampella: Guida Ragionata allo Svolgimento di Esercizi di Chimica Organica (ISBN 9788895122519) - V edizione, Loghia Editore

B. L. Iverson, S. Iverson: Guida alla soluzione dei problemi da "Chimica Organica" di Brown, Iverson, Anslyn, Foote, Edises, ISBN: 9788879598965, edizione 4 - 2016

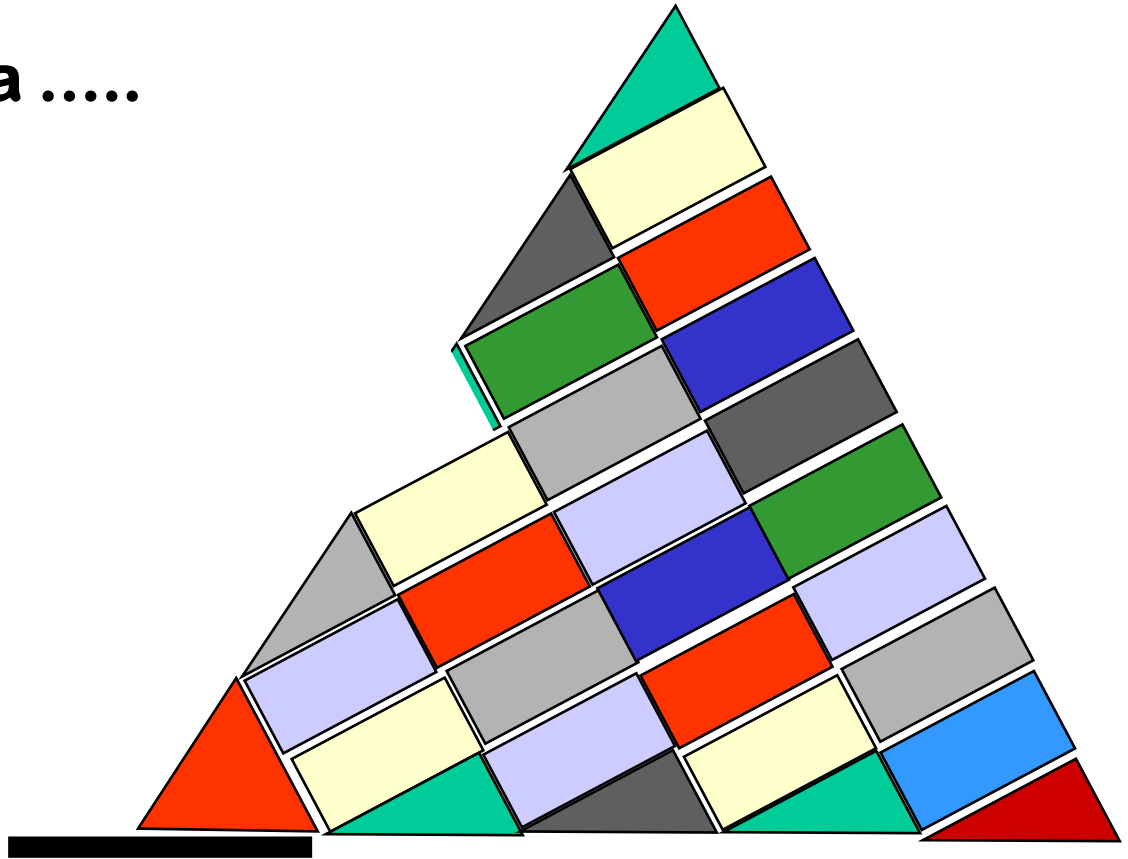


LA CONOSCENZA IN CHIMICA ORGANICA



Se manca qualcosa

**L'intera struttura
collassa !!!**



NON STARE INDIETRO !

Sebbene il materiale iniziale sia semplice e per molti di voi costituisca un ripasso, esso potrebbe diventare complesso molto velocemente



i contenuti di questo corso sono cumulativi

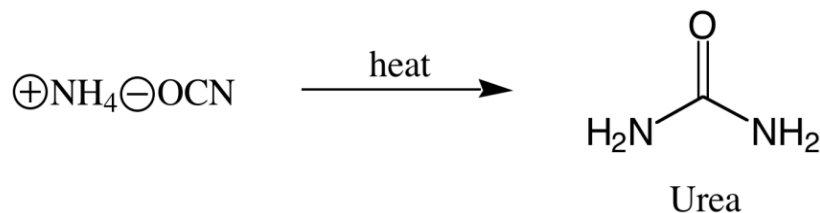
le nuove conoscenze partiranno dai concetti appena appresi

il consiglio è di non lasciare lo studio della chimica organica esclusivamente alla fine del corso

per superare l'esame scritto è necessario saper utilizzare e collegare le conoscenze

Cos'è la chimica organica

- organico: forza speciale negli organismi viventi
- cambio di paradigma: sintesi dell'urea (Wöhler, 1828)

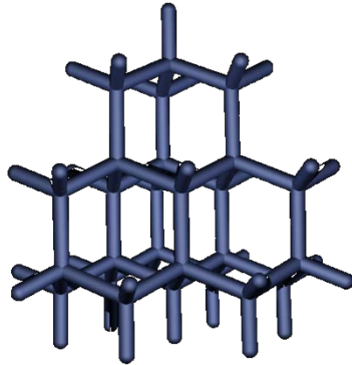


- chimica organica: chimica del carbonio*

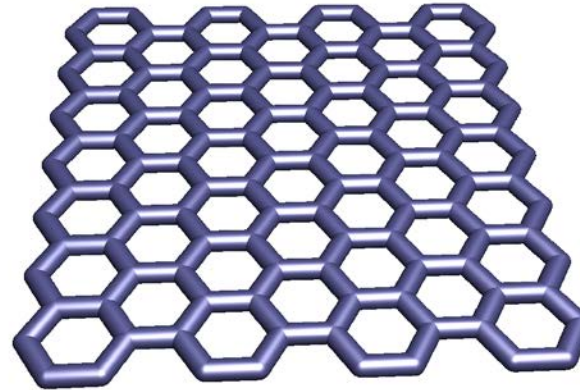
*...ma non solo; ci sono solo poche forme di puro carbonio



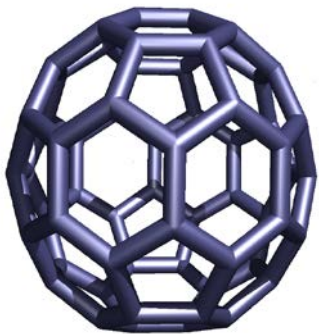
diamante



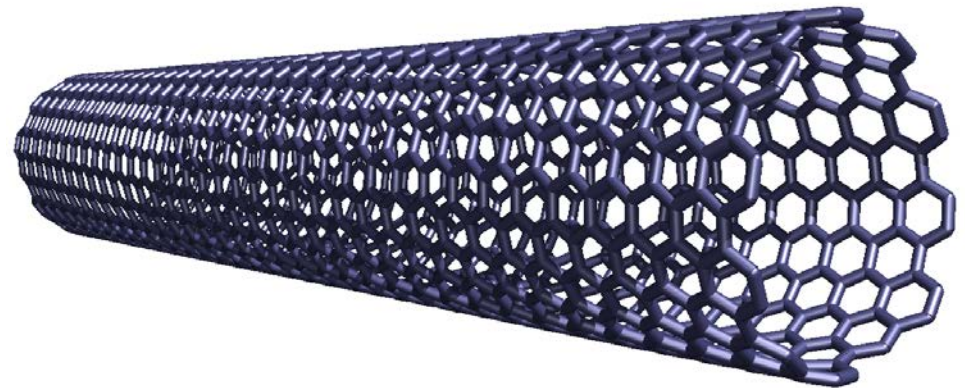
grafene



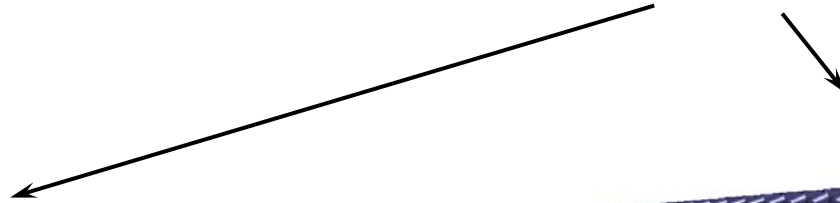
grafite



fullereni



Nanotubi di carbonio



Molecular

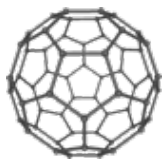
Particellar

1D

2D

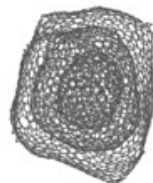
3D

C_{60}



Smalley, R. E. *et al.*
Nature
1985, 318, 162

Nano-onions



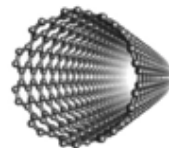
Ugarte, D.
Nature
1992, 359, 707

Nano-diamond



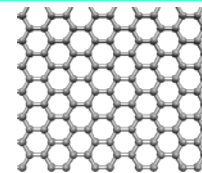
Greiner, N. R. *et al.*
Nature
1990, 343, 244

SWCNTs



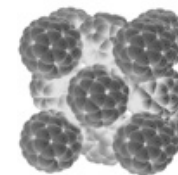
Ichihashi, T. *et al.*
Nature
1993, 363, 603

Graphene



Novoselov, K. S. *et al.*
Science
2004, 306, 666

Fullerite



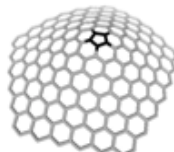
Thiel, F. A. *et al.*
Nature
1991, 351, 380

C_{70}



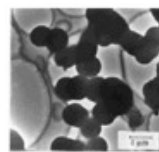
Whetten, R. L. *et al.*
J. Phys. Chem.
1990, 94, 8630

Nano-cones



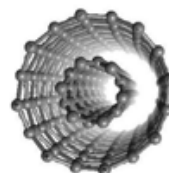
Sattler, K. *et al.*
Chem. Phys. Lett.
1994, 220, 192

Nano-beads



Zhao, X. L. *et al.*
Carbon
1998, 36, 507

DWCNTs



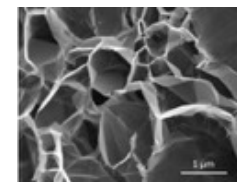
Zakharov, D. N. *et al.*
Carbon
2001, 39, 761

Graphene nanodots



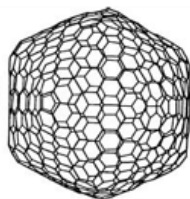
Novoselov, K. S. *et al.*
Science
2008, 320, 356

Graphene nanoplatelets



Thiel, F. A. *et al.*
Nature
1991, 351, 380

C_{540}



Kroto, H.
Pure Appl. Chem.
1990, 62, 407

Nano-horns



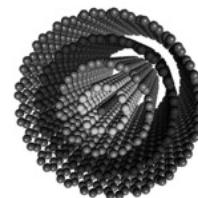
Iijima, S. *et al.*
Chem. Phys. Lett.
2000, 321, 514

Nano-dots



Scrivens, W. A. *et al.*
J. Am. Chem. Soc.
2004, 126, 12736

MWCNTs



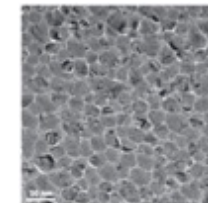
Iijima, S.
Nature
1991, 354, 56

Graphene nanoribbons



Dresselhaus, M. S. *et al.*
Phys. Rev. B
1996, 54, 17954

Nanocrystalline diamond films



Gruen, D. M. *et al.*
Annu. Rev. Mater. Sci.
1999, 29, 211

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<table border="1"> <tr> <td>58</td><td>59</td><td>60</td><td>61</td><td>62</td><td>63</td><td>64</td><td>65</td><td>66</td><td>67</td><td>68</td><td>69</td><td>70</td><td>71</td> </tr> <tr> <td>Ce</td><td>Pr</td><td>Nd</td><td>Pm</td><td>Sm</td><td>Eu</td><td>Gd</td><td>Tb</td><td>Dy</td><td>Ho</td><td>Er</td><td>Tm</td><td>Yb</td><td>Lu</td> </tr> <tr> <td>90</td><td>91</td><td>92</td><td>93</td><td>94</td><td>95</td><td>96</td><td>97</td><td>98</td><td>99</td><td>100</td><td>101</td><td>102</td><td>103</td> </tr> <tr> <td>Th</td><td>Pa</td><td>U</td><td>Np</td><td>Pu</td><td>Am</td><td>Cm</td><td>Bk</td><td>Cf</td><td>Es</td><td>Fm</td><td>Md</td><td>No</td><td>Lr</td> </tr> </table>																		58	59	60	61	62	63	64	65	66	67	68	69	70	71	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu	90	91	92	93	94	95	96	97	98	99	100	101	102	103	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr
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Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr																																																												

