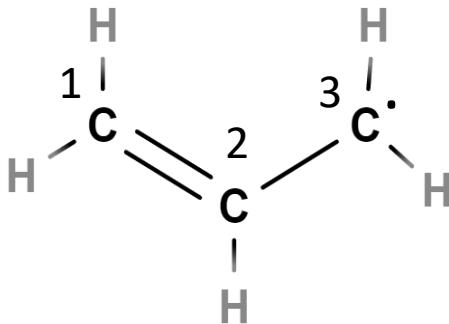


# MO di tipo $\pi$ in idrocarburi Metodo di Hückel

## Radicale allile



Equazione secolare:

$$\begin{vmatrix} \alpha - E & \beta & 0 \\ \beta & \alpha - E & \beta \\ 0 & \beta & \alpha - E \end{vmatrix} = 0$$

$$+ \begin{vmatrix} \alpha - E & \beta & 0 \\ \beta & \alpha - E & \beta \\ 0 & \beta & \alpha - E \end{vmatrix} - \begin{vmatrix} \alpha - E & \beta & 0 \\ \beta & \alpha - E & \beta \\ 0 & \beta & \alpha - E \end{vmatrix} =$$

$$= +(\alpha - E)^3 + 0 + 0 - 0 - \beta^2(\alpha - E) - \beta^2(\alpha - E) =$$

$$= +(\alpha - E)^3 - 2\beta^2(\alpha - E) = 0$$

Divido per  $\beta^3$

$$+ \frac{(\alpha - E)^3}{\beta^3} - 2 \frac{(\alpha - E)}{\beta} = 0$$

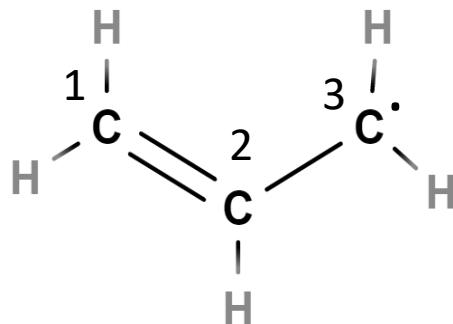
Sostituzione con  $x = \frac{(\alpha - E)}{\beta}$

$$+x^3 - 2x = 0$$

# MO di tipo $\pi$ in idrocarburi

## Metodo di Hückel

### Radicale allile



Equazione secolare:

$$\begin{vmatrix} \alpha - E & \beta & 0 \\ \beta & \alpha - E & \beta \\ 0 & \beta & \alpha - E \end{vmatrix} = 0$$

$$= +\text{Sostituzione con } x = \frac{(\alpha - E)}{\beta}$$

$$+x^3 - 2x = 0$$

$$x_1 = -\sqrt{2}$$

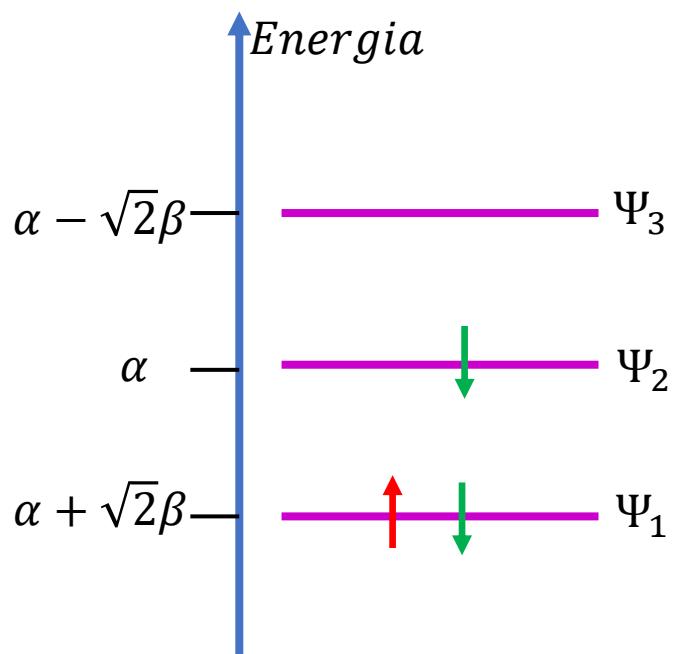
$$E_1 = \alpha + \sqrt{2}\beta$$

$$x_2 = 0$$

$$E_2 = \alpha$$

$$x_3 = +\sqrt{2}$$

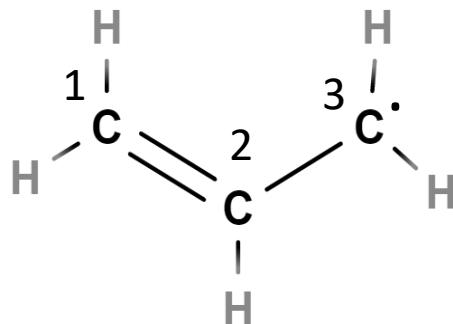
$$E_3 = \alpha - \sqrt{2}\beta$$



# MO di tipo $\pi$ in idrocarburi

## Metodo di Hückel

### Radicale allile



Equazione secolare:

$$\begin{vmatrix} \alpha - E & \beta & 0 \\ \beta & \alpha - E & \beta \\ 0 & \beta & \alpha - E \end{vmatrix} = 0$$