Numero atomico

6

C

Carbonio
12,011

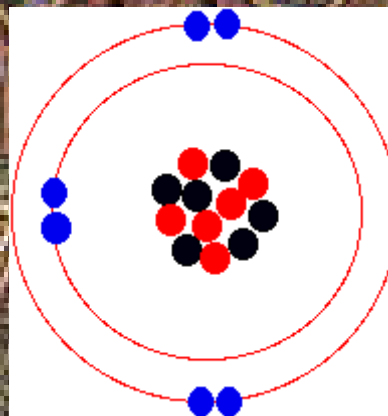
[He] 2s² 2p²

Configurazione
elettronica

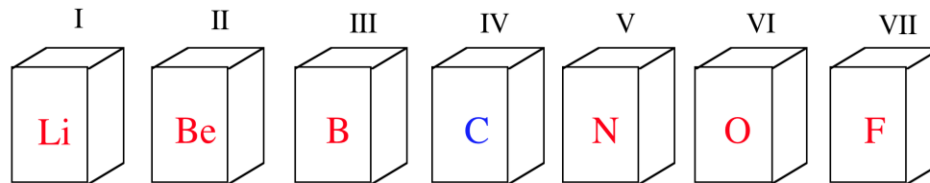
Simbolo atomico

Nome dell'elemento

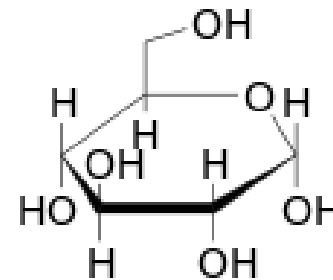
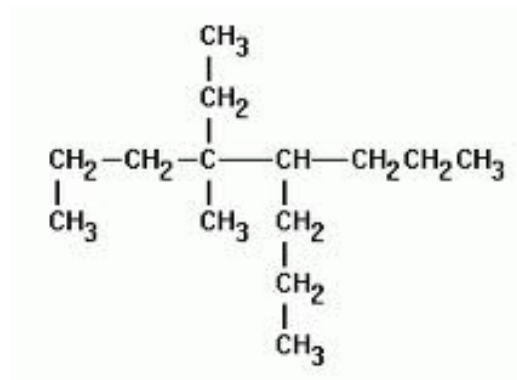
Peso atomico



- la posizione del C nel mezzo del II periodo della TP permette di condividere elettroni con molti elementi e fare legami (**ci interessa molto cosa fanno gli elettroni e come si comportano**)



- il C è uno dei pochi elementi che forma **catene**; può formare **sistemi ramificati e anelli**



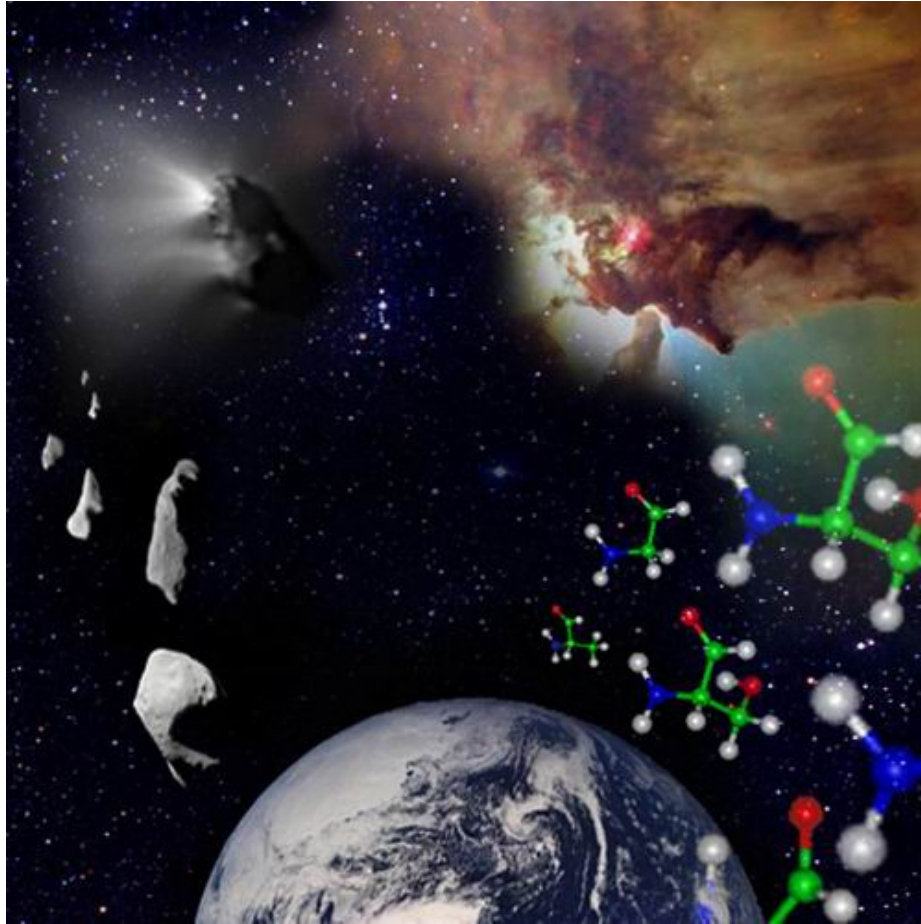
LA GRANDE VARIETA' DEL CARBONIO

Per il fatto che il C può concatenarsi e ramificare
esistono MILIONI di composti a base di carbonio

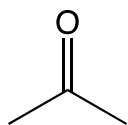
La grande varietà e molteplicità dei derivati rende il carbonio un elemento su cui basare complicati sistemi viventi.

Si possono identificare molecole specifiche per codificare ogni processo biochimico

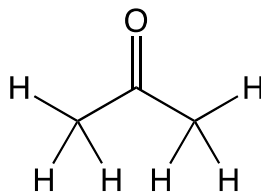
spazio chimico : 10^{60} composti



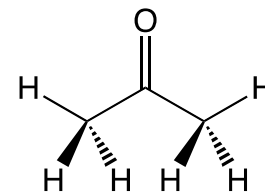
comprendere la chimica organica attraverso la conoscenza dei legami tra gli atomi nelle molecole e la loro disposizione spaziale, e del movimento degli elettroni che li formano.



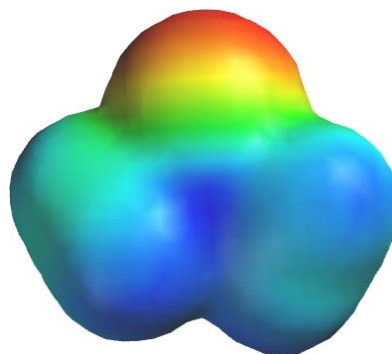
Shorthand drawing



All atoms indicated

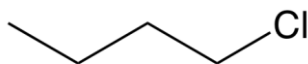
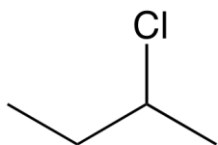


3-dimensional
drawing
wedge and dash
lines have meaning

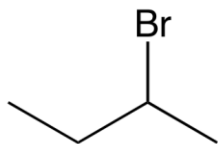
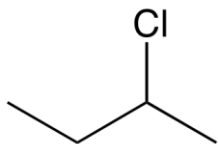


Electron density plot

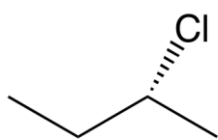
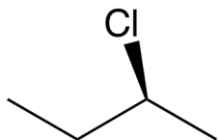
le reazioni organiche si possono anticipare attraverso la comprensione della disposizione e della natura dei legami che si rompono e si formano nelle molecole interagenti



stessa formula C_4H_9Cl ma reattività differente



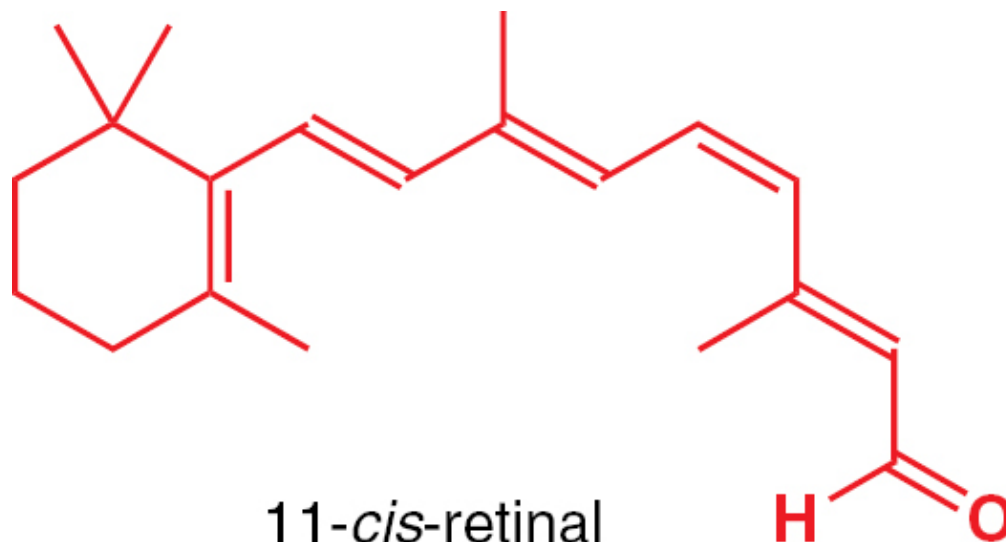
cambia l'alogeno, cambia la reattività



stessa struttura ma differente disposizione spaziale dell'alogeno (stereochemica)

Cos'è la chimica organica

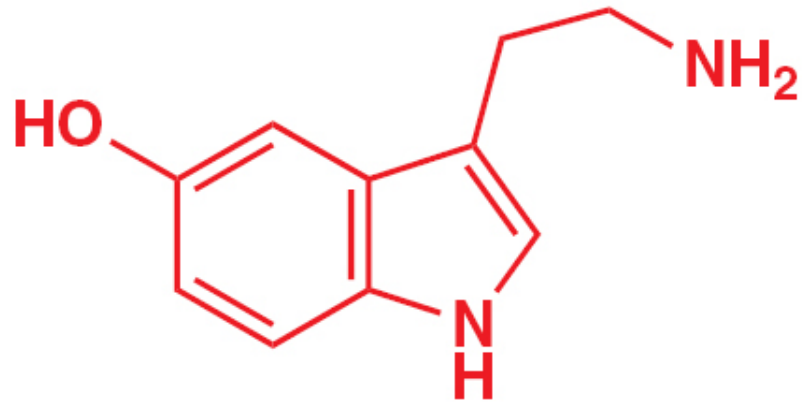
siamo tutti chimici organici



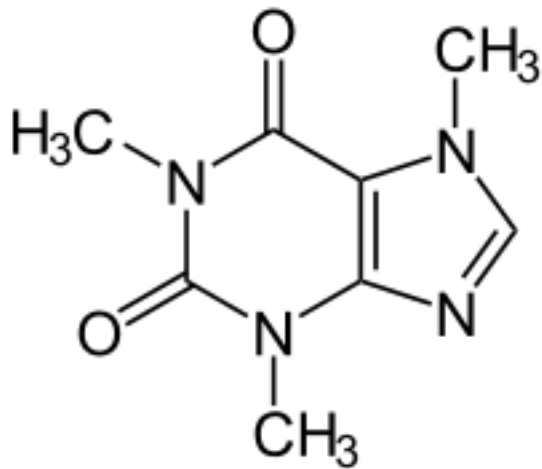
11-*cis*-retinal
absorbs light and allows vision

Nonostante la grande biocomplessità, alla base della natura umana ci sono meccanismi di fondo relativamente semplici. Il flusso di informazioni tra l' inconscio (il corpo) e il conscio (il cervello) è portato avanti da interruttori molecolari regolati da **interazioni deboli** e **variazioni conformazionali**.

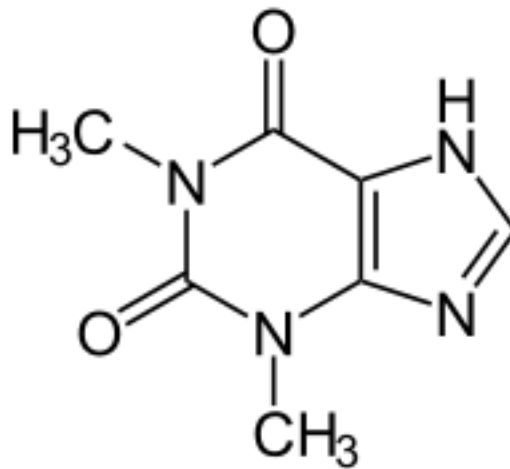
L' intero sistema **EMOZIONALE-SENSORIO-INTELLETTUALE** che noi definiamo umano ha una **base molecolare**.



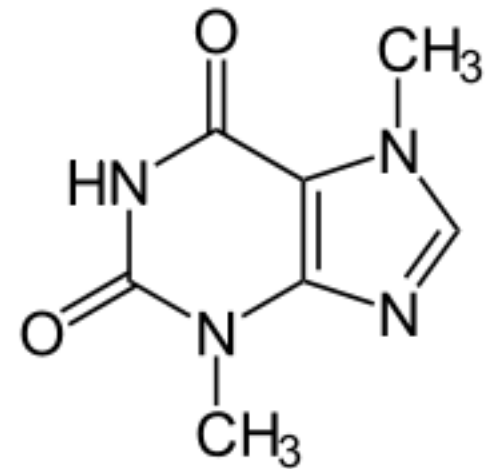
serotonin
human neurotransmitter



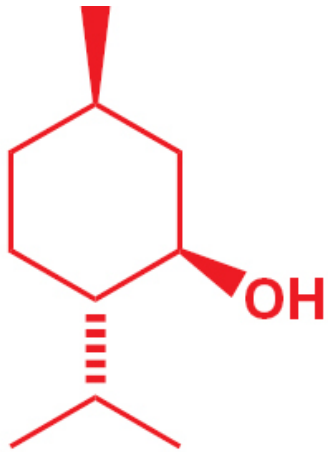
Caffeina



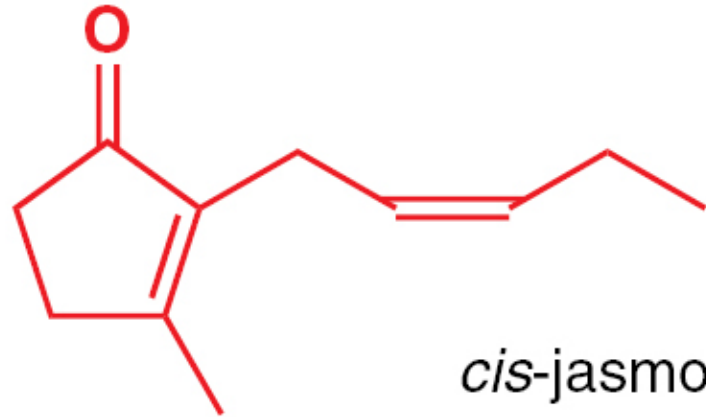
Teofillina



Teobromina

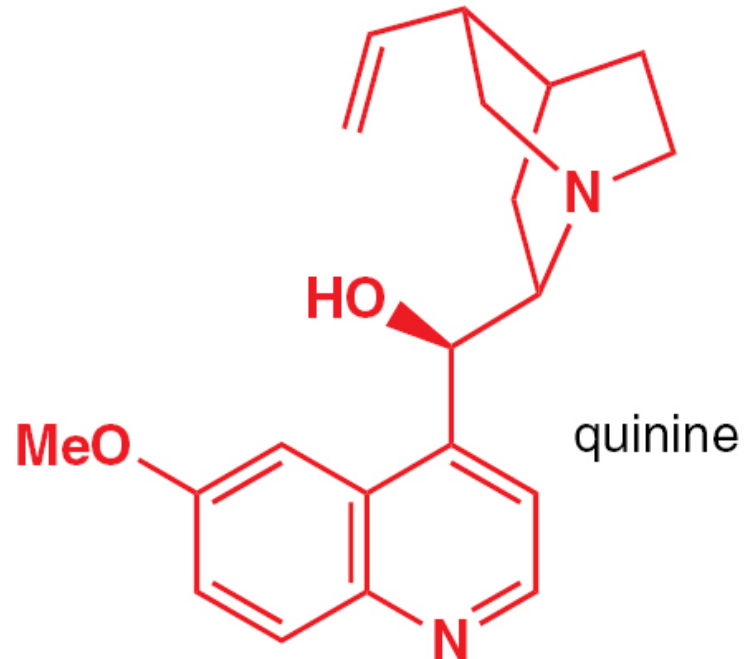


menthol



cis-jasmone

composti organici di origine naturale

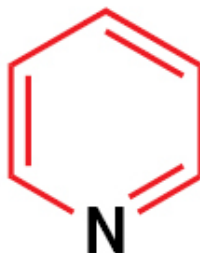


quinine

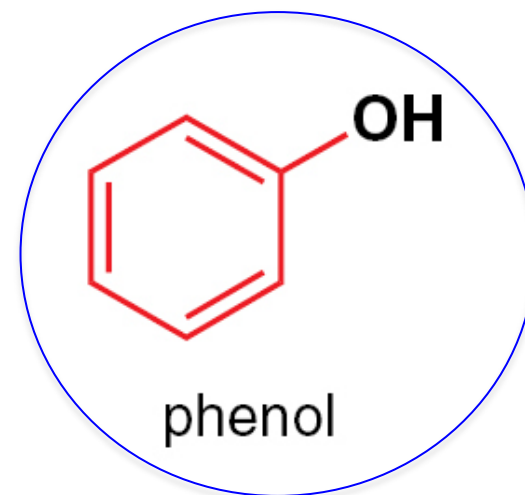
Residui della distillazione del carbone (litantrace)



benzene

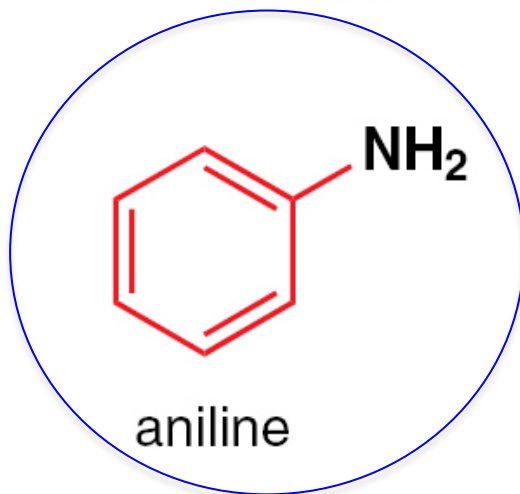


pyridine



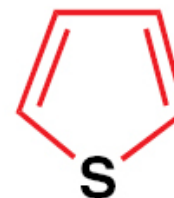
phenol

disinfettante



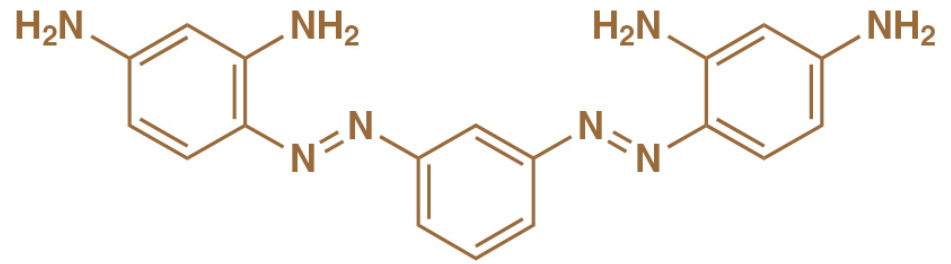
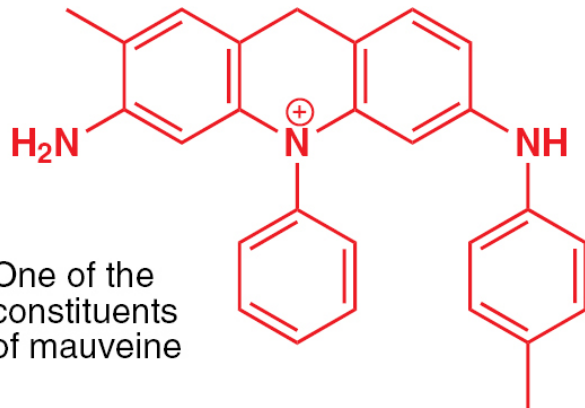
aniline

**coloranti
organici**



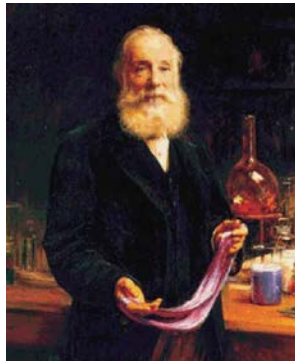
thiophene

composti organici non naturali



Bismarck Brown Y

primo colorante sintetico

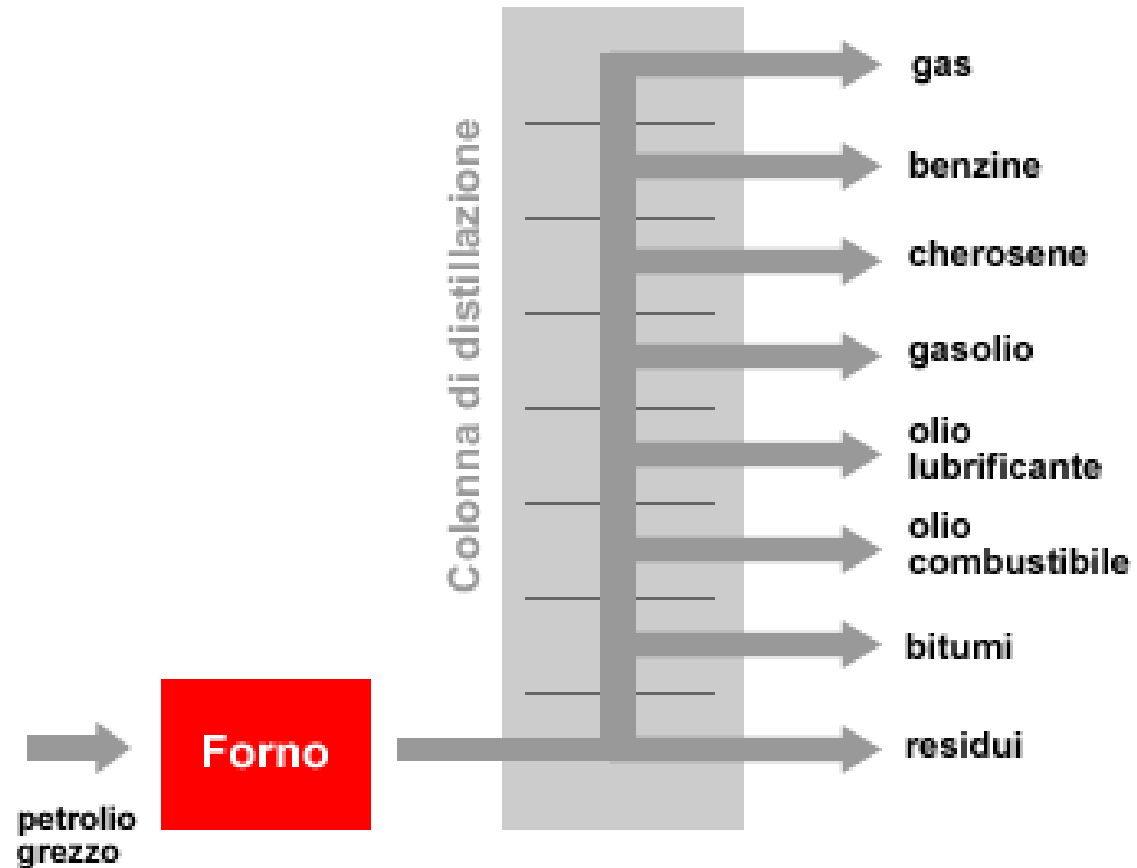


mauveina, porpora di anilina







maggini_CO1_2023

petrolio

Torre di frazionamento



Colore

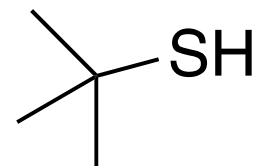
Colour	Description	Compound	Structure
	red	dark red hexagonal plates	3-methoxycycloheptatriene-2-one
	orange	amber needles	dichlorodicyanoquinone (DDQ)
	yellow	toxic yellow explosive gas	diazomethane
	green	green prisms with a steel-blue lustre	9-nitrosojulolidine
	blue	deep blue liquid with a peppery smell	azulene
	purple	deep blue gas condensing to a purple solid	nitrosotrifluoromethane