$$= +\text{Sostituzione con } x = \frac{(\alpha - E)}{\beta}$$

$$+x^3 - 2x = 0$$

$$x_1 = -\sqrt{2}$$

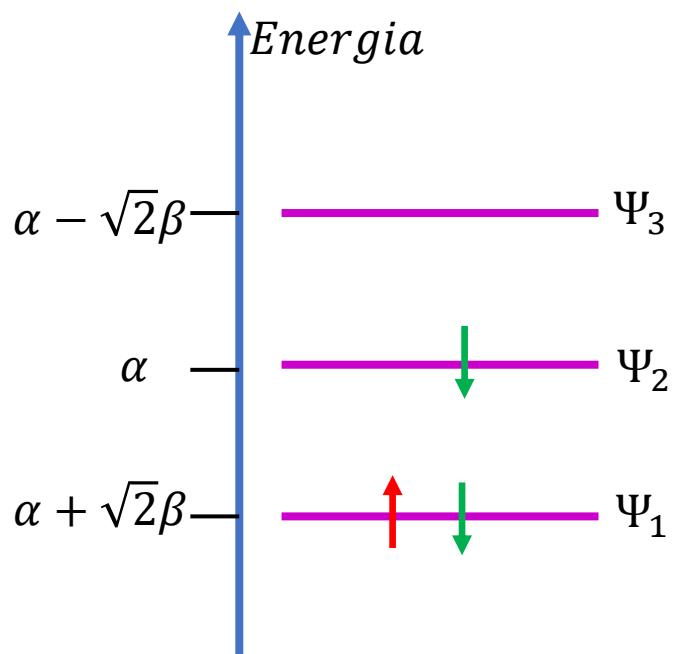
$$E_1 = \alpha + \sqrt{2}\beta$$

$$x_2 = 0$$

$$E_2 = \alpha$$

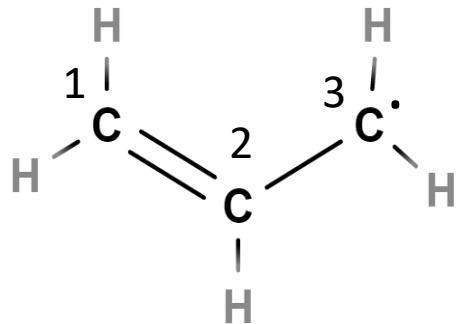
$$x_3 = +\sqrt{2}$$

$$E_3 = \alpha - \sqrt{2}\beta$$



# MO di tipo $\pi$ in idrocarburi Metodo di Hückel

## Radicale allile



Equazione secolare:

$$\begin{vmatrix} \alpha - E & \beta & 0 \\ \beta & \alpha - E & \beta \\ 0 & \beta & \alpha - E \end{vmatrix} = 0$$

$$+ (\alpha - \epsilon)^3 + 0 + 0 - 0 - \beta^2(\alpha - \epsilon) - \beta^2(\alpha - \epsilon) =$$

$$= (\alpha - \epsilon)^3 - 2\beta^2(\alpha - \epsilon) = 0$$

tutta diviso per  $\beta^3$

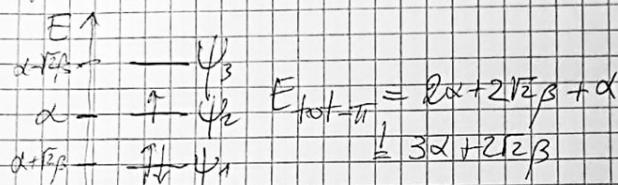
$$\frac{(\alpha - \epsilon)^3}{\beta^3} - 2 \frac{\alpha - \epsilon}{\beta} = 0$$

$$x^3 - 2x = 0$$

$$x_1 = -\sqrt[3]{2} \quad E_1 = \alpha + \sqrt[3]{2}\beta$$

$$x_2 = 0 \quad E_2 = \alpha$$

$$x_3 = \sqrt[3]{2} \quad E_3 = \alpha - \sqrt[3]{2}\beta$$



### ORBITALI MOLECOLARI

$$\psi_b \quad (E_b = \alpha + \sqrt{2}\beta) \quad \text{dividiamo per } \beta$$

$$\begin{pmatrix} \alpha - \sqrt{2}\beta & \beta & 0 \\ \beta & \alpha - \sqrt{2}\beta & \beta \\ 0 & \beta & \alpha + \sqrt{2}\beta \end{pmatrix} \xrightarrow{\text{dividiamo per } \beta} \begin{pmatrix} \alpha - \sqrt{2} & 1 & 0 \\ 1 & \alpha - \sqrt{2} & 1 \\ 0 & 1 & \alpha + \sqrt{2} \end{pmatrix}$$

$$\left\{ \begin{array}{l} -\sqrt{2}c_1 + c_2 = 0 \\ c_1 - \sqrt{2}c_2 + c_3 = 0 \\ +c_2 - \sqrt{2}c_3 = 0 \end{array} \right. \quad \left\{ \begin{array}{l} c_2 = \sqrt{2}c_1 \\ c_3 = \sqrt{2}c_1 \\ -\sqrt{2}c_1 - \sqrt{2}c_3 = 0 \end{array} \right. \quad c_1 = c_3 = 0$$

$$\psi_b = c\varphi_1 + \sqrt{2}c\varphi_2 + c\varphi_3$$

IMPONIAMO LA NORMALIZZAZIONE

$$1 = 1 \cdot c^2 + 2c^2 + 1 \cdot c^2 = 4c^2 \quad c = \pm \sqrt{2} = \pm \frac{1}{2}$$

$$\psi_b = \frac{1}{2}\varphi_1 + \frac{\sqrt{2}}{2}\varphi_2 + \frac{1}{2}\varphi_3$$

$$\int \psi_1^* \psi_1 d\tau = 1 = \int \psi_1^* \psi_1 d\tau + \int \psi_2^* \psi_2 d\tau$$

+ termini nulli perché  
contengono  $\int \psi_3^* \psi_3 d\tau = 0$   
in altra H atomica