Odore

skunk spray contains:



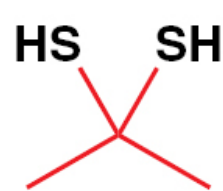
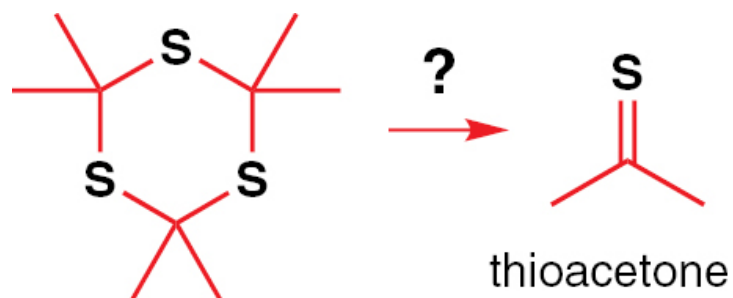
the scent of the black truffle



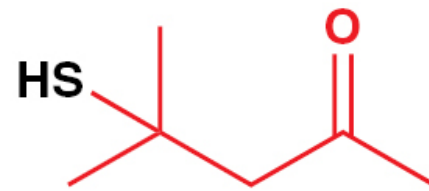
tert-butil tiolo

1 parte su
50 000 000 000
di gas naturale

odore e diluizione

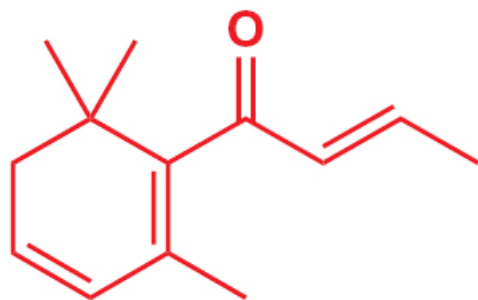


propane-
dithiol



4-methyl-4-
sulfanylpentan-2-one

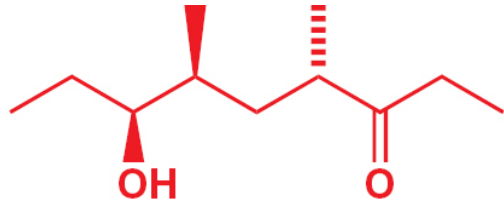
two candidates for
the worst smell in the world
(no-one wants to find the winner)



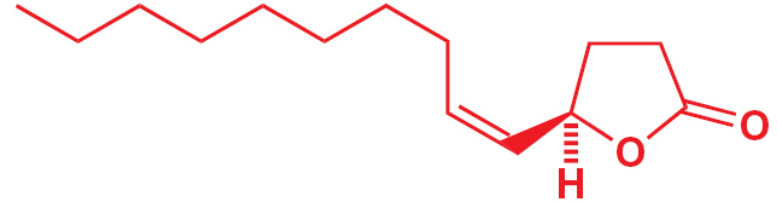
damascenone—the smell of roses



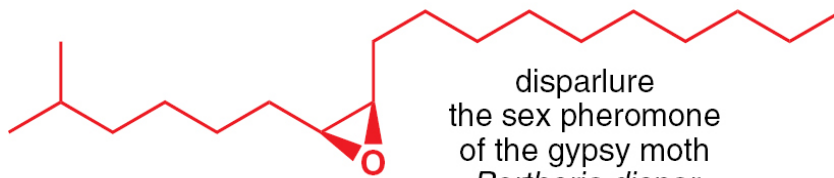
Insetti e trappole chimiche



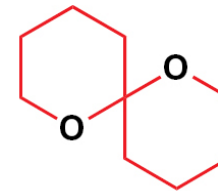
serricornin
the sex pheromone of the cigarette beetle
Lasioderma serricorne



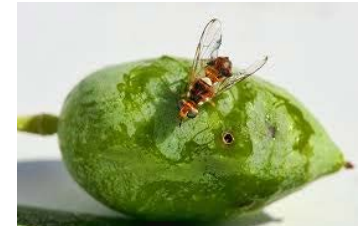
japonilure
the sex pheromone of the Japanese beetle
Popilia japonica



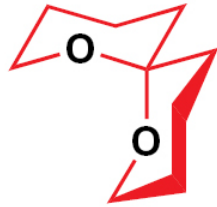
disparlure
the sex pheromone
of the gypsy moth
Portheria dispar



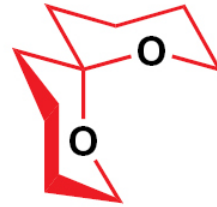
olean
sex pheromone
of the olive fly
Bacrocera oleae



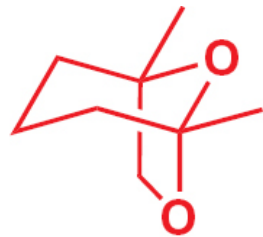
Simmetria e feromoni



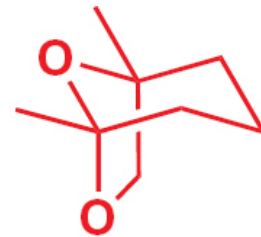
this mirror image
isomer attracts
male olive flies



this mirror image
isomer attracts
female olive flies



this mirror image
isomer smells of
young male elephant*

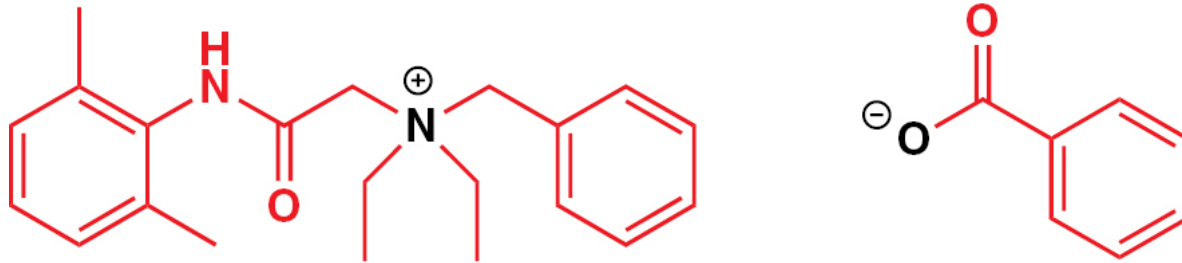


this mirror image
isomer smells of
old male elephant*

*if you are a
female elephant

Nature. 2005 Dec 22;438(7071):1097-8.

gusto

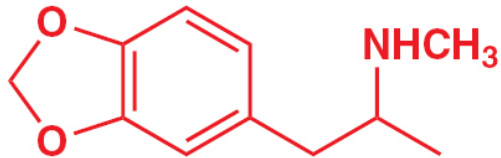


'denatonium benzoate', marked as Bitrex
benzyl-diethyl[(2,6-xylyl-carbamoyl)-methyl]ammonium benzoate

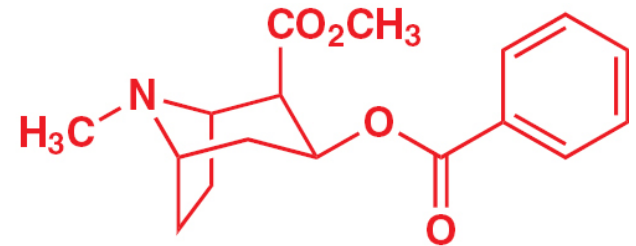




alcohol
(ethanol)

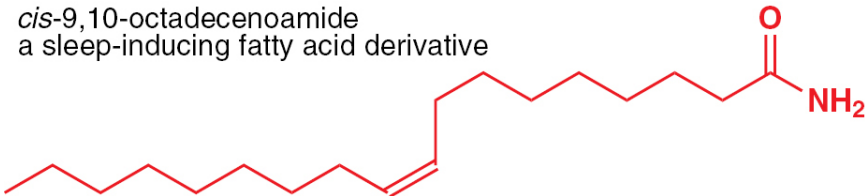


MDMA
(ecstasy)

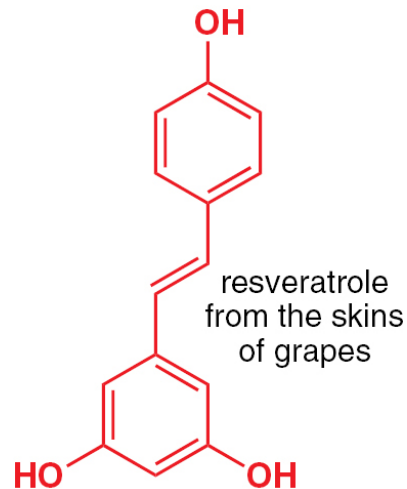
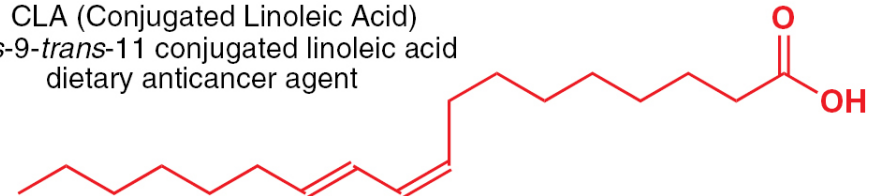


cocaine—an addictive alkaloid

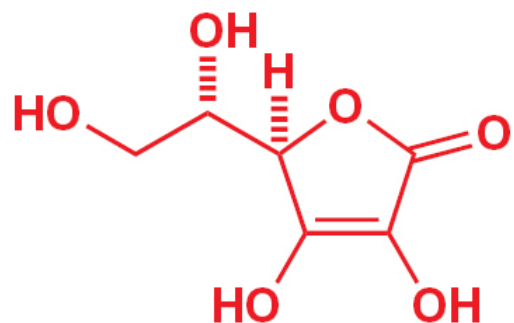
cis-9,10-octadecenoamide
a sleep-inducing fatty acid derivative



CLA (Conjugated Linoleic Acid)
cis-9-*trans*-11 conjugated linoleic acid
dietary anticancer agent

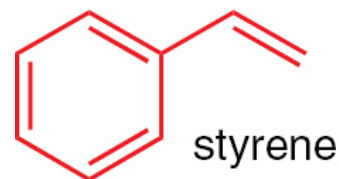


chimica organica e industria

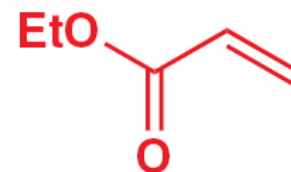


vitamin C (ascorbic acid)

monomers for polymer manufacture



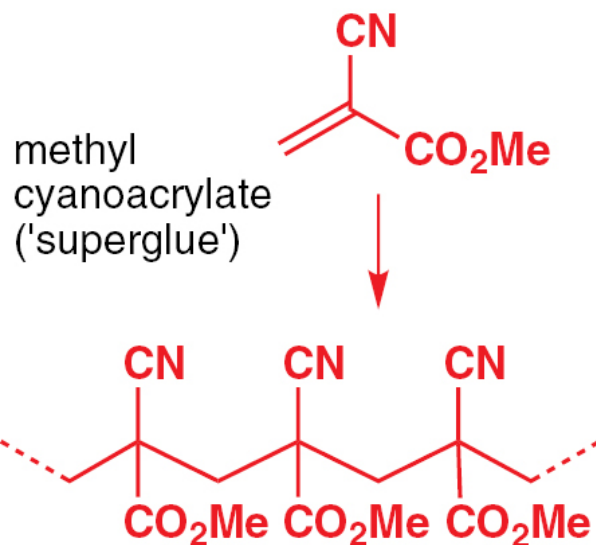
styrene



ethyl acrylate



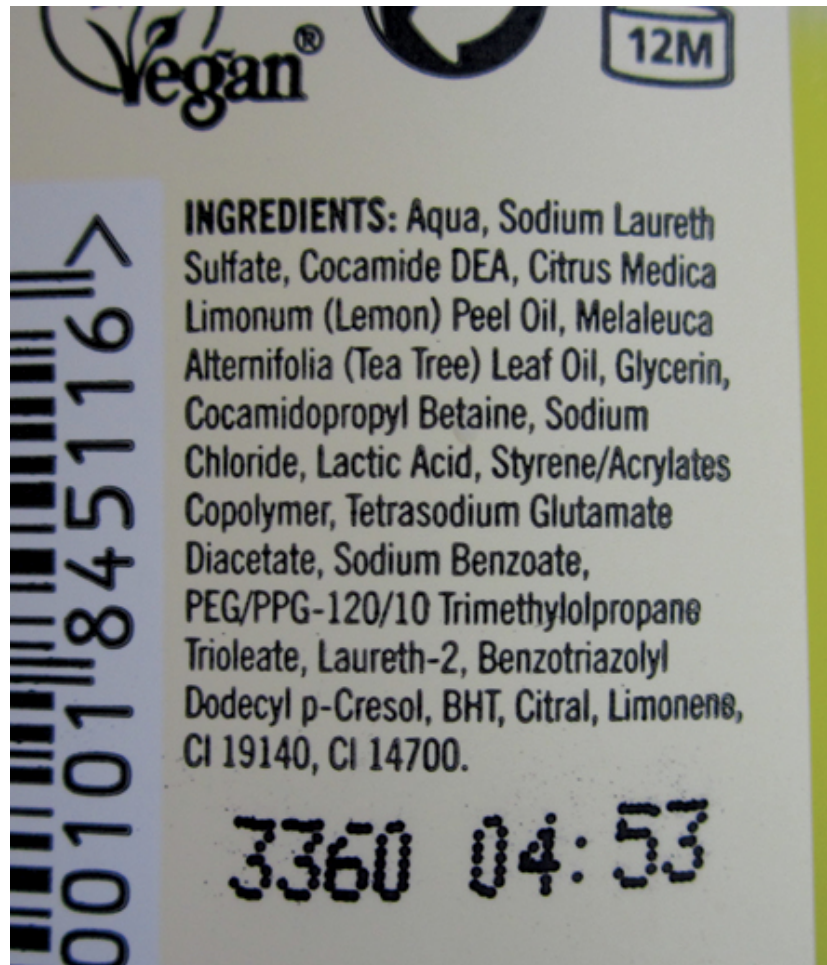
vinyl chloride

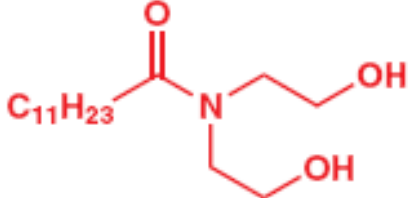




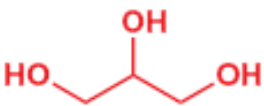
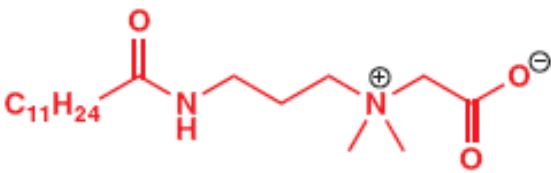
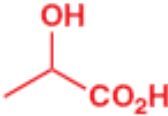
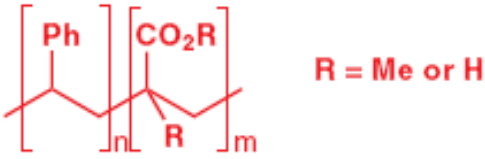
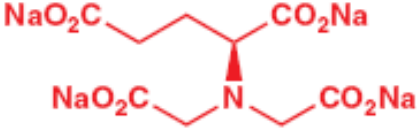
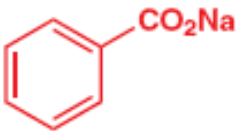


Giulio Natta e il polipropilene (1903-1979)

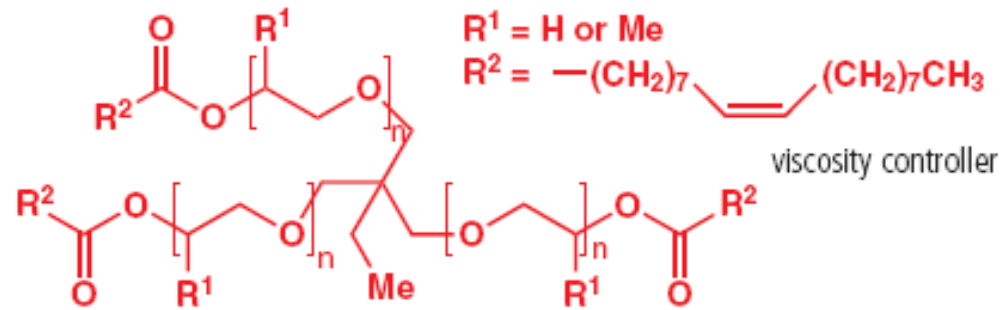
doccia gel



Ingredient	Chemical meaning	Purpose
aqua	$\text{H}-\text{O}-\text{H}$ water	solvent
sodium laureth sulfate	$\text{C}_{12}\text{H}_{25}(\text{O}-\text{CH}_2\text{CH}_2)_n\text{OSO}_3\text{Na}$ typically $n = 3$	detergent
cocamide DEA	 $\text{C}_{11}\text{H}_{23}-\text{C}(=\text{O})-\text{N}(\text{CH}_2\text{CH}_2\text{OH})_2$	foaming agent
<i>Citrus medica limonum</i> peel oil	mainly  α -pinene	scent, appeal to customer
<i>Melaleuca alternifolia</i> leaf oil	mainly  terpinen-4-ol	scent, appeal to customer, possibly antiseptic

glycerin	 <p>glycerol</p>	cosolvent; moisturizer; ensures smoothness
cocamidopropyl betaine		detergent and anti-electrostatic
sodium chloride	NaCl	control solubility of Na ⁺ -based detergents
lactic acid		acidifier
styrene acrylates copolymer	 <p>R = Me or H</p>	film former
tetrasodium glutamate diacetate		chelator, to prevent formation of insoluble scum in hard water
sodium benzoate		preservative

PEG/PPG -120/10
trimethylolpropane
trioleate

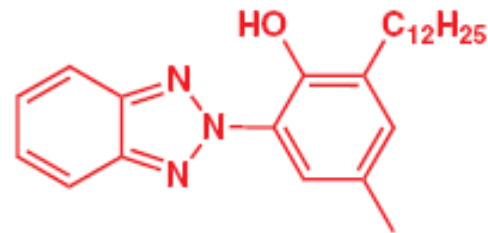


laureth-2



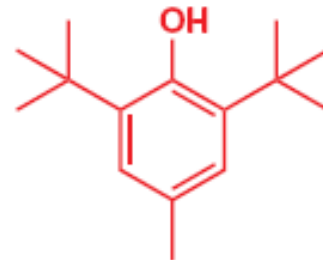
emulsifier

benzotriazolyl
dodecyl *p*-cresol



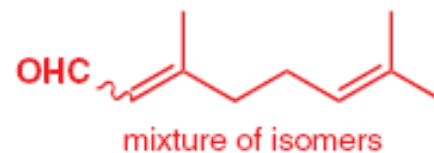
absorbs UV light

BHT

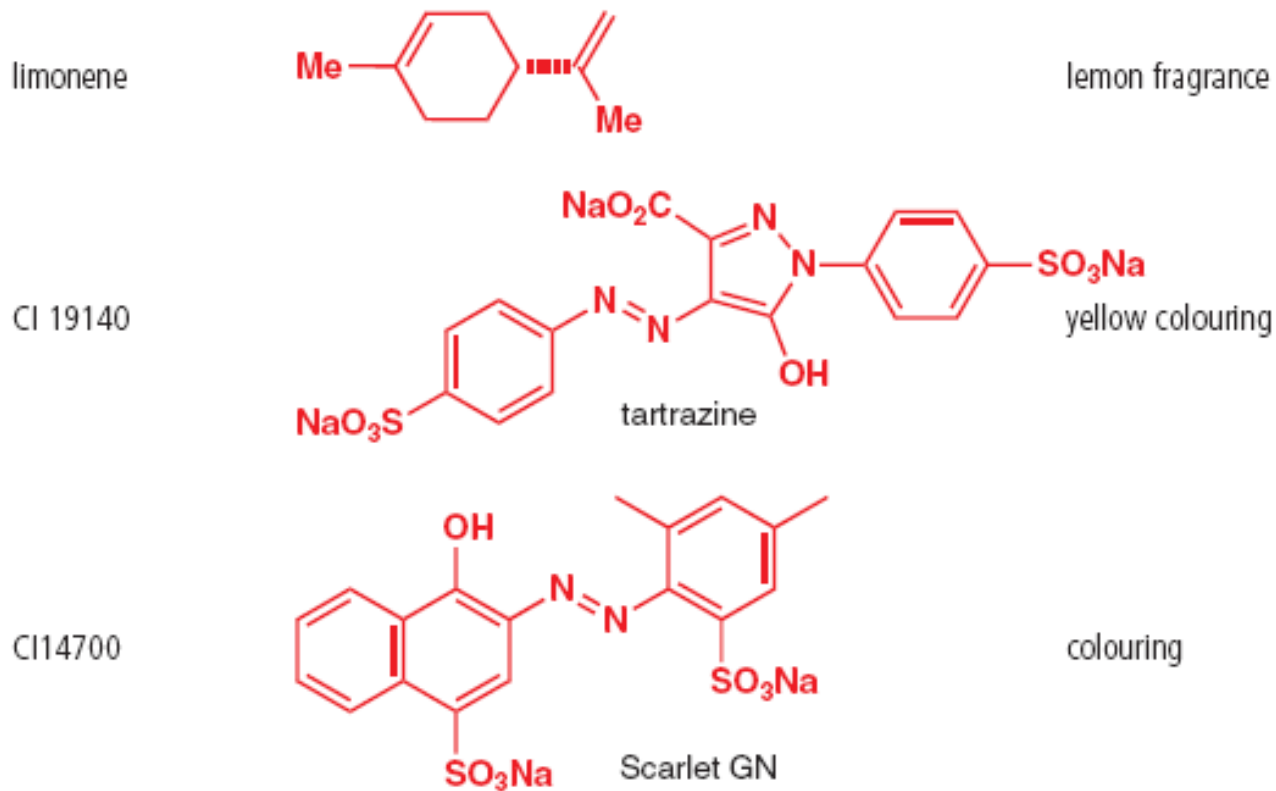


antioxidant

citral



lemon fragrance



l'etichetta dei docciaschiuma nella maggioranza dei casi evoca prodotti naturali e fragranze dalla natura.... in effetti è vero se pensiamo che le fonti naturali da cui derivano le molecole usate per preparare il docciaschiuma provengono dalle foreste carbonifere decomposte e intrappolate nel sottosuolo...

dentifrici



acqua (1), **carbonato di calcio** (2), **cloruro di sodio** (3), **glicerina** (4), **silice idrata** (5),
aroma (6), **idrossietilcellulosa** (7),
lauramide propil betaina (8), **saccarina** (9),
triclosan (10), **eugenolo** (11), **limonene**
(12), CI 77891 (13), CI 73015 (14).



acqua, silice, glicerina, carbonato: ca 70-80%

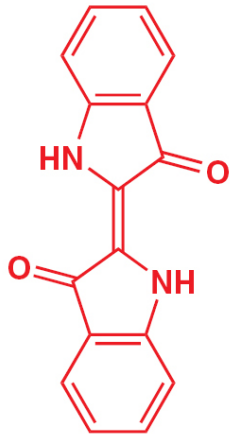
egizi

si lavavano i denti spalmando su di essi, con le dita, una miscela composta di ceneri polverizzate di zoccoli di bue, mirra, gusci d'uova polverizzati e bruciati

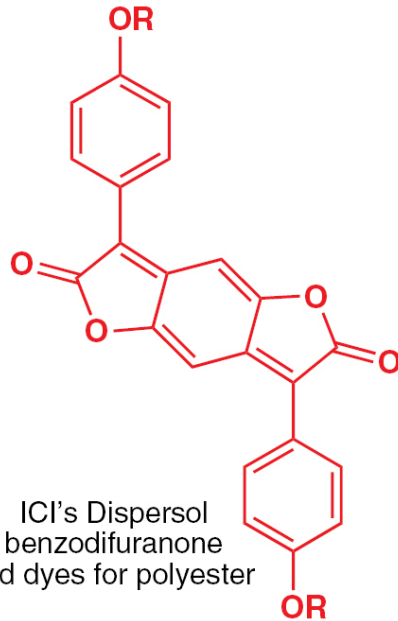
da fine 1800 ad oggi

molte sostanze sono state aggiunte: disinfettanti, tensioattivi, addensanti, dolcificanti, coloranti per migliorare il sapore della pasta dentifricia e fornire additivi come disinfettanti, dolcificanti e coloranti che servono ad aumentare l'accettabilità del prodotto da parte del cliente.

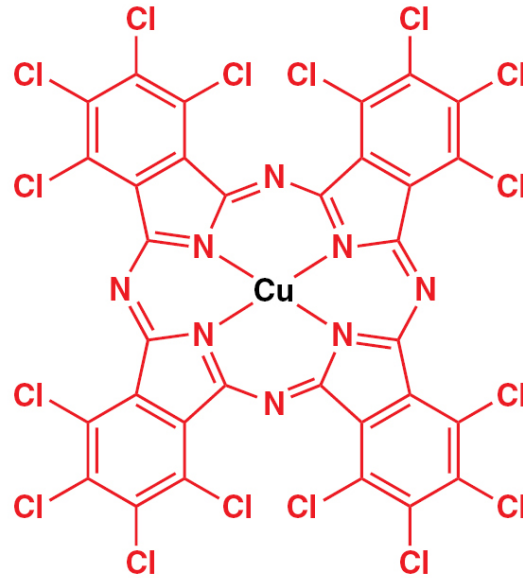
coloranti industriali



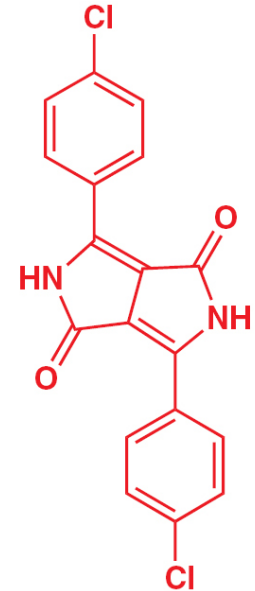
indigo
the colour of blue jeans



ICI's Dispersol
benzodifuranone
red dyes for polyester

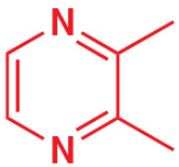


ICI's Monastral Green GNA
a green for plastic objects

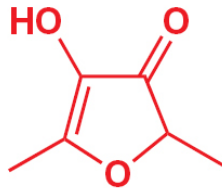


Ciba-Geigy's Pigment Red 254
an intense DPP pigment

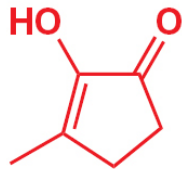
sapori e fragranze



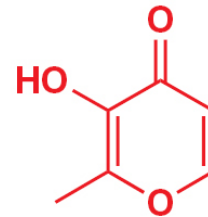
an alkyl pyrazine
from coffee and
roast meat



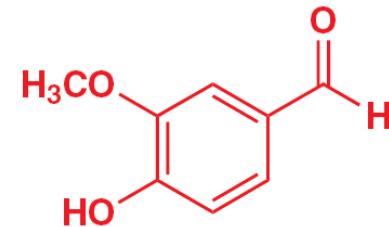
furonol
roast meat



corylone
caramel
roasted taste

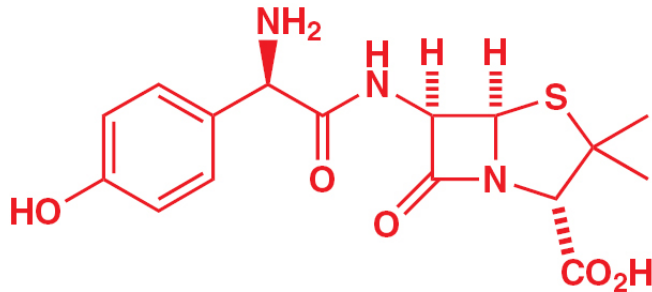


maltol
E-636 for cakes
and biscuits

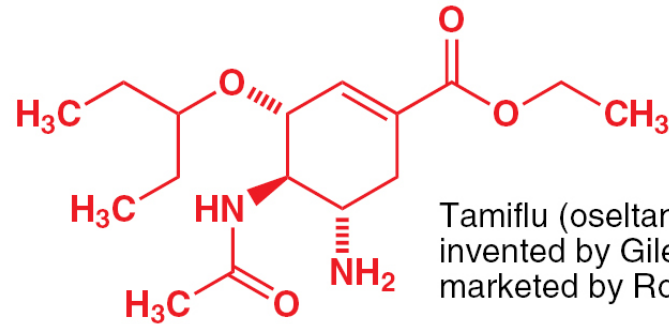


vanillin
found in vanilla pods;
manufactured on a large scale

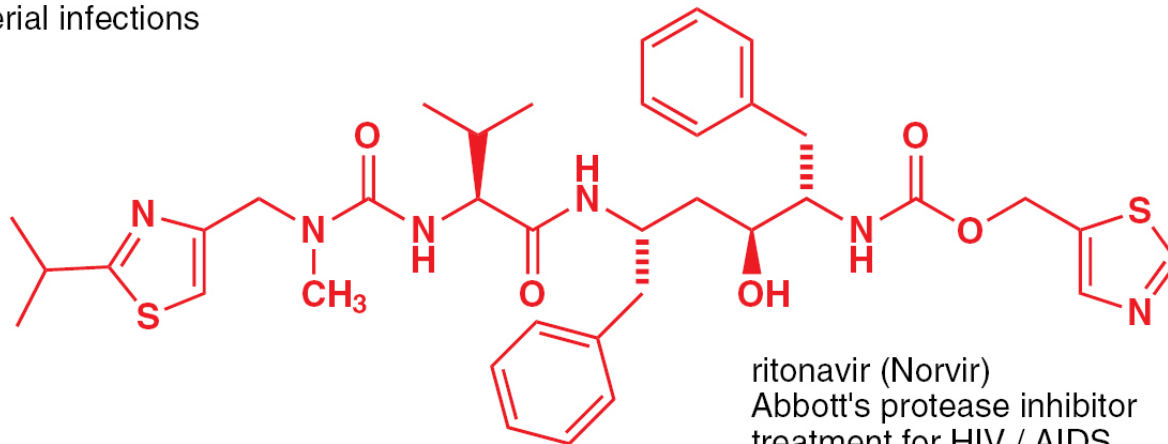
principi attivi farmaceutici



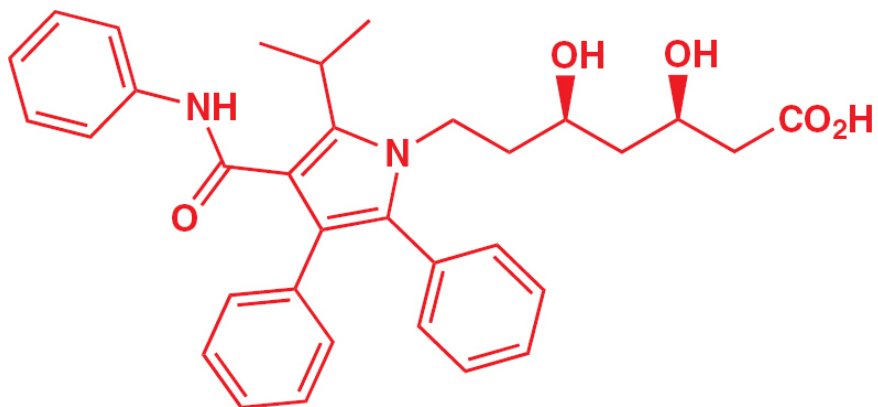
amoxicillin
developed by SmithKline Beecham
 β -lactam antibiotic treatment of
bacterial infections



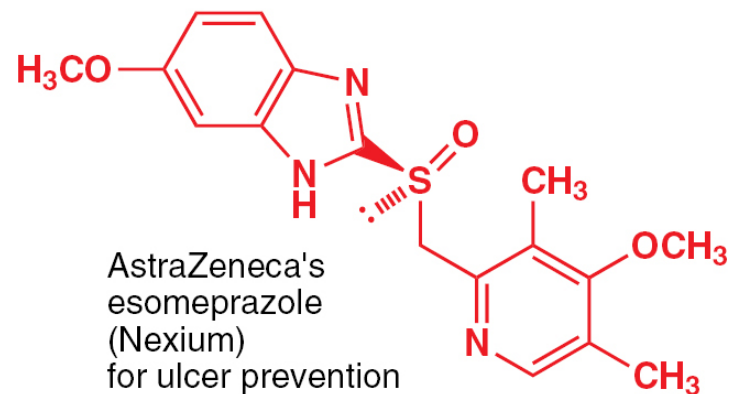
Tamiflu (oseltamivir)
invented by Gilead Sciences
marketed by Roche



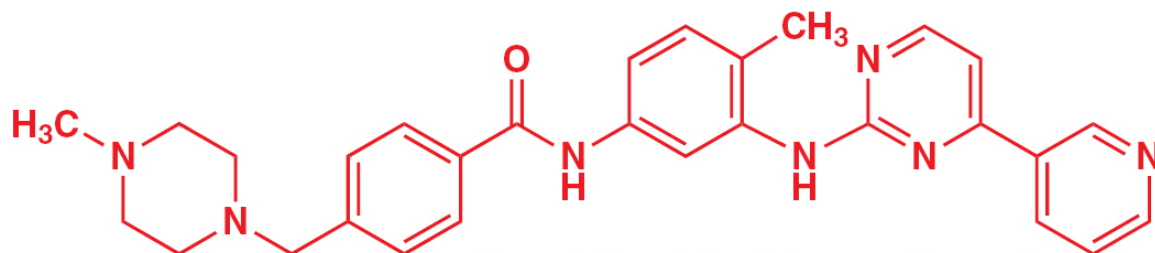
ritonavir (Norvir)
Abbott's protease inhibitor
treatment for HIV / AIDS



Pfizer's atorvastatin (Lipitor)
cholesterol-lowering drug

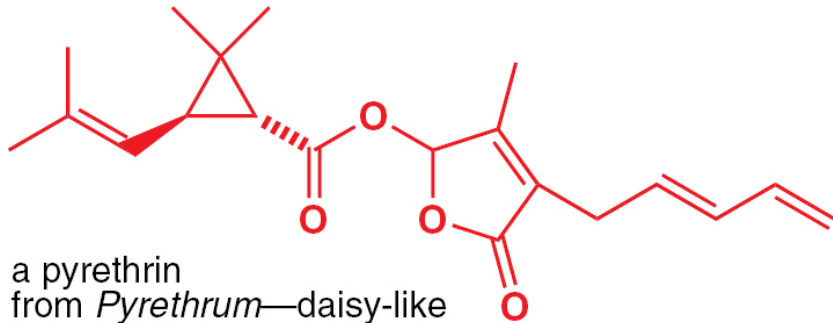


AstraZeneca's
esomeprazole
(Nexium)
for ulcer prevention

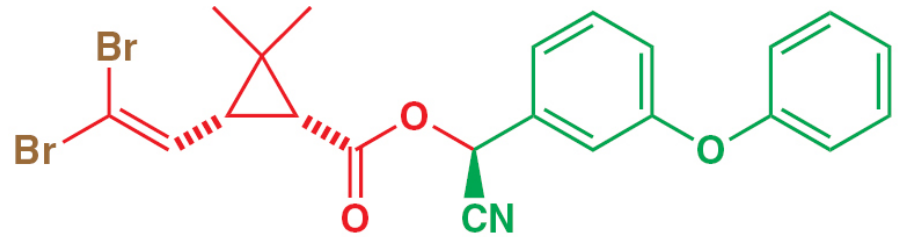


Novartis' imatinib (Glivec or Gleevec)
treatment for cancers such as leukaemia

erbicidi, fungicidi, insetticidi



a pyrethrin
from *Pyrethrum*—daisy-like
flowers from East Africa



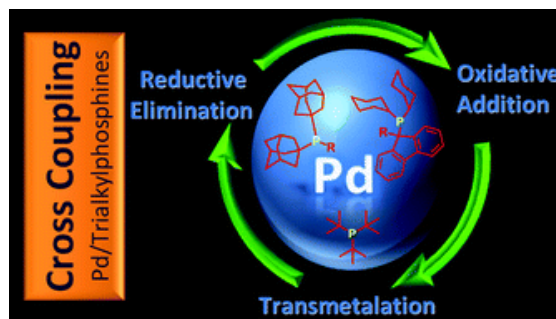
decamethrin


a modified pyrethrin—more active and stable in sunlight

la tavola periodica della chimica organica

the organic chemist's periodic table

1 H													13 B	14 C	15 N	16 O	17 F	18 <input type="text"/>	
Li	<input type="text"/>													Al	Si	P	S	Cl	<input type="text"/>
Na	Mg	3	4	5	6	7	8	9	10	11	12	Al	Si	P	S	Cl	<input type="text"/>		
K	<input type="text"/>	<input type="text"/>	Ti	<input type="text"/>	Cr	<input type="text"/>	Fe	<input type="text"/>	<input type="text"/>	Cu	Zn	<input type="text"/>	<input type="text"/>	<input type="text"/>	Se	Br	<input type="text"/>		
<input type="text"/>	<input type="text"/>	<input type="text"/>	<input type="text"/>	<input type="text"/>	<input type="text"/>	<input type="text"/>	Ru	<input type="text"/>	Pd	Ag	<input type="text"/>	<input type="text"/>	Sn	<input type="text"/>	<input type="text"/>	I	<input type="text"/>		
<input type="text"/>	<input type="text"/>	<input type="text"/>	<input type="text"/>	<input type="text"/>	<input type="text"/>	<input type="text"/>	Os	<input type="text"/>	<input type="text"/>	Au	Hg	<input type="text"/>	<input type="text"/>	<input type="text"/>	<input type="text"/>	<input type="text"/>	<input type="text"/>		
Sm																			





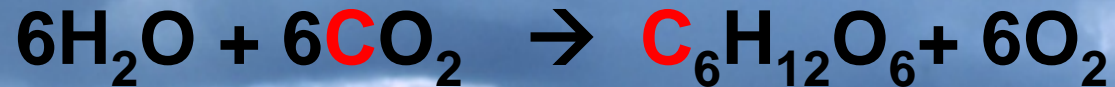
“...The world is not running out of oil - at least not yet. What our society does face, and soon, is the end of the abundant and cheap oil...”

Think gas is expensive now?
Just wait.
You've heard it before,
but this time it's for real.
We're at the beginning of

the end of
cheap

Oil

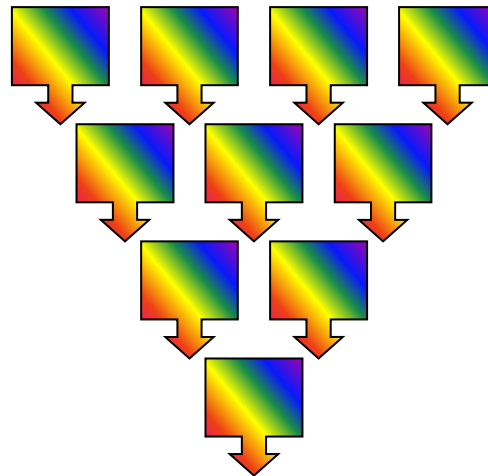
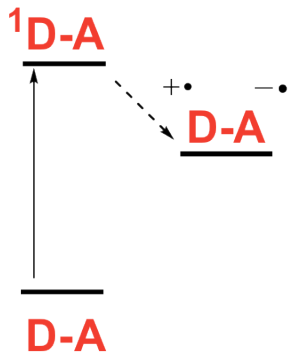
la fotosintesi clorofilliana



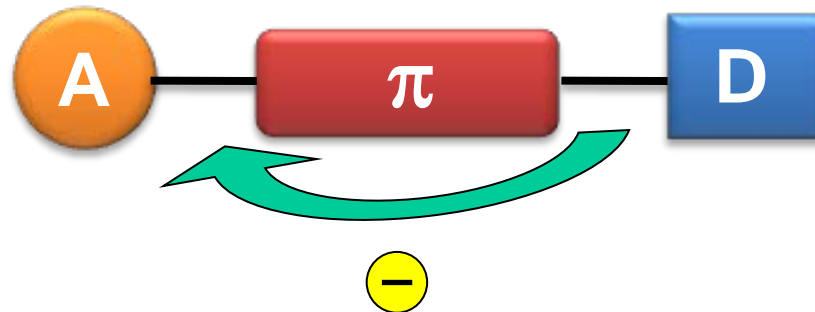
"Entra nella foglia collidendo con altre innumerevoli (ma qui inutili) molecole di azoto e di ossigeno. Aderisce ad una grossa e complicata molecola che lo attiva, e simultaneamente riceve il decisivo messaggio dal cielo, sotto forma di un pacchetto di luce solare: in un istante come un insetto preda del ragno, viene separato dal suo ossigeno, combinato con l'idrogeno e (si crede) fosforo, ed infine inserito in una catena, lunga o breve non importa, ma la catena della vita." (Primo Levi, *Il Sistema Periodico*, Torino: Einaudi, 1975).

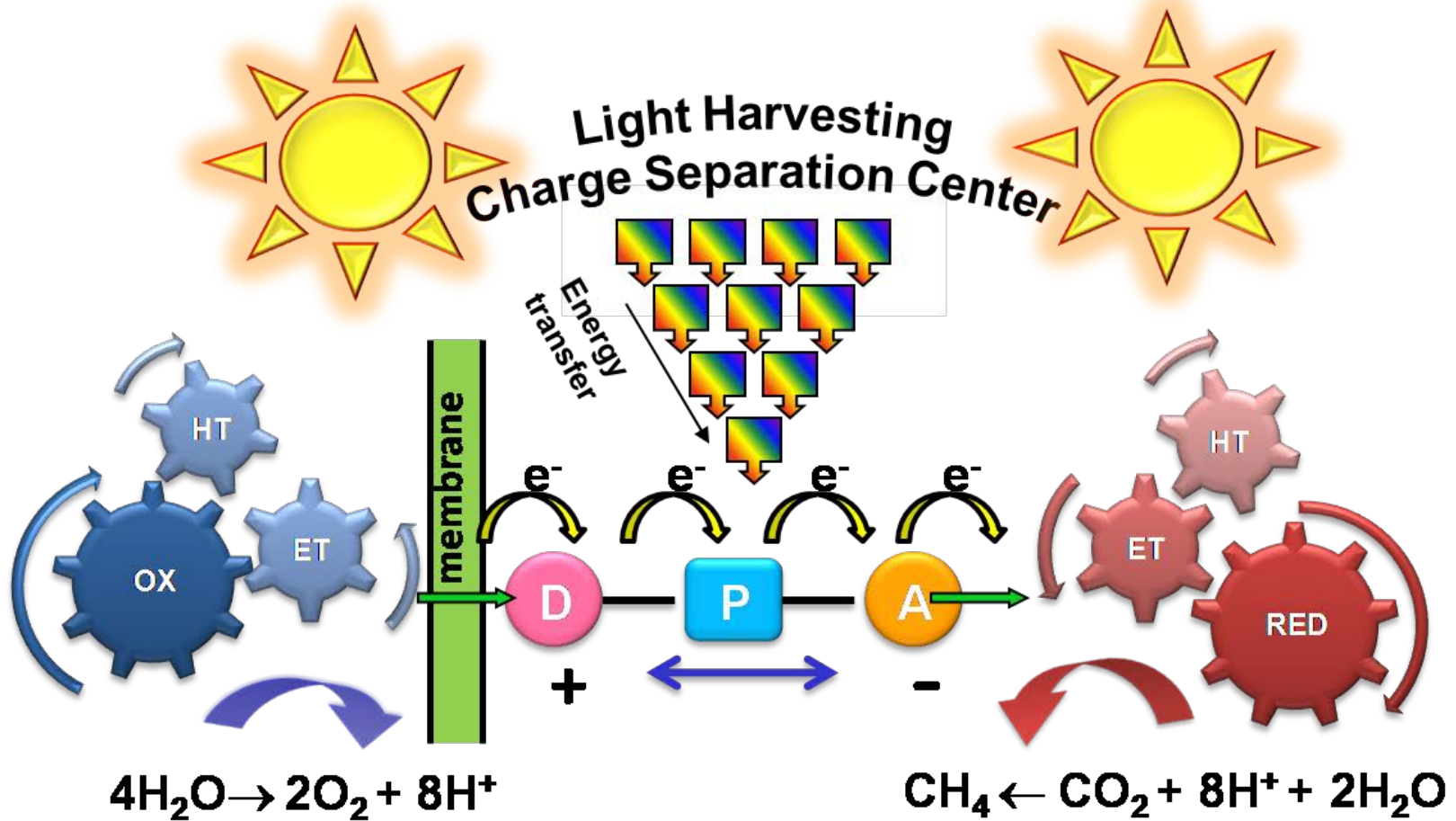


celle solari molecolari organiche a film sottile: molecole che assorbono la luce, trasferiscono elettroni, producono elettricità.



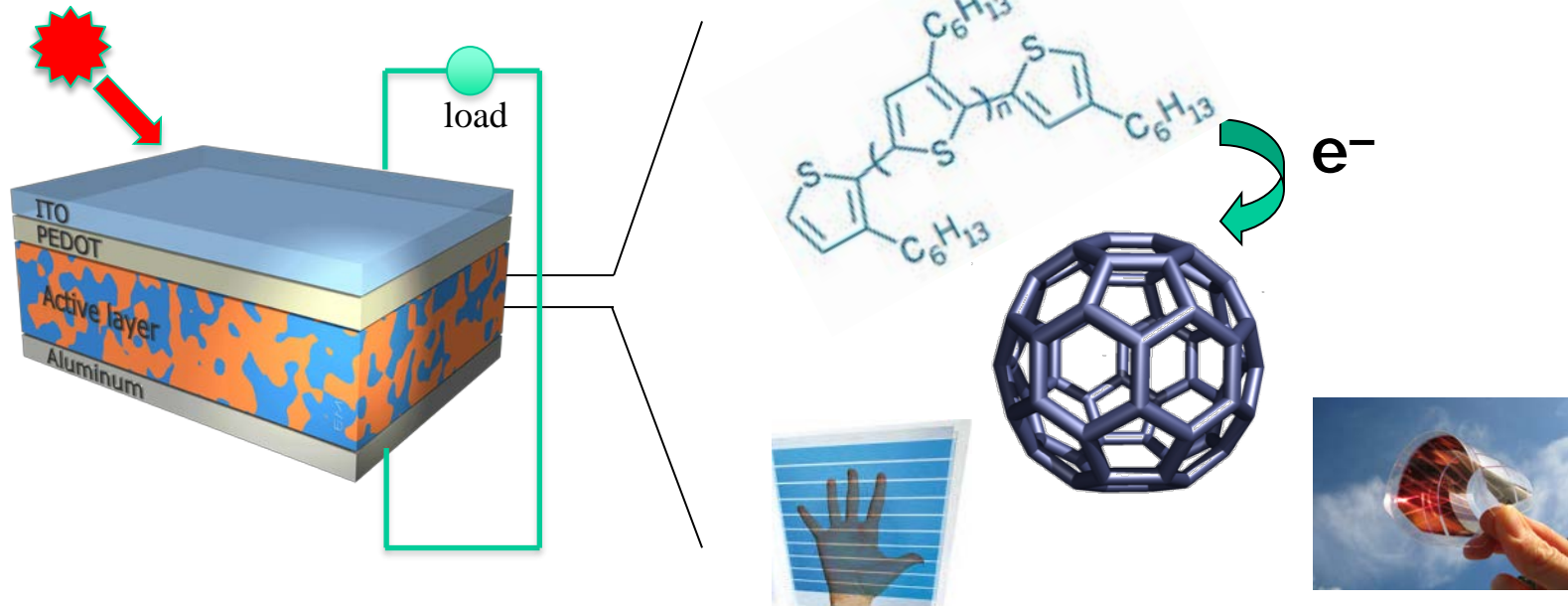
fotosintesi: molecole che assorbono la luce, trasferiscono elettroni, producono energia chimica dal CO_2 e dall'acqua.





Scheme 1. Modular assembly for Artificial Photosynthesis finalized to CO₂ reduction using water as electron and proton donor

celle OPV – leggere e flessibili



- **Film sottili, semitrasparenti, poco materiale necessario (g/m²)**
- **Deposizione da soluzione, supporti flessibili e adattabili**
- **Materiali progettabili, architetture del dispositivo variabili**
- **Tecnologia integrabile in strutture pre-esistenti che diventano produttrici di energia elettrica**

Forme allotropiche del carbonio

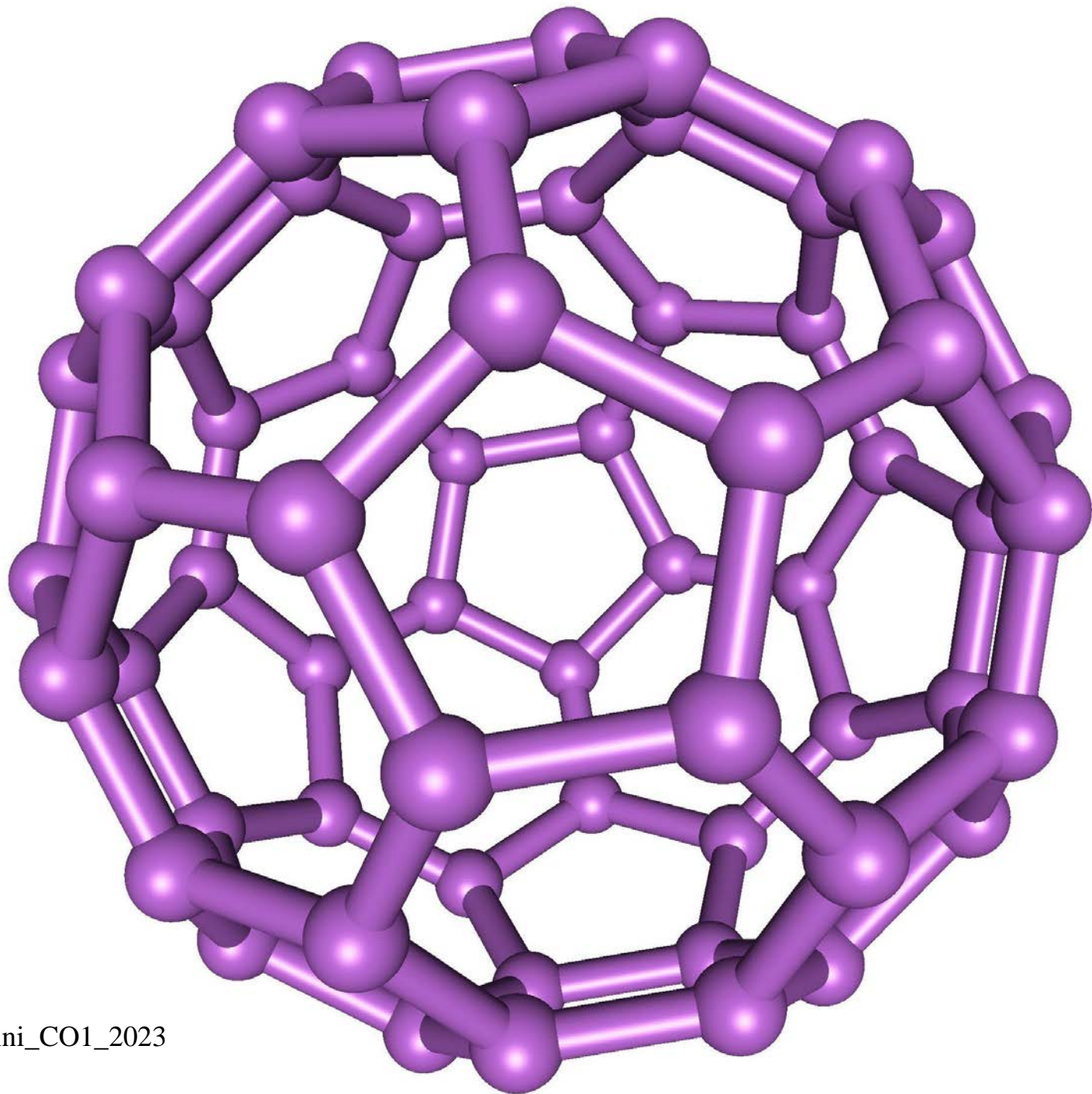
(differente struttura cristallina)

Prima del 1985

Allo stato solido si conoscevano due strutture fondamentali: la **grafite** e il **diamante**.

Dopo il 1985

Si aggiungono i **fullereni** e i **nanotubi di carbonio** derivati dalla grafite mediante processi di condensazione durante i quali il carbonio passa dalla fase liquida a quella solida, in determinate condizioni di pressione.

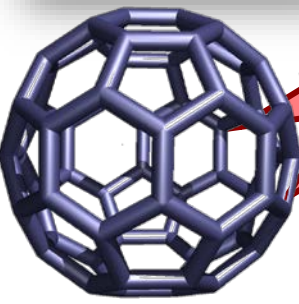


maggini_CO1_2023

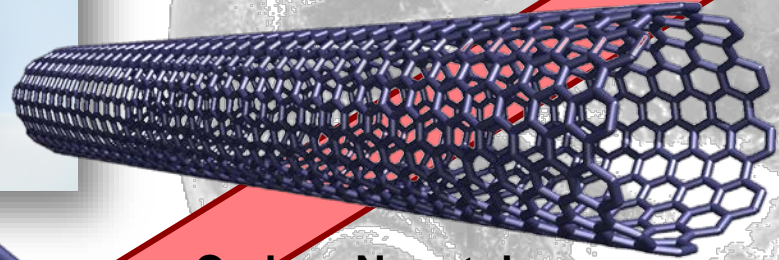
Fullerene



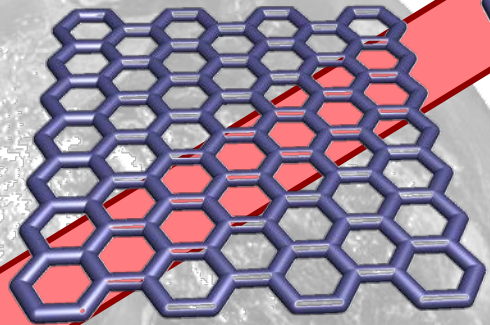




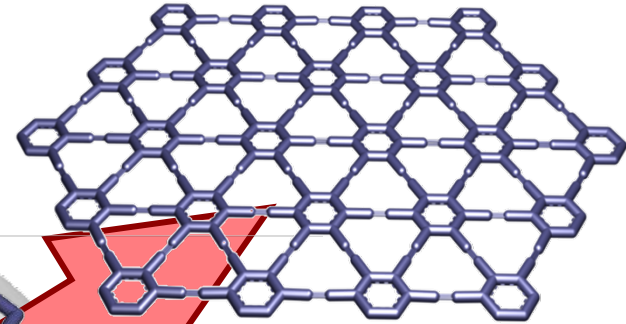
Fullerenes
1990



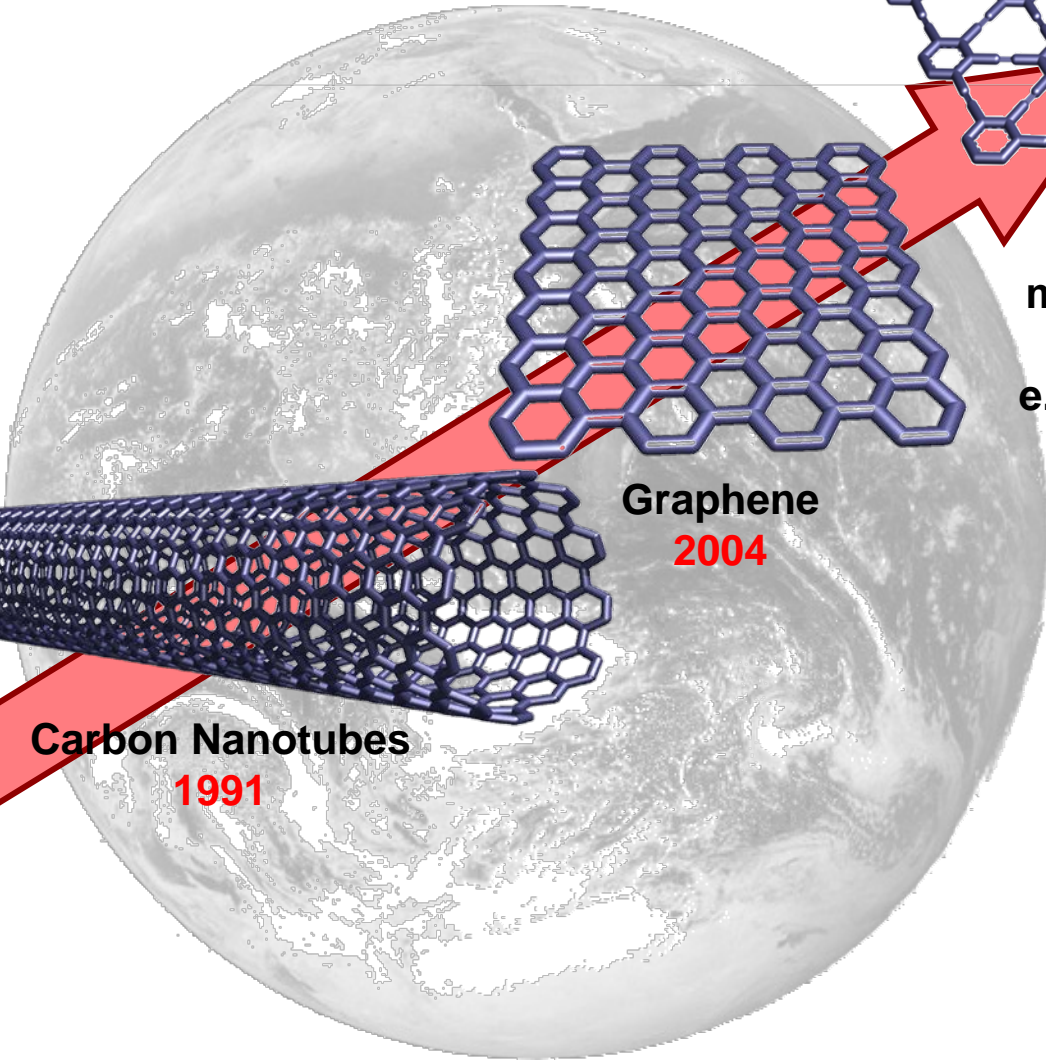
Carbon Nanotubes
1991



Graphene
2004



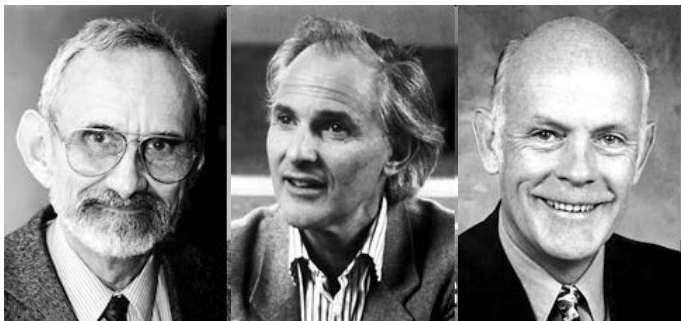
many undiscovered
allotropes
e.g. sp-sp²-Graphyne
> 20??



Properties of graphene

- Electron mobility 100 times that of silicon (seemingly massless activity of electrons)
- 100 times stronger than steel
- Graphene transistors operate at more than twice the speed of the best Si transistor of similar size
- Exceptional heat conduction

1996: Nobel per la chimica

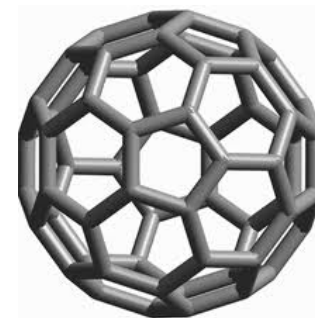


Curl, Kroto, Smalley

2010: anno del carbonio



Heck, Negishi, Suzuki
(legami C-C)



fullerene



Geim, Novoselov

